

**Electrochemical Generation of Glycosyl Dioxalenium Ions and Their  
Application to the Synthesis of Cyclic  $\beta$ -Glucans**

**January 2024**

**Akito Shibuya**

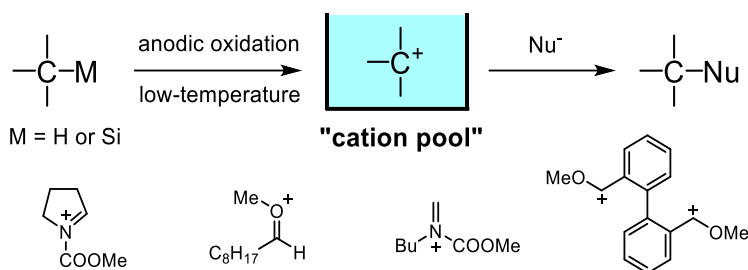


## Contents

General introduction.....	1
Chapter 1. Glycosyl Dioxalenium Ions as Reactive Intermediates of Automated Electrochemical Assembly.....	4
- Introduction	
- Results and Discussion	
- Conclusion	
- Experimental	
- References	
- NMR Spectra	
Chapter 2. Electrochemical Synthesis of the Protected Cyclic (1,3;1,6)- $\beta$ -Glucan Dodecasaccharide.....	60
- Introduction	
- Results and Discussion	
- Conclusion	
- Experimental	
- References	
- NMR Spectra	
Chapter 3. Towards Rational Design of Oligosaccharide Building Blocks of Cyclic $\beta$ -Glucans.....	109
- Introduction	
- Results and Discussion	
- Conclusion	
- Experimental	
- References	
- NMR Spectra	
Summary.....	128
Acknowledgement.....	129
List of publications.....	130
List of other publications.....	131

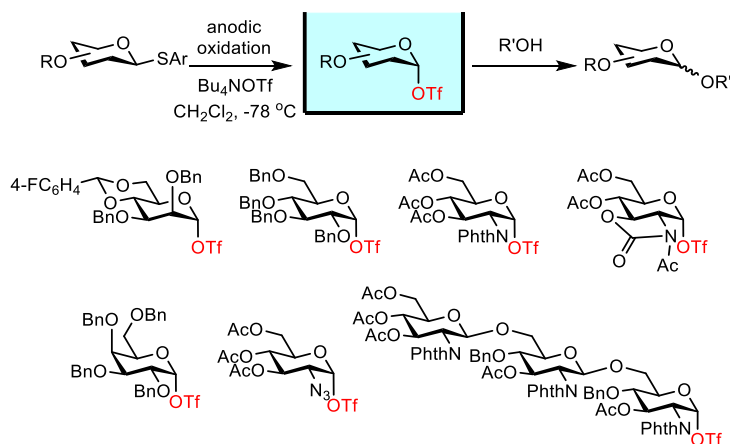
## General introduction

Activation of organic molecules is a key issue in organic synthesis. We often use strong oxidants or reductants to activate substrates with low reactivities. Constant current electrolysis is a powerful alternative that demands electricity instead of oxidant or reductant. Electrochemical reactions can be started and stopped by switching on and off the power supply. Reaction rate can be controlled under the constant current conditions and constant potential electrolysis enables selective activation of organic compounds. Therefore, electrochemical method makes organic reactions controllable, repeatable, and reproducible. For example, 'Cation Pool Method' is a revolutionary method which enables generation and accumulation of carbocations by electrochemical oxidation at very low temperature (Figure 1).<sup>1</sup> Moreover, the method enables reaction with nucleophiles which have lower oxidation potentials than those of precursors of carbocations.



**Figure 1.** Examples of cations generated by the cation pool method.

Glycosyl cations have been proposed as a reactive intermediate for glycosylation; however, no one has been successful in detecting glycosyl cations by spectroscopy.<sup>2</sup> We expected that application of 'Cation Pool Method' to glycosylation would enable glycosyl cations to be spectroscopically observed. Although we tried to generate and accumulate a glycosyl cation by electrochemical oxidation of a thioglycoside using various electrolytes including tetrabutylammonium tetrakis(pentafluorophenyl) borate ( $Bu_4NB(C_6F_5)_4$ ), we could not detect glycosyl cations by NMR even at low temperature.

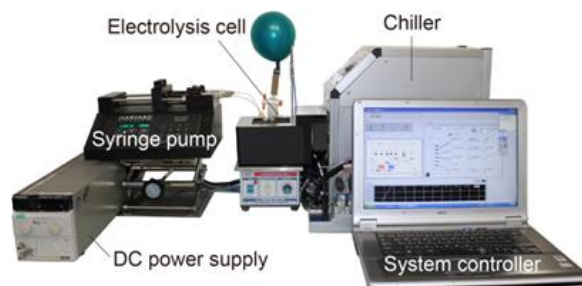


**Figure 2.** Structures of electrochemically generated glycosyl triflates.

Glycosyl triflates have been known as one of the most reactive intermediates for chemical glycosylation and observable by NMR under the low temperature conditions.<sup>3</sup> Thus, we expected that electrochemical activation of a thioglycoside in the presence of tetrabutylammonium triflate ( $Bu_4NOTf$ ) might be useful



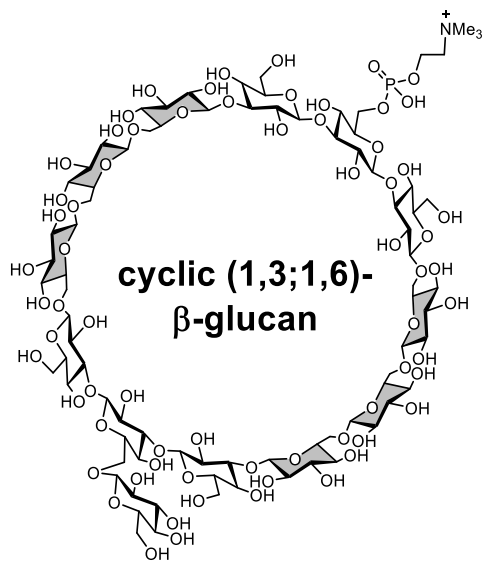
method to generate and accumulate of glycosyl triflates.<sup>4</sup> A variety of glycosyl triflates have been electrochemically generated and utilized in glycosylation with alcohols including hydroxyl groups of carbohydrates (Figure 2). Based on this result, we have developed ‘automated electrochemical assembly’ (AEA), which is an electrochemical glycosylation method controlled by a computer (Figure 3). In AEA experiments, glycosyl triflate intermediates, generated by electrochemical oxidation, are accumulated, and following addition of alcohols afford glycoside products.<sup>5</sup> In principle, reactive intermediates other than glycosyl triflate can be generated and accumulated.



**Figure 3.** Equipments for AEA.

Cyclic oligosaccharides such as cyclodextrins (CDs) have hydrophilic outer region and hydrophobic inner region. Therefore, they can be used as host molecules that include hydrophobic guest molecules. To date,  $\delta$ -CD (nonasaccharide) is the largest CD that has been chemically synthesized.<sup>6</sup> Thus, sizes and linkage types of synthesized cyclic oligosaccharides are still limited.

We then focused on a natural oligosaccharide shown in Figure 4. This compound is a cyclic (1,3;1,6)- $\beta$ -glucan, which has a cyclic dodecasaccharide structure consisting of glucosidic  $\beta$ -(1,3)- and  $\beta$ -(1,6)-linkages,<sup>7</sup> produced by root-nodulating bacteria *Bradyrhizobium japonicum* MTCC 120. The glucan is related to osmotic regulation, nodulation, and suppression of plant defense response.<sup>8</sup> Due to its large-sized ring, the compound is expected to be applied to separation of chiral molecules, drug delivery system (DDS) and catalysts. We intended to efficiently synthesize the cyclic  $\beta$ -glucan with our originally developed AEA method.



**Figure 4.** Cyclic (1,3;1,6)- $\beta$ -glucan isolated from *Bradyrhizobium japonicum* MTCC 120.

## References

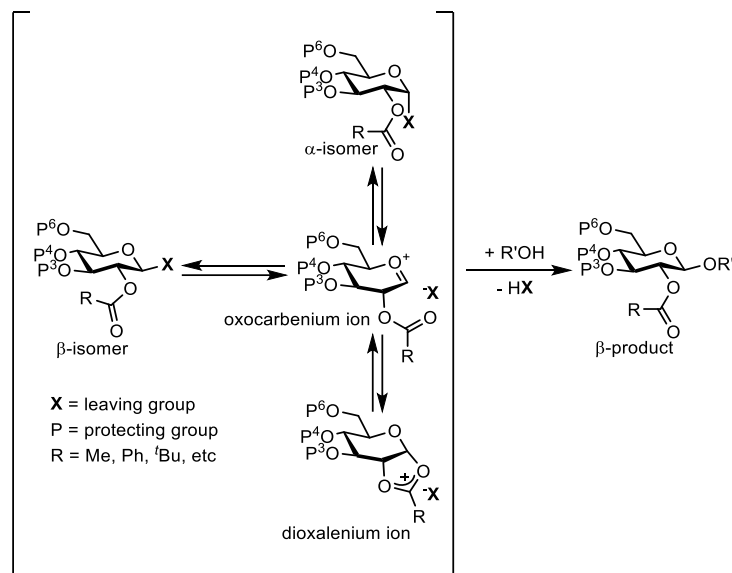
- [1] a) J. Yoshida, A. Shimizu, Y. Ashikari, T. Morofuji, R. Hayashi, T. Nokami, A. Nagaki, *Bull. Chem. Soc. Jpn.* **2015**, *88*, 763–775. b) J. Yoshida, A. Shimizu, R. Hayashi, *Chem. Rev.* **2018**, *118*, 4702–4730.
- [2] S. Suzuki, K. Matsumoto, K. Kawamura, S. Suga, J. Yoshida, *Org. Lett.* **2004**, *6*, 3755–3758.
- [3] D. Crich, S. Sun, *J. Am. Chem. Soc.* **1997**, *119*, 11217–11223.
- [4] a) T. Nokami, A. Shibuya, H. Tsuyama, S. Suga, A. A. Bowers, D. Crich, J. Yoshida, *J. Am. Chem. Soc.* **2007**, *129*, 10922–10928. b) T. Nokami, A. Shibuya, S. Manabe, Y. Ito, J. Yoshida, *Chem. Eur. J.* **2009**, *15*, 2252–2255. c) T. Nokami, Y. Nozaki, Y. Saigusa, A. Shibuya, S. Manabe, Y. Ito, J. Yoshida, *Org. Lett.* **2011**, *13*, 1544–1547. d) T. Nokami, A. Shibuya, Y. Saigusa, S. Manabe, Y. Ito, J. Yoshida, *Beilstein J. Org. Chem.* **2012**, *8*, 456–460.
- [5] a) T. Nokami, R. Hayashi, Y. Saigusa, A. Shimizu, C.-Y. Liu, K.-K. T. Mong, J. Yoshida, *Org. Lett.* **2013**, *15*, 4520–4523. b) T. Nokami, Y. Isoda, N. Sasaki, A. Takaiso, S. Hayase, T. Itoh, A. Shimizu, R. Hayashi, J. Yoshida, *Org. Lett.* **2015**, *17*, 1525–1528.
- [6] M. Wakao, K. Fukase and S. Kusumoto, *J. Org. Chem.* **2002**, *67*, 8182-8190.
- [7] a) M.W. Breedveld and K. J. Miller, *Microbiol. Rev.* **1994**, *58*, 145-161. b) A. V. Nair, S. N. Gummadi and M. Doble, *Biotechnol. Lett.* **2016**, *38*, 1519-1525. c) E. Cho, D. Jeong, Y. Choi and S. Jung, *J. Inclusion Phenom. Macrocyclic Chem.* **2016**, *85*, 175-185.
- [8] V. A. Stanisich and B. A. Stone, *Chemistry, Biochemistry, and Biology of (1-3)- $\beta$ -Glucans and Related Polysaccharides*, **2009**, 327-352.

## Chapter 1.

# Glycosyl Dioxalenium Ions as Reactive Intermediates of Automated Electrochemical Assembly

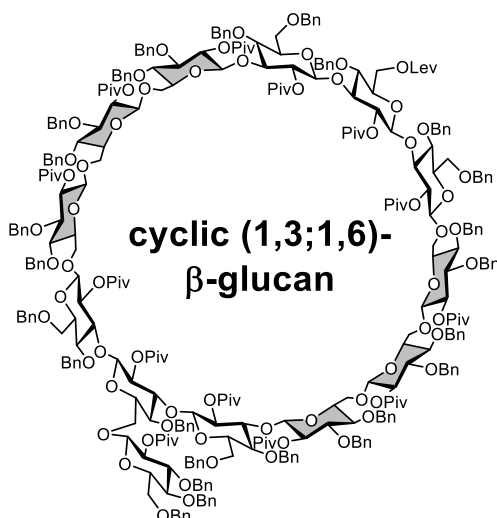
### Introduction

Stereoselectivity in chemical glycosylation is a crucial issue for total synthesis of complex oligosaccharides.<sup>1</sup> One of the most reliable methods is using glycosyl donor with a stereo-controlling group as a protecting group of the hydroxyl group at C-2 position (2-OH). For example, acetyl (R = Me), benzoyl (R = Ph), pivaloyl (R = <sup>t</sup>Bu), and other acyl groups at 2-OH can work as neighboring groups which can form glycosyl dioxalenium ions (Figure 1-1).<sup>2</sup> Thus-generated glycosyl dioxalenium ions have been known as reactive glycosylation intermediates and important chemical species for stereoselective synthesis of 1,2-*trans* glycosidic linkages including  $\beta$ -glucosides and  $\alpha$ -mannosides.



**Figure 1-1.** Possible glycosylation intermediates for  $\beta$ -selective glycosylation of glucoside.

We have been interested in synthesis of cyclic  $\beta$ -(1,3)- $\beta$ -(1,6)-glucan<sup>3</sup> as a target oligosaccharide of automated electrochemical assembly (Figure 1-2).<sup>4</sup> Cyclic  $\beta$ -glucans including the cyclic  $\beta$ -(1,3)- $\beta$ -(1,6)-glucan have potential applications for separation of chiral molecules, drug delivery, and catalyst.<sup>3c</sup> Although synthesis of hexasaccharide, which is a half structure of the cyclic oligosaccharide, has already been achieved, the yield was not high enough to complete the total synthesis.<sup>5</sup> Therefore, we decided to revise the synthetic route for the half structure and re-optimized the reaction conditions. Here, we report optimization of the synthesis of  $\beta$ -1,6-glucan trisaccharide and NMR study of electrochemically generated glycosylation intermediates.<sup>6</sup>

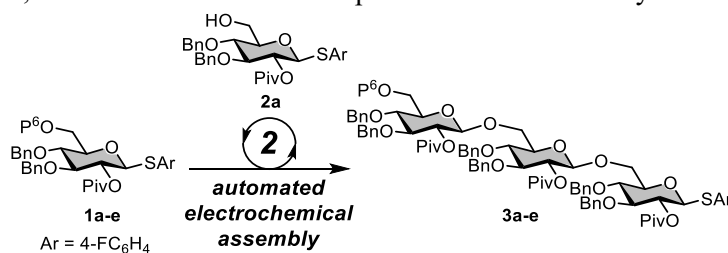


**Figure 1-2.**  $\beta$ -1,6-Glucan trisaccharide as a partial structure of the cyclic oligosaccharide.

## Results and Discussion

We initiated the synthesis by using 3,4,6-tri-*O*-benzyl-2-*O*-pivaloyl-thioglycoside **1a** ( $P^2 = \text{Piv}$ ,  $P^6 = \text{Bn}$ ) as a model compound and the desired trisaccharide **3a** was obtained in 61% yield (78% yield per cycle) (Table 1-1, entry 1). To prepare the trisaccharide unit for the total synthesis of the cyclic oligosaccharide, the reaction of thioglycoside **1b** ( $P^2 = \text{Piv}$ ,  $P^6 = 9$ -fluorenylmethyloxycarbonyl (Fmoc)) with the Fmoc group as a temporary protecting group of the hydroxyl group at C-6 position (6-OH) was tested; however, the yield of trisaccharide **3b** ( $P^6 = \text{H}$ ) was only 6% after one-pot deprotection of the Fmoc group (entry 2). Other thioglycoside **1c** ( $P^2 = \text{Piv}$ ,  $P^6 = \text{Piv}$ ), **1d** ( $P^2 = \text{Piv}$ ,  $P^6 = \text{Ac}$ ), and **1e** ( $P^2 = \text{Piv}$ ,  $P^6 = \text{TBDPS}$ ) also gave the desired trisaccharide **3c-e**; however, the yields of trisaccharide **3c-e** were moderate (entries 3-5).

**Table 1-1.**  $\beta$ -1,6-Glucan trisaccharide as a partial structure of the cyclic oligosaccharide.



entry <sup>[a]</sup>	thioglycoside <b>1</b>	product <b>3</b> (yield)	selectivity ( $\alpha$ : $\beta$ )
1	<b>1a</b> ( $P^6 = \text{Bn}$ )	<b>3a</b> (61%, $P^6 = \text{Bn}$ )	$\beta$ only
2	<b>1b</b> ( $P^6 = \text{Fmoc}$ )	<b>3b</b> (6%, $P^6 = \text{H}$ ) <sup>[b]</sup>	$\beta$ only
3	<b>1c</b> ( $P^6 = \text{Ac}$ )	<b>3c</b> (24%, $P^6 = \text{Ac}$ )	$\beta$ only
4	<b>1d</b> ( $P^6 = \text{Piv}$ )	<b>3d</b> (36%, $P^6 = \text{Piv}$ )	$\beta$ only
5	<b>1e</b> ( $P^6 = \text{TBDPS}$ )	<b>3e</b> (29%, $P^6 = \text{TBDPS}$ )	$\beta$ only

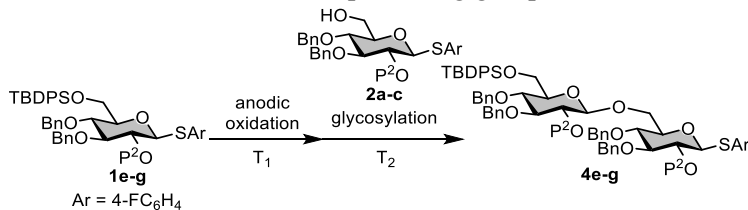
[a] Anodic oxidation was performed using a divided cell at  $-80\text{ }^\circ\text{C}$  ( $T_1$ ) under constant current (12 mA, 1.2 F/mol) in  $\text{CH}_2\text{Cl}_2$  with 0.1 M  $\text{Bu}_4\text{NOTf}$  as electrolyte. The subsequent one-pot glycosylation was carried out at  $-50\text{ }^\circ\text{C}$  ( $T_2$ ) for 1 h. [b] Deprotection of Fmoc group was carried out in one pot.

Optimization of reaction temperature was performed using thioglycoside **1e** ( $P^2 = \text{Piv}$ ,  $P^6 = \text{TBDPS}$ ) because of its better yield among the thioglycosides with a temporary protecting group (Table 1-2). We tested reactions at different temperatures from  $-60$  to  $0$  °C and performed reactions at same temperature for both anodic oxidation and glycosylation. Although the yield of disaccharide **4e** did not change at  $-80$  and  $-60$  °C (entries 1 and 2), the best yield was observed at  $-40$  °C (entry 3). To our surprise, the desired disaccharide **4e** was obtained even at elevated temperature  $-20$  and  $0$  °C (entries 4 and 5).

**Table 1-2.** Temperature effect on glycosylation.

entry	T <sub>1</sub> (°C)	T <sub>2</sub> (°C)	yield	selectivity (α:β)
1	-80	-50	67%	β only
2	-60	-60	67%	β only
3	-40	-40	72%	β only
4	-20	-20	53%	β only
5	0	0	45%	β only

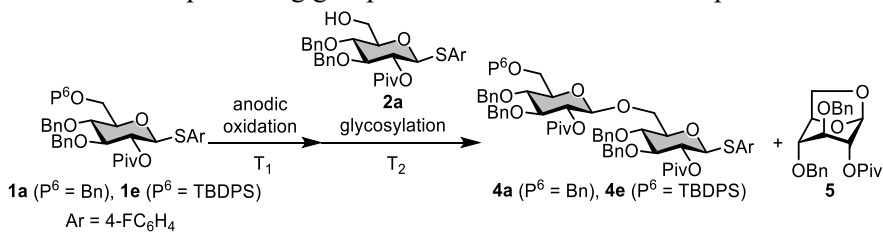
Inspired by successful glycosylation at elevated temperature, further optimization of protecting group at 2-OH ( $P^2$ ) was carried out (Table 1-3). Although both Ac group (entries 3 and 4) and benzoyl (Bz) group (entries 5 and 6) gave the product, their yields were lower than those of Piv group (entries 1 and 2). Therefore, thioglycoside **1e** ( $P^2 = \text{Piv}$ ,  $P^6 = \text{TBDPS}$ ) may provide the most stable intermediate which does not decompose during accumulation at elevated temperature. These results encouraged us to observe glycosylation intermediates by NMR, because reactive glycosyl triflates, which have been generated by anodic oxidation, seemed to be unstable at elevated temperatures.

**Table 1-3.** Effect of protecting group at 2-OH.

entry	thioglycoside <b>1</b>	thioglycoside <b>2</b>	T <sub>1</sub> , T <sub>2</sub>	product (yield)	selectivity (α:β)
1	<b>1e</b> (P <sub>2</sub> = Piv)	<b>2a</b> (P <sub>2</sub> = Piv)	-40 °C	<b>4e</b> (72%)	β only
2	<b>1e</b> (P <sub>2</sub> = Piv)	<b>2a</b> (P <sub>2</sub> = Piv)	0 °C	<b>4e</b> (45%)	β only
3	<b>1f</b> (P <sub>2</sub> = Bz)	<b>2b</b> (P <sub>2</sub> = Bz)	-40 °C	<b>4f</b> (65%)	β only
4	<b>1f</b> (P <sub>2</sub> = Bz)	<b>2b</b> (P <sub>2</sub> = Bz)	0 °C	<b>4f</b> (25%) <sup>a</sup>	β only
5	<b>1g</b> (P <sub>2</sub> = Ac)	<b>2c</b> (P <sub>2</sub> = Ac)	-40 °C	<b>4g</b> (56%)	β only
6	<b>1g</b> (P <sub>2</sub> = Ac)	<b>2c</b> (P <sub>2</sub> = Ac)	0 °C	<b>4g</b> (36%)	β only

<sup>a</sup>1,6-anhydrosugar **9** was obtained as a by-product in 7% yield.

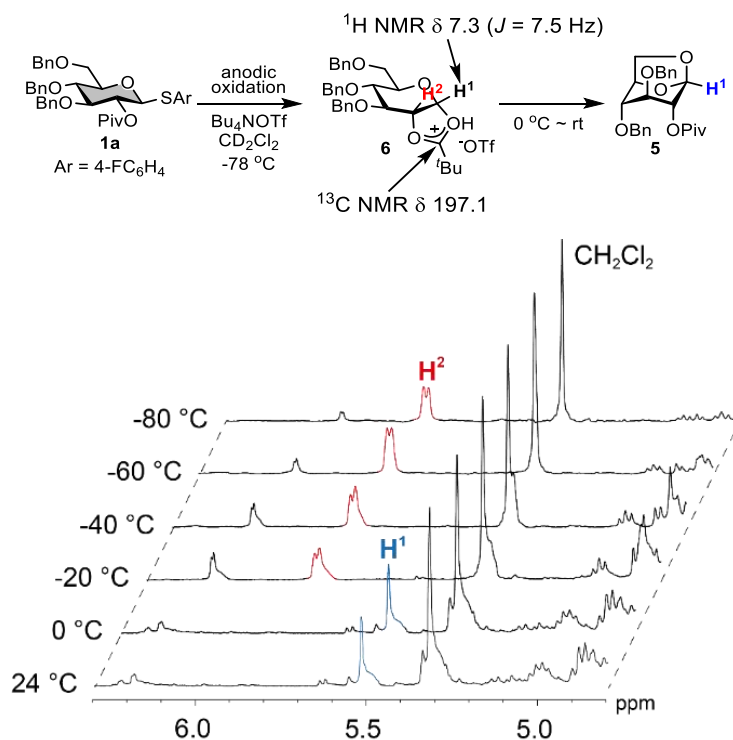
Prior to the NMR analysis, we again investigated effect of protecting group at 6-OH (P<sup>6</sup>) at elevated temperature (Table 1-4). Thioglycoside **1a** (P<sup>2</sup> = Piv, P<sup>6</sup> = Bn) gave the desired disaccharide **4a** in lower yield than that with TBDPS group (entries 1 and 2). Decrease of the yields of disaccharide **4a** and **4e** was significant at 0 °C and 1,6-anhydrosugar **5** was obtained in 33% yield together with disaccharide **4a** (entries 3 and 4). Therefore, TBDPS group at 6-OH must contribute to prevent intramolecular glycosylation at elevated temperature.

**Table 1-4.** Effect of protecting group at 6-OH under elevated temperature conditions.

entry	thioglycoside <b>1</b>	T <sub>1</sub> , T <sub>2</sub>	product (yield)	selectivity (α:β)
1	<b>1a</b> (P <sup>6</sup> = Bn)	-40 °C	<b>4a</b> (61%)	β only
2	<b>1e</b> (P <sup>6</sup> = TBDPS)	-40 °C	<b>4e</b> (72%)	β only
3	<b>1a</b> (P <sup>6</sup> = Bn)	0 °C	<b>4a</b> (12%), <b>5</b> (33%)	β only
4	<b>1e</b> (P <sup>6</sup> = TBDPS)	0 °C	<b>4e</b> (45%)	β only

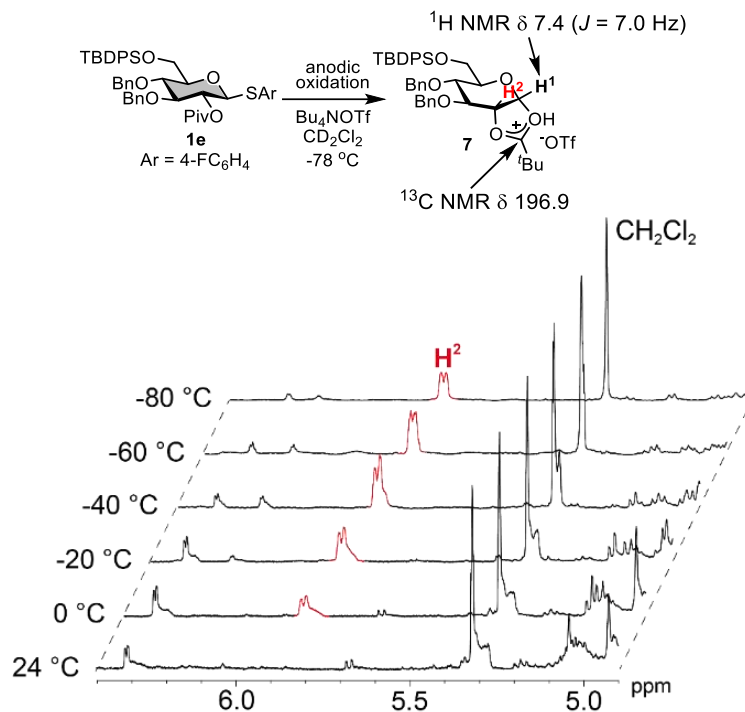
Low temperature <sup>1</sup>H NMR, <sup>13</sup>C NMR, and VT-NMR measurements were performed using three thioglycosides **1a**, **1e**, and **1f** equipped with different protecting groups at 2-OH and 6-OH (Figures 1-3, 1-4 and 1-5).<sup>7</sup> At first, thioglycoside **1a** (P<sup>2</sup> = Piv, P<sup>6</sup> = Bn) was activated, and low temperature NMR spectra were measured at -80 °C. Downfield shift of the anomeric proton (δ 7.3) and the carbonyl carbon of Piv group (δ 197.1) suggested that the major accumulated species was glycosyl dioxalenium ion, not glycosyl

triflate. Then, temperature was gradually raised from -80 °C to 24 °C with 15 min interval. The peaks derived from glycosyl dioxalenium ion was detected up to -20 °C; however, the corresponding peaks of glycosyl dioxalenium ion **6** was disappeared and the new sets of peaks, which were derived from 1,6-anhydrosugar **5**, appeared at 0 °C. These VT-NMR experiments revealed that 1,6-anhydrosugar **5** derived from **1a** and 6-OH of **6** with benzyl protecting group was reactive enough at elevated temperature. This result was consistent with the lower yield of glycosylation using thioglycoside **1a** at 0 °C (Table 1-4, entry 3).



**Figure 1-3.** VT-NMR of glycosyl dioxalenium ion **6** ( $P^2 = \text{Piv}$ ,  $P^6 = \text{Bn}$ )

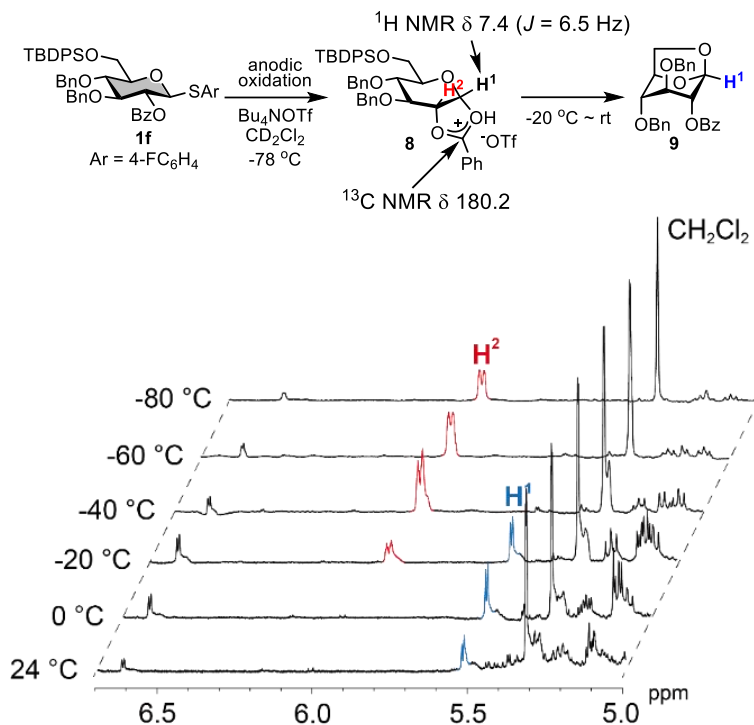
By changing protecting group  $P^6$  at 6-OH of thioglycoside **1** from Bn group to TBDPS group, glycosyl dioxalenium ion **7** ( $P^2 = \text{Piv}$ ,  $P^6 = \text{TBDPS}$ ) was again observed (Figure 1-4). Although the spectra became complex by raising temperature, the corresponding peaks of glycosyl dioxalenium ion **7** were still observed at 0 °C without peaks derived from 1,6-anhydrosugar **5**. This result clearly indicates that glycosyl dioxalenium ion was accumulated reactive intermediate at 0 °C (Table 1-2, entry 5) and the TBDPS group at 6-OH contributed higher stability of glycosyl dioxalenium ion **7**. We presume that the steric effect of TBDPS group may prevent the nucleophilic attack of 6-OH to the anomeric carbon because similar chemical shifts of the cationic carbons of glycosyl dioxalenium ions (197.1 ppm for **6** and 196.9 ppm for **7**) suggest the small electronic effect of protecting groups at 6-OH.



**Figure 1-4.** VT-NMR of glycosyl dioxalenium ion **7** ( $P^2 = \text{Piv}$ ,  $P^6 = \text{TBDPS}$ )

Other thioglycoside **1f** with Bz group at 2-OH also afforded the corresponding glycosyl dioxalenium ion **8** ( $P^2 = \text{Bz}$ ,  $P^6 = \text{TBDPS}$ ); however, stability of glycosyl dioxalenium ion **8** was lower than that of **7** (Figure 1-5). Formation of 1,6-anhydrosugar **9** was observed at  $-20\text{ }^\circ\text{C}$  and glycosyl dioxalenium ion **8** was completely disappeared at  $0\text{ }^\circ\text{C}$ . These results suggest that not only TBDPS group at 6-OH but Piv group at 2-OH also contribute to higher stability of glycosyl dioxalenium ion intermediate. Based on the formation of 1,6-anhydrosugar as a decomposition product of glycosyl dioxalenium ions, the order of stability of glycosyl dioxalenium ions **6-8** was as follows; **7** ( $P^2 = \text{Piv}$ ,  $P^6 = \text{TBDPS}$ ) > **6** ( $P^2 = \text{Piv}$ ,  $P^6 = \text{Bn}$ ) > **8** ( $P^2 = \text{Bz}$ ,  $P^6 = \text{TBDPS}$ ). We assume that bulky protecting groups such as Piv group at 2-OH and TBDPS group at 6-OH may kinetically stabilize the dioxalenium ion **7** by preventing the formation of 1,6-anhydrosugar **5** via intramolecular glycosylation. Although further NMR analysis is required to identify other chemical species than glycosyl dioxalenium ion observed at low temperature,  $\alpha$ - and/or  $\beta$ -isomer of glycosyl triflate is the most plausible intermediate.<sup>8</sup>

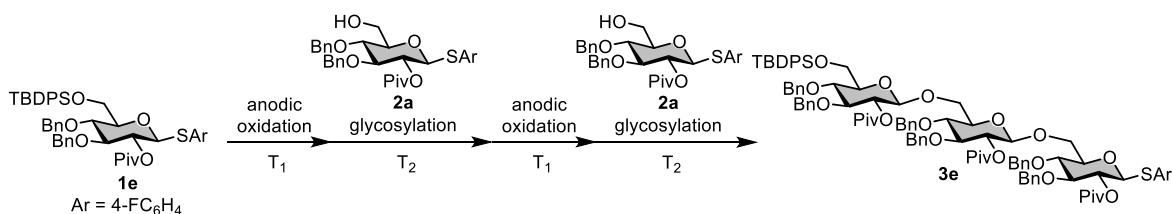




**Figure 1-5.** VT-NMR of glycosyl dioxalenium ion **8** ( $P^2 = \text{Bz}$ ,  $P^6 = \text{TBDPS}$ )

Finally, we reinvestigated the trisaccharide synthesis at elevated temperature (Table 1-5). Based on optimization of disaccharide synthesis and VT-NMR study, both anodic oxidation and glycosylation were performed at  $-40\text{ }^\circ\text{C}$  (entry 2). The desired trisaccharide **3e** was obtained in 46% yield, which is 1.5 times higher than the yield under conventional reaction conditions (entry 1). Although further optimization of reaction conditions and the temporary protecting group at 6-OH must be necessary, higher thermal stability of glycosyl dioxalenium ions enables both anodic oxidation and glycosylation at elevated temperature.

**Table 1-5.** Synthesis of trisaccharide **3e** at elevated temperature.



entry	$T_1$ ( $^\circ\text{C}$ )	$T_2$ ( $^\circ\text{C}$ )	yield	average yield per cycle	selectivity ( $\alpha:\beta$ )
1	-80	-50	29%	54%	$\beta$ only
2	-40	-40	46%	68%	$\beta$ only

## Conclusion

In conclusion, we improved the yield of synthesis  $\beta$ -1,6-glucan trisaccharide based on temperature optimization and VT-NMR experiment of glycosylation intermediate. Electrochemically generated glycosyl dioxalenium ions were more stable than glycosyl triflates which decompose at elevated

temperature such as 0 °C.<sup>9</sup> VT-NMR study clearly indicated that not only protecting groups at 2-OH but also protecting group at 6-OH influenced thermal stability of glycosyl dioxalenium ions. Based on these findings, further optimization of synthesis of  $\beta$ -1,3-glucan trisaccharide and synthesis of the linear precursor of cyclic dodecasaccharide are in progress in our laboratory.

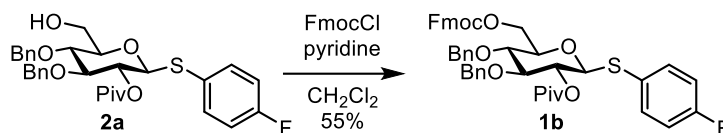
## Experimental

### 1. General

$^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were recorded on Bruker AVANCE II 600 ( $^1\text{H}$  600 MHz,  $^{13}\text{C}$  150 MHz) and JEOL JNM-ECZ500R ( $^1\text{H}$  500 MHz,  $^{13}\text{C}$  125 MHz). Electro-spray ionization mass spectra (ESI-MS) were recorded on Thermo Scientific Exactive spectrometer. Preparative recycling gel permeation chromatography (PR-GPC) was performed on Japan Analytical Industry LC-5060. Kanto silica gel 60 N (spherical, neutral, 63-210  $\mu\text{m}$ ) was used for silica gel column chromatography. The automated synthesizer is consisting of the commercially available instruments such as the chiller with a cooling bath (UCR-150, Techno Sigma), the power supply for constant current electrolysis (PMC 350-0.2 A, KIKUSUI), the syringe pump (PHD 2000 infusion, Harvard apparatus), and the system controller (LabVIEW, National Instruments). Optical rotation was recorded on JASCO DIP-1000 digital polarimeter in chloroform. Merck TLC (silica gel 60 F254) was used for TLC analysis. Starting materials **S1**,<sup>5</sup> **1a**,<sup>5</sup> and **2a**<sup>5</sup> were prepared according to the reported procedures. Unless otherwise noted, all materials were obtained from commercial suppliers and used without further purification.

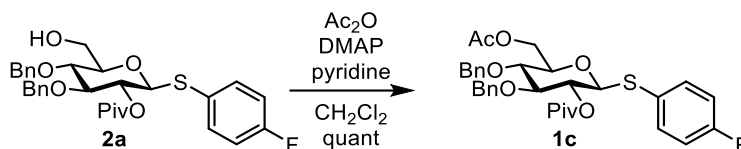
### 2. Preparation of building blocks

Preparation of 4-fluorophenyl 3,4-di-*O*-benzyl-6-*O*-(9-fluorenylmethyloxycarbonyl)-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**1b**)



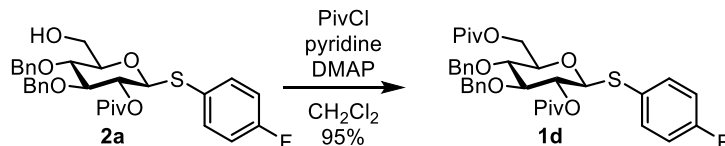
To the solution of **2a** (1.00 mmol, 555 mg) in  $\text{CH}_2\text{Cl}_2$  (6.4 mL), pyridine (13.0 mmol, 1.0 mL) and FmocCl (2.00 mmol, 523 mg) were added at room temperature and the reaction mixture was stirred for another 2 days. After the completion of the reaction determined by TLC (hexane/EtOAc 4:1). The reaction was quenched with 1 N HCl. The reaction mixture was washed with  $\text{H}_2\text{O}$  for three times and dried over  $\text{Na}_2\text{SO}_4$ . After filtration and removal of solvent under reduced pressure the crude product was purified with silica gel chromatography to obtain **1b** in 55% yield (0.553 mmol, 430 mg) as a pale-yellow solid. **4-Fluorophenyl 3,4-di-*O*-benzyl-6-*O*-(9-fluorenylmethyloxycarbonyl)-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**1b**)**; TLC (Hexane/EtOAc 4:1):  $R_f$  0.52;  $[\alpha]_D = -6.4$  ( $c = 1.0$ ,  $\text{CHCl}_3$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.76 (d,  $J = 7.2$  Hz, 2 H), 7.61 (t,  $J = 7.2$  Hz, 2 H), 7.51–7.48 (m, 2 H), 7.40 (t,  $J = 7.2$  Hz, 2 H), 7.32–7.22 (m, 12 H), 6.95 (*pseudo-t*,  $J = 8.4$  Hz, 2 H), 5.04 (*pseudo-t*,  $J = 9.6$  Hz, 1 H, H-2), 4.79 (d,  $J = 11.4$  Hz, 1 H, benzylic-H), 4.75 (d,  $J = 10.8$  Hz, 2 H, benzylic-H), 4.70 (d,  $J = 10.8$  Hz, 1 H, benzylic-H), 4.56–4.52 (m, 2 H, H-1 and benzylic-H), 4.47–4.40 (m, 3 H, H-6' and Fmoc-2H), 4.28–4.23 (m, 2 H, H-6 and Fmoc-1H), 3.73 (*pseudo-t*,  $J = 9.0$  Hz, 1 H, H-3), 3.60–3.58 (m, 2 H, H-4 and H-5), 1.25 (s, 9 H, Piv);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  176.7, 163.0 (d,  $J = 246.0$  Hz, ArS), 155.0, 143.4, 143.3, 141.4, 137.8, 137.5, 135.7 (d,  $J = 7.5$  Hz, ArS), 128.6, 128.5, 128.1, 128.0, 127.8, 127.4, 127.3, 125.2, 125.1, 120.2, 116.0 (d,  $J = 21.0$  Hz, ArS), 86.3 (C-1), 84.7 (C-3), 77.2 (C-4 or C-5), 77.0 (C-4 or C-5), 75.4 (benzylic-C), 75.2 (benzylic-C), 71.4 (C-2), 70.0 (Fmoc- $\text{CH}_2$ ), 66.6 (C-6), 46.8 (Fmoc-CH), 38.9 (Piv-1C), 27.2 (Piv-3C); HRMS (ESI)  $m/z$  calculated for  $\text{C}_{46}\text{H}_{45}\text{FKO}_8\text{S}$   $[\text{M}+\text{K}]^+$  815.2456; found 815.2463.

Preparation of 4-fluorophenyl 6-*O*-acetyl-3,4-di-*O*-benzyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**1c**)



To the solution of **2a** (0.533 mmol, 296 mg) in CH<sub>2</sub>Cl<sub>2</sub> (3.00 mL), DMAP (0.066 mmol, 8.1 mg), pyridine (0.30 mL), and Ac<sub>2</sub>O (1.8 mmol, 0.17 mL) were added at 0 °C and the reaction mixture was stirred at room temperature overnight. After the completion of the reaction determined by TLC (hexane/EtOAc 4:1). The reaction was quenched with 1 N HCl. The reaction mixture was washed with H<sub>2</sub>O for three times and dried over Na<sub>2</sub>SO<sub>4</sub>. After filtration and removal of solvent under reduced pressure the crude product was purified with silica gel chromatography to obtain **1c** in quantitative yield (0.544 mmol, 324 mg) as a white solid. **4-Fluorophenyl 6-*O*-acetyl-3,4-di-*O*-benzyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**1c**)**; TLC (Hexane/EtOAc 4:1): R<sub>f</sub> 0.50; [ $\alpha$ ]<sub>D</sub> = -8.1 (*c* = 1.1, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz)  $\delta$  7.50–7.48 (m, 2 H), 7.33–7.22 (m, 10 H), 6.99 (*pseudo-t*, *J* = 9.0 Hz, 2 H), 5.01 (*pseudo-t*, *J* = 9.6 Hz, 1 H, H-2), 4.78 (d, *J* = 11.4 Hz, 1 H, benzylic-H), 4.77 (d, *J* = 11.4 Hz, 1 H, benzylic-H), 4.70 (d, *J* = 11.4 Hz, 1 H, benzylic-H), 4.55–4.51 (m, 2 H, H-1 and benzylic-H), 4.43 (d, *J* = 12.0 Hz, 1 H, H-6'), 4.17 (dd, *J* = 12.0, 1.2 Hz, 1 H, H-6), 3.75–3.71 (m, 1 H, H-3), 3.58–3.54 (m, 2 H, H-4 and H-5), 2.04 (s, 3 H, Ac), 1.25 (s, 9 H, Piv); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz)  $\delta$  176.7, 170.6, 163.1 (d, *J* = 245.7 Hz, ArS), 137.8, 137.5, 135.9 (d, *J* = 7.5 Hz, ArS), 128.6, 128.5, 128.1, 127.8, 127.4, 127.02, 127.0, 115.9 (d, *J* = 19.5 Hz, ArS), 86.0 (C-1), 84.7 (C-3), 75.4 (benzylic-C), 75.1 (benzylic-C), 71.4 (C-2), 77.1 (C-4 or C-5), 77.0 (C-4 or C-5), 62.9 (C-6), 38.8 (Piv-1C), 27.2 (Piv-3C), 20.9 (Ac); HRMS (ESI) *m/z* calculated for C<sub>33</sub>H<sub>37</sub>FKO<sub>7</sub>S [M+K]<sup>+</sup> 635.1881; found 635.1879.

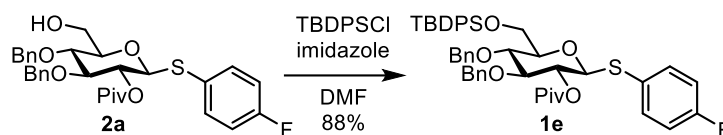
Preparation of 4-fluorophenyl 3,4-di-*O*-benzyl-2,6-di-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**1d**)



To the solution of **2a** (0.542 mmol, 301 mg) were added pyridine (0.30 mL) and DMAP (0.61 mmol, 74 mg) in CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL), and pivaloyl chloride (0.82 mmol, 0.10 mL) was added at room temperature. The reaction mixture was stirred at room temperature overnight. After the completion of the reaction determined by TLC (hexane/EtOAc 4:1), the reaction was quenched with aqueous hydrochloric acid solution. The reaction mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> for three times and dried over Na<sub>2</sub>SO<sub>4</sub>. After filtration and removal of solvent under reduced pressure the crude product was purified with silica gel chromatography to obtain **1d** in 95% yield (0.516 mmol, 329 mg) as a white solid. **4-Fluorophenyl 3,4-di-*O*-benzyl-2,6-di-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**1d**)** TLC (hexane/EtOAc 4:1): R<sub>f</sub> 0.72; [ $\alpha$ ]<sub>D</sub> = -15.1 (*c* = 1.3, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz)  $\delta$  7.49 (dd, *J* = 5.4, 4.4 Hz, 2 H), 7.34–7.27 (m, 6 H), 7.25–7.22 (m, 4 H), 6.98 (*pseudo-t*, *J* = 9.0 Hz, 2 H), 4.99 (d, *J* = 9.4 Hz, 1 H, H-2), 4.78 (d, *J* = 10.7 Hz, 1 H, benzylic-H), 4.75 (d, *J* = 10.9 Hz, 1 H, benzylic-H), 4.68 (d, *J* = 10.9 Hz, 1 H, benzylic-H), 4.54 (d, *J* = 10.0 Hz, 1 H, H-1), 4.53 (d, *J* = 10.7 Hz, 1 H, benzylic-H), 4.47 (dd, *J* = 12.0, 1.7 Hz, 1 H, H-6'), 4.18 (dd, *J* = 12.0, 4.8 Hz, 1 H, H-6), 3.73 (*pseudo-t*, *J* = 8.8 Hz, 1 H, H-3), 3.58 (ddd, *J* = 9.8, 4.8, 1.8 Hz, 1 H, H-5), 3.55 (dd, *J* = 9.7, 8.5 Hz, 1H, H-4), 1.25 (s, 9 H, Piv), 1.21 (s, 9 H, Piv); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz)  $\delta$  178.1 (Piv-CO), 176.6 (Piv-CO), 163.0 (d, *J* = 246.9 Hz, ArS), 137.7, 137.5 (d, *J* = 8.5 Hz, ArS), 128.6,

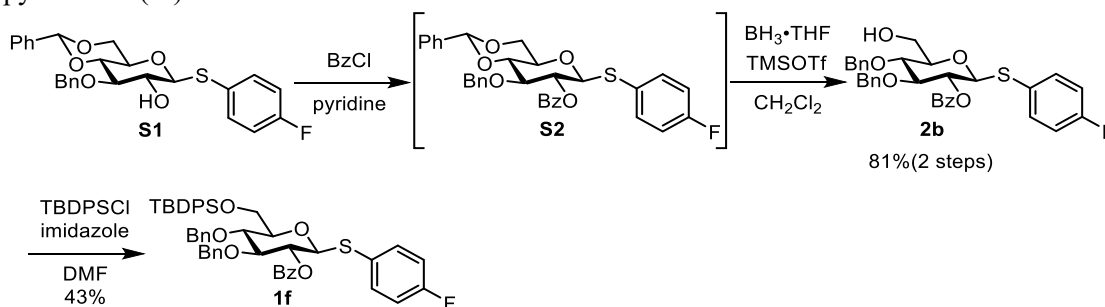
128.5, 128.1, 128.0, 127.8, 127.5, 126.88, 126.86, 115.9 (d,  $J=21.5$  Hz, ArS), 86.0 (C-1), 84.7 (C-3), 77.49 (C-4 or C-5), 77.47 (C-4 or C-5), 75.5 (benzylic-C), 75.3 (benzylic-C), 71.4 (C-2), 62.9 (C-6), 38.9 (Piv-1C), 38.8 (Piv-1C), 27.24 (Piv-3C), 27.22 (Piv-3C); HRMS (ESI)  $m/z$  calculated for  $C_{33}H_{37}FKO_7S$ ,  $[M+K]^+$  677.2345; found 677.2318.

Preparation of 4-Fluorophenyl 3,4-di-*O*-benzyl-6-*O*-*tert*-butyldiphenylsilyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**1e**)



To the solution of **2a** (1.22 mmol, 700 mg) in DMF (4.0 mL), imidazole (2.44 mmol, 166 mg), and *tert*-butylchlorodiphenylsilane (1.83 mmol, 0.469 mL) were added at 0 °C and the reaction mixture was stirred at room temperature overnight. After the completion of the reaction determined by TLC (Hexane/EtOAc 4:1), the reaction was quenched with sat. aqueous  $NaHCO_3$ . The reaction mixture was washed with  $H_2O$  for three times and dried over  $Na_2SO_4$ . After filtration and removal of solvent under reduced pressure the crude product was purified with silica gel chromatography to obtain **1e** in 88% yield (1.10 mmol, 876 mg) as colorless oil. **4-Fluorophenyl 3,4-di-*O*-benzyl-6-*O*-*tert*-butyldiphenylsilyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**1e**)**; TLC (hexane/EtOAc 4:1):  $R_f$  0.69;  $[\alpha]_D = -18.9$  ( $c = 1.1$ ,  $CHCl_3$ );  $^1H$  NMR ( $CDCl_3$ , 600 MHz)  $\delta$  7.77 (dd,  $J = 8.4, 1.2$  Hz, 2 H), 7.70 (dd,  $J = 7.8, 1.2$  Hz, 2 H), 7.52 (dd,  $J = 7.2, 5.4$  Hz, 2 H), 7.43–7.24 (m, 14 H), 7.10 (dd,  $J = 7.8, 4.2$  Hz, 2 H), 6.89 (*pseudo-t*,  $J = 9.0$  Hz, 2 H), 5.09 (*pseudo-t*,  $J = 9.6$  Hz, 1 H, H-2), 4.79 (d,  $J = 10.7$  Hz, 1 H, benzylic-H), 4.78 (d,  $J = 10.9$  Hz, 1 H, benzylic-H), 4.70 (d,  $J = 10.9$  Hz, 1 H, benzylic-H), 4.62 (d,  $J = 10.7$  Hz, 1 H, benzylic-H), 4.58 (d,  $J = 10.1$  Hz, 1 H, H-1), 3.99 (dd,  $J = 11.4, 1.8$  Hz, 1 H, H-6'), 3.93 (dd,  $J = 11.4, 4.2$  Hz, 1 H, H-6), 3.81 (*pseudo-t*,  $J = 9.6$  Hz, 1 H, H-4), 3.74 (*pseudo-t*,  $J = 9.0$  Hz, 1 H, H-3), 3.44 (ddd,  $J = 15.6, 3.6, 1.2$  Hz, 1 H, H-5), 1.26 (s, 9 H, pivaloyl-H), 1.08 (s, 9 H,  $t$ Bu);  $^{13}C$  NMR ( $CDCl_3$ , 150 MHz)  $\delta$  176.8, 174.1, 162.9 (d,  $J = 246.5$  Hz, ArS), 138.1, 138.0, 135.9, 135.7, 135.1 (d,  $J = 8.0$  Hz, ArS), 133.4, 133.0, 129.8, 128.5, 128.03, 128.01, 127.9, 127.8, 127.5, 115.9 (d,  $J = 21.8$  Hz, ArS), 86.6 (C-1), 84.9 (C-3), 80.3 (C-5), 77.3 (C-4), 75.5 (benzylic-C), 75.2 (benzylic-C), 71.6 (C-2), 62.7 (C-6), 38.9 (pivaloyl-1C), 27.3 (pivaloyl-3C), 26.6 ( $t$ Bu-3C), 19.4 ( $t$ Bu-1C); HRMS (ESI)  $m/z$  calculated for  $C_{47}H_{53}FNaO_6SSi$   $[M+Na]^+$  815.3208; found 815.3206.

Preparation of 4-Fluorophenyl 2-*O*-benzoyl-3,4-di-*O*-benzyl-6-*O*-*tert*-butyldiphenylsilyl-1-thio- $\beta$ -D-glucopyranoside (**1f**)

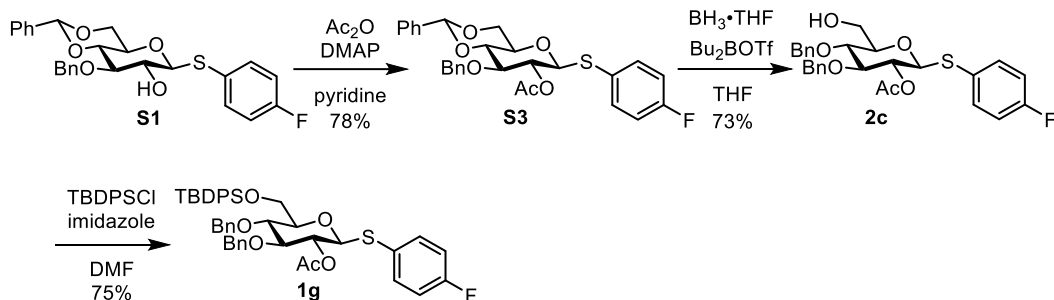


To the solution of **S1** (4.27 mmol, 2.00 g) in pyridine (15.0 mL), benzoyl chloride (4.27 mmol, 2.00 g) was added, and the reaction mixture was stirred at 50 °C overnight. After the completion of the reaction

determined by TLC (Hexane/EtOAc 9:1), the reaction was quenched with 1 N HCl. The reaction mixture was washed with H<sub>2</sub>O for three times and dried over Na<sub>2</sub>SO<sub>4</sub>. Solid was removed by filtration and the reaction mixture was evaporated to obtain crude product of **S2**. To the solution of **S2** in CH<sub>2</sub>Cl<sub>2</sub> (22.0 mL), tetrahydrofuran borane (23.5 mmol, 25.0 mL) was added at 0 °C and the reaction mixture was stirred at 0 °C for 10 min. And trimethylsilyl triflate (0.676 mmol, 0.125 mL) was added at 0 °C and the reaction mixture was stirred at room temperature for 6 h. After the completion of the reaction determined by TLC (eluent: Hexane/EtOAc 4:1). The reaction was quenched with sat. aqueous NaHCO<sub>3</sub>. The reaction mixture was washed with H<sub>2</sub>O for three times and dried over Na<sub>2</sub>SO<sub>4</sub>. After filtration and removal of solvent under reduced pressure the crude product was purified with silica gel chromatography to obtain **2b** in 81% yield (3.44 mmol, 1.98 g) as a white solid. **4-Fluorophenyl 2-O-benzoyl-3,4-di-O-benzyl-1-thio-β-D-glucopyranoside (2b)**; TLC (hexane/EtOAc 4:1): R<sub>f</sub> 0.14; [α]<sub>D</sub> = 37.6 (*c* = 1.1, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz) δ 8.05 (d, *J* = 7.2 Hz, 2 H), 7.59–7.55 (m, 1 H), 7.47–7.42 (m, 4 H), 7.33–7.27 (m, 5 H), 7.11 (bs, 5 H), 6.95 (*pseudo-t*, *J* = 8.4 Hz, 2 H), 5.22 (*pseudo-t*, *J* = 9.6 Hz, 1 H, H-2), 4.84 (d, *J* = 11.0 Hz, 1 H, benzylic-H), 4.74 (d, *J* = 11.4 Hz, 1 H, benzylic-H), 4.73 (d, *J* = 10.8 Hz, 1 H, benzylic-H), 4.64 (d, *J* = 11.0 Hz, 2 H, H-1 and benzylic-H), 3.92 (dd, *J* = 11.9, 2.2 Hz, 1 H, H-6'), 3.86 (*pseudo-t*, *J* = 9.1 Hz, 1 H, H-3), 3.74 (dd, *J* = 12.1, 4.7 Hz, 1 H, H-6), 3.67 (*pseudo-t*, *J* = 9.4 Hz, 1 H, H-4), 3.52–3.48 (m, 1 H, H-5); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz) δ 165.2, 162.6 (d, *J* = 247.2 Hz, ArS), 137.7, 137.6, 135.7 (d, *J* = 8.1 Hz, ArS), 133.4, 130.2, 129.9, 128.6, 128.58, 128.54, 128.37, 128.30, 128.19, 128.13, 127.8, 127.0, 126.9, 116.1 (d, *J* = 21.8 Hz, ArS), 86.1 (C-1), 84.0 (C-3), 79.7 (C-5), 77.4 (C-4), 75.4 (benzylic-C), 75.2 (benzylic-C), 72.4 (C-2), 62.0 (C-6); HRMS (ESI) *m/z* calculated for C<sub>33</sub>H<sub>31</sub>FKO<sub>6</sub>S [M+K]<sup>+</sup> 597.1718; found 597.1714.

To the solution of **2b** (1.39 mmol, 800 mg) in DMF (4.3 mL), imidazole (2.78 mmol, 190 mg), and *tert*-butylchlorodiphenylsilane (2.09 mmol, 0.536 mL) were added at 0 °C and the reaction mixture was stirred at room temperature overnight. After the completion of the reaction determined by TLC (Hexane/EtOAc 4:1), the reaction was quenched with sat. aqueous NaHCO<sub>3</sub>. The reaction mixture was washed with H<sub>2</sub>O for three times and dried over Na<sub>2</sub>SO<sub>4</sub>. After filtration and removal of solvent under reduced pressure the crude product was purified with silica gel chromatography to obtain **1f** in 43% yield (0.602 mmol, 490 mg) as a white solid. **4-Fluorophenyl 2-O-benzoyl-3,4-di-O-benzyl-6-O-tert-butylidiphenylsilyl-1-thio-β-D-glucopyranoside (1f)**; TLC (Hexane/EtOAc 4:1): R<sub>f</sub> 0.59; [α]<sub>D</sub> = 13.1 (*c* = 5.2, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz) δ 8.06 (d, *J* = 7.2 Hz, 2 H), 7.79 (d, *J* = 7.2 Hz, 2 H), 7.71 (d, *J* = 7.2 Hz, 2 H), 7.58 (t, *J* = 7.2 Hz, 1 H), 7.51–7.48 (m, 3 H), 7.47–7.41 (m, 4 H), 7.38 (t, *J* = 7.8 Hz, 2 H), 7.35 (*pseudo-t*, *J* = 7.8 Hz, 2 H), 7.26 (dd, *J* = 6.6, 3.6 Hz, 3 H), 7.15–7.13 (m, 2 H), 6.86 (*pseudo-t*, *J* = 8.4 Hz, 2 H), 5.27 (*pseudo-t*, *J* = 9.0 Hz, 1 H, H-2), 4.86 (d, *J* = 10.7 Hz, 1 H, benzylic-H), 4.75 (d, *J* = 11.0 Hz, 1 H, benzylic-H), 4.71 (d, *J* = 10.0 Hz, 1 H, H-1), 4.67 (d, *J* = 10.7 Hz, 1 H, benzylic-H), 4.64 (d, *J* = 11.0 Hz, 1 H, benzylic-H), 4.02 (dd, *J* = 4.0, 1.6 Hz, 1 H, H-6'), 3.97 (dd, *J* = 11.4, 4.2 Hz, 1 H, H-6), 3.87 (dt, *J* = 14.4, 9.0 Hz, 2 H, H-4), 3.48 (ddd, *J* = 9.3, 3.6, 1.6 Hz, 1 H, H-5), 1.09 (s, 9 H, <sup>t</sup>Bu); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz) δ 165.2, 162.9 (d, *J* = 264.4 Hz, ArS), 137.9, 137.6, 135.9, 135.7, 135.5 (d, *J* = 8.0 Hz, ArS), 133.38, 133.27, 132.9, 130.0, 129.9, 129.8, 128.5, 128.3, 128.2, 128.0, 127.9, 128.8, 127.78, 127.76, 127.4 (d, *J* = 3.1 Hz, ArS), 115.9 (d, *J* = 21.7 Hz, ArS), 86.2 (C-1), 84.4 (C-3), 80.3 (C-5), 77.4 (C-4), 75.5 (benzylic-C), 75.2 (benzylic-C), 72.4 (C-2), 62.6 (C-6), 26.7 (<sup>t</sup>Bu), 19.3 (<sup>t</sup>Bu); HRMS (ESI) *m/z* calculated for C<sub>49</sub>H<sub>49</sub>FNaO<sub>6</sub>SSi [M+Na]<sup>+</sup> 835.2895; found 835.2889.

Preparation of 4-Fluorophenyl 2-O-acetyl-3,4-di-O-benzyl-6-O-*tert*-butylidiphenylsilyl-1-thio-β-D-glucopyranoside (**1g**)

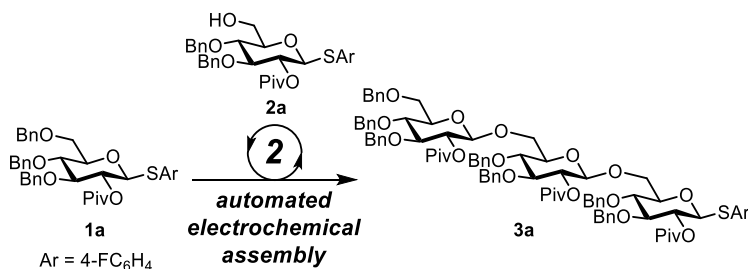


To the solution of **S1** (2.77 mmol, 1.31 g) in pyridine (12.0 mL), DMAP (0.83 mmol, 100 mg) and acetic anhydride (3.6 mmol, 0.34 mL) were added, and the reaction mixture was stirred at 25 °C overnight. After the completion of the reaction determined by TLC (Hexane/EtOAc 4:1), the reaction was quenched with 1 N HCl. The reaction mixture was washed with H<sub>2</sub>O for three times and dried over Na<sub>2</sub>SO<sub>4</sub>. Solid was removed by filtration and the reaction mixture was evaporated to obtain crude product of **S3**. Thus-obtained crude product was purified with silica gel chromatography to obtain **S3** in 78% yield (2.15 mmol, 1.10 g) as a white solid. **4-Fluorophenyl 2-O-acetyl-3-O-benzyl-4,6-O-benzylidene-1-thio-β-D-glucopyranoside (S3)**; TLC (Hexane/EtOAc 4:1): R<sub>f</sub> 0.30; [α]<sub>D</sub> = -5.6 (*c* = 1.0, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz) δ 7.49–7.46 (m, 4 H), 7.41–7.37 (m, 3 H), 7.32–7.25 (m, 4 H), 7.01 (*pseudo-t*, *J* = 8.4 Hz, 2 H), 5.56 (s, 1 H, benzylidene-H), 4.95 (dd, *J* = 10.2, 8.4 Hz, 1 H, H-2), 4.85 (d, *J* = 11.4 Hz, 1 H, benzylic-H), 4.65 (d, *J* = 12.0 Hz, 1 H, benzylic-H), 4.58 (d, *J* = 11.4 Hz, 1 H, H-1), 4.38 (dd, *J* = 10.8, 4.8 Hz, 1 H, H-6'), 3.78 (*pseudo-t*, *J* = 10.8 Hz, 1 H, H-6), 3.74 (*pseudo-t*, *J* = 9.0 Hz, 1 H, H-3), 3.69 (*pseudo-t*, *J* = 9.0 Hz, 1 H, H-4), 3.48 (td, *J* = 10.2, 5.4 Hz, 1 H, H-5), 2.04 (s, 3 H, Ac); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz) δ 169.3 (Ac-CO), 163.1 (*J* = 247.5 Hz, ArS), 138.0, 137.1, 136.1 (d, *J* = 9.0 Hz, ArS), 129.1, 128.6, 128.34, 128.29, 128.1, 127.93, 127.85, 127.7, 126.3 (d, *J* = 3.0 Hz, ArS), 126.0, 116.0 (d, *J* = 21.0 Hz, ArS), 101.2 (benzylidene-C), 86.4 (C-1), 81.3 (C-4), 79.7 (C-3), 74.4 (benzylic-C), 71.2 (C-2), 70.5 (C-5), 68.5 (C-6), 21.0 (Ac); HRMS (ESI) *m/z* calculated for C<sub>28</sub>H<sub>27</sub>FKO<sub>6</sub>S [M+K]<sup>+</sup> 549.1144; found 549.1122.

To the solution of **S3** (2.15 mmol, 1.10 g) in THF (16.0 mL), BH<sub>3</sub>-THF (14.0 mmol, 16.0 mL) was added at 0 °C and the reaction mixture was stirred at 0 °C for 15 min. And 1.0 M dibutyl boron triflate dichloromethane solution (3.0 mmol, 3.0 mL) was added dropwise at 0 °C and the reaction mixture was stirred at room temperature for 3 h. After the completion of the reaction determined by TLC (eluent: Hexane/EtOAc 7:3). The reaction was quenched with sat. aqueous NaHCO<sub>3</sub>. The reaction mixture was washed with H<sub>2</sub>O for three times and dried over Na<sub>2</sub>SO<sub>4</sub>. After filtration and removal of solvent under reduced pressure the crude product was purified with silica gel chromatography to obtain **2c** in 73% yield (1.56 mmol, 0.80 g) as a white solid. **4-Fluorophenyl 2-O-acetyl-3,4-di-O-benzyl-1-thio-β-D-glucopyranoside (2c)**; TLC (Hexane/EtOAc 7:3): R<sub>f</sub> 0.15; [α]<sub>D</sub> = 2.7 (*c* = 1.1, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz) δ 7.46 (dd, *J* = 9.0, 5.4 Hz, 2 H), 7.35–7.25 (m, 9 H), 7.01 (*pseudo-t*, *J* = 8.4 Hz, 2 H), 4.93 (dd, *J* = 9.6, 9.0 Hz, 1 H, H-2), 4.811 (d, *J* = 10.8 Hz, 1 H, benzylic-H), 4.807 (d, *J* = 11.4 Hz, 1 H, benzylic-H), 4.67 (d, *J* = 11.4 Hz, 1 H, benzylic-H), 4.63 (d, *J* = 10.8 Hz, 1 H, benzylic-H), 4.54 (d, *J* = 10.2 Hz, 1 H, H-1), 3.87 (ddd, *J* = 12.0, 5.4, 2.4 Hz, 1 H, H-6'), 3.71–3.67 (m, 2 H, H-3 and H-6), 3.59 (*pseudo-t*, *J* = 9.0 Hz, 1 H, H-4), 3.40 (ddd, *J* = 9.6, 4.2, 2.4 Hz, 1 H, H-5), 2.01 (s, 1 H, Ac), 1.78 (t, *J* = 6.6 Hz, 1 H, OH); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz) δ 169.5 (Ac-CO), 163.0 (d, *J* = 247.5 Hz, ArS), 138.0, 137.7, 135.5 (d, *J* = 7.5 Hz, ArS), 128.6, 128.5, 128.43, 128.35, 128.12, 128.09, 128.0, 127.88, 126.9, 116.1 (d, *J* = 21.0 Hz, ArS), 85.9 (C-1), 84.1 (C-3), 79.6 (C-5), 77.4 (C-4), 75.3 (benzylic-C), 75.2 (benzylic-C), 71.8 (C-2), 61.9 (C-6), 21.0 (Ac); HRMS (ESI) *m/z* calculated for C<sub>28</sub>H<sub>29</sub>FKO<sub>6</sub>S [M+K]<sup>+</sup> 551.1300; found 551.1282.

To the solution of **2c** (0.977 mmol, 501 mg) in DMF (3.0 mL), imidazole (1.95 mmol, 134 mg), and *tert*-butylchlorodiphenylsilane (1.5 mmol, 0.40 mL) were added at 0 °C and the reaction mixture was stirred at room temperature overnight. After the completion of the reaction determined by TLC (Hexane/EtOAc 9:1), the reaction was quenched with sat. aqueous NaHCO<sub>3</sub>. The reaction mixture was diluted with EtOAc and washed with H<sub>2</sub>O for three times and dried over Na<sub>2</sub>SO<sub>4</sub>. After filtration and removal of solvent under reduced pressure the crude product was purified with silica gel chromatography to obtain **1g** in 75% yield (0.732 mmol, 550 mg) as colorless oil. **4-Fluorophenyl 2-O-acetyl-3,4-di-O-benzyl-6-O-*tert*-butyldiphenylsilyl-1-thio-β-D-glucopyranoside (1g)** TLC (Hexane/EtOAc 4:1): R<sub>f</sub> 0.72; [α]<sub>D</sub> = -15.1 (*c* = 1.3, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz) δ 7.75 (dd, *J* = 7.4, 1.3 Hz, 1 H), 7.69 (dd, *J* = 7.4, 1.1 Hz, 2 H), 7.52–7.49 (m, 2 H), 7.42 (td, *J* = 7.4, 1.5 Hz, 2 H), 7.37–7.31 (m, 6 H), 7.30–7.25 (m, 6 H), 7.13–7.11 (m, 2 H), 6.89 (*pseudo-t*, *J* = 8.7 Hz, 2 H), 4.98 (*pseudo-t*, *J* = 9.6 Hz, 1 H, H-2), 4.83 (d, *J* = 11.3 Hz, 1 H, benzylic-H), 4.82 (d, *J* = 10.6 Hz, 1 H, benzylic-H), 4.67 (d, *J* = 11.5 Hz, 1 H, benzylic-H), 4.65 (d, *J* = 10.7 Hz, 1 H, benzylic-H) 4.53 (d, *J* = 10.1 Hz, 1 H, H-1), 3.98 (dd, *J* = 11.4, 1.4 Hz, 1 H, H-6'), 3.92 (dd, *J* = 11.4, 3.8 Hz, 1 H, H-6), 3.81 (*pseudo-t*, *J* = 9.4 Hz, 1 H, H-4), 3.68 (*pseudo-t*, *J* = 9.2 Hz, 1 H, H-3), 3.40 (ddd, *J* = 9.8, 3.5, 1.3 Hz, 1 H, H-5), <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz) δ 169.6 (Ac-CO), 162.8 (d, *J* = 246.9 Hz, ArS), 135.9, 135.7, 135.3 (d, *J* = 8.5 Hz, ArS), 133.3, 132.9, 129.8, 128.54, 128.49, 127.98, 127.89, 127.82, 127.7, 127.5, 127.48, 115.9 (d, *J* = 21.5 Hz, ArS), 86.1 (C-1), 84.5 (C-3), 80.2 (C-5), 77.3 (C-4), 75.5 (benzylic-C), 75.2 (benzylic-C), 71.7 (C-2), 62.5 (C-6), 26.8 (<sup>t</sup>Bu-3C), 21.1 (Ac), 17.3 (<sup>t</sup>Bu-1C); HRMS (ESI) *m/z* calculated for C<sub>44</sub>H<sub>47</sub>FKO<sub>6</sub>SSi, [M+K]<sup>+</sup> 789.2448; found 789.2477.

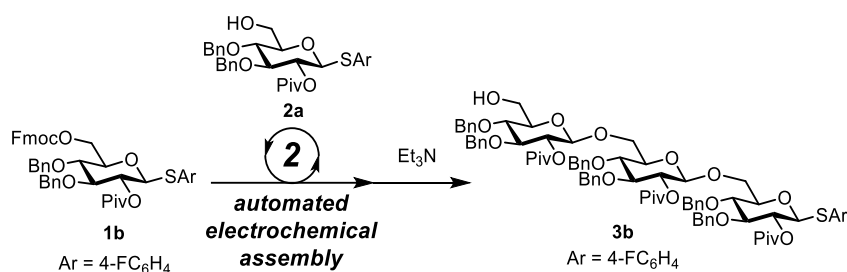
### 3. Synthesis of trisaccharides



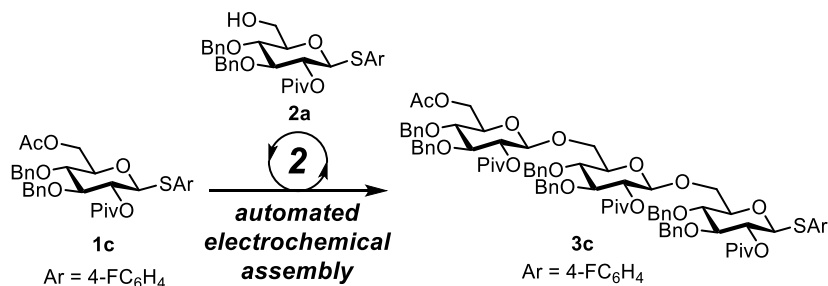
The automated synthesis of trisaccharide **3a** was carried out in an H-type divided cell (4G glass filter) equipped with a carbon felt anode (Nippon Carbon JF-20-P7) and a platinum plate cathode (10 mm×20 mm). In the anodic chamber were placed terminal building block **1a** (0.300 mmol, 196 mg), Bu<sub>4</sub>NOTf (1.55 mmol, 613 mg) and CH<sub>2</sub>Cl<sub>2</sub> (15.0 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.63 mmol, 55 μL), Bu<sub>4</sub>NOTf (1.55 mmol, 611 mg) and CH<sub>2</sub>Cl<sub>2</sub> (15.0 mL). The constant current electrolysis (8.0 mA) was carried out at -80 °C with magnetic stirring until 1.0 F/mol of electricity was consumed. After the electrolysis, building block **2a** (0.210 mmol, 353 mg) dissolved in CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL) was subsequently added by the syringe pump (1.0 mL (0.105 mmol) for one cycle) under an argon atmosphere at -50 °C and kept for 60 min. Then the second cycle started automatically. After the 2nd cycle, Et<sub>3</sub>N (0.5 mL) was added, and the mixture was filtered through a short column (4×3 cm) of silica gel to remove Bu<sub>4</sub>NOTf. Removal of the solvent under reduced pressure and the crude product was purified with silica gel chromatography and the mixture was purified by PR-GPC with CHCl<sub>3</sub> as an eluent and trisaccharide **3a** was obtained in 61% isolated yield (0.184 mmol, 276 mg) as a white solid. **4-Fluorophenyl 3,4,6-tri-O-benzyl-2-O-pivaloyl-β-D-glucopyranosyl-(1→6)-3,4-di-O-benzyl-2-O-pivaloyl-β-D-glucopyranosyl-(1→6)-3,4-di-O-benzyl-2-O-pivaloyl-1-thio-β-D-glucopyranoside (3a)**; TLC



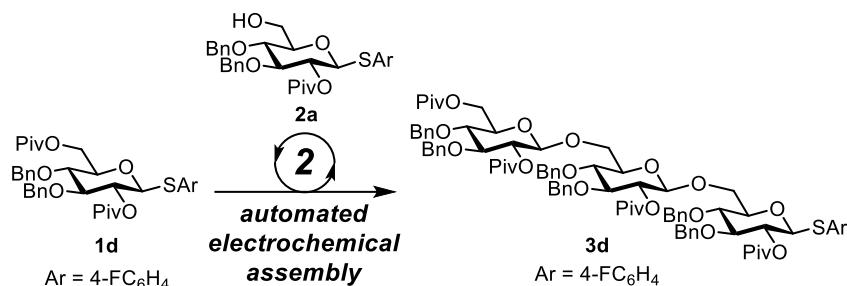
(Hexane/EtOAc 4:1):  $R_f = 0.66$ ;  $[\alpha]_D = -0.6$  ( $c = 1.1$ ,  $\text{CHCl}_3$ );  $^1\text{H NMR}$  ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.50–7.47 (m, 2 H), 7.32–7.21 (m, 3 H), 7.18 (d,  $J = 7.3$  Hz, 2 H), 7.13–7.11 (m, 2 H), 7.03 (*pseudo-t*,  $J = 11.2$  Hz, 2 H), 5.10–5.06 (m, 2 H), 4.99 (*pseudo-t*,  $J = 9.6$  Hz, 1 H), 4.76 (d,  $J = 11.0$  Hz, 1 H), 4.74–4.54 (m, 13 H), 4.51 (d,  $J = 9.8$  Hz, 2 H), 3.97 (dd,  $J = 11.4$ , 1.8 Hz, 1 H), 3.91 (dd,  $J = 11.4$ , 1.2 Hz, 1 H), 3.73–3.55 (m, 12 H), 3.48–3.45 (m, 1 H), 3.40 (*pseudo-t*,  $J = 9.4$  Hz, 1 H), 1.23 (s, 9 H), 1.15, (s, 9 H), 1.13 (s, 9 H);  $^{13}\text{C NMR}$  ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  176.8, 176.73, 176.61, 162.8 (d,  $J = 246.5$  Hz), 138.3, 138.28, 138.21, 138.13, 138.07, 137.9, 135.3 (d,  $J = 8.3$  Hz), 128.6, 128.5, 128.45, 128.41, 128.39, 128.3, 128.1, 128.0, 127.9, 127.78, 127.75, 127.6, 127.4, 127.3, 116.1 (d,  $J = 21.7$  Hz), 101.8, 101.1, 86.5, 84.7, 83.3, 83.27, 79.0, 78.1, 78.0, 75.3, 75.2, 75.09, 74.05, 74.9, 74.89, 74.85, 73.5, 73.2, 72.9, 72.0, 71.5, 69.0, 68.9, 68.3, 38.85, 38.83, 38.80, 27.27, 27.24, 27.22; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{88}\text{H}_{101}\text{FKO}_{18}\text{S}$ ,  $[\text{M}+\text{K}]^+$  1535.6324; found 1535.6309.



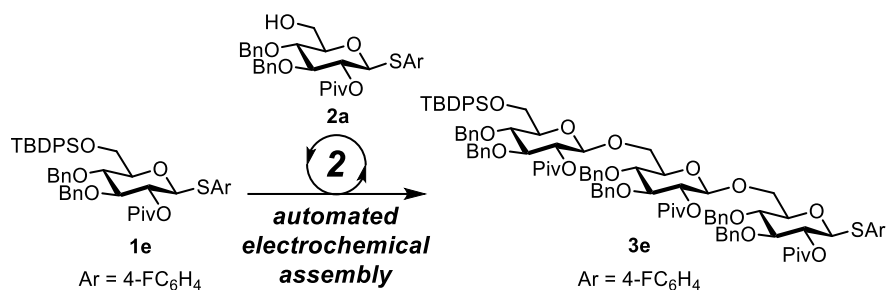
Automated electrochemical glycosylation of building blocks **1b** (0.308 mmol, 239 mg) and **2a** (0.629 mmol, 349 mg) afforded **3b** (0.0178 mmol, 25.0 mg) in 6% yield as a white solid, following the same procedure as that of compound **3a** except deprotection of Fmoc group by adding excess amount of  $\text{Et}_3\text{N}$  (3.0 mL). **4-Fluorophenyl 3,4-di-O-benzyl-2-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-O-benzyl-2-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-O-benzyl-2-O-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (3b)**; TLC (Hexane/EtOAc 4:1):  $R_f$  0.3;  $[\alpha]_D = -6.7$  ( $c = 1.0$ ,  $\text{CHCl}_3$ )  $^1\text{H NMR}$  ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.48 (dd,  $J = 8.4$ , 6.0 Hz, 2 H), 7.34–7.21 (m, 30 H), 7.01 (*pseudo-t*,  $J = 8.4$  Hz, 2 H), 5.04–4.99 (m, 3 H), 4.80–4.68 (m, 8 H), 4.65 (d,  $J = 11.5$  Hz, 1 H), 4.62–4.56 (m, 3 H), 4.41 (d,  $J = 7.7$  Hz, 1 H), 4.39 (d,  $J = 7.9$  Hz, 1 H), 3.86 (d,  $J = 10.8$  Hz, 1 H), 3.82–3.79 (m, 1 H), 3.80 (d,  $J = 10.8$  Hz, 1 H), 3.75 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.70–3.60 (m, 7 H), 3.59 (*pseudo-t*,  $J = 9.0$  Hz, 2 H), 3.49 (dd,  $J = 9.0$  Hz, 1 H), 3.44 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.36 (bs, 1 H), 2.50 (bs, 1 H), 1.23 (s, 9 H), 1.13 (s, 9 H), 1.13 (s, 9 H);  $^{13}\text{C NMR}$  ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  176.8, 176.6, 176.5, 161.9, 138.2, 138.04, 137.97 137.93, 137.7, 135.0 (d,  $J = 8.25$  Hz), 128.48, 128.46, 128.43, 128.31, 128.30, 128.2, 128.1, 127.98, 127.96, 127.9, 127.64, 127.59, 127.55, 127.4, 127.32, 127.30, 116.0 (d,  $J = 21.6$  Hz), 102.1, 101.4, 86.5, 84.6, 83.2, 83.0, 78.7, 77.8, 77.6, 75.8, 75.1, 74.93, 74.89, 74.86, 74.7, 73.2, 73.0, 71.4, 69.6, 68.9, 61.9, 38.79, 38.76, 38.7, 27.2, 27.1; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{81}\text{H}_{95}\text{FKO}_{18}\text{S}$ ;  $[\text{M}+\text{K}]^+$  1445.5860; found 1445.5852.



Automated electrochemical glycosylation of building blocks **1c** (0.100 mmol, 59.7 mg) and **2a** (0.204 mmol, 113 mg) afforded **3c** (0.0239 mmol, 34.7 mg) in 24% yield as a white solid, following the same procedure as that of compound **3a**. **4-Fluorophenyl 6-O-acetyl-3,4-di-O-benzyl-2-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-O-benzyl-2-O-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (3c)**; TLC (Hexane/EtOAc 4:1):  $R_f$  0.27;  $[\alpha]_D = -9.2$  ( $c = 2.8$ ,  $\text{CHCl}_3$ );  $^1\text{H NMR}$  ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.49 (dd,  $J = 8.4, 5.4$  Hz, 2 H), 7.32–7.23 (m, 26 H), 7.21 (d,  $J = 7.2$  Hz, 2 H), 7.18 (d,  $J = 7.2$  Hz, 2 H), 7.04 (*pseudo-t*,  $J = 8.4$  Hz, 2 H), 5.07 (dd,  $J = 9.0, 7.8$  Hz, 1 H), 5.04 (dd,  $J = 9.6, 7.8$  Hz, 1 H), 4.98 (*pseudo-t*,  $J = 10.2$  Hz, 1 H), 4.78 (d,  $J = 11.4$  Hz, 1 H), 4.75–4.65 (m, 8 H), 4.58 (dd,  $J = 10.8, 5.4$  Hz, 2 H), 4.53 (d,  $J = 10.2$  Hz, 1 H), 4.44–4.42 (m, 3 H), 4.28 (dd,  $J = 11.8, 1.8$  Hz, 1 H), 4.19 (dd,  $J = 12.0, 5.4$  Hz, 1 H), 3.96 (d,  $J = 10.2$  Hz, 1 H), 3.89 (d,  $J = 9.6$  Hz, 1 H), 3.73 (*pseudo-t*,  $J = 8.4$  Hz, 1 H), 3.67–3.60 (m, 5 H), 3.55 (*pseudo-t*,  $J = 9.6$  Hz, 2 H), 3.52–3.47 (m, 2 H), 3.40 (*pseudo-t*,  $J = 9.6$  Hz, 1 H), 1.96 (s, 3 H), 1.23 (s, 9 H), 1.15 (s, 9 H), 1.13 (s, 9 H);  $^{13}\text{C NMR}$  ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  176.71, 176.67, 176.5, 170.7, 162.8 (d,  $J = 246.0$  Hz), 138.04, 138.02, 137.95, 137.9, 137.8, 137.6, 135.3 (d,  $J = 9.0$  Hz), 128.5, 128.4, 128.3, 128.2, 128.1, 128.05, 128.01, 127.68, 127.66, 127.6, 127.4, 127.4, 127.2, 116.0 (d,  $J = 22.5$  Hz), 101.3, 101.0, 86.4, 84.7, 83.2, 83.2, 79.0, 78.1, 77.9, 77.5, 75.3, 74.9, 74.9, 74.8, 73.2, 72.9, 71.4, 68.6, 68.2, 63.2, 38.79, 38.78, 38.75, 27.18, 27.17, 27.15, 20.9; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{83}\text{H}_{97}\text{FKO}_{19}\text{S}$ ;  $[\text{M}+\text{K}]^+$  1487.5966; found 1487.5952.



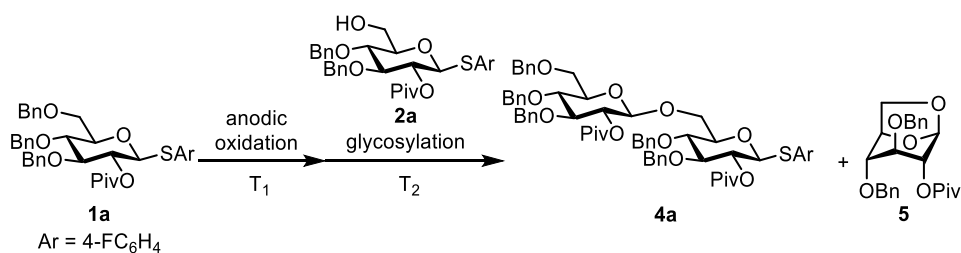
Automated electrochemical glycosylation of building blocks **1d** (0.106 mmol, 67.5 mg) and **2a** (0.211 mmol, 117 mg) afforded **3b** (0.0383 mmol, 57.1 mg) in 36% yield as a white solid, following the same procedure as that of compound **3a**. **4-Fluorophenyl 3,4-di-O-benzyl-2,6-di-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-O-benzyl-2-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-O-benzyl-2-O-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (3d)** TLC (Hexane/EtOAc 4:1):  $R_f$  0.42;  $[\alpha]_D = -11.0$  ( $c = 1.2$ ,  $\text{CHCl}_3$ );  $^1\text{H NMR}$  ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.48 (dd,  $J = 9.0, 5.4$  Hz, 2 H), 7.33–7.16 (m, 30 H), 7.04 (*pseudo-t*,  $J = 8.4$  Hz, 2 H), 5.07–5.02 (m, 4 H), 4.98 (*pseudo-t*,  $J = 9.6$  Hz, 1 H), 4.79–4.63 (m, 9 H), 4.57 (d,  $J = 12.0$  Hz, 1 H), 4.47–4.43 (m, 3 H), 4.38 (dd,  $J = 12.0, 1.8$  Hz, 1 H), 4.16 (dd,  $J = 12.4, 4.8$  Hz, 1 H), 3.97 (d,  $J = 10.7$  Hz, 1 H), 3.90 (dd,  $J = 11.3, 1.4$  Hz, 1 H), 3.74 (*pseudo-t*,  $J = 9.0$  Hz, 1H), 3.67–3.61 (m, 5 H), 3.57–3.54 (m, 2 H), 3.46–3.44 (m, 1 H), 3.41 (*pseudo-t*,  $J = 9.1$  Hz, 1 H), 1.23 (s, 9 H), 1.17 (s, 9 H), 1.15 (s, 9 H), 1.12 (s, 9 H);  $^{13}\text{C NMR}$  ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  178.1, 176.71, 176.66, 176.6, 162.8 (d,  $J = 246.3$  Hz), 138.04, 137.93, 137.79, 137.69, 135.2 (d,  $J = 7.6$  Hz), 128.6, 128.5, 128.48, 128.39, 128.35, 128.2, 127.98, 127.94, 127.91, 127.7, 127.64, 127.58, 127.45, 127.4, 127.21, 116.1 (d,  $J = 21.7$  Hz), 101.27, 101.0, 86.4, 84.7, 83.2, 83.1, 79.0, 78.2, 77.9, 77.8, 77.3, 77.1, 76.9, 75.3, 75.1, 74.99, 74.96, 74.8, 73.5, 72.8, 38.79, 38.76, 27.3, 27.2, 27.18, 27.15; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{86}\text{H}_{103}\text{FNaO}_{19}\text{S}$ ;  $[\text{M}+\text{Na}]^+$  1513.6691; found 1513.6671.



Automated electrochemical glycosylation of building blocks **1e** (0.300 mmol, 238 mg) and **2a** (0.633 mmol, 351 mg) afforded **3e** (0.0875 mmol, 144 mg) in 29% yield as a white solid, following the same procedure as that of compound **3e**. **4-Fluorophenyl 3,4-di-O-benzyl-6-O-tert-butyl-diphenylsilyl-2-O-pivaloyl-β-D-glucopyranosyl-(1→6)-3,4-di-O-benzyl-2-O-pivaloyl-1-thio-β-D-glucopyranosyl-(1→6)-3,4-di-O-benzyl-2-O-pivaloyl-1-thio-β-D-glucopyranoside (3e)**; TLC (Hexane/EtOAc 4:1);  $R_f$  0.56;  $[\alpha]_D = -13.3$  ( $c = 1.0$ ,  $\text{CHCl}_3$ )  $^1\text{H NMR}$  ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.71 (dd,  $J = 7.2, 3.6$  Hz, 2 H), 7.67 (dd,  $J = 7.8, 1.2$  Hz, 2 H), 7.49 (dd,  $J = 9.0, 5.4$  Hz, 2 H), 7.38 (tt,  $J = 7.2, 1.8$  Hz, 1 H), 7.36–7.15 (m, 33 H), 7.08 (dd,  $J = 7.2, 4.2$  Hz, 2 H), 7.03 (*pseudo-t*,  $J = 9.0$  Hz, 2 H), 5.11–5.05 (m, 2 H), 4.97 (*pseudo-t*,  $J = 9.6$  Hz, 1 H), 4.77–4.46 (m, 16 H), 4.02 (*pseudo-t*,  $J = 9.0$  Hz, 2 H), 3.89–3.83 (m, 2 H), 3.75–3.60 (m, 7 H), 3.45 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.41 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.31–3.28 (m, 1 H), 1.22 (s, 9 H), 1.17 (s, 9 H), 1.13 (s, 9 H), 1.03 (s, 9 H);  $^{13}\text{C NMR}$  ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  176.70, 176.68, 162.9 (d,  $J = 246.5$  Hz), 138.2, 138.1, 137.9, 137.8, 135.8, 135.6, 135.3 (d,  $J = 8.25$  Hz), 129.8, 129.7, 128.49, 128.45, 128.4, 128.3, 128.2, 128.1, 127.9, 127.84, 127.79, 127.76, 127.7, 127.65, 127.59, 127.5, 127.4, 127.2, 116.1 (d,  $J = 21.6$  Hz), 101.3, 101.0, 86.4, 84.7, 83.30, 83.27, 79.1, 78.5, 77.9, 77.7, 77.3, 77.1, 76.9, 76.3, 75.33, 75.25, 75.0, 75.0, 74.9, 74.8, 73.1, 72.9, 71.4, 68.1, 67.9, 38.85, 38.82, 38.79, 27.3, 27.2, 26.9, 19.3; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{97}\text{H}_{113}\text{FNaO}_{18}\text{SSi}$ ;  $[\text{M}+\text{Na}]^+$  1667.7293; found 1667.7330.

#### 4. Synthesis of disaccharides

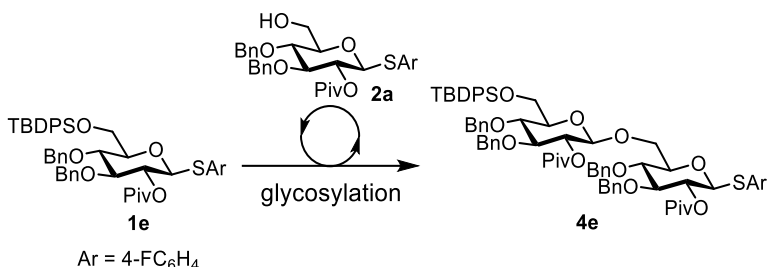
Preparation of 4-Fluorophenyl 3,4,6-tri-O-benzyl-2-O-pivaloyl-β-D-glucopyranosyl-(1→6)-3,4-di-O-benzyl-2-O-pivaloyl-1-thio-β-D-glucopyranoside (**4a**)



The automated synthesis of trisaccharide **4a** was carried out in an H-type divided cell (4G glass filter) equipped with a carbon felt anode (Nippon Carbon JF-20-P7) and a platinum plate cathode (10 mm×20 mm). In the anodic chamber were placed terminal building block **1a** (0.300 mmol, 193 mg),  $\text{Bu}_4\text{NOTf}$  (1.48 mmol, 578 mg) and  $\text{CH}_2\text{Cl}_2$  (15.0 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.296 mmol, 26  $\mu\text{L}$ ),  $\text{Bu}_4\text{NOTf}$  (1.51 mmol, 590 mg) and  $\text{CH}_2\text{Cl}_2$  (15.0 mL). The constant current electrolysis (12.0 mA) was carried out at  $-40$  °C with magnetic stirring until 1.2 F/mol of electricity was consumed. After the electrolysis, building block **2a** (0.315 mmol, 176 mg) dissolved in  $\text{CH}_2\text{Cl}_2$  (2.0 mL)

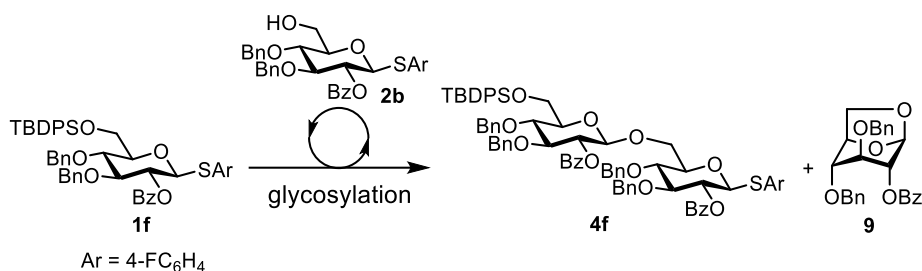
was subsequently added by the syringe pump under an argon atmosphere at  $-40\text{ }^{\circ}\text{C}$ , and kept for 60 min. After the cycle,  $\text{Et}_3\text{N}$  (0.5 mL) was added, and the mixture was filtered through a short column (4×3 cm) of silica gel to remove  $\text{Bu}_4\text{NOTf}$ . Removal of the solvent under reduced pressure and the crude product was purified with silica gel chromatography to obtain **4a** in 61% isolated yield (0.183 mmol, 196 mg) as a white solid. **4-Fluorophenyl 3,4,6-tri-O-benzyl-2-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-O-benzyl-2-O-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (4a)** TLC (hexane/EtOAc 4:1):  $R_f$ ;  $[\alpha]_D = -12.6$  ( $c = 1.5$ ,  $\text{CHCl}_3$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.48 (dd,  $J = 8.8, 5.3$  Hz, 2 H), 7.40–7.24 (m, 22 H), 7.23–7.21 (m, 4 H), 7.15 (dd,  $J = 7.3, 2.4$  Hz, 2 H), 7.04 (*pseudo-t*,  $J = 8.8$  Hz, 2 H), 5.10 (dd,  $J = 9.2, 8.0$  Hz, 1 H), 5.01 (dd,  $J = 9.8, 9.4$  Hz, 1 H), 4.77–4.71 (m, 4 H), 4.69 (d,  $J = 11.1$  Hz, 1 H), 4.66 (d,  $J = 11.0$  Hz, 1 H), 4.58–4.52 (m, 3 H), 4.49–4.47 (m, 2 H), 4.46 (d,  $J = 7.9$  Hz, 1 H), 4.00 (dd,  $J = 11.2, 1.5$  Hz, 1 H), 3.75–3.70 (m, 2 H), 3.70–3.65 (m, 4 H), 3.61–3.57 (m, 1 H), 3.48 (ddd,  $J = 9.7, 4.4, 1.9$  Hz, 1 H), 3.44 (*pseudo-t*,  $J = 9.3$  Hz, 1 H), 1.23 (s, 9 H), 1.13 (s, 9 H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  176.7, 162.8 (d,  $J = 246.7$  Hz), 138.13, 138.09, 138.0, 137.8, 135.0 (d,  $J = 8.4$  Hz), 128.45, 128.43, 128.40, 128.05, 128.01, 127.9, 127.83, 127.81, 127.67, 127.64, 127.5, 127.3, 116.2 (d,  $J = 21.7$  Hz), 101.2, 86.7, 84.6, 83.4, 78.3, 77.8, 77.5, 75.23, 75.21, 75.0, 74.95, 73.5, 72.9, 71.4, 68.8, 68.1, 38.8, 38.77, 27.20, 27.15; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{63}\text{H}_{71}\text{FKO}_{12}\text{S}$ ,  $[\text{M}+\text{K}]^+$  1109.4283; found 1109.4279.

1,6-Anhydrosugar **5** was obtained as a major product (33% yield) when the reaction was performed at elevated temperature ( $T_1 = T_2 = 0\text{ }^{\circ}\text{C}$ ).  $^1\text{H}$  NMR and HRMS of **5** were compared with those reported in a literature to confirm its formation.<sup>10</sup> **1,6-Anhydro-3,4-di-O-benzyl-2-O-pivaloyl- $\beta$ -D-glucopyranose (5)**  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.33–7.25 (m, 10 H), 5.42 (s, 1 H), 4.78 (d,  $J = 12.0$  Hz, 1 H), 4.71 (s, 1 H), 4.62 (d,  $J = 5.4$  Hz, 1 H), 4.57 (d,  $J = 12.0$  Hz, 1 H), 4.50 (d,  $J = 12.6$  Hz, 1 H), 4.37 (d,  $J = 12.0$  Hz, 1 H), 4.04 (d,  $J = 7.2$  Hz, 1 H), 3.72 (*pseudo-t*,  $J = 6.0$  Hz, 1 H), 3.50 (brs, 1 H), 3.33 (brs, 1 H), 1.22 (s, 9 H); HRMS (ESI)  $m/z$  calculated for  $\text{C}_{25}\text{H}_{30}\text{NaO}_6$ ,  $[\text{M}+\text{Na}]^+$  465.1674; found 465.1669.



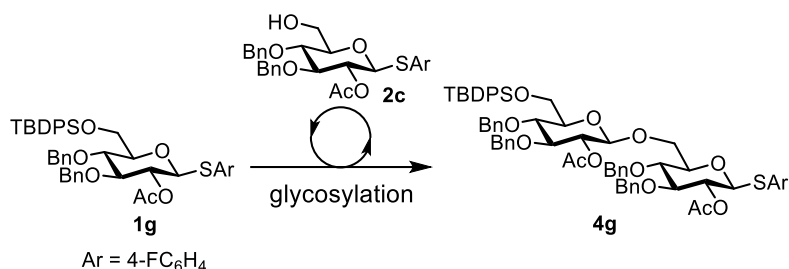
Automated electrochemical glycosylation of building blocks **1e** (0.299 mmol, 237 mg) and **2a** (0.315 mmol, 176 mg) afforded **3e** (0.216 mmol, 263 mg) in 72% yield as a white solid, following the same procedure as that of compound **4a**. **4-Fluorophenyl 3,4-di-O-benzyl-6-O-tert-butylidiphenylsilyl-2-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-O-benzyl-2-O-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (4e)**; TLC (Hexane/EtOAc 4:1);  $R_f$  0.64;  $[\alpha]_D = -11.7$  ( $c = 1.3$ ,  $\text{CHCl}_3$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.72 (dd,  $J = 6.6, 3.6$  Hz, 2 H), 7.67 (dd,  $J = 7.8, 1.2$  Hz, 2 H), 7.49 (dd,  $J = 8.4, 4.8$  Hz, 2 H), 7.39 (t,  $J = 7.8$  Hz, 2 H), 7.33–7.23 (m, 18 H), 7.21 (d,  $J = 7.2$  Hz, 2 H), 7.18 (dd,  $J = 6.0, 2.4$  Hz, 2 H), 7.15 (dd,  $J = 6.0, 2.4$  Hz, 2 H), 7.03 (*pseudo-t*,  $J = 9.0$  Hz, 2 H), 5.09 (dd,  $J = 9.6, 8.4$  Hz, 1 H), 4.97 (*pseudo-t*,  $J = 9.6$  Hz, 1 H), 4.82 (d,  $J = 20.4$  Hz, 1 H), 4.80 (d,  $J = 20.4$  Hz, 1 H), 4.75–4.65 (m, 5 H), 4.53–4.48 (m, 3 H), 4.01 (d,  $J = 10.2$  Hz, 1 H), 3.95–3.90 (m, 2 H), 3.87 (*pseudo-t*,  $J = 9.6$  Hz, 1 H), 3.72–3.61 (m, 4 H), 3.36 (ddd,  $J = 9.6, 3.0, 1.8$  Hz, 1 H), 3.32 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 1.23 (s, 9 H), 1.16 (s, 9 H), 1.04 (s, 9 H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  176.8, 176.6, 162.1, 138.2, 138.1, 137.9, 137.7, 135.9, 135.6, 135.4 (d,  $J = 8.1$  Hz), 133.6,

133.0, 129.69, 129.67, 128.43, 128.41, 128.39, 127.93, 127.88, 127.8, 127.7, 127.6, 127.5, 127.3, 116.2 (d,  $J = 21.8$  Hz), 101.1, 86.3, 84.7, 83.4, 79.7, 78.0, 77.6, 76.2, 75.2, 75.2, 75.1, 74.9, 73.1, 71.4, 67.8, 62.7, 38.8, 29.7, 27.2, 26.9, 19.3; HRMS (ESI)  $m/z$  calculated for  $C_{72}H_{83}FNaO_{12}SSi$ ;  $[M+Na]^+$  1241.5251; found 1241.5277.



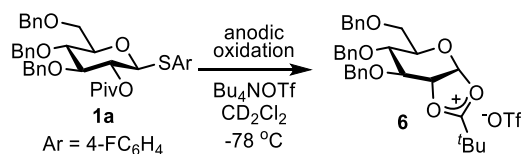
Automated electrochemical glycosylation of building blocks **1f** (0.299 mmol, 243 mg) and **2b** (0.315 mmol, 182 mg) afforded **4f** (0.194 mmol, 245 mg) in 65% yield as a white solid, following the same procedure as that of compound **4a**. **4-Fluoropheny 2-O-benzoyl-3,4-di-O-benzyl-6-O-tert-butylidiphenylsilyl-β-D-glucopyranosyl-(1→6)-2-O-benzoyl-3,4-di-O-benzyl-1-thio-β-D-glucopyranoside (4f)**. TLC (Hexane/EtOAc 4:1):  $R_f = 0.39$ ;  $[\alpha]_D = 10.4$  ( $c = 12.1$ , CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz)  $\delta$  8.00 (d,  $J = 8.4$  Hz, 2 H), 7.96 (d,  $J = 8.3$  Hz, 2 H), 7.77 (dd,  $J = 7.8$ , 1.2 Hz, 2 H), 7.71 (dd,  $J = 7.9$ , 1.2 Hz, 2 H), 7.58 (*pseudo-t*,  $J = 7.4$  Hz, 1 H), 7.46–7.34 (m, 11 H), 7.30–7.28 (m, 5 H), 7.25–7.21 (m, 5 H), 7.15–7.13 (m, 5 H), 7.10–7.06 (m, 5 H), 6.95 (*pseudo-t*,  $J = 8.7$  Hz, 1 H), 5.37 (dd,  $J = 8.4$ , 8.1 Hz, 1 H), 5.06 (*pseudo-t*,  $J = 9.3$  Hz, 1 H), 4.91 (d,  $J = 10.8$  Hz, 1 H), 4.78 (d,  $J = 11.0$  Hz, 1 H), 4.75 (d,  $J = 10.9$  Hz, 1 H), 4.69 (d,  $J = 11.5$  Hz, 1 H), 4.63 (d,  $J = 7.8$  Hz, 1 H), 4.57 (d,  $J = 11.0$  Hz, 1 H), 4.56 (d,  $J = 9.8$  Hz, 1 H), 4.49 (d,  $J = 11.0$  Hz, 1 H), 4.47 (d,  $J = 11.1$  Hz, 1 H), 4.38 (d,  $J = 11.4$  Hz, 1 H), 4.00–3.97 (m, 3 H), 3.84 (*pseudo-t*,  $J = 9.2$  Hz, 1 H), 3.70 (*pseudo-t*,  $J = 8.9$  Hz, 1 H), 3.65 (dd,  $J = 11.6$  Hz, 6.0 Hz, 1 H), 3.58–3.56 (m, 1 H), 3.44 (d,  $J = 9.7$  Hz, 1 H), 3.40 (*pseudo-t*,  $J = 9.2$  Hz, 1 H), 1.06 (s, 9 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz)  $\delta$  165.2, 165.0, 163.0 (d,  $J = 246.9$  Hz), 138.1, 137.84, 137.83, 137.6, 136.0, 135.9 (d,  $J = 7.4$  Hz), 135.6, 133.6, 123.7, 133.2, 132.99, 132.96, 130.0, 129.9, 129.7, 128.5, 128.48, 128.39, 128.36, 128.2, 128.16, 128.1, 127.99, 127.88, 127.83, 127.8, 127.74, 127.65, 126.65, 126.63, 116.0 (d,  $J = 21.7$  Hz), 101.2, 85.6, 84.13, 83.0, 79.1, 77.8, 77.5, 77.3, 77.1, 76.9, 76.2, 75.3, 75.2, 75.1, 74.8, 73.9, 72.0, 67.7, 62.6, 26.9, 19.4; HRMS (ESI)  $m/z$  calculated for  $C_{76}H_{75}FKO_{12}SSi$ ,  $[M+K]^+$  1297.4364; found 1297.4343.

1,6-Anhydrosugar **9** was obtained as a minor product (7%) when the reaction was performed at elevated temperature ( $T_1 = T_2 = 0$  °C). <sup>1</sup>H NMR and HRMS of **9** were compared with those reported in a literature to confirm its formation.<sup>11</sup> **1,6-Anhydro-2-O-benzoyl-3,4-di-O-benzyl-β-D-glucopyranose (9)** <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz)  $\delta$  8.10–8.06 (m, 2 H), 7.60–7.55 (m, 1 H), 7.45–7.40 (m, 2 H), 7.35–7.23 (m, 10 H), 5.59 (s, 1 H), 5.01 (s, 1 H), 4.83 (d,  $J = 12.0$  Hz, 1 H), 4.69 (d,  $J = 5.4$  Hz, 1 H), 4.61 (d,  $J = 12.0$  Hz, 1 H), 4.50 (d,  $J = 12.0$  Hz, 1 H), 4.46 (d,  $J = 12.0$  Hz, 1 H), 4.13 (d,  $J = 7.2$  Hz, 1 H), 3.78 (*pseudo-t*,  $J = 6.6$  Hz, 1 H), 3.70 (*pseudo-t*,  $J = 1.8$  Hz, 1 H), 3.42 (s, 1 H); HRMS (ESI)  $m/z$  calculated for  $C_{27}H_{26}NaO_6$ ,  $[M+Na]^+$  469.1622; found 469.1606.



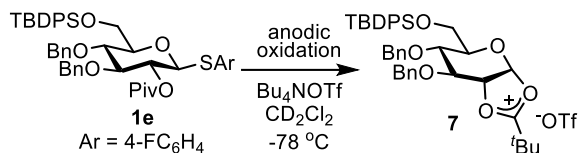
Automated electrochemical glycosylation of building blocks **1g** (0.305 mmol, 229 mg) and **2c** (0.315 mmol, 161 mg) afforded **4g** (0.172 mmol, 195 mg) in 56% yield as a white solid, following the same procedure as that of compound **4a**. **4-Fluoropheny 2-O-acetyl-3,4-di-O-benzyl-6-O-tert-butylidiphenylsilyl-β-D-glucopyranosyl-(1→6)-2-O-acetyl-3,4-di-O-benzyl-1-thio-β-D-glucopyranoside (4g)**. TLC (Hexane/EtOAc 4:1):  $R_f$ ;  $[\alpha]_D = 0.3$  ( $c = 4.1$ , CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz)  $\delta$  7.74 (d,  $J = 7.3$  Hz, 2 H), 7.68 (d,  $J = 7.7$  Hz, 2 H), 7.47 (dd,  $J = 7.9, 5.4$  Hz, 2 H), 7.41–7.25 (m, 22 H), 7.24–7.19 (m, 6 H), 5.28 (s, 1 H), 5.06 (*pseudo-t*,  $J = 8.6$  Hz, 1 H), 4.89–4.87 (m, 2 H), 4.84 (d,  $J = 11.3$  Hz, 1 H), 4.76 (d,  $J = 11.0$  Hz, 1 H), 4.75 (d,  $J = 11.3$  Hz, 1 H), 4.73 (d,  $J = 11.3$  Hz, 1 H), 4.69 (d,  $J = 11.3$  Hz, 1 H), 4.64 (d,  $J = 11.3$  Hz, 1 H), 4.55 (d,  $J = 11.1$  Hz, 1 H), 4.50 (d,  $J = 10.0$  Hz, 1 H), 4.42 (d,  $J = 8.0$  Hz, 1 H), 4.13 (d,  $J = 9.8$  Hz, 1 H), 3.95–3.90 (m, 3 H), 3.68–3.64 (m, 2 H), 3.60–3.55 (m, 2 H), 3.39 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.34 (d,  $J = 9.6$  Hz, 1 H), 1.99 (s, 3 H), 1.96 (s, 3 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz)  $\delta$  169.45, 169.42, 162.9 (d,  $J = 247.1$  Hz), 137.9, 137.7, 135.9, 135.6, 135.3 (d,  $J = 8.5$  Hz), 133.6, 132.9, 129.71, 129.70, 128.52, 128.49, 128.0, 127.97, 127.94, 127.87, 127.84, 127.8, 127.7, 127.1, 116.1 (d,  $J = 21.8$  Hz), 85.8, 84.3, 83.1, 79.0, 77.8, 77.7, 76.0, 75.4, 75.3, 75.2, 75.0, 73.1, 71.5, 67.8, 62.4, 53.5, 26.8, 21.0, 19.3; HRMS (ESI)  $m/z$  calculated for C<sub>66</sub>H<sub>71</sub>FN<sub>2</sub>O<sub>12</sub>SSi,  $[M+Na]^+$  1157.4312; found 1157.4324.

## 5. NMR analysis of glycosyl dioxalenium ions

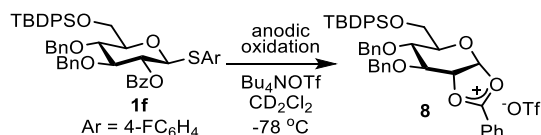


The anodic oxidation of thioglycoside **1a** was carried out in an H-type divided cell (4G glass filter) equipped with a carbon felt anode (Nippon Carbon JF-20-P7) and a platinum plate cathode (10 mm×20 mm). In the anodic chamber were placed terminal building block **1a** (0.100 mmol, 64.7 mg), Bu<sub>4</sub>NOTf (0.49 mmol, 191 mg) and CD<sub>2</sub>Cl<sub>2</sub> (5.0 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.091 mmol, 8  $\mu$ L), Bu<sub>4</sub>NOTf (0.50 mmol, 195 mg) and CD<sub>2</sub>Cl<sub>2</sub> (5.0 mL). The constant current electrolysis (4.0 mA) was carried out at -80 °C with magnetic stirring until 1.05 F/mol of electricity was consumed. After the electrolysis, the reaction mixture of anodic chamber was transferred to 5 mm  $\phi$  NMR tubes with a septum cap under argon atmosphere at -78 °C. The NMR measurement was carried out at low temperature. Chemical shifts were reported using signals of CH<sub>2</sub>Cl<sub>2</sub> at 5.32 ppm for <sup>1</sup>H NMR and CD<sub>2</sub>Cl<sub>2</sub> at 53.8 ppm for <sup>13</sup>C NMR as standard. Selected data for **3,4,6-tri-O-benzyl-D-glucopyranosyl pivaloxonium triflate (6)**. <sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>, 500 MHz, at -40 °C)  $\delta$  5.77 (d,  $J = 7.5$  Hz, 1 H, H-2), 4.54 (d,  $J = 12.0$  Hz, 1 H, benzylic-H), 4.47 (d,  $J = 11.5$  Hz, 1 H, benzylic-H), 4.46 (d,  $J = 12.0$  Hz, 1 H, benzylic-H), 4.43 (d,  $J = 11.5$  Hz, 1 H, benzylic-H), 4.39 (d,  $J = 11.5$  Hz, 1 H, benzylic-H), 4.32 (d,  $J = 11.5$  Hz, 1 H, benzylic-H), 3.99 (brs, 1 H, H-3), 3.86 (dd,  $J = 10.0, 5.5$  Hz, 1 H, H-5), 3.74 (brs, 1 H, H-4), 3.58 (dd,  $J$

= 10.5, 5.5 Hz, 1 H, H-6), 3.49 (dd,  $J = 10.0, 5.0$  Hz, 1 H, H-6);  $^{13}\text{C}$  NMR ( $\text{CD}_2\text{Cl}_2$ , 125 MHz, at  $-80^\circ\text{C}$ )  $\delta$  197.1 (cationic-C), 111.5 (C-1), 77.5 (C-2), 74.1 (C-5), 72.6 (benzylic-C), 71.6 (benzylic-C), 71.5 (benzylic-C), 69.1 (C-4), 67.8 (C-3 and C-6).



**3,4-Di-*O*-benzyl-6-*O*-*tert*-butyldiphenylsilyl-D-glucopyranosyl pivaloxonium triflate (7).** Anodic oxidation of thioglycoside **1e** (0.102 mmol, 81.0 mg) afforded **7**. Selected data for **7**.  $^1\text{H}$  NMR ( $\text{CD}_2\text{Cl}_2$ , 500 MHz, at  $-40^\circ\text{C}$ )  $\delta$  5.79 (d,  $J = 7.0$  Hz, 1 H, H-2), 4.60 (d,  $J = 12.0$  Hz, 1 H, benzylic-H), 4.51 (d,  $J = 12.0$  Hz, 1 H, benzylic-H), 4.46 (d,  $J = 11.0$  Hz, 1 H, benzylic-H), 4.28 (d,  $J = 11.5$  Hz, 1 H, benzylic-H), 4.00 (brs, 1 H, H-3), 3.87 (brs, 1 H, H-4), 3.76–3.71 (m, 1 H, H-6), 3.67–3.62 (m, 2 H, H-5 and H-6);  $^{13}\text{C}$  NMR ( $\text{CD}_2\text{Cl}_2$ , 125 MHz, at  $-80^\circ\text{C}$ )  $\delta$  196.9 (cationic-C), 112.2 (C-1), 78.4 (C-2), 75.1 (C-5), 72.0 (benzylic-C), 71.6 (benzylic-C), 69.7 (C-4), 69.3 (C-3), 61.3 (C-6).



**3,4-Di-*O*-benzyl-6-*O*-*tert*-butyldiphenylsilyl-D-glucopyranosyl benzoxonium triflate (8).** Anodic oxidation of thioglycoside **1f** (0.101 mmol, 82.2 mg) afforded **8**. Selected data for **8**.  $^1\text{H}$  NMR ( $\text{CD}_2\text{Cl}_2$ , 500 MHz, at  $-80^\circ\text{C}$ )  $\delta$  5.89 (d,  $J = 7.0$  Hz, 1 H, H-2), 4.70 (d,  $J = 11.5$  Hz, 1 H, benzylic-H), 4.59 (d,  $J = 12.0$  Hz, 1 H, benzylic-H), 4.36 (d,  $J = 10.5$  Hz, 1 H, benzylic-H), 4.27 (d,  $J = 10.5$  Hz, 1 H, benzylic-H), 4.26 (brs, 1 H, H-3), 4.07 (brs, 2 H, H-4 and H-5), 3.88–3.80 (m, 1 H, H-6), 3.76–3.68 (m, 1 H, H-6);  $^{13}\text{C}$  NMR ( $\text{CD}_2\text{Cl}_2$ , 125 MHz, at  $-80^\circ\text{C}$ )  $\delta$  180.2 (cationic-C), 110.7 (C-1), 77.4 (C-2), 75.2 (C-5), 72.2 (benzylic-C), 71.8 (benzylic-C), 70.6 (C-4), 69.5 (C-3), 61.6 (C-6).

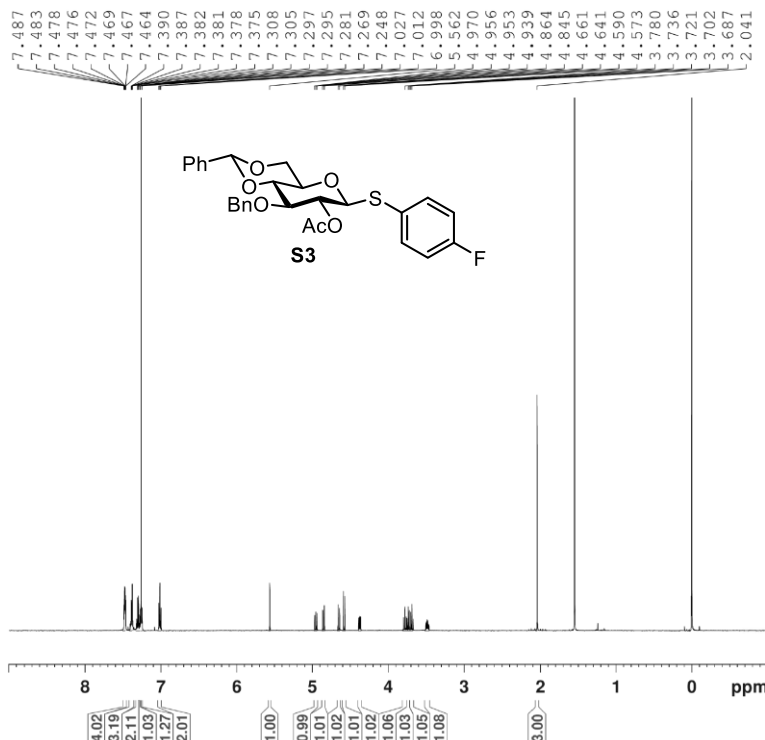
## References

- [1] a) Selective Glycosylations, C. S. Bennett, Ed. Wiley-VCH, Weinheim, Germany **2017**. b) W.-L. Leng, H. Yao, J.-X. He, X.-W. Liu, *Acc. Chem. Res.* **2018**, *51*, 628–639. c) M. Guberman, P. H. Seeberger, *J. Am. Chem. Soc.* **2019**, *141*, 5581–5592.
- [2] a) D. Crich, Z. Dai, S. Gastaldi, *J. Org. Chem.* **1999**, *64*, 5224–5229. b) Y. Zeng, Z. Wang, D. Whitfield, X. Huang, *J. Org. Chem.* **2008**, *73*, 7952–7962. c) T. Hamsen, H. Elferink, J. M. A. van Hengst, K. J. Houthuijs, W. A. Remmerswaal, A. Kromm, G. Berden, S. van der Vorm, A. M. Rijs, H. S. Overkleef, D. V. Filippov, F. P. J. T. Rutjes, G. A. van der Marel, J. Martens, J. Oomens, J. D. C. Codée, T. J. Boltje, *Nature. Commun.* **2020**, *11*:2664. d) A. A. Hettikankanamalage, R. Lassfolk, F. S. Ekholm, R. Leino, D. Crich, *Chem. Rev.* **2020**, *120*, 7104–7151.
- [3] a) M. W. Breedveld, K. J. Miller, *Microbiol. Rev.* **1994**, *58*, 145–161. b) A. V. Nair, S. N. Gummadi, M. Doble, *Biotechnol. Lett.* **2016**, *38*, 1519–1525. c) E. Cho, D. Jeong, Y. Choi, S. Jung, *J. Incl. Phenom. Macrocycl. Chem.* **2016**, *85*, 175–185.
- [4] a) S. Manmode, K. Matsumoto, T. Itoh, T. Nokami, *Asian J. Org. Chem.* **2018**, *7*, 1719–1729; b) A. Shibuya, T. Nokami, *Chem. Rec.* **2021**, *21*, 2389–2396. c) K. Yano, N. Sasaki, T. Itoh, T. Nokami, *J. Synth. Org. Chem. Jpn* **2021**, *79*, 839–848.
- [5] S. Manmode, M. Kato, T. Ichiyangi, T. Nokami, T. Itoh, *Asian J. Org. Chem.* **2018**, *7*, 1802–1805.
- [6] a) J. Yoshida, S. Suga, *Chem. Eur. J.* **2002**, *8*, 2650–2658. b) J. Yoshida, Y. Ashikari, K. Matsumoto, T. Nokami, *J. Synth. Chem. Soc. Jpn* **2013**, *71*, 1136–1144. c) J. Yoshida, A. Shimizu, Y. Ashikari, T. Morofuji, R. Hayashi, T. Nokami, A. Nagaki, *Bull. Chem. Soc. Jpn* **2015**, *88*, 763–775.
- [7] <sup>1</sup>H NMR spectra of the VT-NMR experiments is normalized by the peak height of CH<sub>2</sub>Cl<sub>2</sub> at 5.32 ppm.
- [8] a) D. Crich, S. Sun, *J. Am. Chem. Soc.* **1997**, *119*, 11217–11223. b) T. G. Frihed, M. Bols, C. M. Pedersen, *Chem. Rev.* **2015**, *115*, 4963–5013.
- [9] a) T. Nokami, A. Shibuya, H. Tsuyama, S. Suga, A. A. Bowers, D. Crich, J. Yoshida, *J. Am. Chem. Soc.* **2007**, *129*, 10922–10928. b) T. Nokami, A. Shibuya, S. Manabe, Y. Ito, J. Yoshida, *Chem. Eur. J.* **2009**, *15*, 2252–2255. c) T. Nokami, Y. Nozaki, Y. Saigusa, A. Shibuya, S. Manabe, Y. Ito, J. Yoshida, *Org. Lett.* **2011**, *13*, 1544–1547. d) T. Nokami, A. Shibuya, Y. Saigusa, S. Manabe, Y. Ito, J. Yoshida, *Beilstein J. Org. Chem.* **2012**, *8*, 456–460.
- [10] X. Zhu, R. T. Dere, J. Jiang, L. Zhang, X. Wang, *J. Org. Chem.* **2011**, *76*, 10187–10197.
- [11] H. Hori, Y. Nishida, H. Ohru, H. Meguro, *J. Org. Chem.* **1989**, *54*, 1346–1353.



# <sup>1</sup>H and <sup>13</sup>C NMR spectra of synthetic intermediate S3 and building blocks

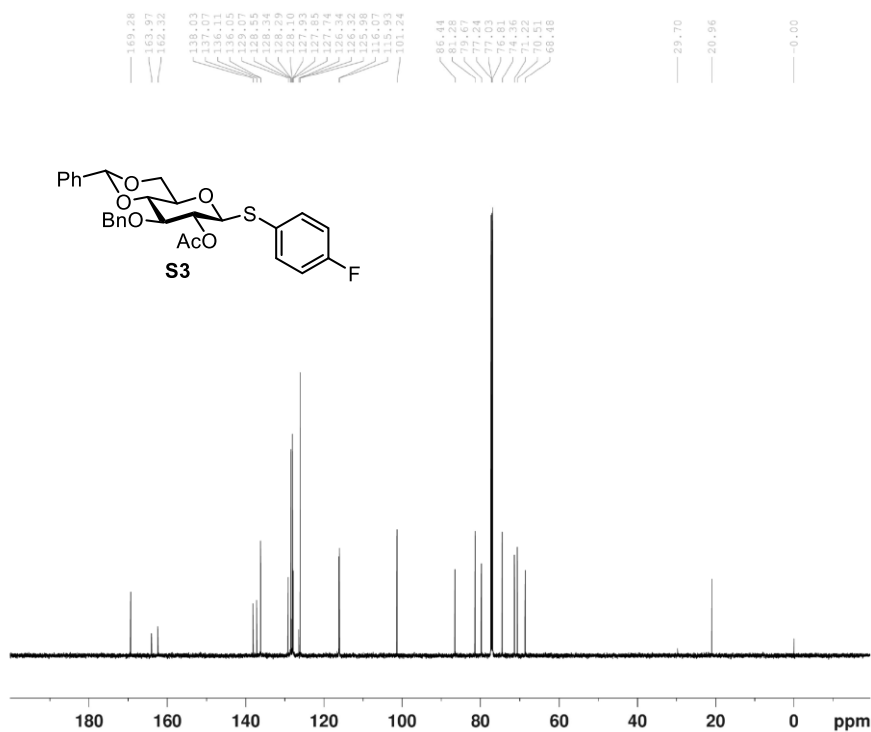
## <sup>1</sup>H NMR



NAME KTM-203  
EXPNO 10  
PROCNO 1  
Date\_ 20210908  
Time 16.03  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zg30  
TD 65536  
SOLVENT CDCl3  
NS 16  
DS 2  
SWH 12335.526 Hz  
FIDRES 0.188225 Hz  
AQ 2.6564426 sec  
RG 203  
DW 40.533 usec  
DE 6.50 usec  
TE 297.6 K  
D1 1.00000000 sec  
TD0 1

===== CHANNEL f1 =====  
NUC1 1H  
P1 13.00 usec  
PL1 -2.00 dB  
PL1W 18.91009140 W  
SFO1 600.1337060 MHz  
SI 32768  
SF 600.1300162 MHz  
WDW EM  
SSB 0  
LB 0.30 Hz  
GB 0  
PC 1.00

## <sup>13</sup>C NMR

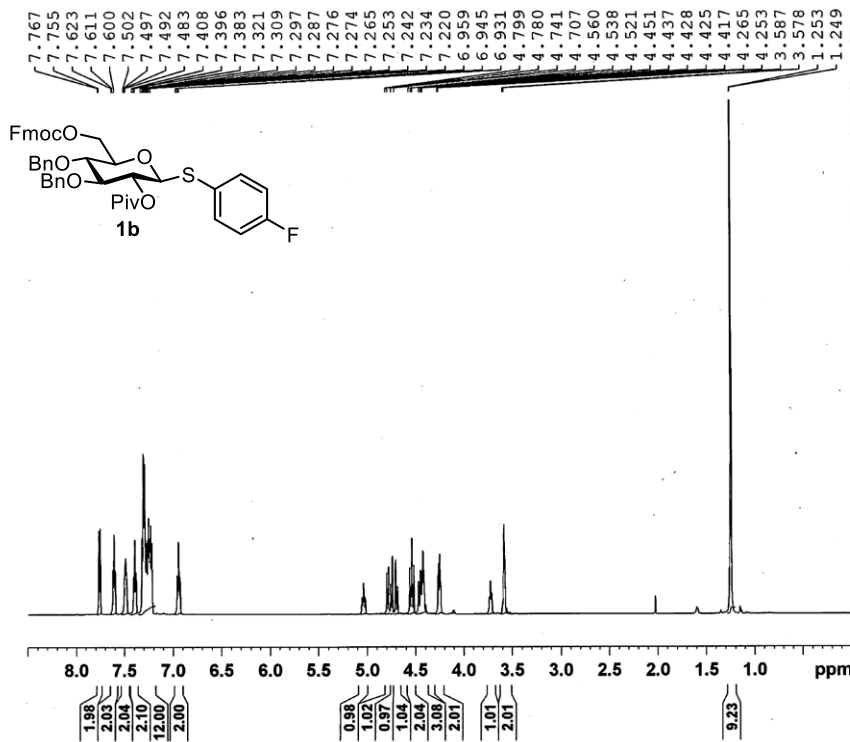


NAME KTM-203  
EXPNO 20  
PROCNO 1  
Date\_ 20210909  
Time 11.31  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zgpg30  
TD 65536  
SOLVENT CDCl3  
NS 500  
DS 2  
SWH 36057.691 Hz  
FIDRES 0.550197 Hz  
AQ 0.9088159 sec  
RG 203  
DW 13.867 usec  
DE 6.50 usec  
TE 298.5 K  
D1 2.00000000 sec  
D11 0.03000000 sec  
TD0 1

===== CHANNEL f1 =====  
NUC1 13C  
P1 10.00 usec  
PL1 -1.00 dB  
PL1W 125.22619629 W  
SFO1 150.9178988 MHz

===== CHANNEL f2 =====  
CPDPRG2 waltz16  
NUC2 1H  
PCPD2 80.00 usec  
PL2 -2.00 dB  
PL12 13.78 dB  
PL13 14.00 dB  
PL2W 18.91009140 W  
PL12W 0.49968192 W  
PL13W 0.47499999 W  
SFO2 600.1324005 MHz  
SI 32768  
SF 150.9028124 MHz  
WDW EM  
SSB 0  
LB 1.00 Hz  
GB 0  
PC 1.40

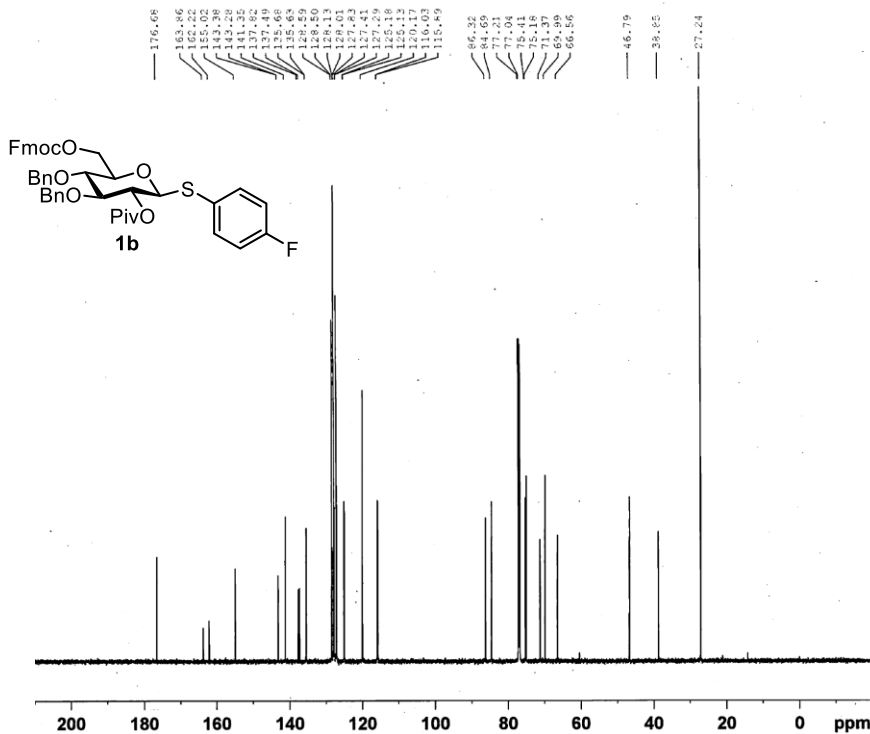
# <sup>1</sup>H NMR



NAME KTM-131 data  
EXPNO 11  
PROCNO 1  
Date\_ 20200220  
Time\_ 17.28  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zg30  
TD 65536  
SOLVENT CDCl3  
NS 16  
DS 2  
SWH 12335.526 Hz  
FIDRES 0.188225 Hz  
AQ 2.6564426 sec  
RG 36  
DW 40.533 usec  
DE 6.50 usec  
TE 295.2 K  
D1 1.00000000 sec  
TDO 1

===== CHANNEL f1 =====  
NUC1 1H  
P1 13.00 usec  
PL1 -2.00 dB  
PL1W 18.91009140 W  
SFO1 600.1337060 MHz  
SI 32768  
SF 600.1300404 MHz  
WDW EM  
SSB 0  
LB 0.30 Hz  
GB 0  
PC 1.00

# <sup>13</sup>C NMR

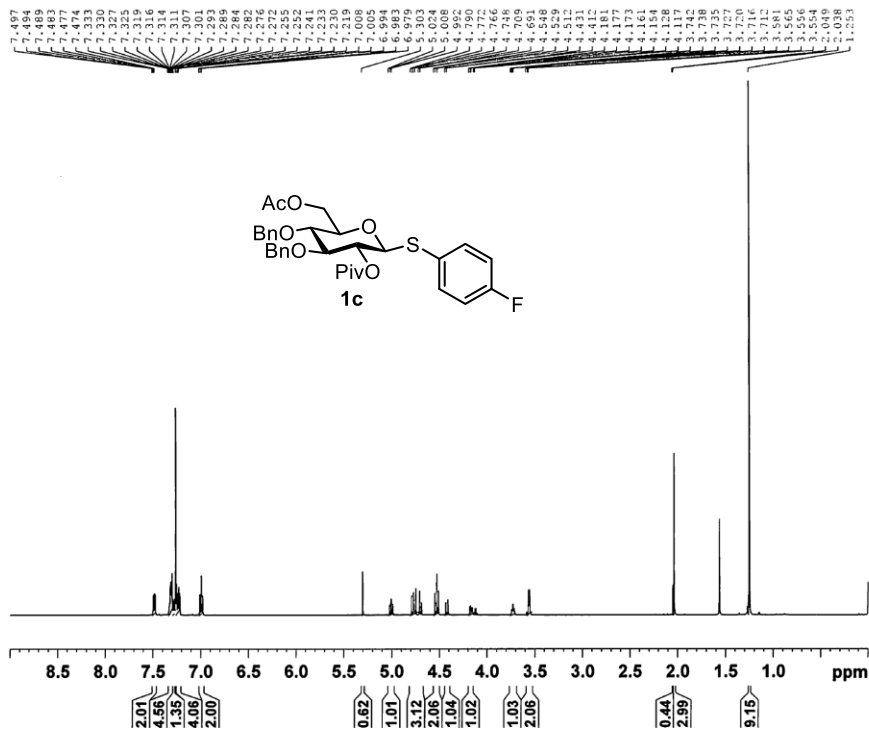


NAME KTM-131 data  
EXPNO 10  
PROCNO 1  
Date\_ 20200220  
Time\_ 17.27  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zgpg30  
TD 65536  
SOLVENT CDCl3  
NS 500  
DS 2  
SWH 36057.691 Hz  
FIDRES 0.550197 Hz  
AQ 0.9088159 sec  
RG 203  
DW 13.867 usec  
DE 6.50 usec  
TE 296.3 K  
D1 2.00000000 sec  
D11 0.03000000 sec  
TDO 1

===== CHANNEL f1 =====  
NUC1 13C  
P1 10.00 usec  
PL1 -1.00 dB  
PL1W 125.22619629 W  
SFO1 150.9178988 MHz

===== CHANNEL f2 =====  
CPDPRG2 waltz16  
NUC2 1H  
PCPD2 80.00 usec  
PL2 -2.00 dB  
PL12 13.78 dB  
PL13 14.00 dB  
PL2W 18.91009140 W  
PL12W 0.45968192 W  
PL13W 0.47499999 W  
SFO2 600.1324005 MHz  
SI 32768  
SF 150.9028090 MHz  
WDW EM  
SSB 0  
LB 1.00 Hz  
GB 0  
PC 1.40

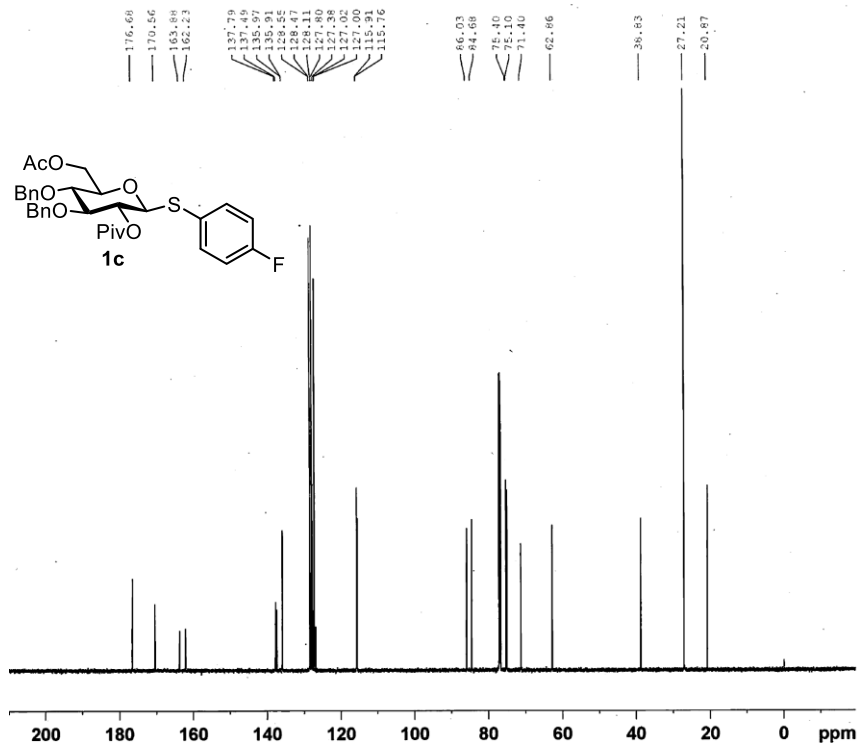
# <sup>1</sup>H NMR



NAME KTM-226 fr.12-30  
EXPNO 10  
PROCNO 1  
Date\_ 20190225  
Time 21.01  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zg30  
TD 65536  
SOLVENT CDCl3  
NS 16  
DS 2  
SWH 12335.526 Hz  
FIDRES 0.188225 Hz  
AQ 2.6564426 sec  
RG 203  
DW 40.533 usec  
DE 6.50 usec  
TE 292.5 K  
D1 1.0000000 sec  
TDO 1

===== CHANNEL f1 =====  
NUC1 1H  
P1 13.00 usec  
PL1 -2.00 dB  
PL1W 18.91009140 W  
SFO1 600.1337060 MHz  
SI 32768  
SF 600.1300159 MHz  
WDW EM  
SSB 0  
LB 0.30 Hz  
GB 0  
PC 1.00

# <sup>13</sup>C NMR

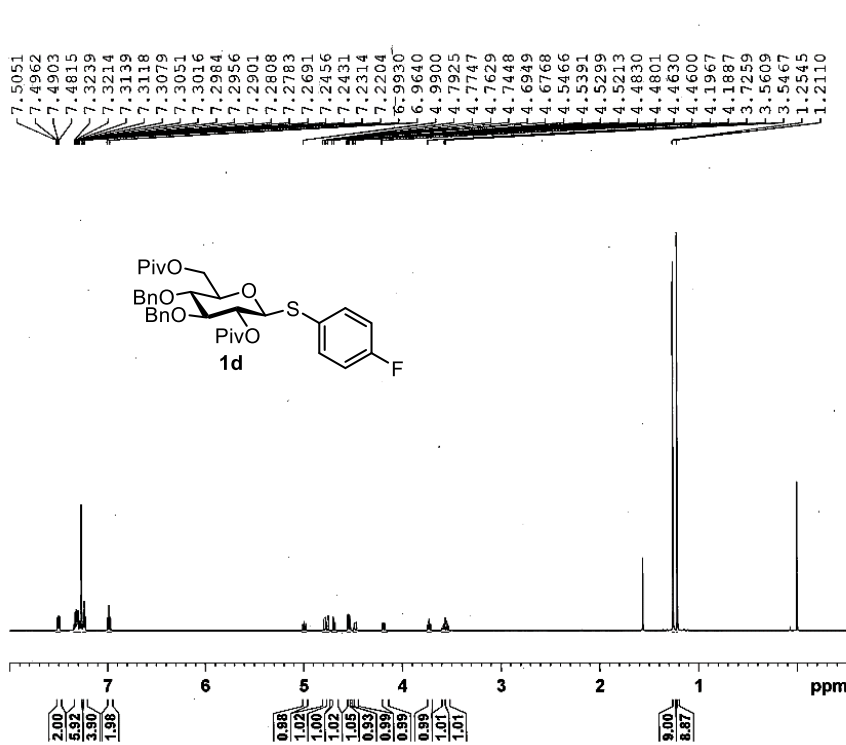


NAME KTM-226 fr.12-30 data  
EXPNO 10  
PROCNO 1  
Date\_ 20200218  
Time\_ 19.31  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zg30  
TD 65536  
SOLVENT CDCl3  
NS 500  
DS 2  
SWH 36057.691 Hz  
FIDRES 0.550197 Hz  
AQ 0.9088159 sec  
RG 203  
DW 13.857 usec  
DE 6.50 usec  
TE 296.6 K  
D1 2.0000000 sec  
D11 0.0300000 sec  
TDO 1

===== CHANNEL f1 =====  
NUC1 13C  
P1 10.00 usec  
PL1 -1.00 dB  
PL1W 125.22619629 W  
SFO1 150.9178988 MHz

===== CHANNEL f2 =====  
CPDPRG2 waltz16  
NUC2 1H  
PCPD2 80.00 usec  
PL2 -2.00 dB  
PL12 13.78 dB  
PL13 14.00 dB  
PL2W 18.91009140 W  
PL12W 0.49968192 W  
PL13W 0.47499999 W  
SFO2 600.1324005 MHz  
SI 32768  
SF 150.9028090 MHz  
WDW EM  
SSB 0  
LB 1.00 Hz  
GB 0  
PC 1.40

<sup>1</sup>H NMR

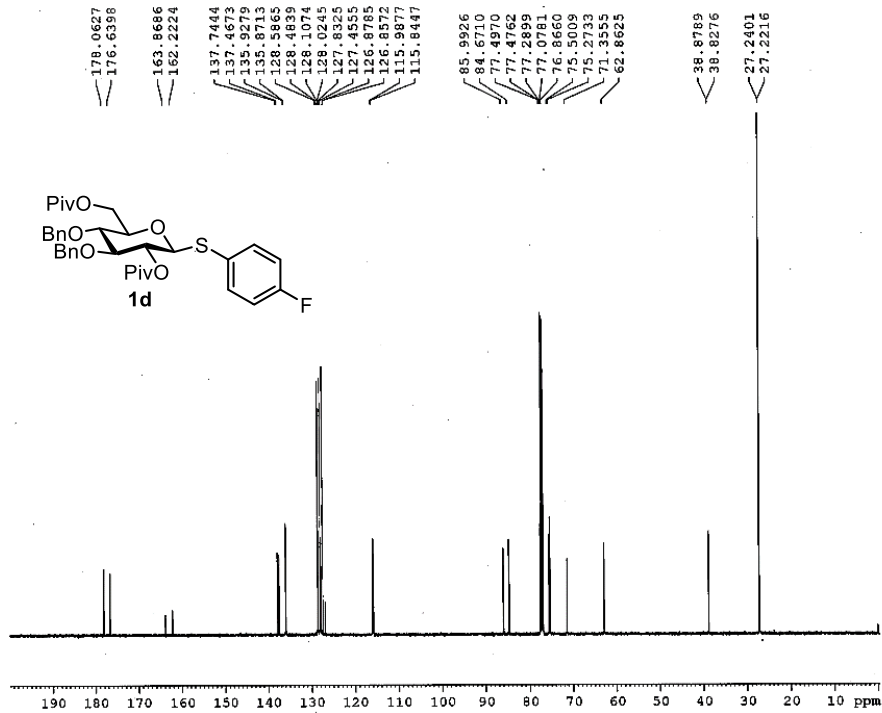


```

NAME      KTM-227 H
EXNO     10
PROCNO   1
Date_    20210105
Time     16.47
INSTRUM  spect
PROBHD   5 mm PABBO BB-
PULPROG  zg30
TD       65536
SOLVENT  CDCl3
NS       2
DS       2
SWH      12335.526 Hz
FIDRES   0.188225 Hz
AQ       2.6561426 sec
RG       203
DM       40.533 usec
DE       6.50 usec
TE       294.0 K
D1       1.00000000 sec
TDO      1

===== CHANNEL f1 =====
NUC1     1H
P1       13.00 usec
PL1      -2.00 dB
PL1W    18.91009140 W
SFO1    600.1337060 MHz
SI       32768
SF       600.1300150 MHz
WDW      EM
SSB      0
LB       0.30 Hz
GB       0
PC       1.00
    
```

<sup>13</sup>C NMR



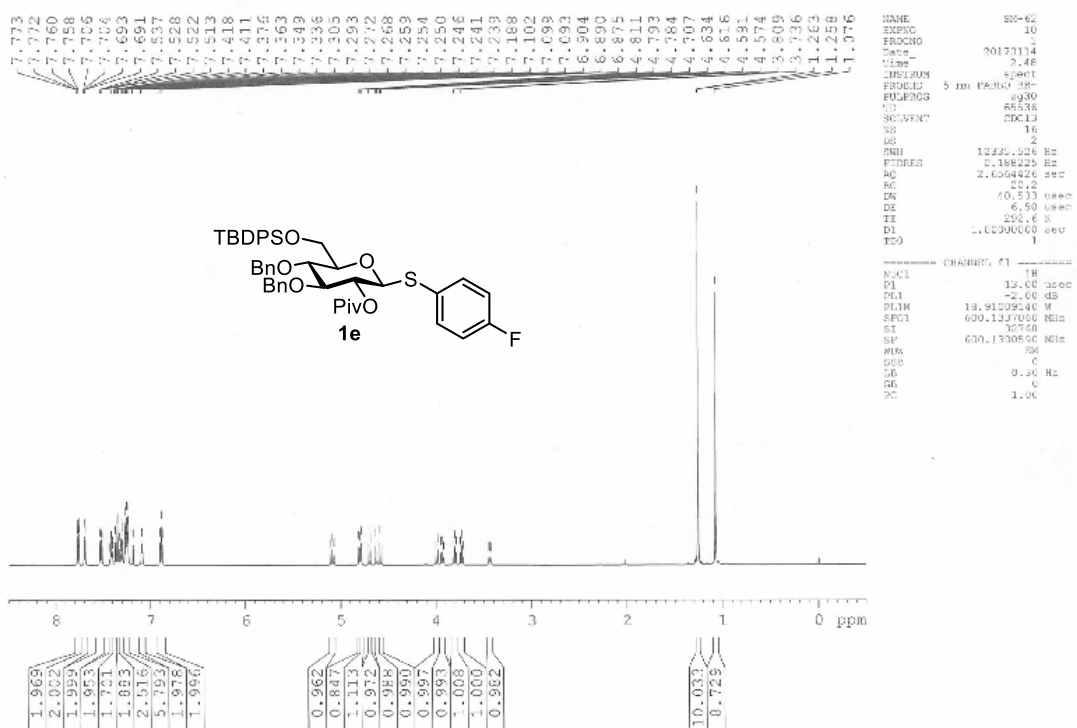
```

NAME      KTM-227
EXNO     10
PROCNO   1
Date_    20210105
Time     16.26
INSTRUM  spect
PROBHD   5 mm PABBO BB-
PULPROG  zg30
TD       65536
SOLVENT  CDCl3
NS       500
DS       2
SWH      36057.691 Hz
FIDRES   0.550197 Hz
AQ       0.9088159 sec
RG       203
DM       13.667 usec
DE       6.50 usec
TE       295.0 K
D1       2.00000000 sec
D11      0.03000000 sec
TDO      1

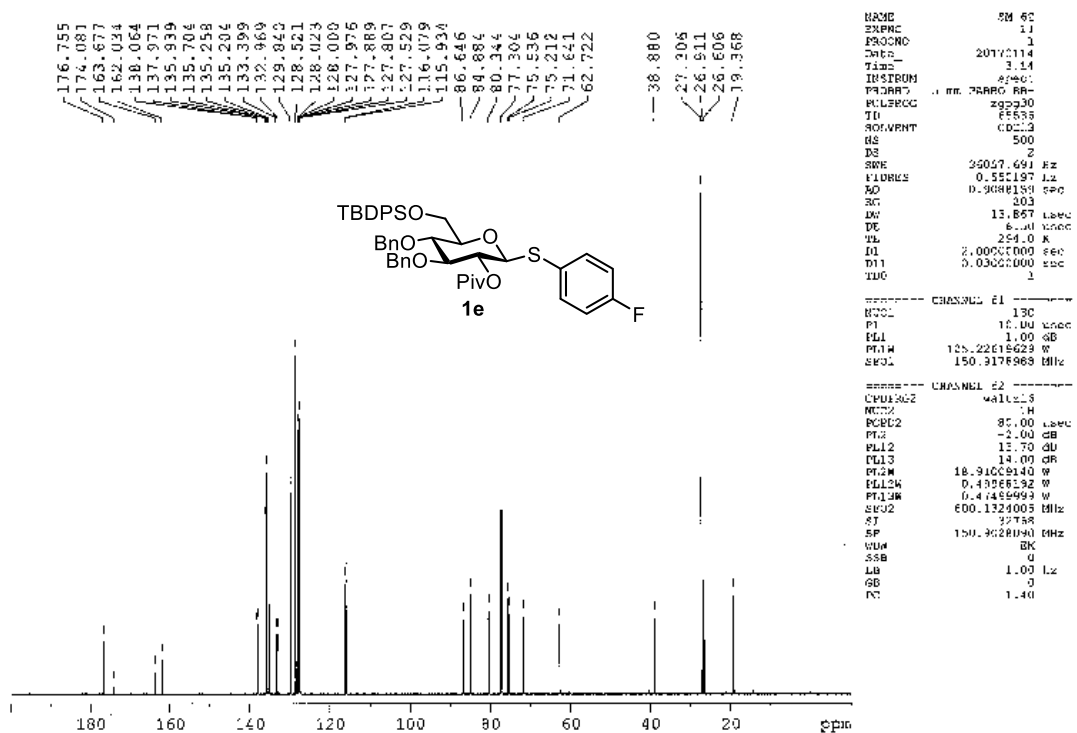
===== CHANNEL f1 =====
NUC1     13C
P1       10.00 usec
PL1      -1.00 dB
PL1W    125.22619629 W
SFO1    150.9178988 MHz

===== CHANNEL f2 =====
CPDPRG2  waltz16
NUC2     1H
PCPD2    80.00 usec
PL2      -2.00 dB
PL12     13.78 dB
PL13     14.00 dB
PL2W    18.91009140 W
PL12W   0.49968192 W
PL13W   0.47499999 W
SFO2    600.1324005 MHz
SI       32768
SF       150.9028090 MHz
WDW      EM
SSB      0
LB       1.00 Hz
GB       0
PC       1.40
    
```

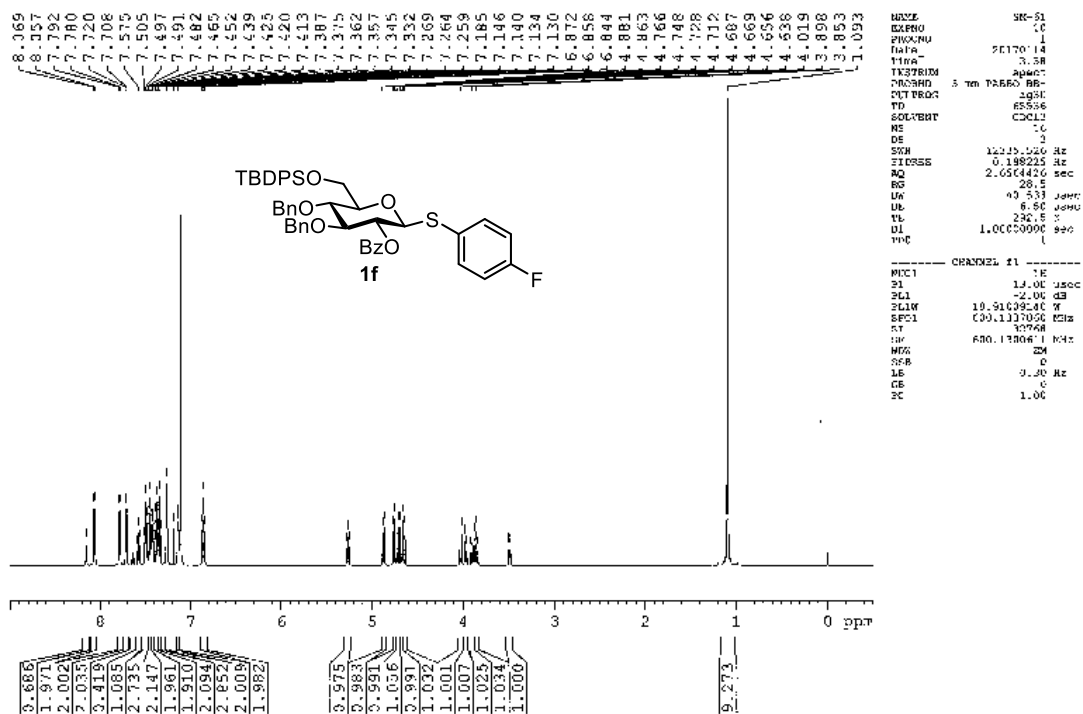
# <sup>1</sup>H NMR



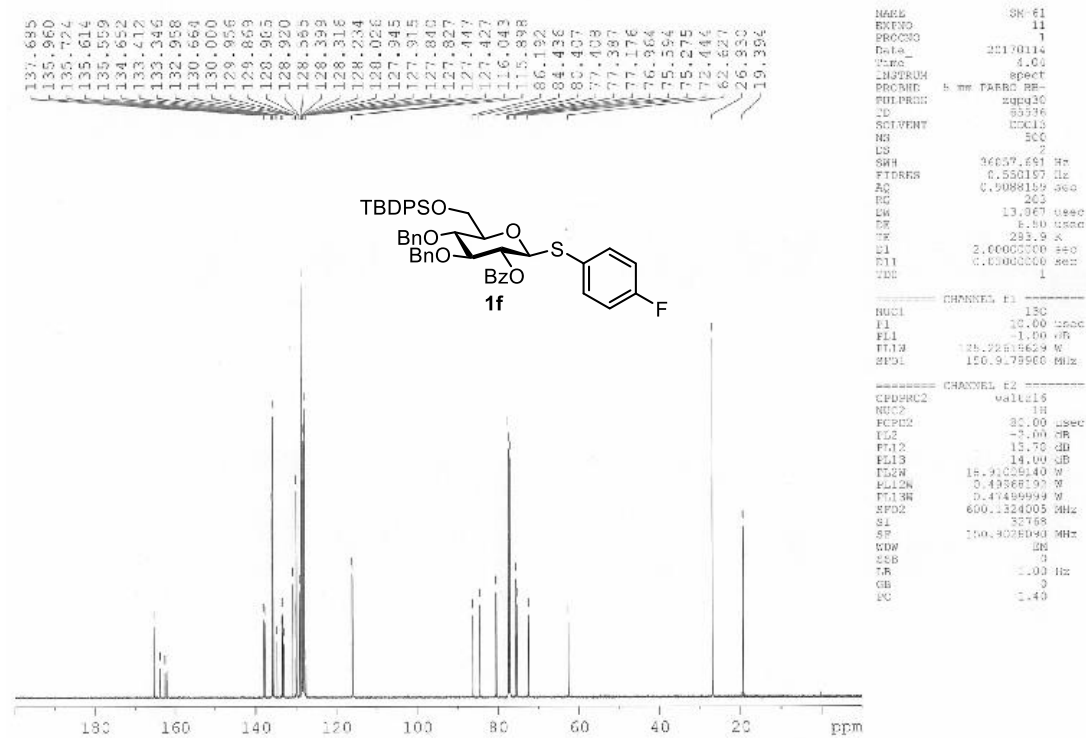
# <sup>13</sup>C NMR



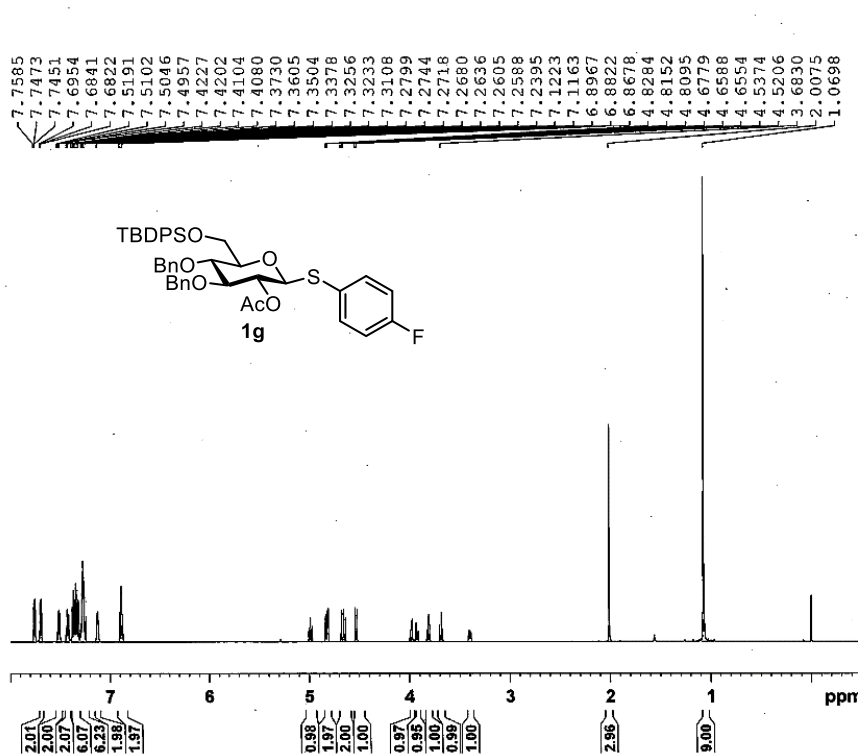
# <sup>1</sup>H NMR



# <sup>13</sup>C NMR



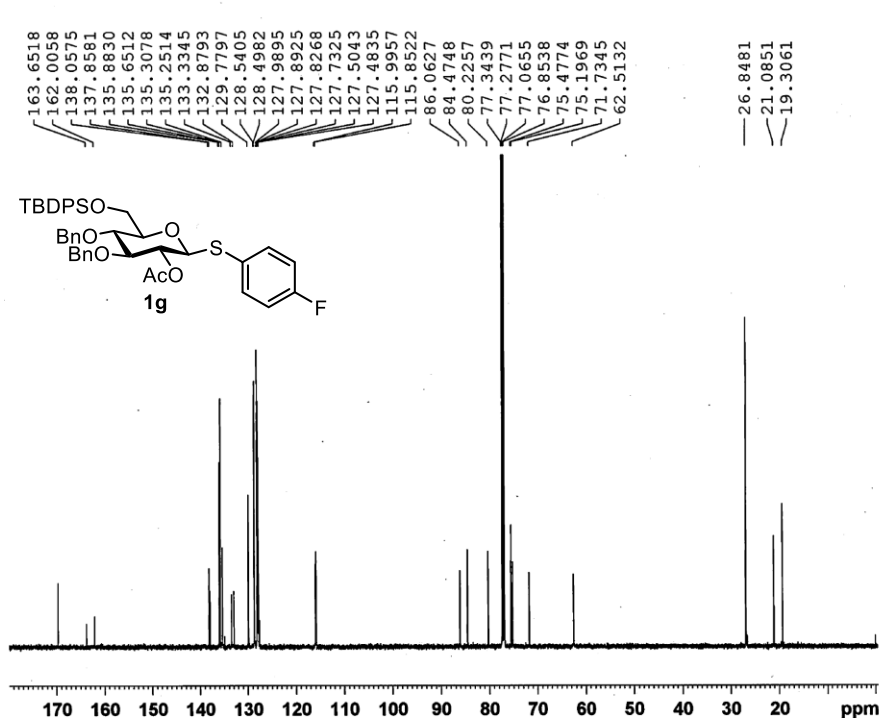
# <sup>1</sup>H NMR



```

NAME           KTM-205
EXPNO          1
PROCNO         1
Date_          20210107
Time           9.26
INSTRUM        spect
PROBHD         5 mm PABBO BB-
PULPROG        zg30
TD             65536
SOLVENT        CDCl3
NS             16
DS             2
SWH            12335.526 Hz
FIDRES         0.168225 Hz
AQ            2.6564426 sec
RG            114
DW            40.533 usec
DE            6.50 usec
TE            294.9 K
D1            1.00000000 sec
D10           1
===== CHANNEL f1 =====
NUC1           1H
P1            13.00 usec
PL1           -2.00 dB
PL1W         18.91009140 W
SFO1         600.1337060 MHz
SI           32768
SF           600.1300284 MHz
MW           EM
SSB          0
LB           0.30 Hz
GB           0
PC           1.00
    
```

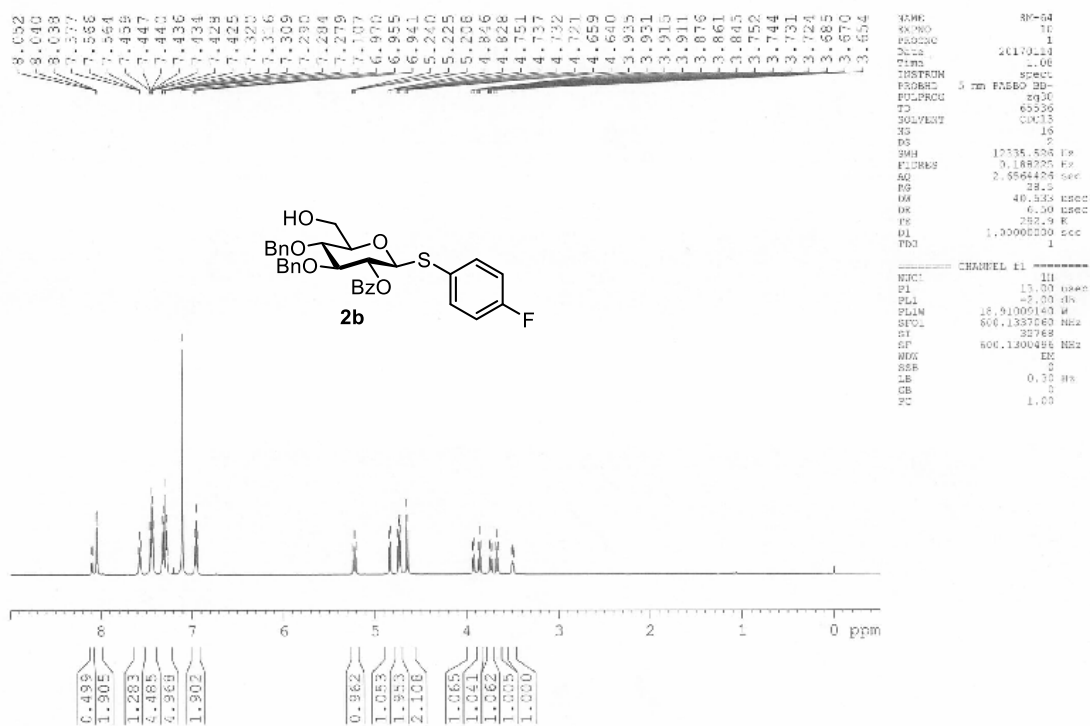
# <sup>13</sup>C NMR



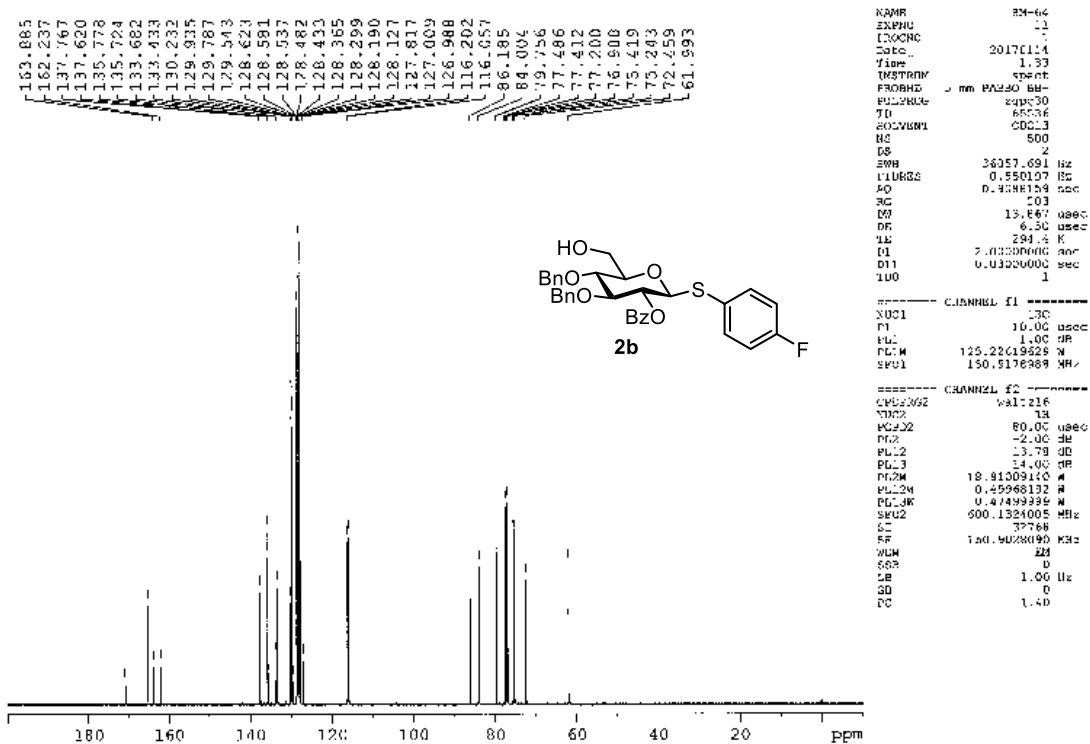
```

NAME           KTM-205
EXPNO          10
PROCNO         1
Date_          20210107
Time           9.25
INSTRUM        spect
PROBHD         5 mm PABBO BB-
PULPROG        zgpg30
TD             65536
SOLVENT        CDCl3
NS             500
DS             2
SWH            36057.691 Hz
FIDRES         0.550197 Hz
AQ            0.9088159 sec
RG            203
DW            13.867 usec
DE            6.50 usec
TE            295.6 K
D1            2.00000000 sec
D11           0.03000000 sec
D10           1
===== CHANNEL f1 =====
NUC1           13C
P1            10.00 usec
PL1           -1.00 dB
PL1W         125.22619629 W
SFO1         150.9178988 MHz
===== CHANNEL f2 =====
CPDPRG2       waltz16
NUC2           1H
PCPD2         80.00 usec
PL2           -2.00 dB
PL12         13.78 dB
PL13         14.00 dB
PL2W         18.91009140 W
PL12W        0.49968192 W
PL13W        0.47499999 W
SFO2         600.1324005 MHz
SI           32768
SF           150.9028090 MHz
MW           EM
SSB          0
LB           1.00 Hz
GB           0
PC           1.40
    
```

<sup>1</sup>H NMR

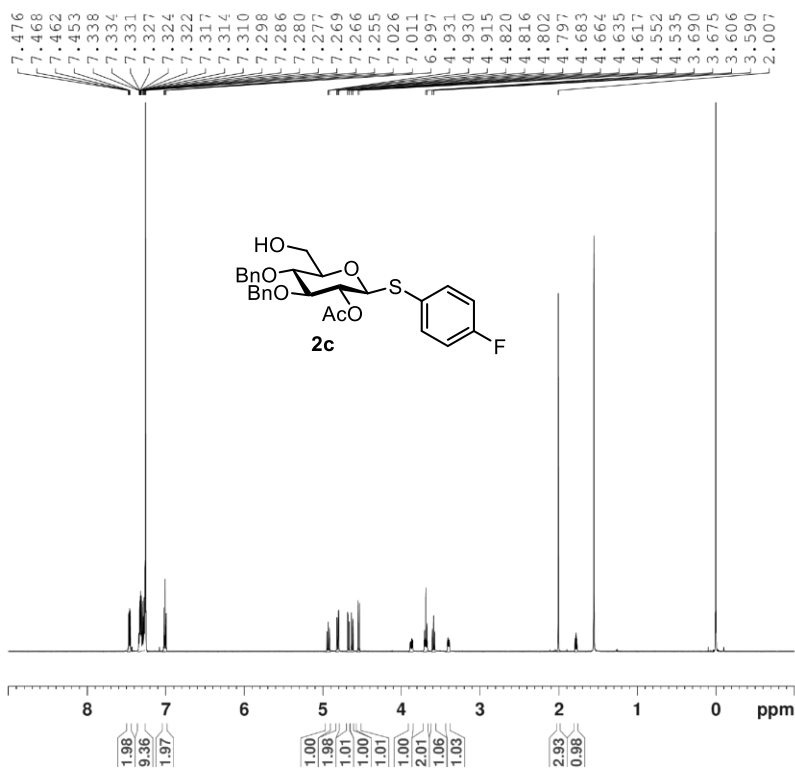


<sup>13</sup>C NMR





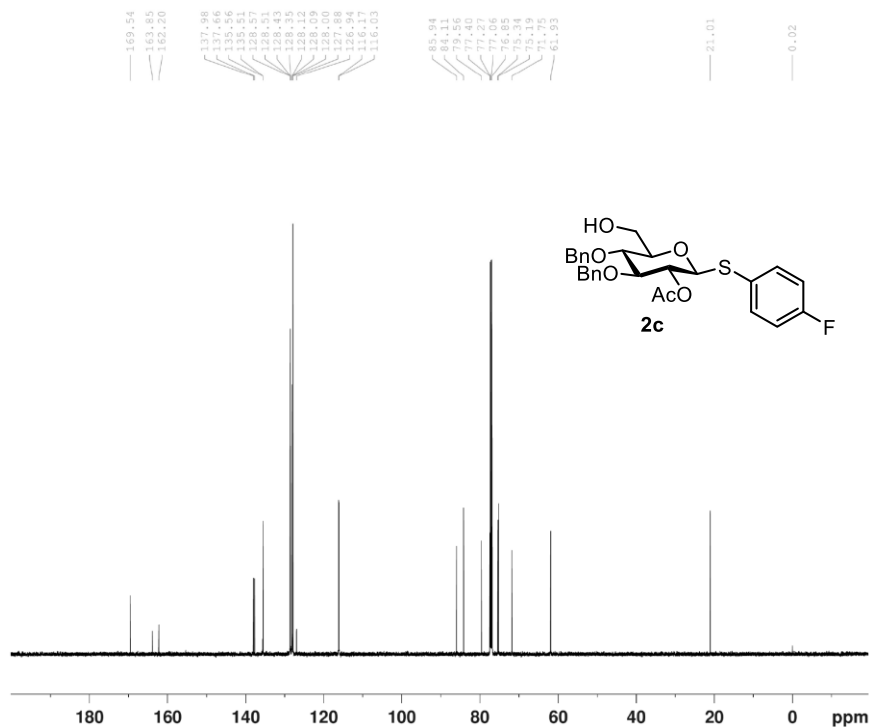
# <sup>1</sup>H NMR



NAME KIM-288  
EXPNO 10  
PROCNO 1  
Date\_ 20210908  
Time 16.12  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zg30  
TD 65536  
SOLVENT CDCl3  
NS 16  
DS 2  
SWH 12335.526 Hz  
FIDRES 0.188225 Hz  
AQ 2.6564426 sec  
RG 203  
DW 40.533 usec  
DE 6.50 usec  
TE 297.6 K  
D1 1.0000000 sec  
TD0 1

===== CHANNEL f1 =====  
NUC1 1H  
P1 13.00 usec  
PL1 -2.00 dB  
PL1W 18.91009140 W  
SF01 600.1337060 MHz  
SI 32768  
SF 600.1300161 MHz  
WDW EM  
SSB 0  
LB 0.30 Hz  
GB 0  
PC 1.00

# <sup>13</sup>C NMR



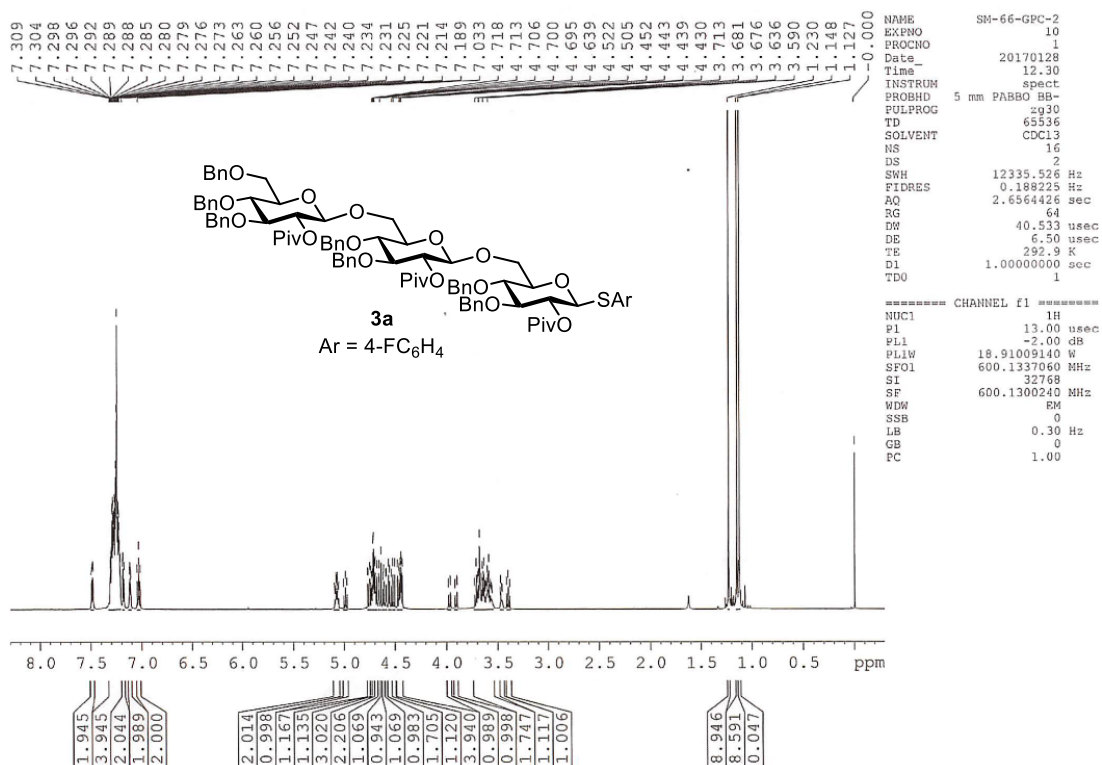
NAME KIM-288  
EXPNO 20  
PROCNO 1  
Date\_ 20210909  
Time 11.58  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zgpg30  
TD 65536  
SOLVENT CDCl3  
NS 500  
DS 2  
SWH 36057.691 Hz  
FIDRES 0.550197 Hz  
AQ 0.9088159 sec  
RG 203  
DW 13.867 usec  
DE 6.50 usec  
TE 298.6 K  
D1 2.00000000 sec  
D11 0.03000000 sec  
TD0 1

===== CHANNEL f1 =====  
NUC1 13C  
P1 10.00 usec  
PL1 -1.00 dB  
PL1W 125.22619629 W  
SF01 150.9178988 MHz

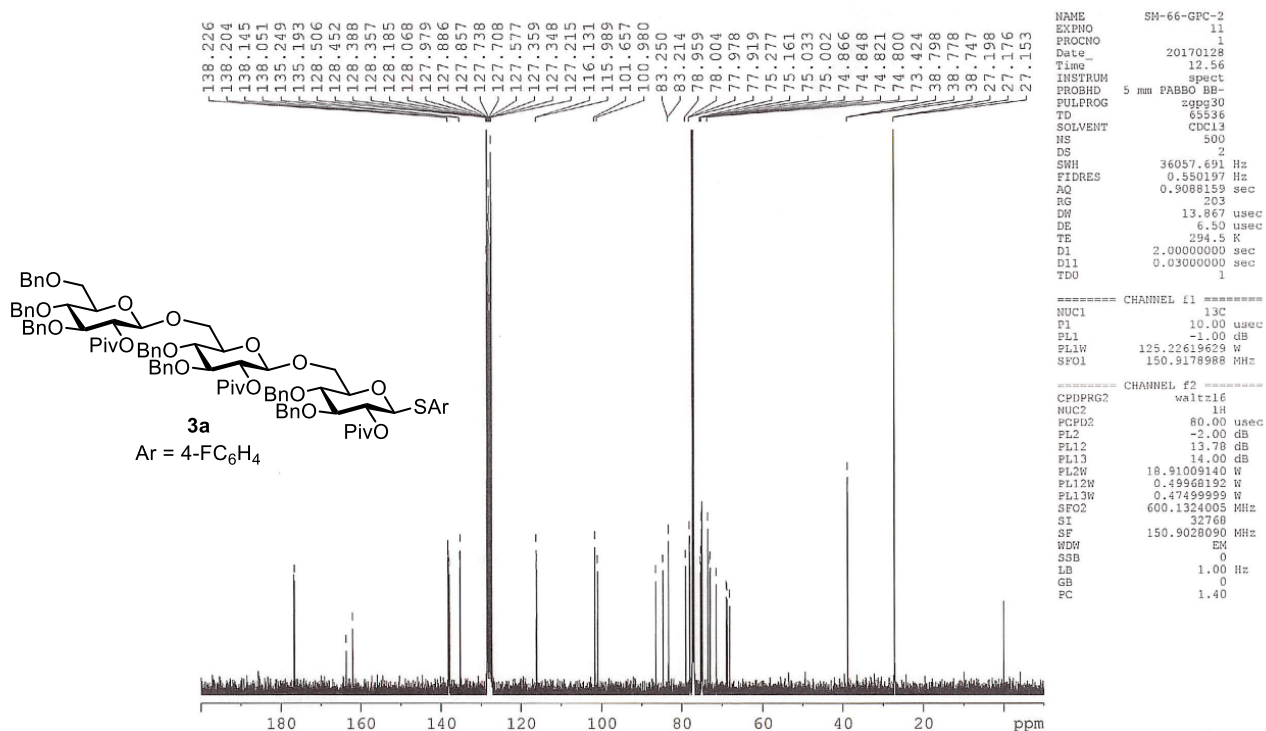
===== CHANNEL f2 =====  
CPDPRG2 waltz16  
NUC2 1H  
PCPD2 80.00 usec  
PL2 -2.00 dB  
PL12 13.78 dB  
PL13 14.00 dB  
PL2W 18.91009140 W  
PL12W 0.49968192 W  
PL13W 0.47499999 W  
SF02 600.1324005 MHz  
SI 32768  
SF 150.9028090 MHz  
WDW EM  
SSB 0  
LB 1.00 Hz  
GB 0  
PC 1.40

# <sup>1</sup>H, <sup>13</sup>C NMR, H-H COSY and HMQC spectra of trisaccharides

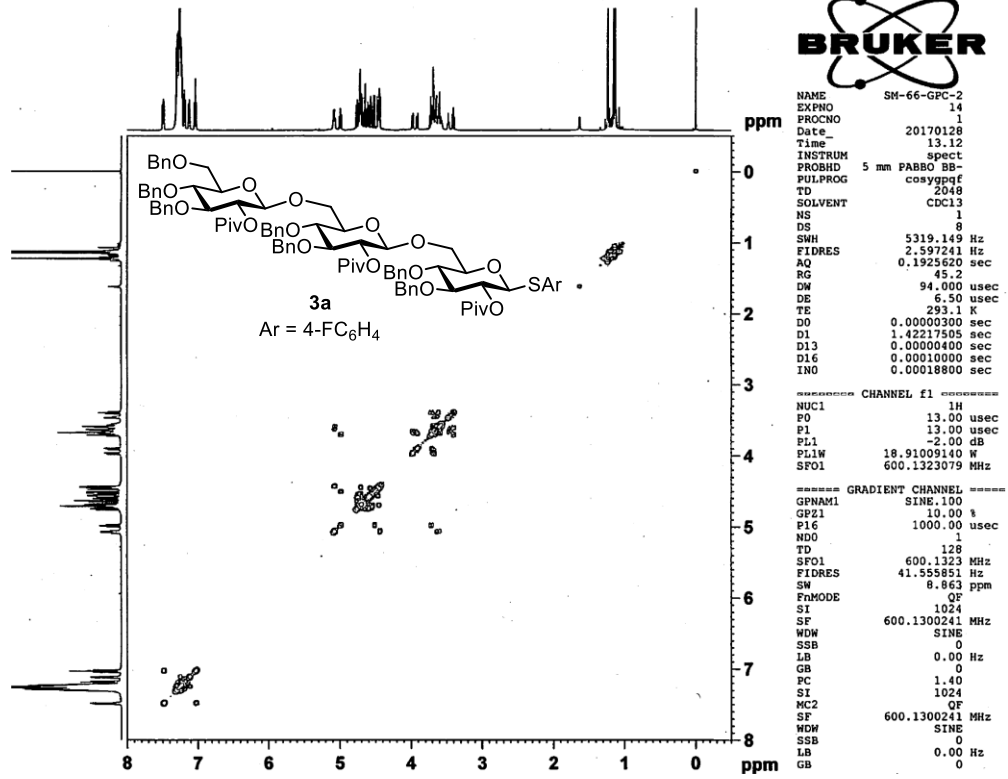
## <sup>1</sup>H NMR



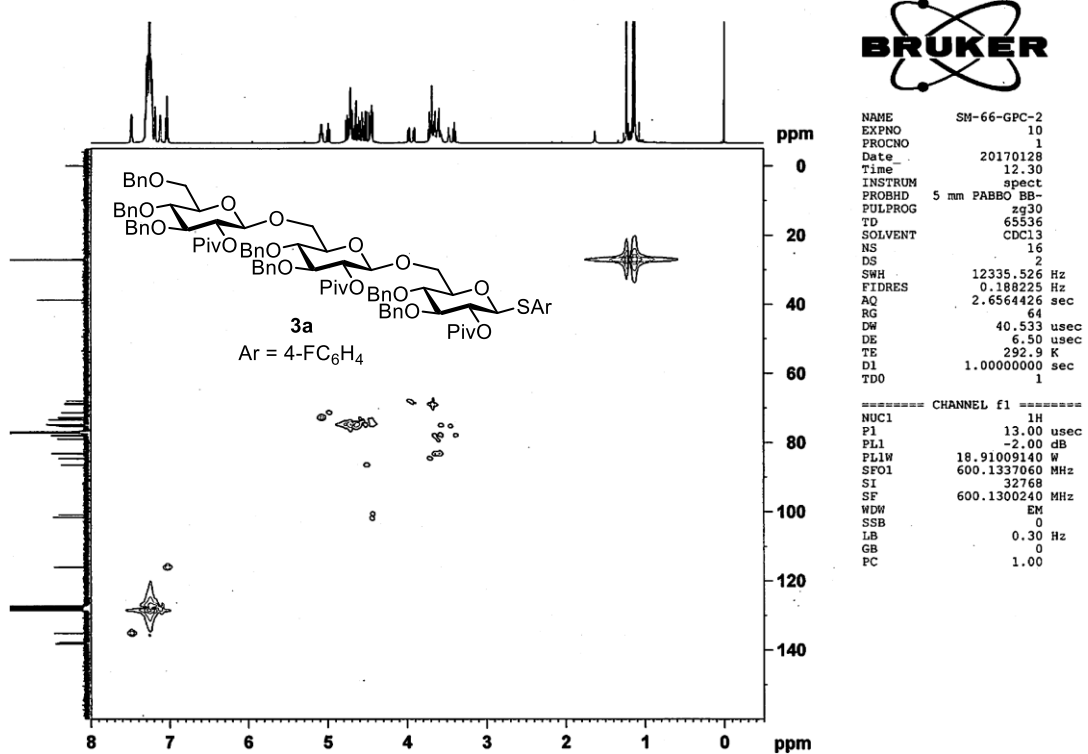
## <sup>13</sup>C NMR



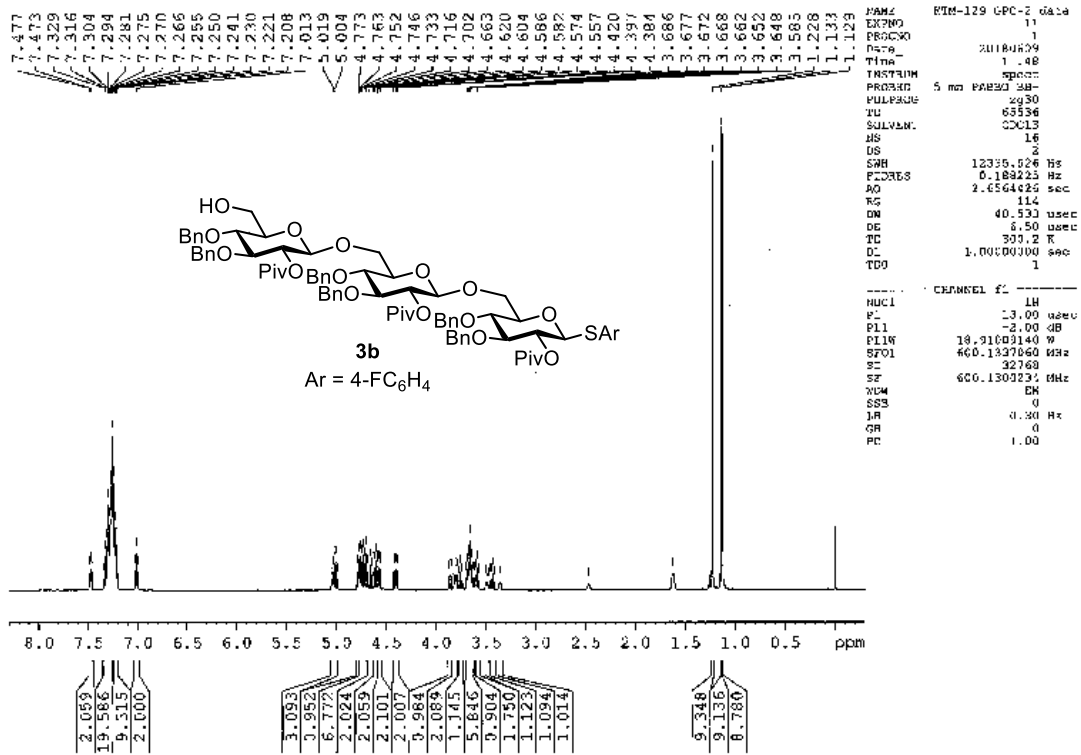
H-H cosy



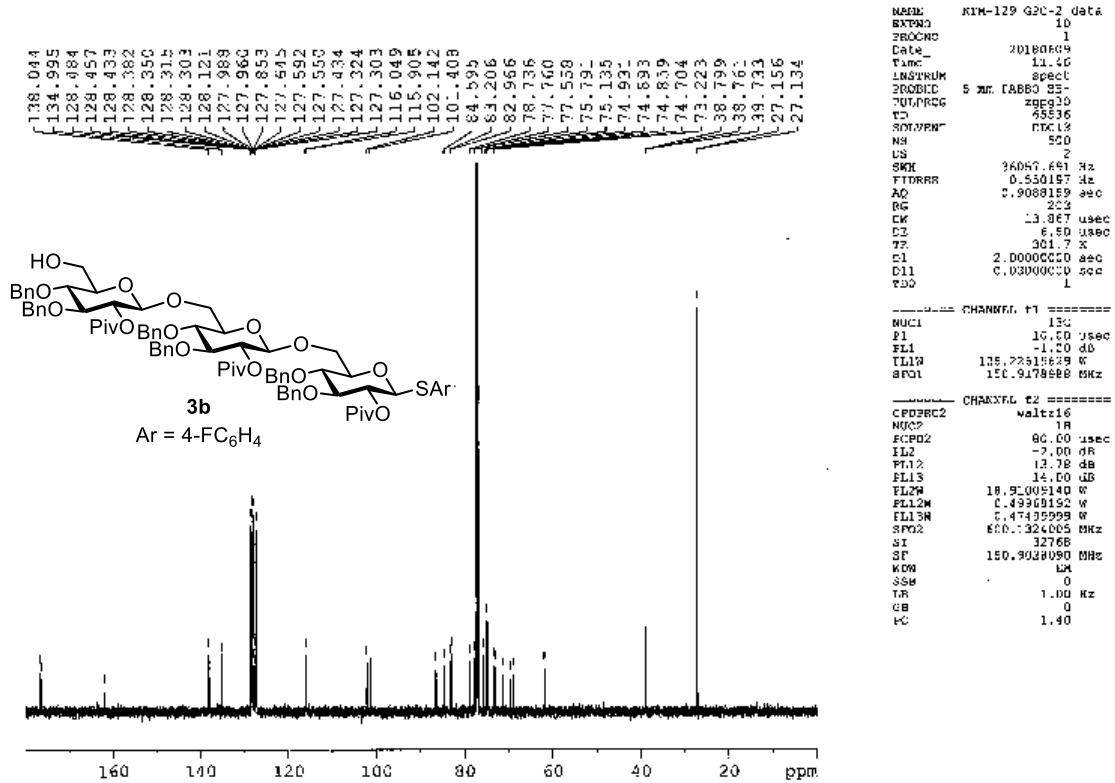
HMQC



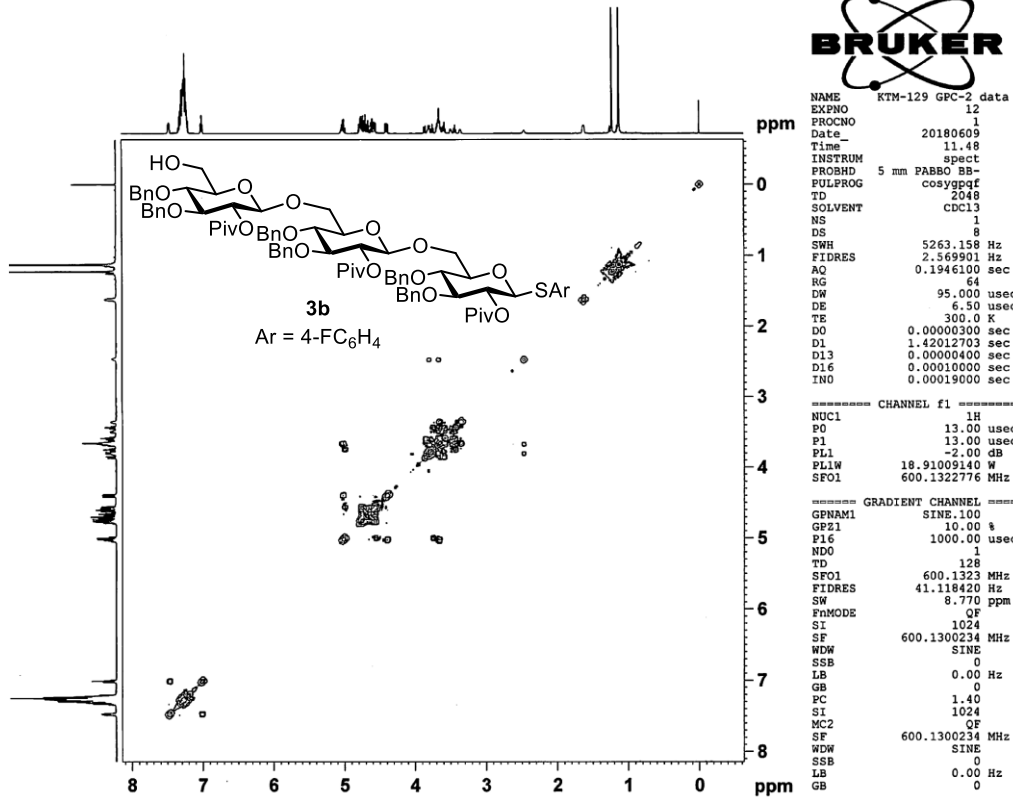
<sup>1</sup>H NMR



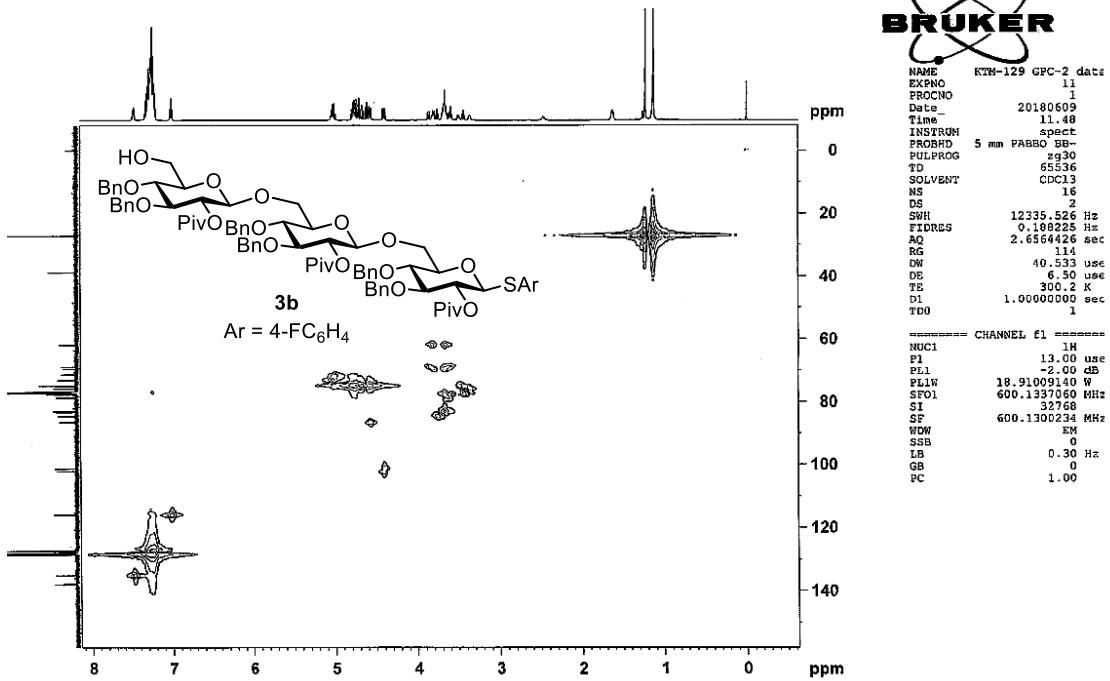
<sup>13</sup>C NMR



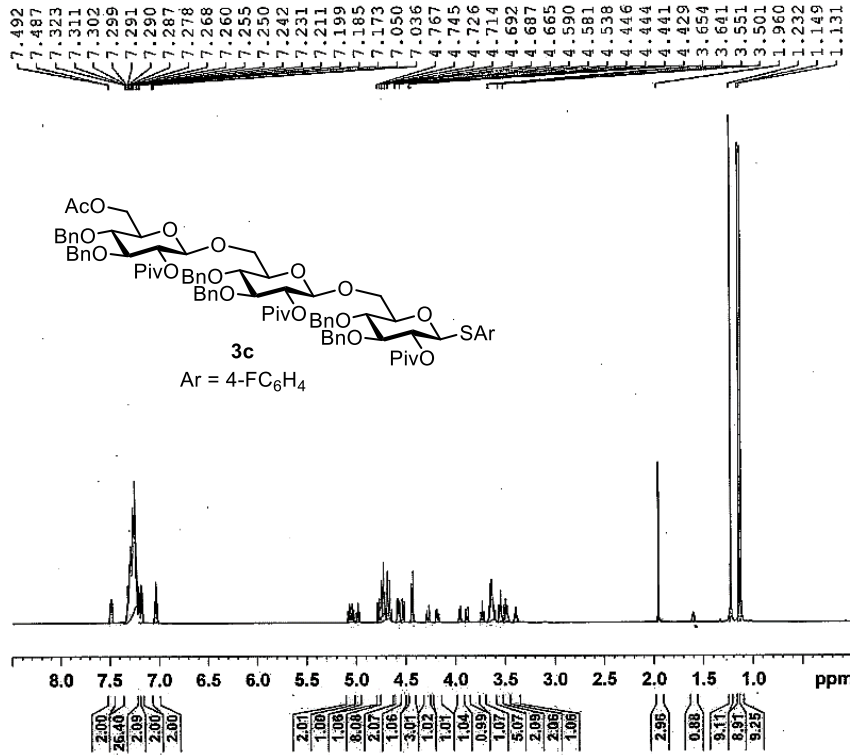
H-H cosy



HMQC



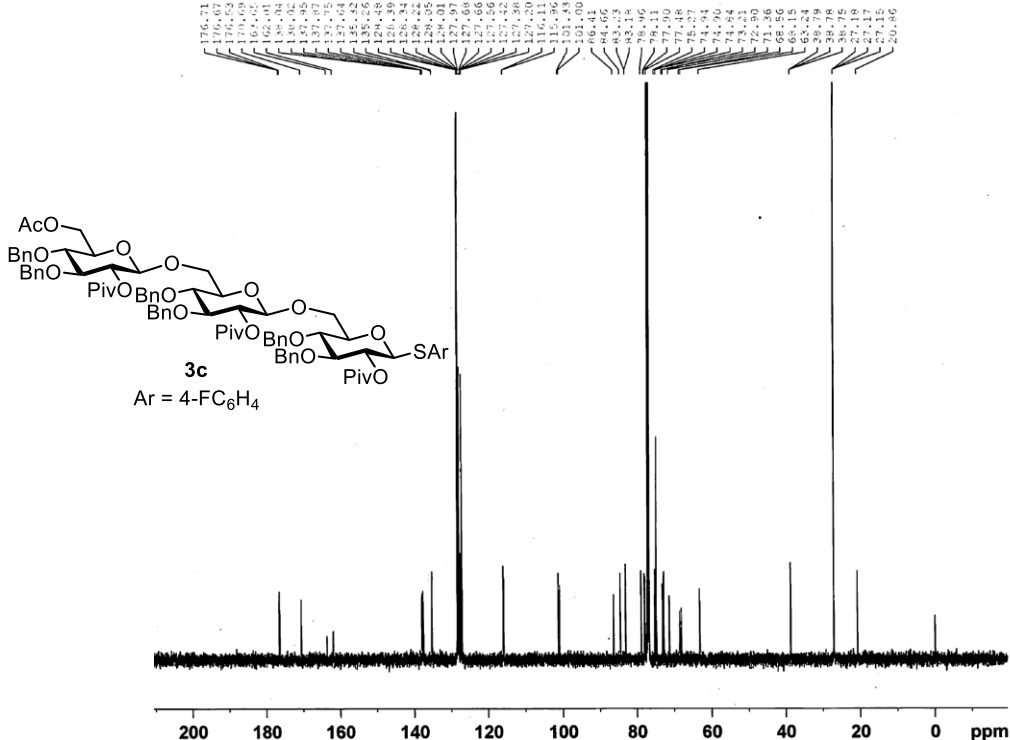
<sup>1</sup>H NMR



NAME KTM-232 GPC-1 data  
EXPNO 11  
PROCNO 1  
Date\_ 20200218  
Time\_ 21.10  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zg30  
TD 65536  
SOLVENT CDC13  
NS 16  
DS 2  
SWH 12335.526 Hz  
FIDRES 0.188225 Hz  
AQ 2.6564426 sec  
RG 90.5  
EW 40.333 usec  
DE 6.50 usec  
TE 295.6 K  
D1 1.00000000 sec  
TDO 1

===== CHANNEL f1 =====  
NUC1 1H  
P1 13.00 usec  
PL1 -2.00 dB  
PL1W 18.91009140 W  
SFO1 600.1337060 MHz  
SI 32768  
SF 600.1300221 MHz  
WDW EM  
SSB 0  
LB 0.30 Hz  
GB 0  
PC 1.00

<sup>13</sup>C NMR

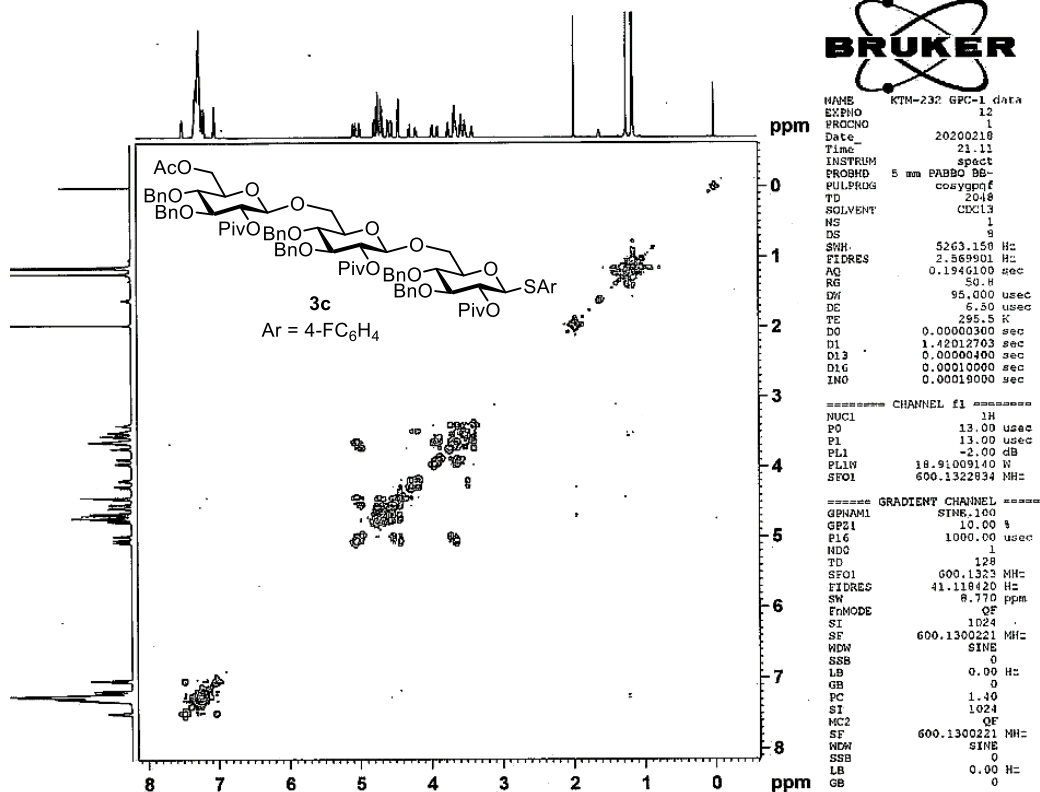


NAME KTM-232 GPC-1 data  
EXPNO 10  
PROCNO 1  
Date\_ 20200218  
Time\_ 21.09  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG -gpg30  
TD 65536  
SOLVENT CDC13  
NS 500  
DS 2  
SWH 36057.691 Hz  
FIDRES 0.550197 Hz  
AQ 0.9088159 sec  
RG 203  
DW 13.867 usec  
DE 6.50 usec  
TE 297.0 K  
D1 2.00000000 sec  
D11 0.03000000 sec  
TDO 1

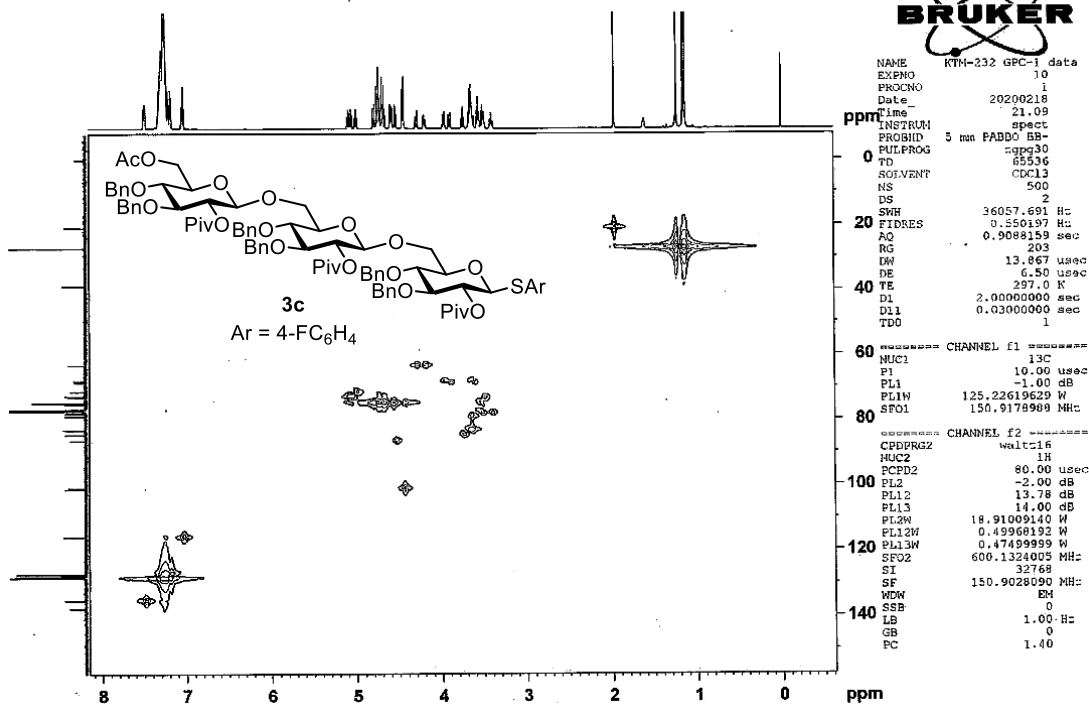
===== CHANNEL f1 =====  
NUC1 13C  
P1 10.00 usec  
PL1 -1.00 dB  
PL1W 125.22619629 W  
SFO1 150.9178988 MHz

===== CHANNEL f2 =====  
CPDPRG2 waltz16  
NUC2 1H  
PCPD2 80.00 usec  
PL2 -2.00 dB  
PL12 13.78 dB  
PL13 14.00 dB  
PL2W 18.91009140 W  
PL12W 0.49968192 W  
PL13W 0.47499999 W  
SFO2 600.1324005 MHz  
SI 32768  
SF 150.9028090 MHz  
WDW EM  
SSB 0  
LB 1.00 Hz  
GB 0  
PC 1.40

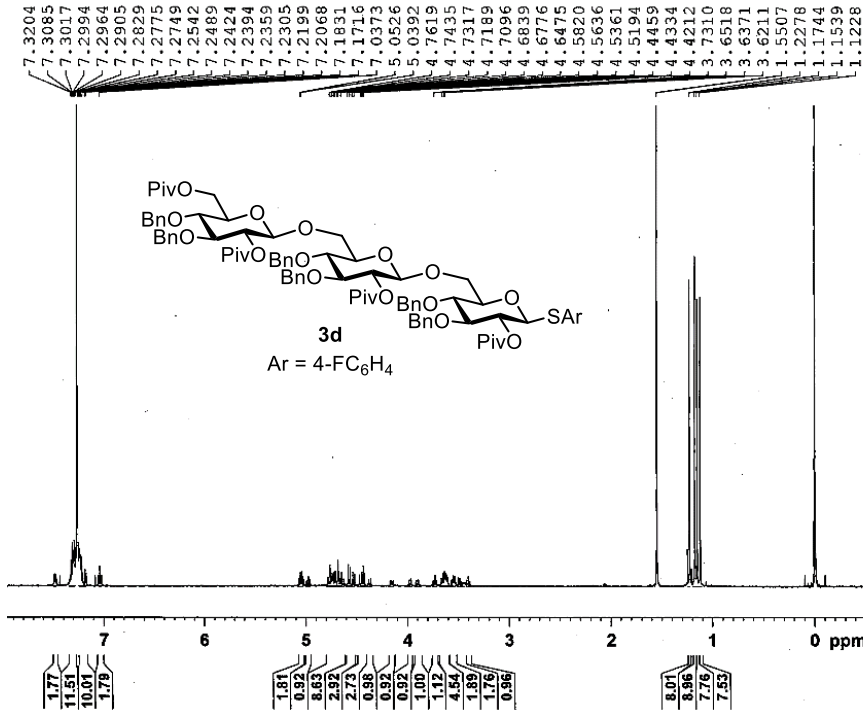
H-H cosy



HMQC



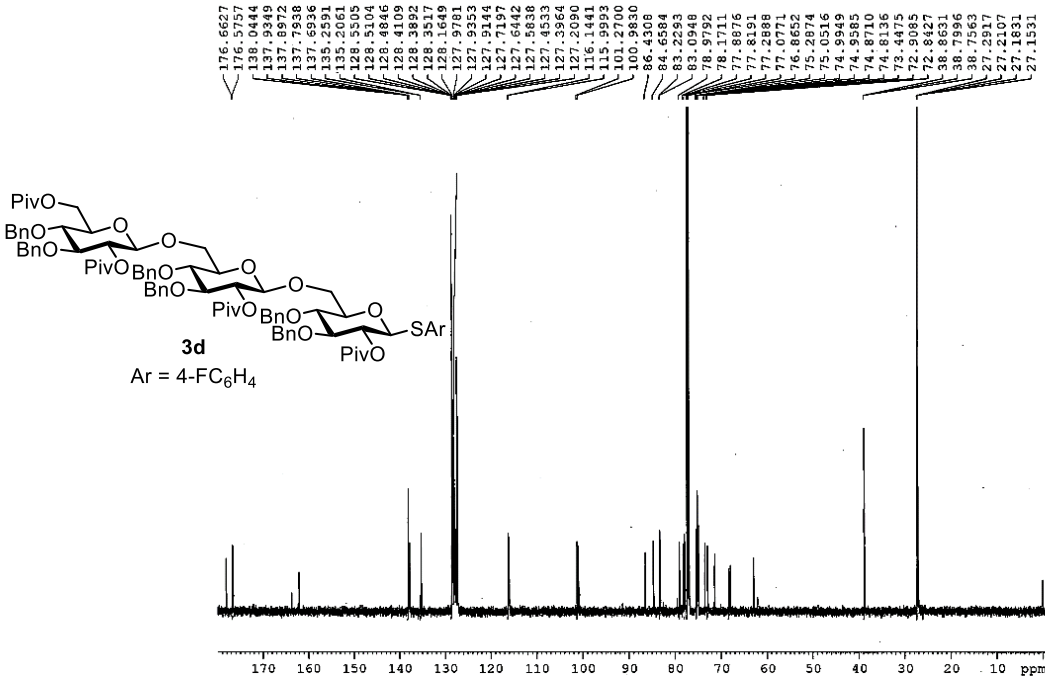
<sup>1</sup>H NMR



```

NAME      KTM-233 fr.54-60
EXPNO    10
PROCNO   1
Date_    20210106
Time     16.42
INSTRUM  spect
PROBHD   5 mm PABBO BB-
PULPROG  zg30
TD       65536
SOLVENT  CDCl3
NS       16
DS       2
SWH      12335.526 Hz
FIDRES   0.188225 Hz
AQ       2.6564426 sec
RG       203
DW       40.533 usec
DE       6.50 usec
TE       294.6 K
D1       1.0000000 sec
TDO      1
----- CHANNEL f1 -----
NUC1     1H
P1       13.00 usec
PL1      -2.00 dB
PL1W    18.91009140 W
SFO1    600.1337060 MHz
SI       32768
SF       600.1300154 MHz
WDW      EM
SSB      0
LB       0.30 Hz
GB       0
PC       1.00
    
```

<sup>13</sup>C NMR

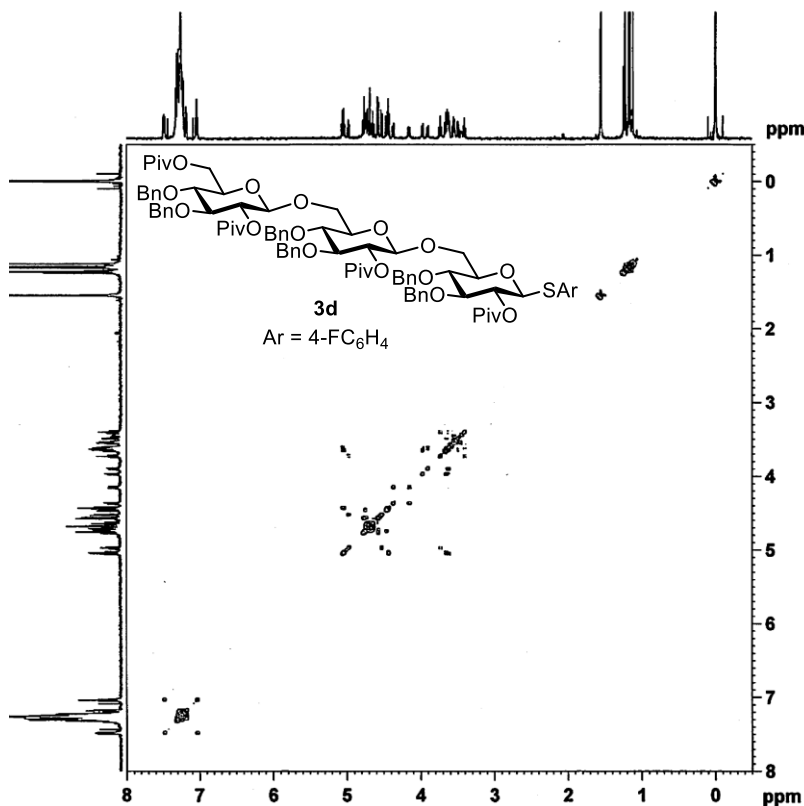


```

NAME      KTM-233 fr.54-60
EXPNO    20
PROCNO   1
Date_    20210112
Time     15.31
INSTRUM  spect
PROBHD   5 mm PABBO BB-
PULPROG  zgpg30
TD       65536
SOLVENT  CDCl3
NS       500
DS       2
SWH      36057.691 Hz
FIDRES   0.550197 Hz
AQ       0.9068159 sec
RG       203
DW       13.867 usec
DE       6.50 usec
TE       295.5 K
D1       2.0000000 sec
D11      0.0300000 sec
TDO      1
----- CHANNEL f1 -----
NUC1     13C
P1       10.00 usec
PL1      -1.00 dB
PL1W    125.22619629 W
SFO1    150.9176988 MHz
----- CHANNEL f2 -----
CPDPRG2  waltz16
NUC2     1H
PCPD2    80.00 usec
PL2      -2.00 dB
PL12     13.78 dB
PL13     14.00 dB
PL2W    18.91009140 W
PL12W   0.49968192 W
PL13W   0.47499999 W
SFO2    600.1324005 MHz
SI       32768
SF       150.9028090 MHz
WDW      EM
SSB      0
LB       1.00 Hz
GB       0
PC       1.40
    
```



H-H cosy



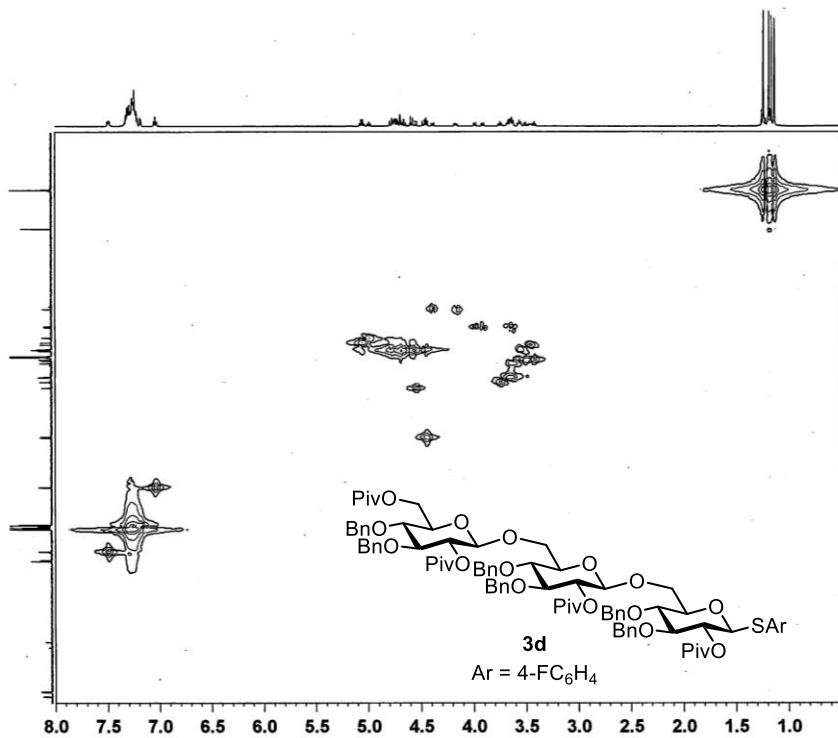
```

NAME      KTM-233 fr.54-60
EXPNO     11
PROCNO    1
Date_     20210126
Time      16.42
INSTRUM   spect
PROBHD    5 mm PABBO BB-
PULPROG   cosyppq4
TD         2048
SOLVENT   CDCl3
NS         1
DS         8
SWH        5208.333 Hz
FIDRES     2.543132 Hz
AQ         0.1966580 sec
RG         203
DM         96.000 usec
DE         6.50 usec
TE         294.6 K
DO         0.00000300 sec
D1         1.41807902 sec
D13        0.00000400 sec
D16        0.00010000 sec
INO        0.00019200 sec

===== CHANNEL f1 =====
NUC1       1H
FO         13.00 usec
P1         13.00 usec
PL1        -2.00 dB
PL1W       18.91009140 W
SFO1       600.1322427 MHz

===== GRADIENT CHANNEL =====
GPNAM1     SINE.100
GP21       10.00 usec
P16        1000.00 usec
ND0        1
TD         128
SFO1       600.1322 MHz
FIDRES     40.690105 Hz
SW         8.679 ppm
FhMODE     QF
SI         1024
SF         600.1300154 MHz
WDW        SINE
SSB        0
LB         0.00 Hz
GB         0
PC         1.40
SI         1024
MC2        QF
SF         600.1300154 MHz
WDW        SINE
SSB        0
LB         0.00 Hz
GB         0
    
```

HMQC



```

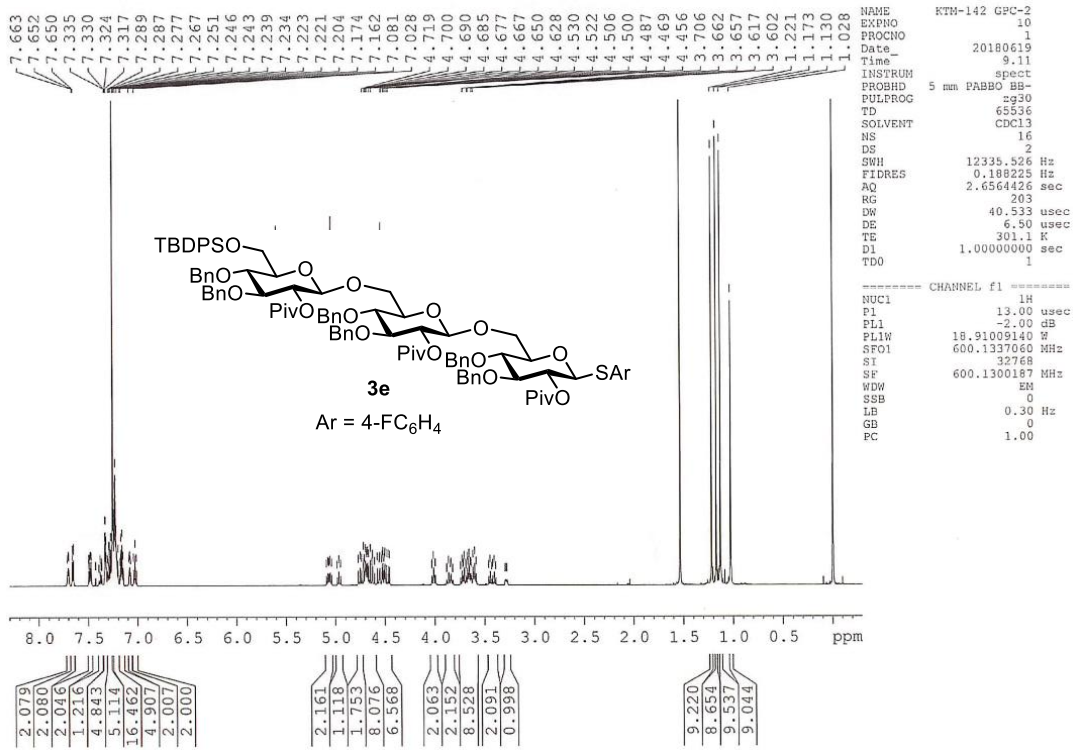
NAME      KTM-233 fr.54-60
EXPNO     12
PROCNO    1
Date_     20210112
Time      15.34
INSTRUM   spect
PROBHD    5 mm PABBO BB-
PULPROG   hmqcpgpgf
TD         1024
SOLVENT   CDCl3
NS         4
DS         16
SWH        4901.961 Hz
FIDRES     4.787071 Hz
AQ         0.1044980 sec
RG         203
DM         102.000 usec
DE         6.50 usec
TE         294.9 K
DO         0.00000300 sec
D1         1.47952857 sec
D2         0.00344828 sec
D12        0.00002000 sec
D13        0.00000400 sec
D16        0.00010000 sec
INO        0.00002000 sec

===== CHANNEL f1 =====
NUC1       1H
FO         13.00 usec
P1         26.00 usec
PL1        -2.00 dB
PL1W       18.91009140 W
SFO1       600.1325185 MHz

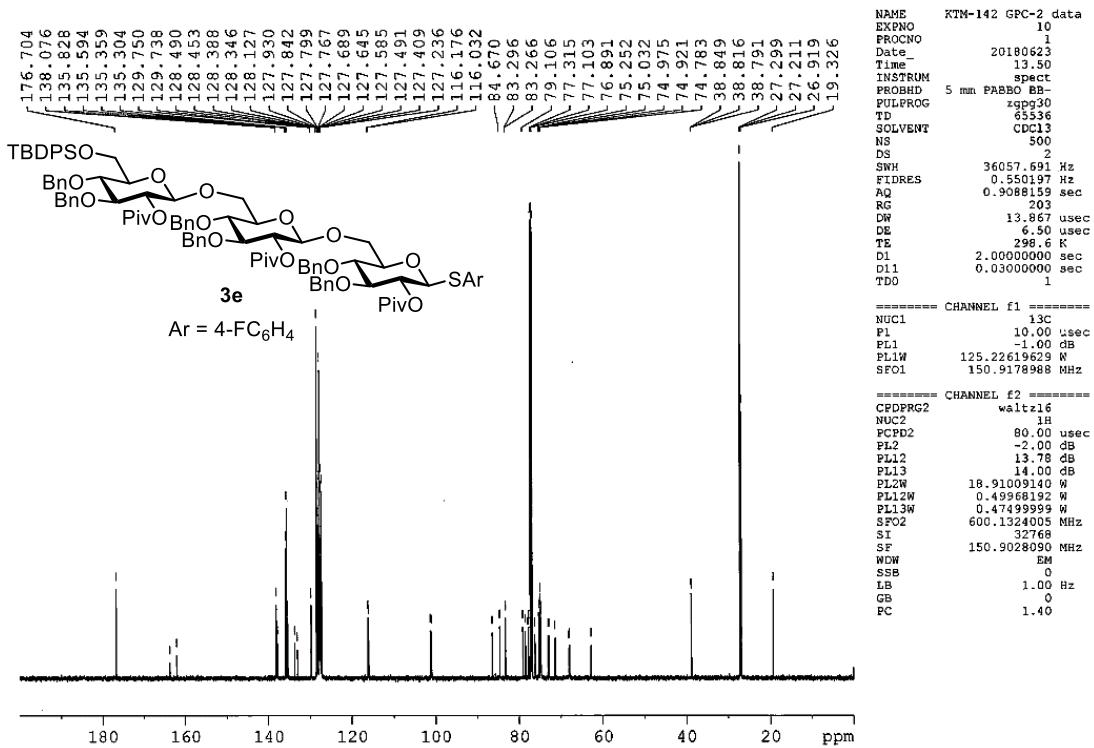
===== CHANNEL f2 =====
CPDPRG2   gmpg
NUC2       13C
FO         10.00 usec
P2         60.00 usec
PL2        -1.00 dB
PL2W       15.10 dB
PL2W       125.22619629 W
PL1W       2.86876655 W
SFO2       150.9141255 MHz

===== GRADIENT CHANNEL =====
GPNAM1     SINE.100
GPNAM2     SINE.100
GP21       10.00 usec
GP22       30.00 usec
GP23       40.10 usec
P16        1000.00 usec
ND0        2
TD         128
SFO1       150.9141 MHz
FIDRES     195.251351 Hz
SW         165.639 ppm
FhMODE     QF
SI         1024
SF         600.1300272 MHz
WDW        SINE
SSB        2
LB         0.00 Hz
GB         0
PC         1.40
SI         1024
MC2        QF
SF         150.9028070 MHz
WDW        SINE
SSB        2
LB         0.00 Hz
GB         0
    
```

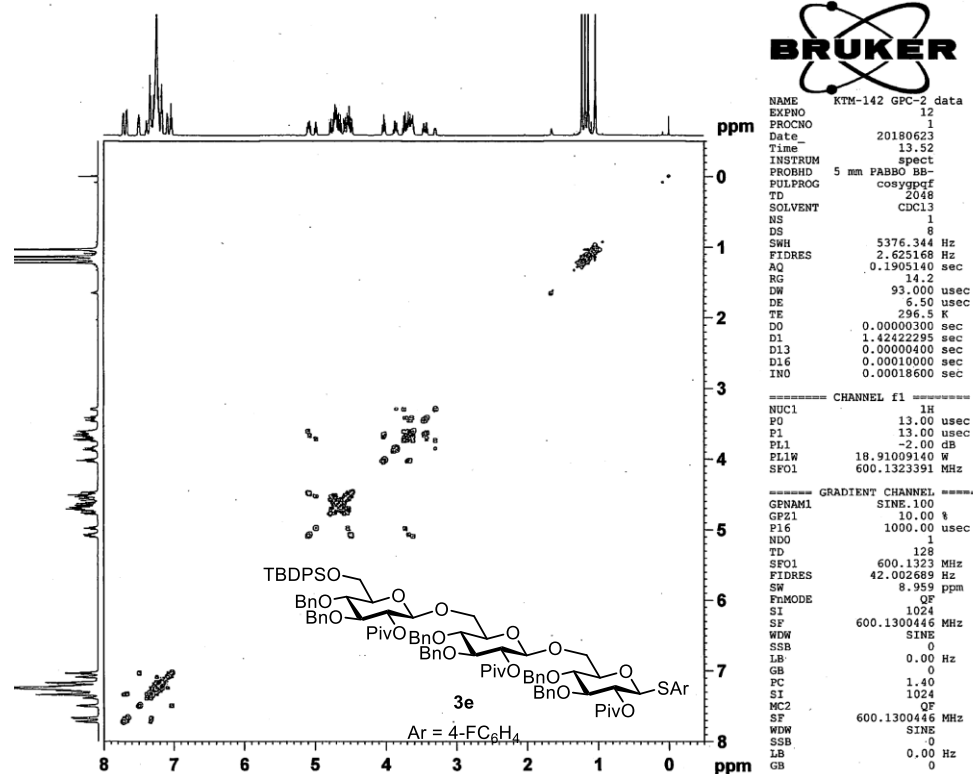
# <sup>1</sup>H NMR



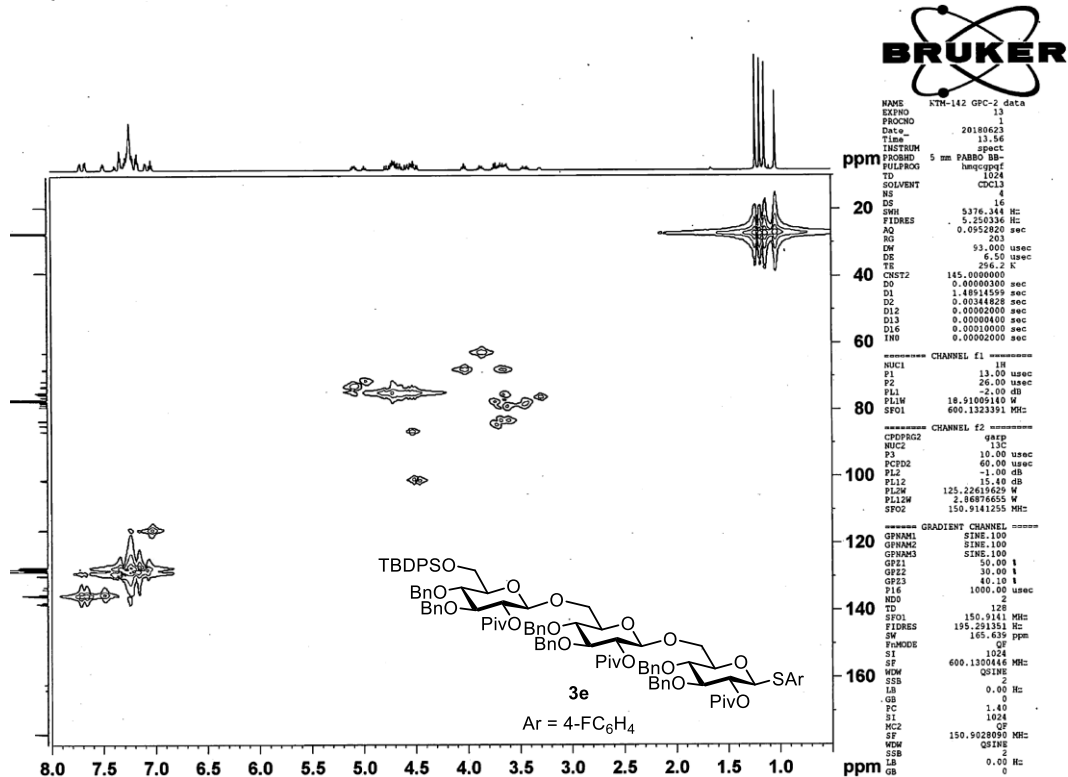
# <sup>13</sup>C NMR



H-H cosy

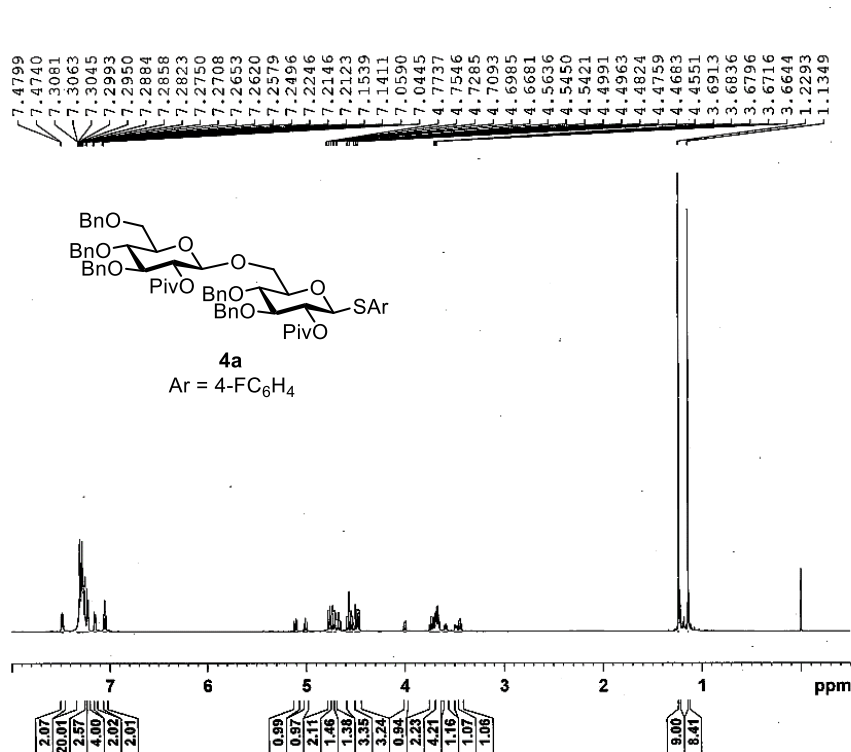


HMQC



# 9. <sup>1</sup>H, <sup>13</sup>C NMR, H-H COSY and HMQC spectra of disaccharides

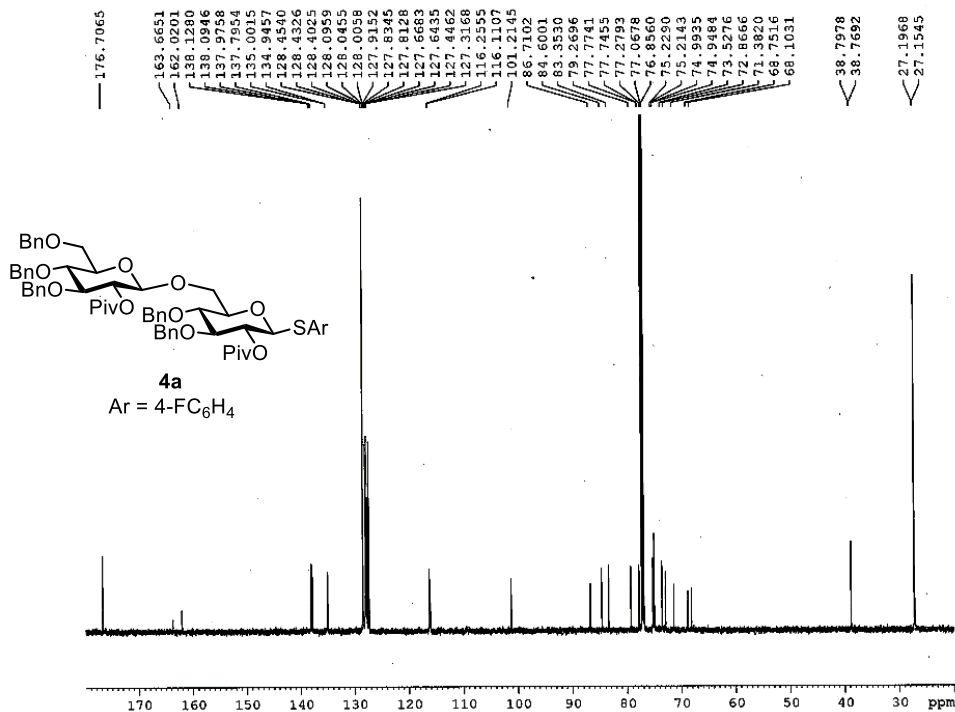
## <sup>1</sup>H NMR



```

NAME      KTM-212 H
EXPNO    10
PROCNO   1
Date_    20210118
Time     18.37
INSTRUM  spect
PROBHD   5 mm PABBO BB-
PULPROG  zg30
TD        65536
SOLVENT  CDCl3
NS        16
DS        2
SWH       12335.526 Hz
FIDRES   0.188225 Hz
AQ        2.6564426 sec
RG        114
DW        40.533 usec
DE        6.50 usec
TE        294.5 K
D1        1.0000000 sec
D11       1
D12       1
D13       1
D14       1
D15       1
D16       1
D17       1
D18       1
D19       1
D20       1
D21       1
D22       1
D23       1
D24       1
D25       1
D26       1
D27       1
D28       1
D29       1
D30       1
===== CHANNEL f1 =====
NUC1      1H
P1        13.00 usec
PL1       -2.00 dB
PL1W      18.91009140 W
SFO1      600.1337060 MHz
SI        32768
SF        600.1300259 MHz
WDW       EM
SSB       0
LB        0.30 Hz
GB        0
PC        1.00
    
```

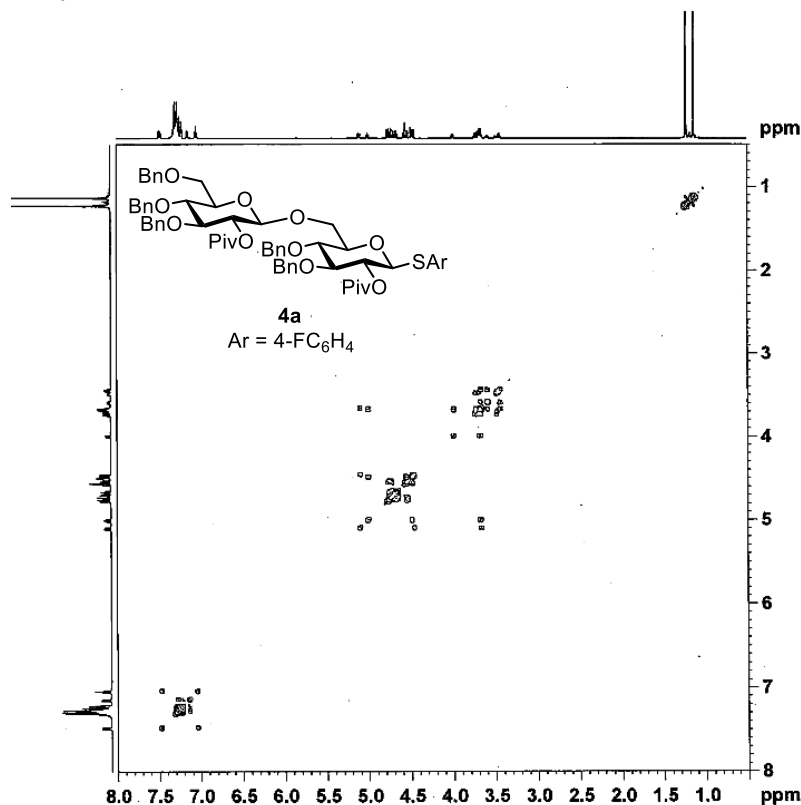
## <sup>13</sup>C NMR



```

NAME      KTM-212
EXPNO    10
PROCNO   1
Date_    20210118
Time     11.28
INSTRUM  spect
PROBHD   5 mm PABBO BB-
PULPROG  zgpg30
TD        65536
SOLVENT  CDCl3
NS        500
DS        2
SWH       36057.691 Hz
FIDRES   0.550197 Hz
AQ        0.9088159 sec
RG        203
DW        13.867 usec
DE        6.50 usec
TE        295.3 K
D1        2.0000000 sec
D11       0.0300000 sec
D12       1
D13       1
D14       1
D15       1
D16       1
D17       1
D18       1
D19       1
D20       1
D21       1
D22       1
D23       1
D24       1
D25       1
D26       1
D27       1
D28       1
D29       1
D30       1
===== CHANNEL f1 =====
NUC1      13C
P1        10.00 usec
PL1       -1.00 dB
PL1W      125.22619629 W
SFO1      150.9178988 MHz
===== CHANNEL f2 =====
CPDPRG2  waltz16
NUC2      1H
PCPD2    80.00 usec
PL2       -2.00 dB
PL12      13.78 dB
PL13      14.00 dB
PL2W      18.91009140 W
PL12W     0.49968192 W
PL13W     0.47499999 W
SFO2      600.1324005 MHz
SI        32768
SF        150.9028090 MHz
WDW       EM
SSB       0
LB        1.00 Hz
GB        0
PC        1.40
    
```

H-H cosy



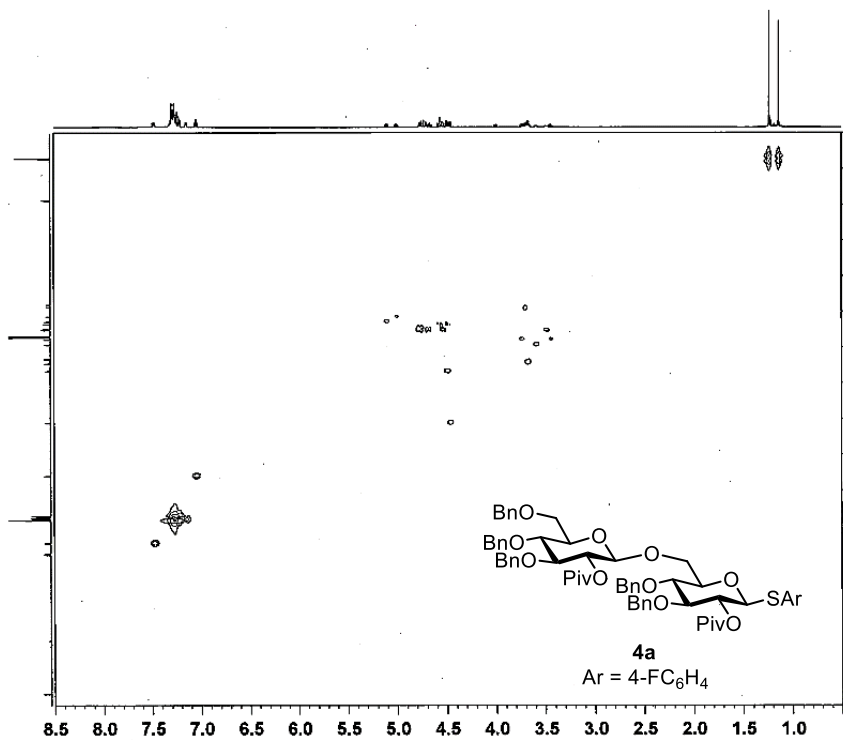
```

NAME      KTM-212 H
EXPNO     11
PROCNO    1
Date_     20210118
Time      18.37
INSTRUM   spect
PROBHD    5 mm PABBO BB-
PULPROG   cosyppg4
TD         4048
SOLVENT   CDCl3
NS         1
DS         0
SMH       5319.149 Hz
FIDRES    2.597241 Hz
AQ         0.1925620 sec
RG         45.2
DQ         94.000 usec
DE         6.50 usec
TE         294.5 K
DO         0.00000300 sec
D1         1.42217505 sec
D13        0.00000400 sec
D16        0.00010000 sec
INO        0.00018900 sec

===== CHANNEL f1 =====
NUC1       1H
P1         13.00 usec
P2         13.00 usec
PL1        -2.00 dB
PL12       18.91009140 W
SFO1       600.1322631 MHz

===== GRADIENT CHANNEL =====
GPNAM1     SINE.100
GPZ1       10.00 %
P15        1000.00 usec
TD         1
TE         128
SFO1       600.1323 MHz
FIDRES     41.558851 Hz
SW         8.863 ppm
FQMODE     QF
SI         1024
SF         600.1300259 MHz
WDW        SINE
SSB        0
LB         0.00 Hz
GB         0
PC         1.40
SI         1024
WC2        QF
SF         600.1300259 MHz
WDW        SINE
SSB        0
LB         0.00 Hz
GB         0
  
```

HMQC



```

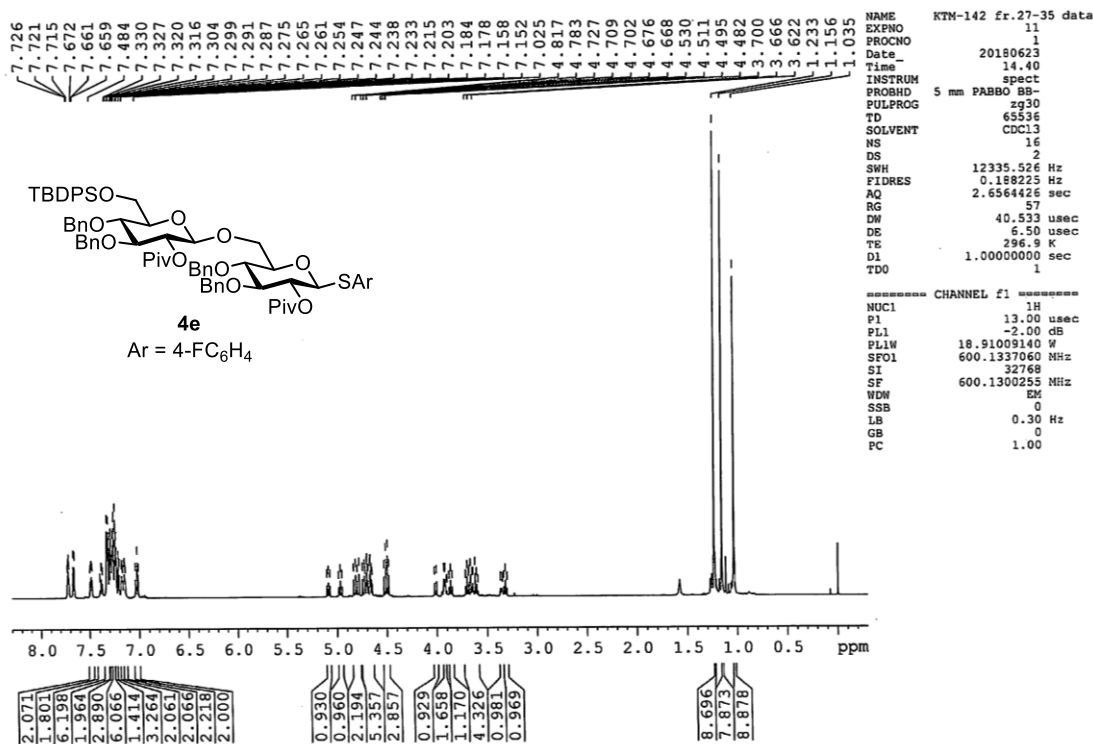
NAME      KTM-212
EXPNO     12
PROCNO    1
Date_     20210118
Time      11.30
INSTRUM   spect
PROBHD    5 mm PABBO BB-
PULPROG   hmqcpg4
TD         1024
SOLVENT   CDCl3
NS         4
DS         16
SMH       5363.158 Hz
FIDRES    3.139802 Hz
AQ         0.3973309 sec
RG         303
DQ         95.000 usec
DE         6.50 usec
TE         294.5 K
DO         0.00000300 sec
D1         1.48789796 sec
D12        0.00244828 sec
D13        0.00002000 sec
D16        0.00000000 sec
INO        0.00002000 sec

===== CHANNEL f1 =====
NUC1       1H
P1         13.00 usec
P2         26.00 usec
PL1        -2.00 dB
PL12       18.91009140 W
SFO1       600.1322672 MHz

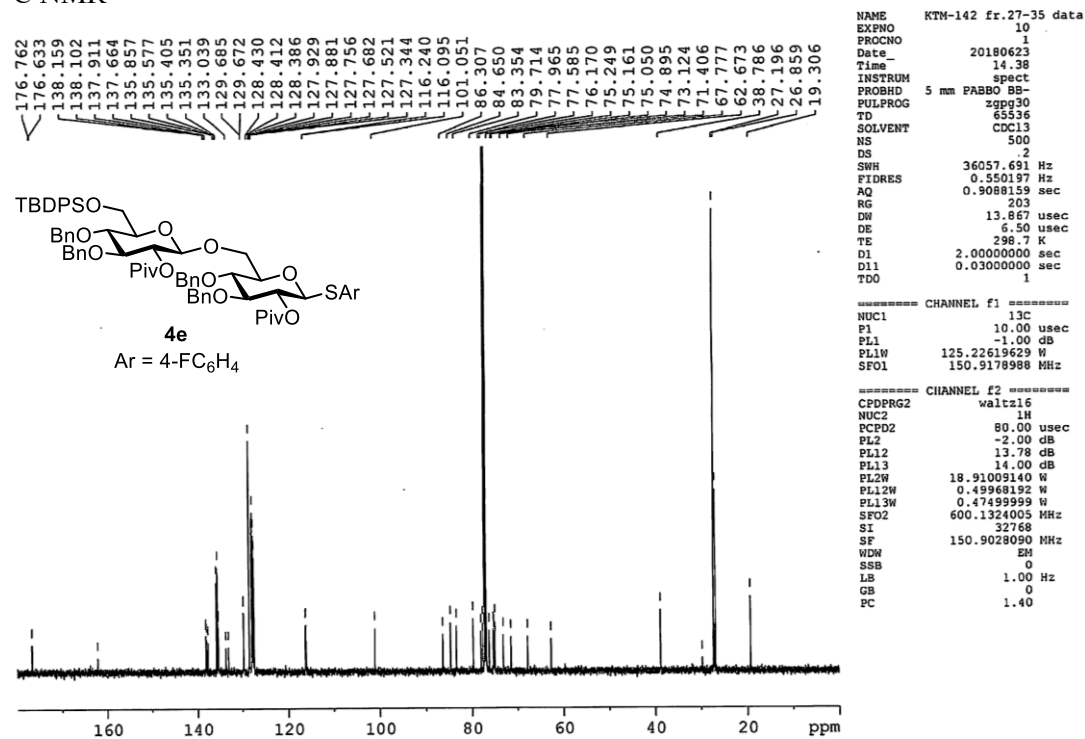
===== CHANNEL f2 =====
CPDPRG2    gpcp
NUC2       13C
P1         10.00 usec
P2         60.00 usec
PL2        -1.00 dB
PL12       15.40 dB
SFO2       125.2261825 MHz
D122W      2.86976655 W
SFO2       150.9141251 MHz

===== GRADIENT CHANNEL =====
GPNAM1     SINE.100
GPNAM2     SINE.100
GPNAM3     SINE.100
GPZ1       50.00 %
GPZ2       50.00 %
GPZ3       40.10 %
P15        1000.00 usec
TD         2
TE         128
SFO1       150.9141 MHz
FIDRES     150.261391 Hz
SW         150.030 ppm
FQMODE     QF
SI         1024
SF         600.1300260 MHz
WDW        SINE
SSB        2
LB         0.00 Hz
GB         0
PC         1.40
SI         1024
WC2        QF
SF         150.9038070 MHz
WDW        SINE
SSB        2
LB         0.00 Hz
GB         0
  
```

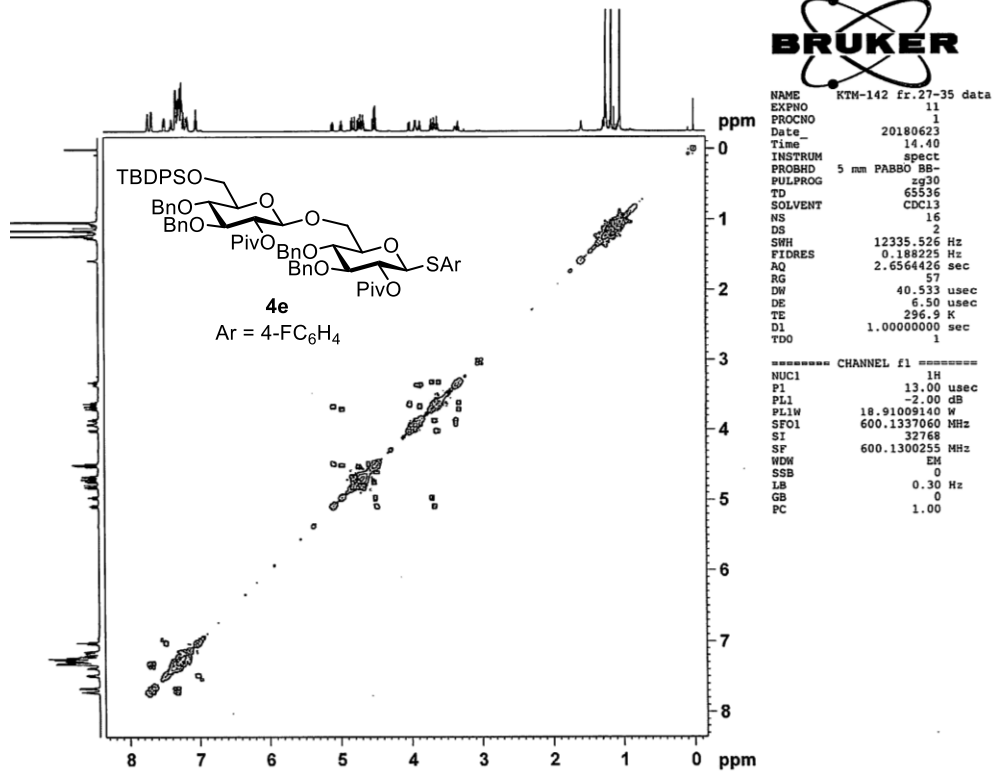
# <sup>1</sup>H NMR



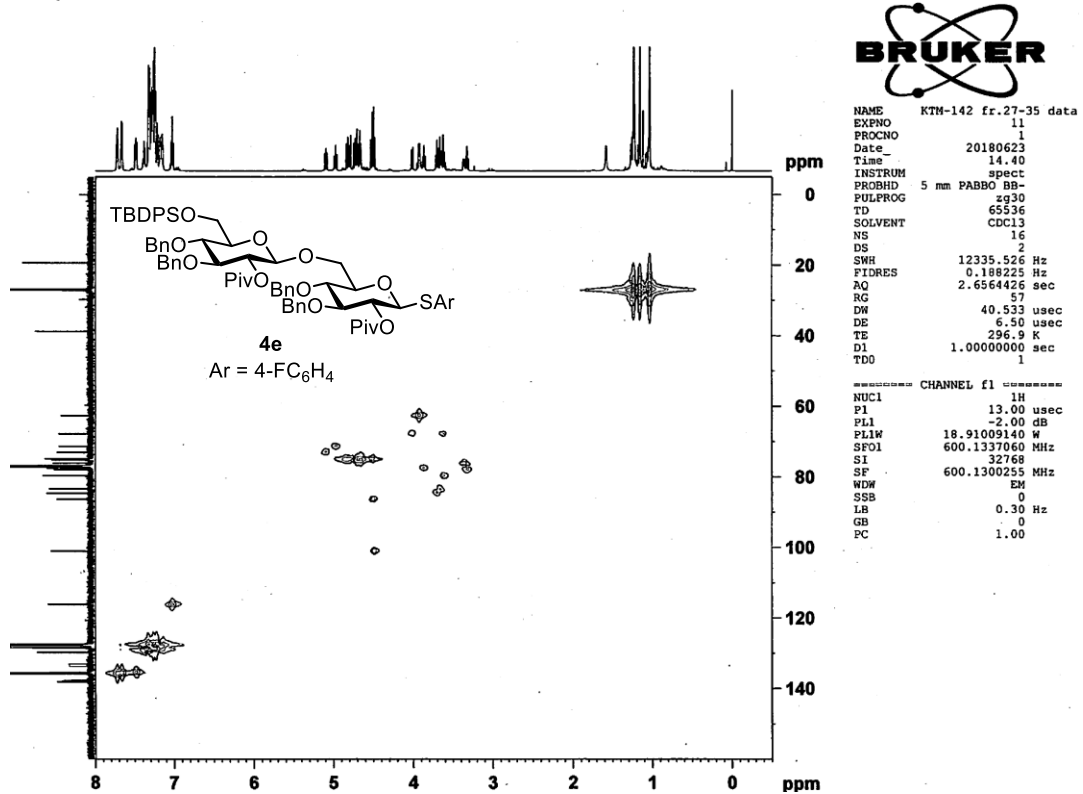
# <sup>13</sup>C NMR



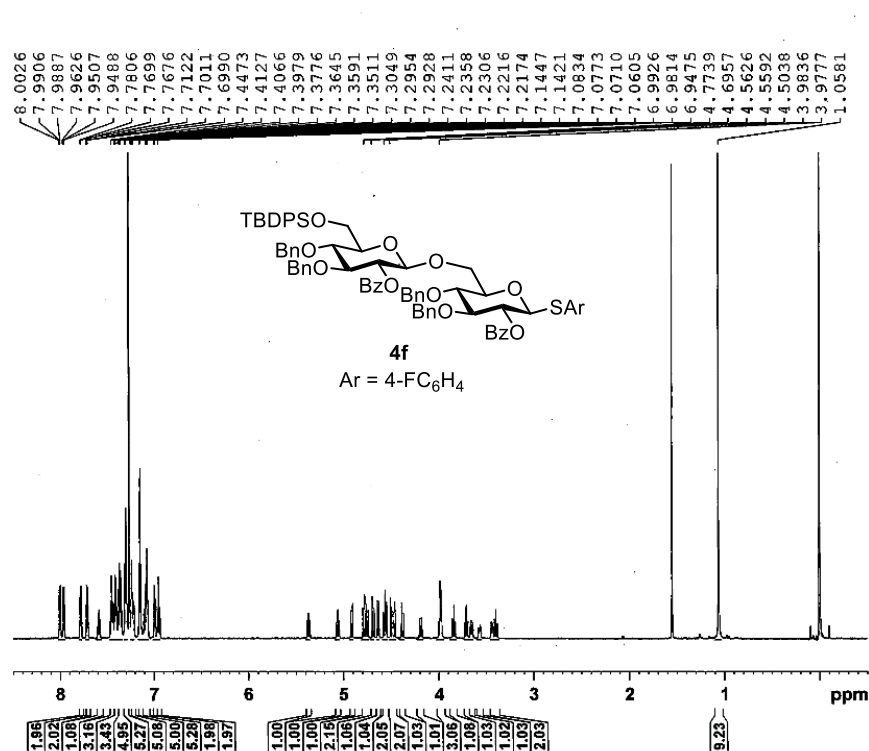
H-H cosy



HMQC



<sup>1</sup>H NMR

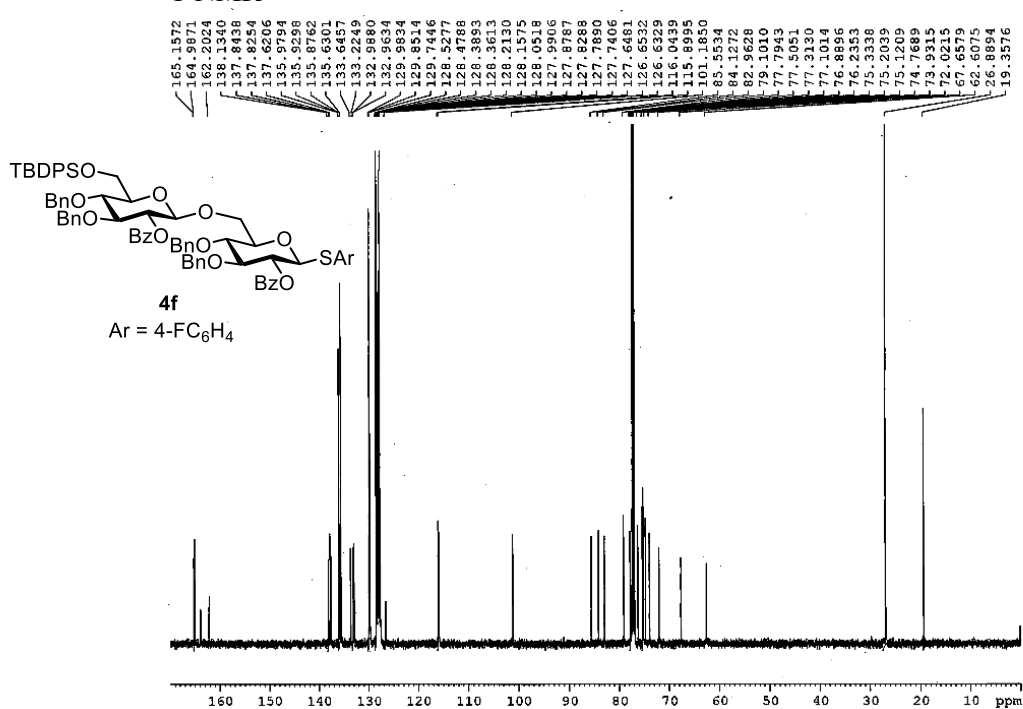


```

NAME      KTM-207
EXPNO    10
PROCNO   1
Date_    20210120
Time     9.28
INSTRUM  spect
PROBHD   5 mm PABBO BB-
PULPROG  zg30
TD        65536
SOLVENT  CDCl3
NS        16
DS        2
SWH       12335.526 Hz
FIDRES    0.189225 Hz
AQ        2.6564426 sec
RG         203
DW        40.533 usec
DE        6.50 usec
TE        293.6 K
D1        1.0000000 sec
TDO       1

----- CHANNEL f1 -----
NUC1      1H
P1        13.00 usec
PL1       -2.00 dB
PL1W      18.91009140 W
SFO1      600.1337060 MHz
SI        32768
SF        600.1300160 MHz
WDW       EM
SSB       0
LB        0.10 Hz
GB        0
PC        1.00
    
```

<sup>13</sup>C NMR



```

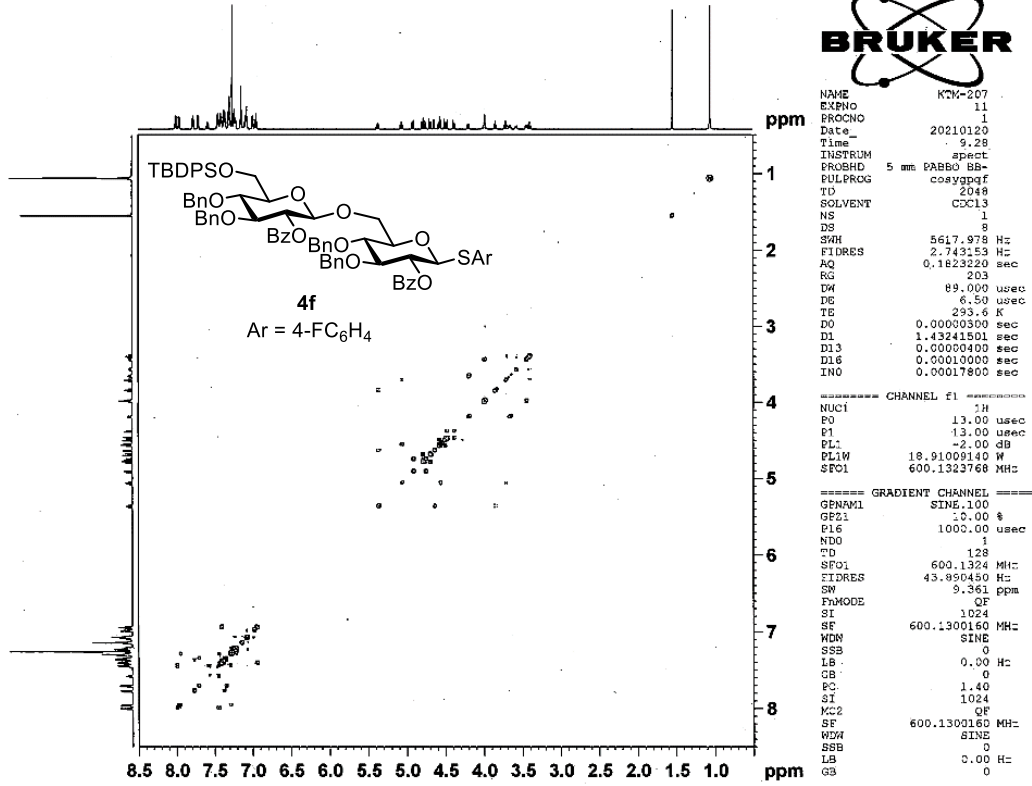
NAME      KTM-207 GPC-1
EXPNO    10
PROCNO   1
Date_    20210212
Time     14.00
INSTRUM  spect
PROBHD   5 mm PABBO BB-
PULPROG  zgpg30
TD        65536
SOLVENT  CDCl3
NS        500
DS        2
SWH       36057.691 Hz
FIDRES    0.550197 Hz
AQ        0.9088159 sec
RG         203
DW        13.967 usec
DE        6.50 usec
TE        297.2 K
D1        2.0000000 sec
D11       0.0300000 sec
TDO       1

----- CHANNEL f1 -----
NUC1      13C
P1        10.00 usec
PL1       -1.00 dB
PL1W      125.22619629 W
SFO1      150.9178988 MHz

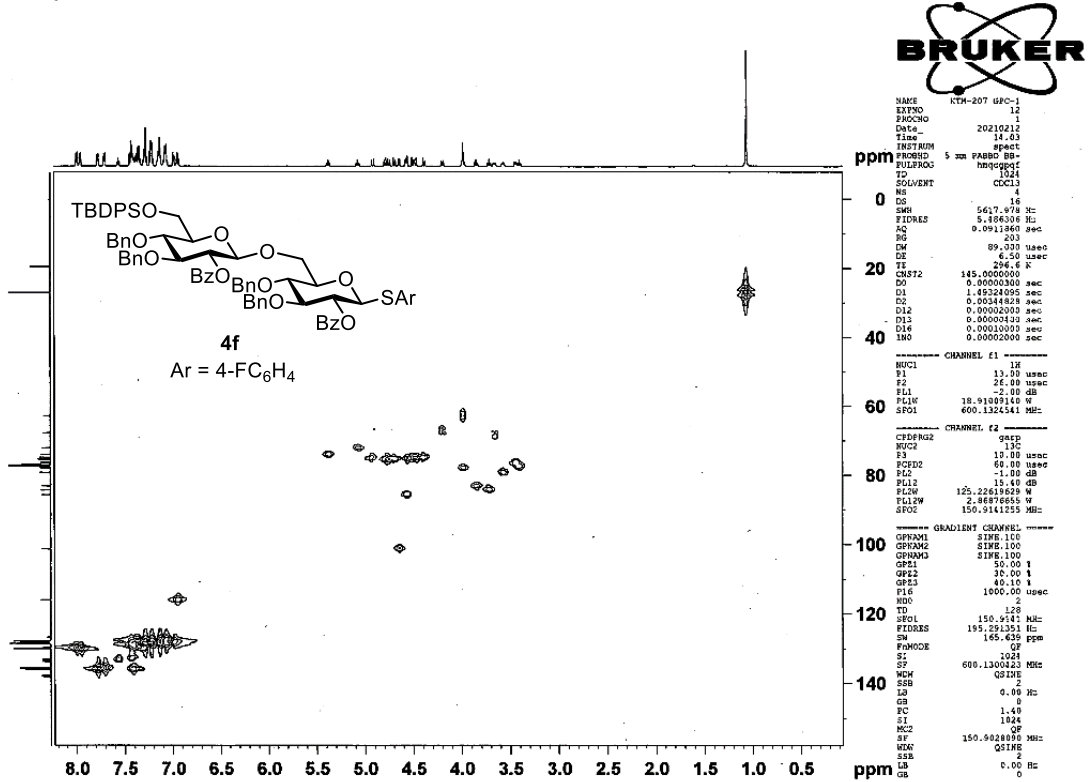
----- CHANNEL f2 -----
CPDPRG2   waltz16
NUC2      1H
PCPD2     80.00 usec
PL2       -2.00 dB
PL12      13.78 dB
PL13      14.00 dB
PL2W      18.91009140 W
PL12W     0.49968192 W
PL13W     0.47499999 W
SFO2      600.1324005 MHz
SI        32768
SF        150.9028090 MHz
WDW       EM
SSB       0
LB        1.00 Hz
GB        0
PC        1.40
    
```



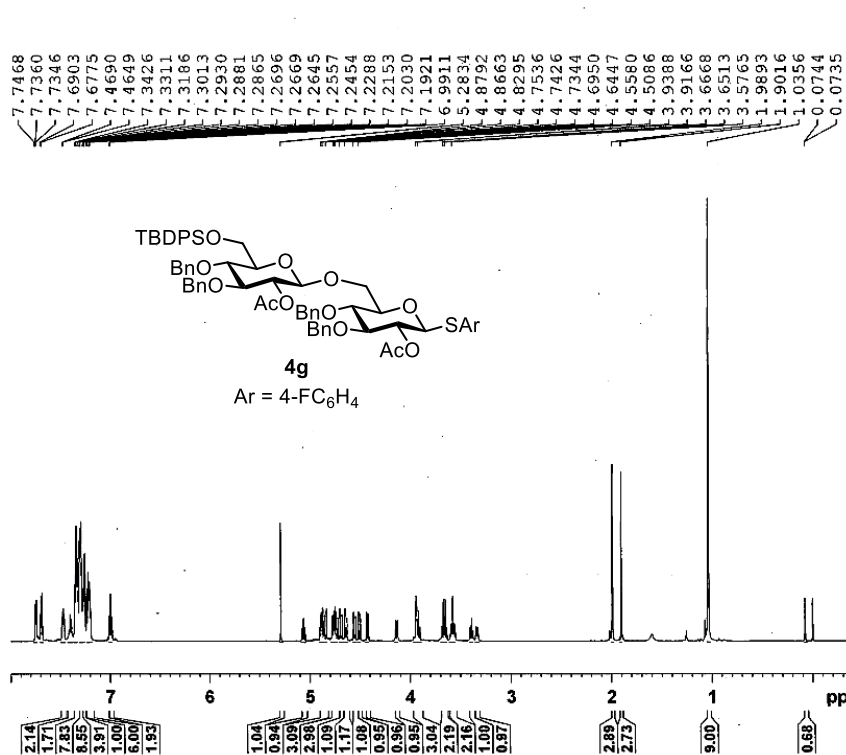
H-H cosy



HMQC



<sup>1</sup>H NMR

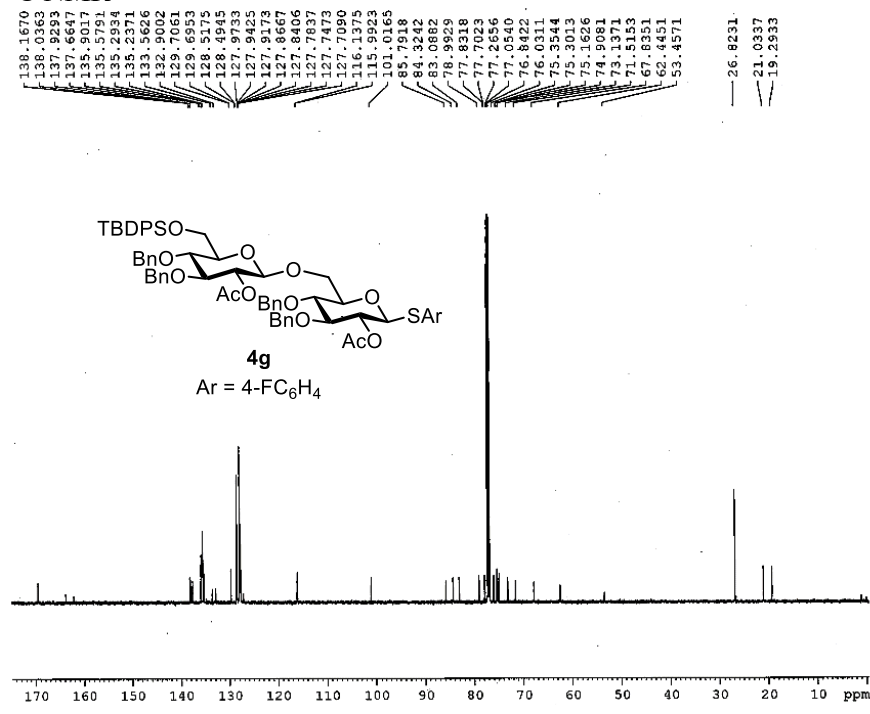


```

NAME          NI-208
EXPNO         10
PROCNO        1
Date_         20210115
Time_         17.05
INSTRUM       spect
PROBHD        5 mm PABBO BB-
PULPROG       zg30
TD            65536
SOLVENT       CDCl3
NS            16
DS            2
SWH           12335.526 Hz
FIDRES        0.188225 Hz
AQ            2.6564426 sec
RG            128
DM            40.933 usec
DE            6.50 usec
TE            295.6 K
D1            1.00000000 sec
D10           1

----- CHANNEL f1 -----
NUC1          1H
P1            13.00 usec
PL1           -2.00 dB
PL1W         18.91009140 W
SFO1         600.1337060 MHz
SI            32768
SF           600.1300250 MHz
WDW           EM
SSB           0
LB            0.30 Hz
GB            0
PC            1.00
    
```

<sup>13</sup>C NMR



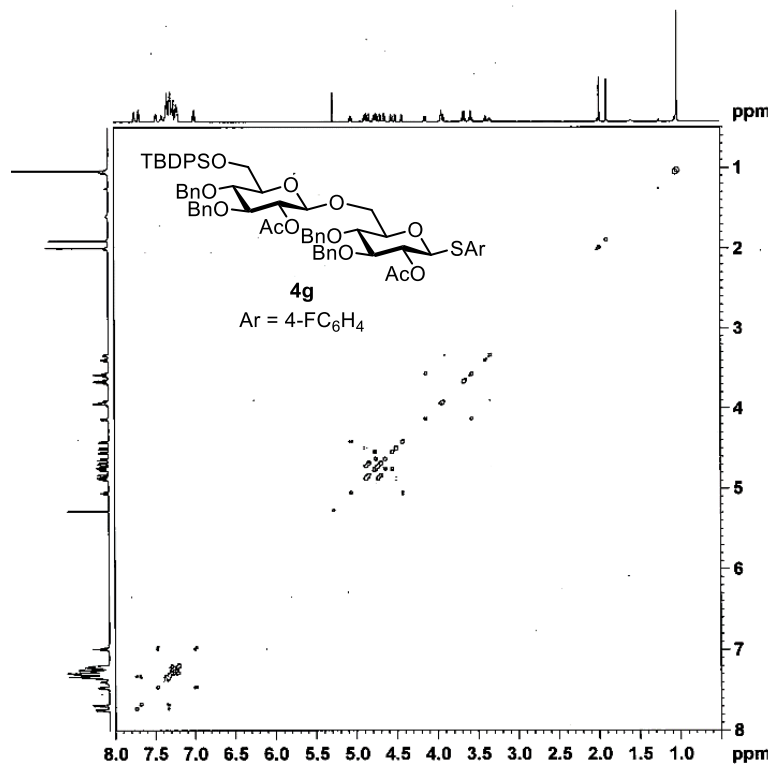
```

NAME          NI-208
EXPNO         12
PROCNO        1
Date_         20210115
Time_         17.35
INSTRUM       spect
PROBHD        5 mm PABBO BB-
PULPROG       zgpg30
TD            65536
SOLVENT       CDCl3
NS            500
DS            2
SWH           36057.691 Hz
FIDRES        0.550197 Hz
AQ            0.9088159 sec
RG            203
DM            13.867 usec
DE            6.50 usec
TE            297.0 K
D1            2.00000000 sec
D11           0.03000000 sec
D10           1

----- CHANNEL f1 -----
NUC1          13C
P1            10.00 usec
PL1           -1.00 dB
PL1W         125.22619629 W
SFO1         150.9178988 MHz

----- CHANNEL f2 -----
CPDPRG2      waltz16
NUC2          1H
PCPD2        80.00 usec
PL2           -2.00 dB
PL12         13.78 dB
PL13         14.00 dB
PL2W         18.91009140 W
PL12W        0.49968192 W
PL13W        0.47499999 W
SFO2         600.1324005 MHz
SI            32768
SF           150.9028090 MHz
WDW           EM
SSB           0
LB            1.00 Hz
GB            0
PC            1.40
    
```

H-H cosy



```

NAME NI-206
EXENO 11
PROCNO 1
Date 20210115
Time 17.06
INSTRUM spect
PROBHD 5 mm PABBO BB-
PULPROG cosypprf
TD 2048
SOLVENT CDCl3
NS 1
DS 8
SWH 5434.783 Hz
FIDRES 2.653702 Hz
AQ 0.1884660 sec
RG 64
DM 92.000 usec
DE 6.50 usec
TE 295.6 K
DO 0.0000300 sec
D1 1.42627096 sec
D13 0.0000400 sec
D16 0.0001000 sec
IN0 0.00018400 sec
  
```

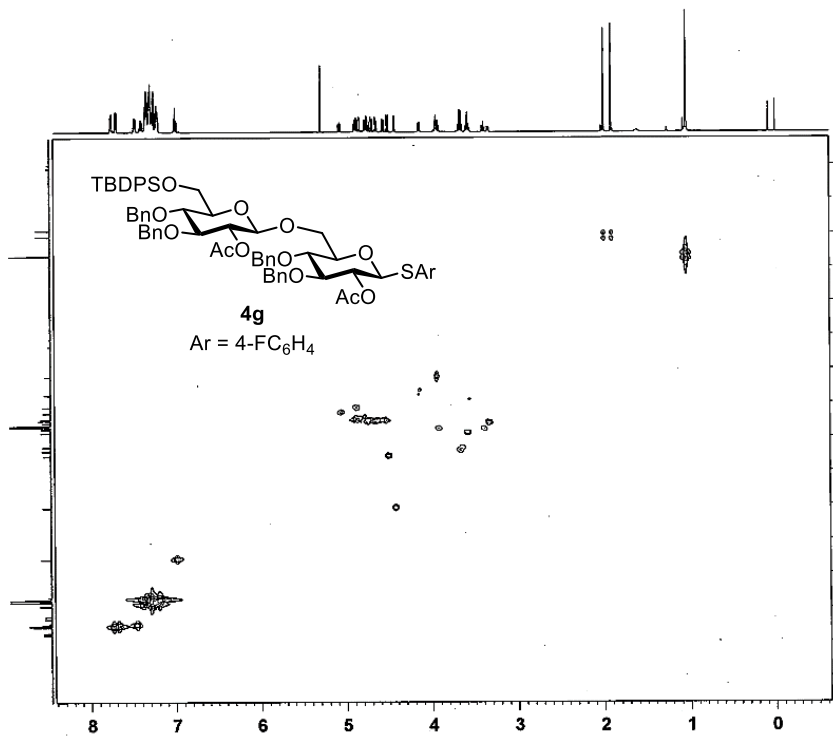
```

----- CHANNEL f1 -----
NUC1 1H
P0 13.00 usec
P1 13.00 usec
PL1 -2.00 dB
PL1W 18.91009140 W
SF01 600.1323880 MHz
  
```

```

----- GRADIENT CHANNEL -----
GENAMI SINE 100
GPZ1 10.00 %
P16 1000.00 usec
ND0 1
TD 128
SF01 600.1324 MHz
FIDRES 42.459240 Hz
SW 9.056 ppm
FMODE QF
SI 1024
SF 600.1300250 MHz
WDM SINE
SSB 0
LB 0.00 Hz
GB 0
FC 1.40
SI 1024
MC2 QF
SF 600.1300250 MHz
WDM SINE
SSB 0
LB 0.00 Hz
GB 0
  
```

HMQC



```

NAME NTH-206 C
EXENO 12
PROCNO 1
Date 20210118
Time 10.33
INSTRUM spect
PROBHD 5 mm PABBO BB-
PULPROG hmccpprf
TD 1624
SOLVENT cdcl3
NS 4
DS 16
SWH 5434.783 Hz
FIDRES 5.307405 Hz
AQ 0.0942580 sec
RG 203
DM 92.000 usec
DE 6.50 usec
TE 295.6 K
DO 0.0000300 sec
D1 1.42616005 sec
D13 0.00348828 sec
D16 0.00002000 sec
D17 0.00004000 sec
D18 0.00010000 sec
D19 0.00002000 sec
  
```

```

----- CHANNEL f1 -----
NUC1 1H
P1 13.00 usec
P2 26.00 usec
PL1 -2.00 dB
PL1W 18.91009140 W
SF01 600.1323880 MHz
  
```

```

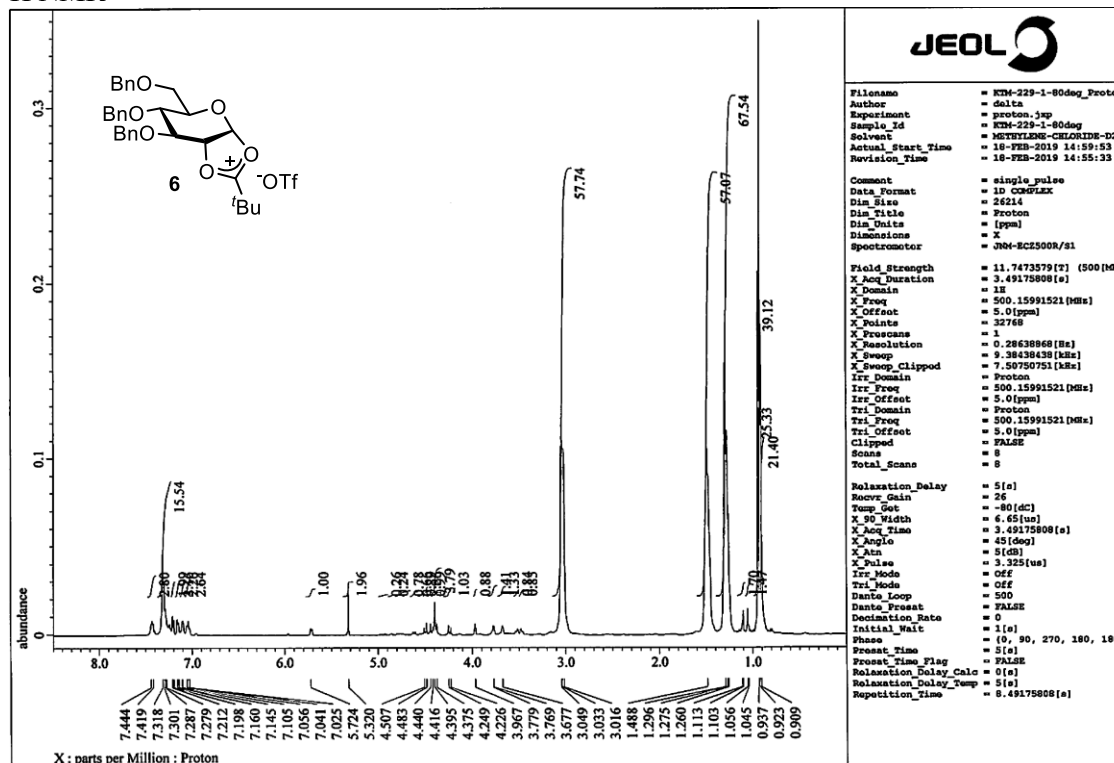
----- CHANNEL f2 -----
CPDPR2 g3p
NUC2 13C
P3 10.00 usec
P4 60.00 usec
PL2 -1.00 dB
PL2 15.40 dB
PL1W 125.22619820 W
PL1W 2.06876655 W
SF02 150.9141255 MHz
  
```

```

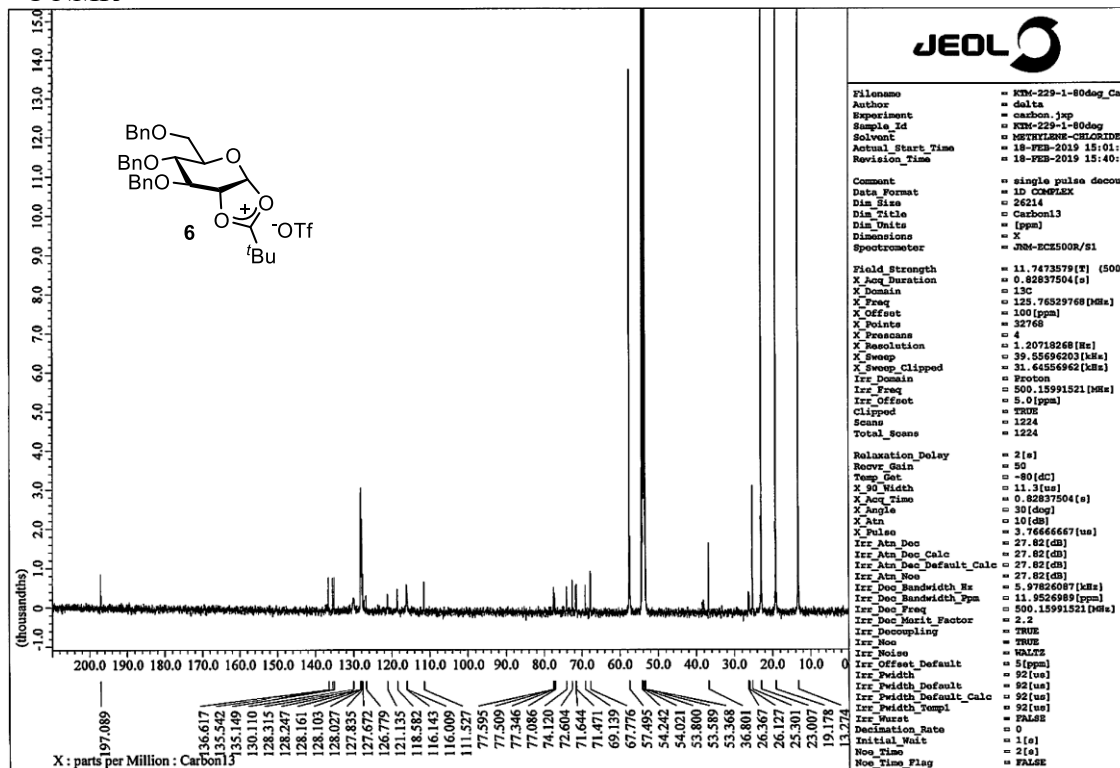
----- GRADIENT CHANNEL -----
GENAM1 SINE 100
GENAM2 SINE 100
GENAM3 SINE 100
GPZ1 50.00 %
GPZ2 38.00 %
GPZ3 40.10 %
P16 1000.00 usec
ND0 2
TD 128
SF01 150.9141 MHz
FIDRES 155.291331 Hz
SW 145.633 ppm
FMODE QF
SI 1024
SF 600.1300250 MHz
WDM QSINE
SSB 2
LB 0.00 Hz
GB 0
FC 1.40
SI 1024
MC2 QF
SF 150.9028090 MHz
WDM QSINE
SSB 2
LB 0.00 Hz
GB 0
  
```

# <sup>1</sup>H, <sup>13</sup>C NMR, H-H COSY and HMQC spectra of glycosyl dioxalenium ions

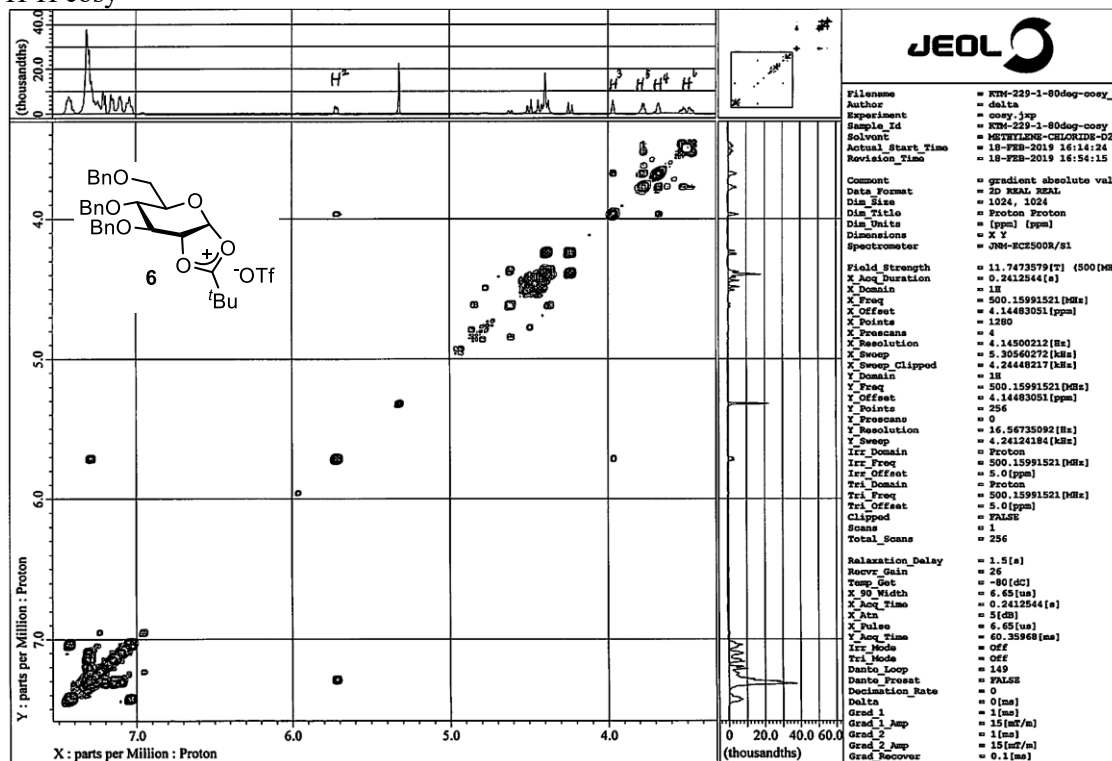
## <sup>1</sup>H NMR



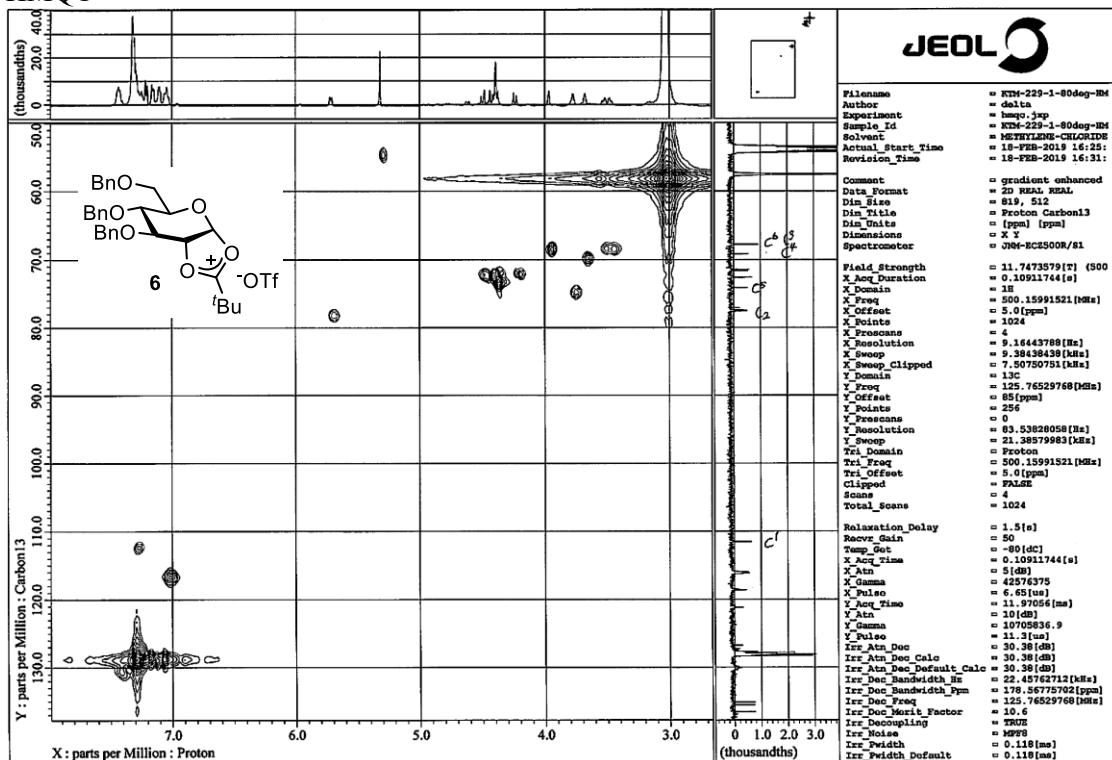
## <sup>13</sup>C NMR



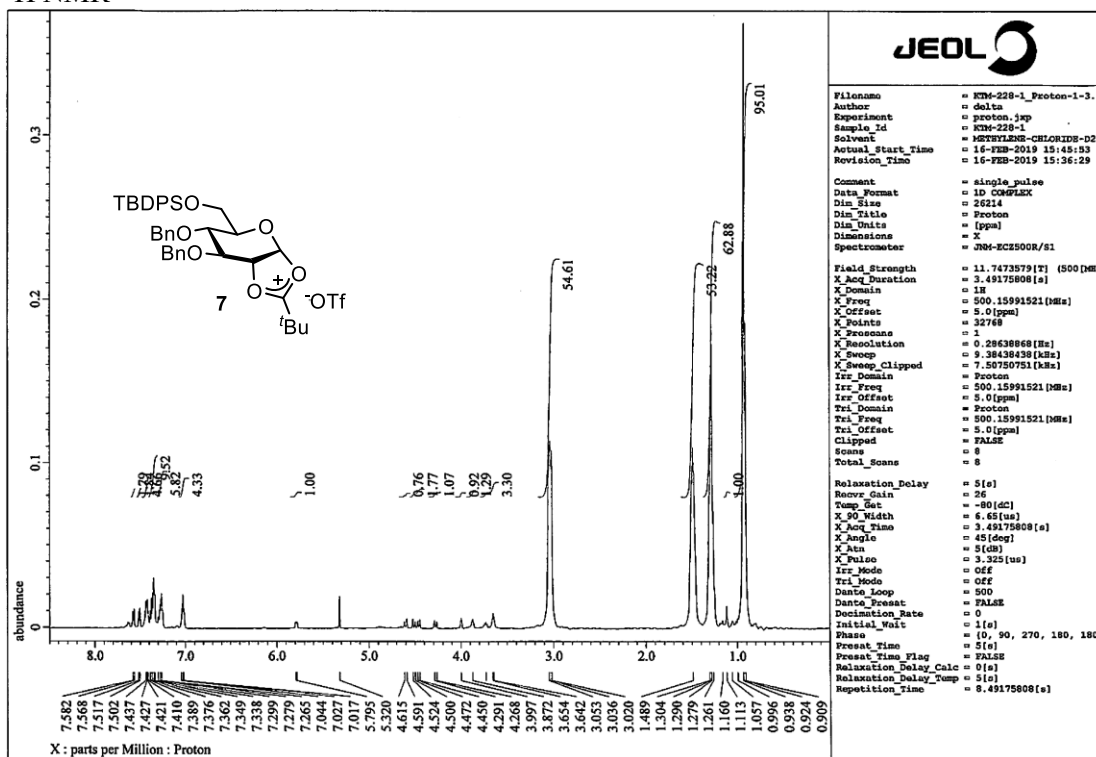
H-H cosy



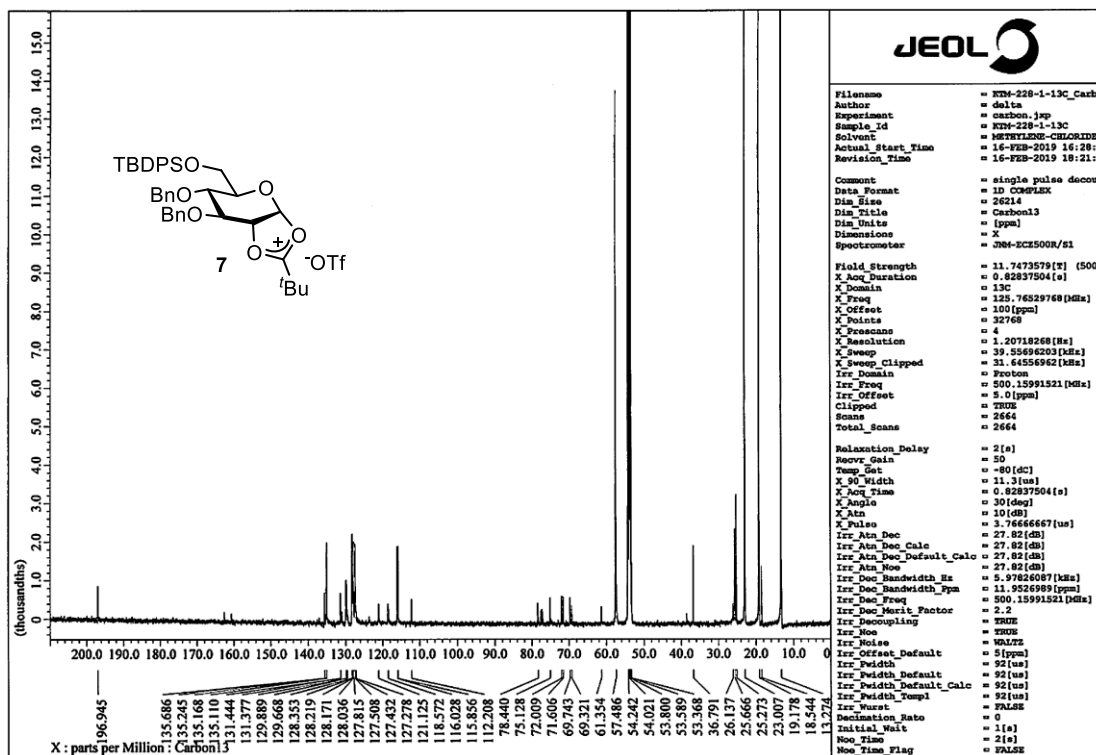
HMQC



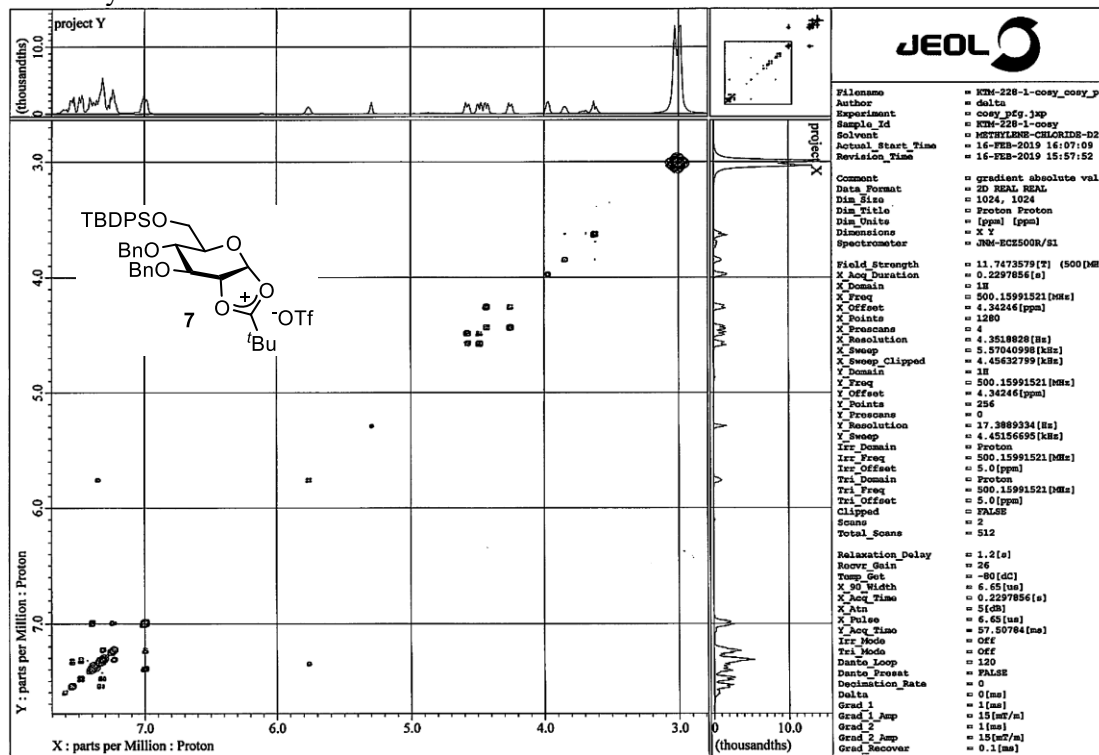
<sup>1</sup>H NMR



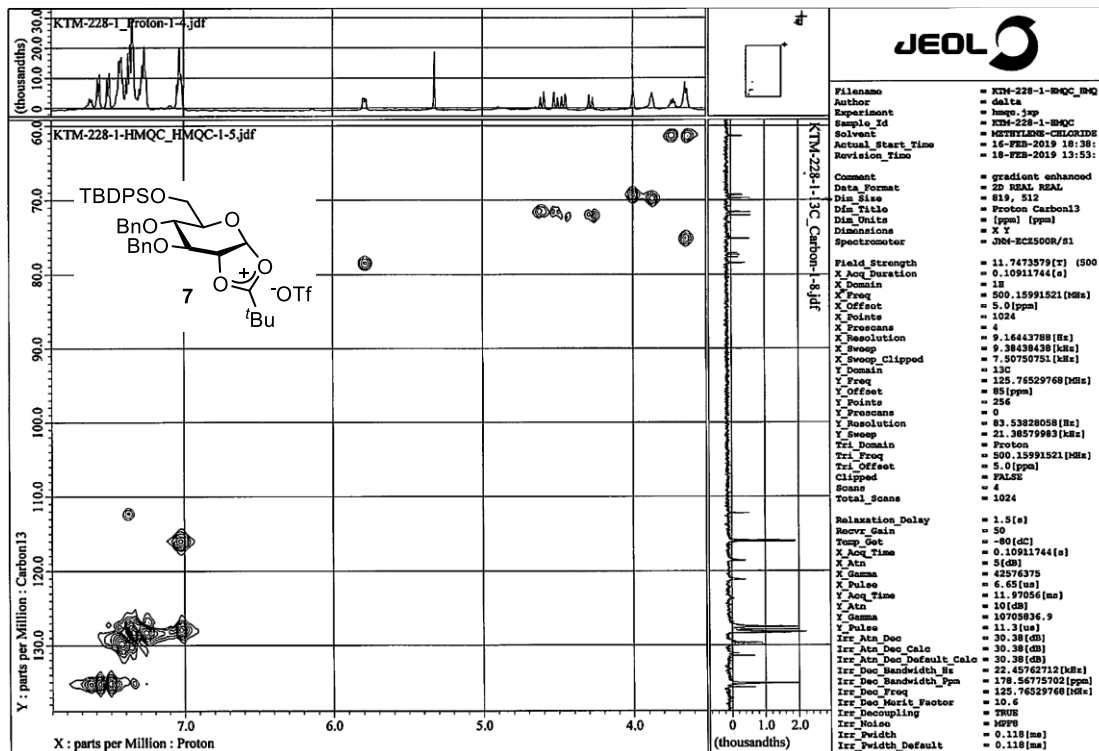
<sup>13</sup>C NMR



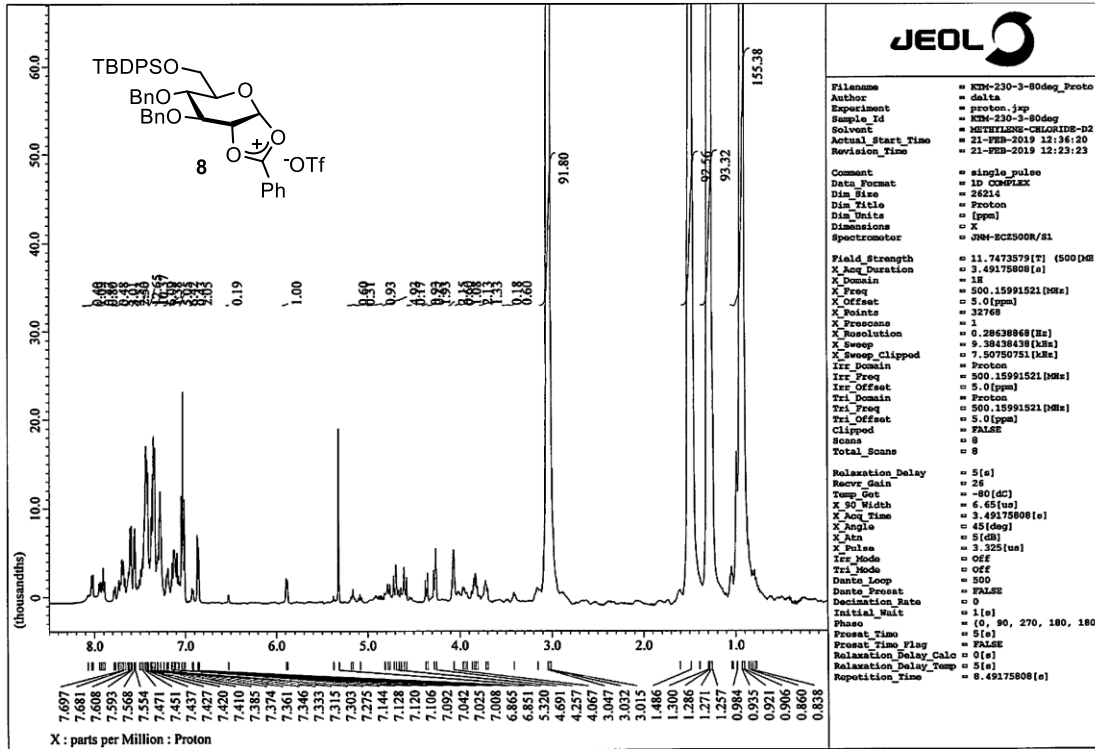
### H-H cosy



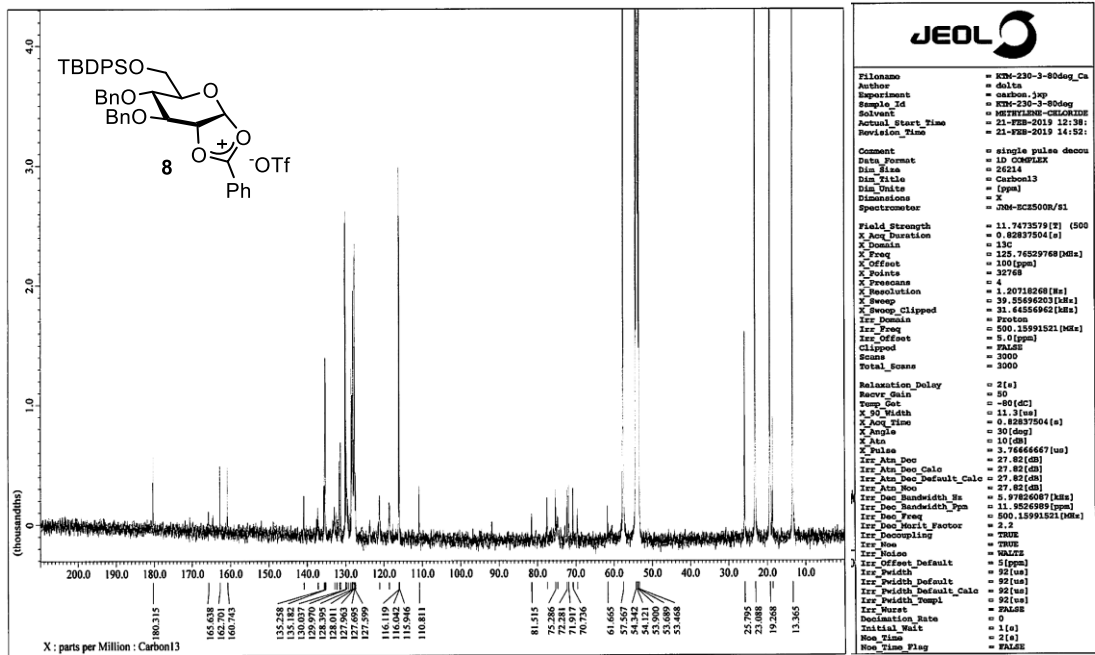
### HMQC



# <sup>1</sup>H NMR

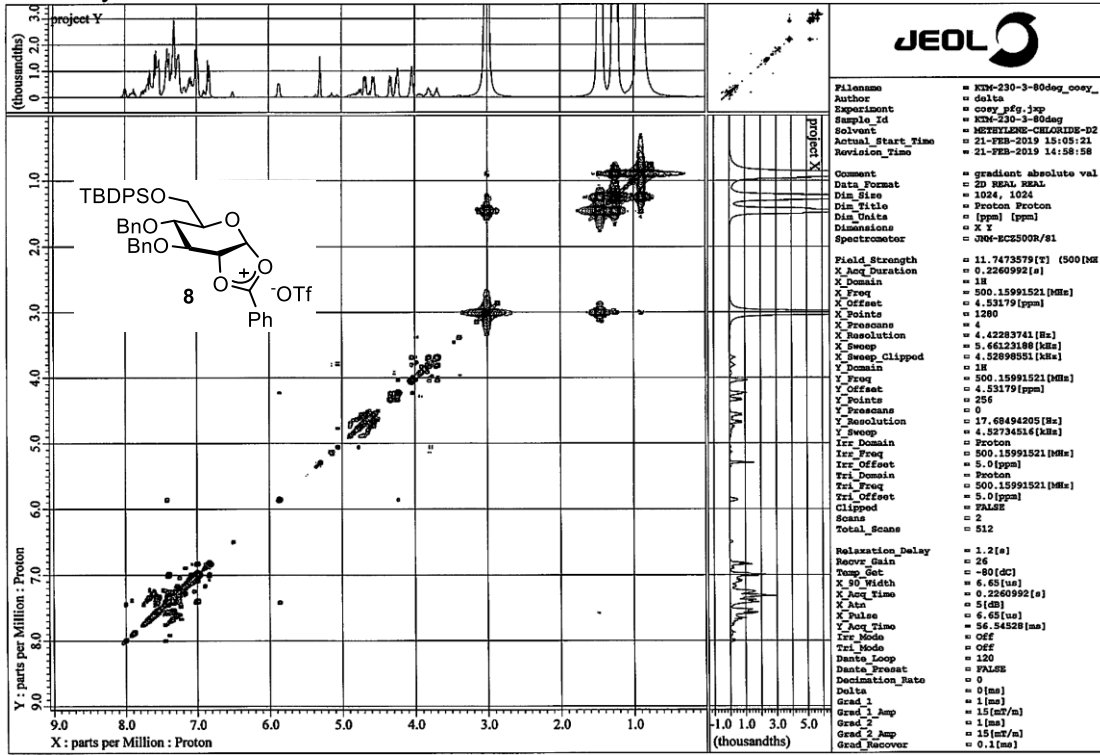


# <sup>13</sup>C NMR

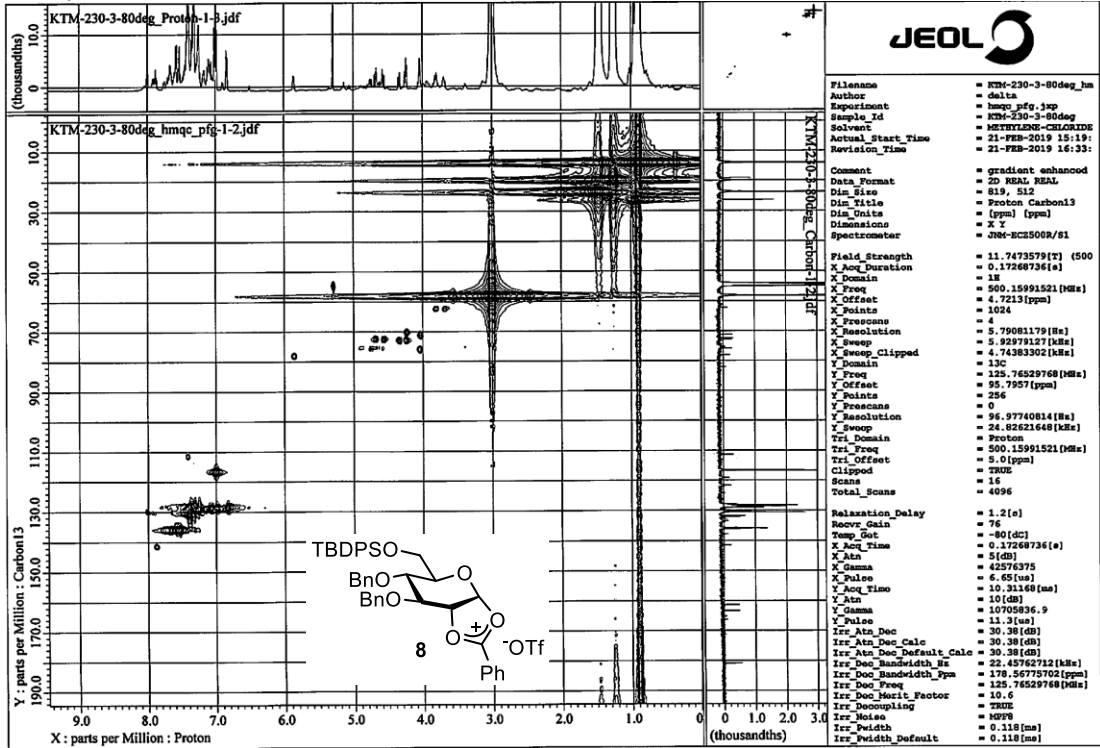




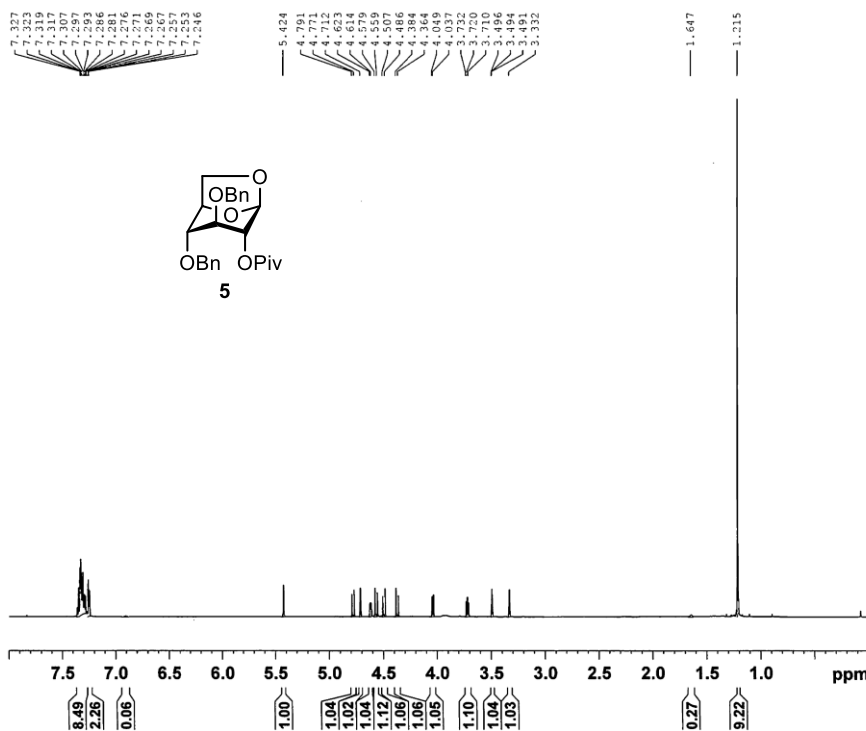
### H-H cosy



### HMQC



# <sup>1</sup>H NMR spectra of 1,6-anhydrosugars

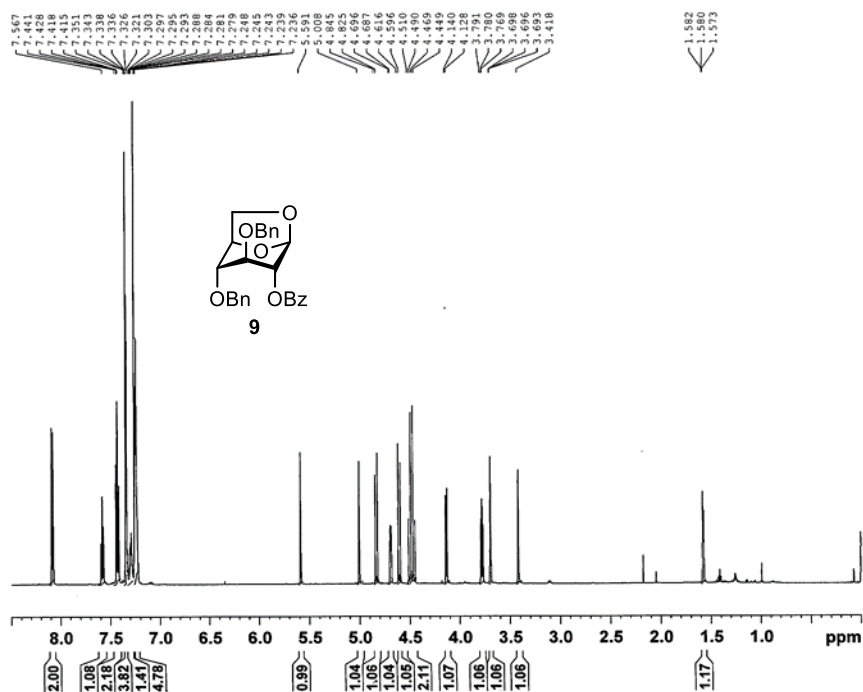


```

NAME      KTM-209 GPC-2
EXPNO     10
PROCNO    1
Date_     20181219
Time      3.22
INSTRUM   spect
PROBHD    5 mm PABBO BB-
PULPROG   zg30
TD         65536
SOLVENT   CDC13
NS         16
DS         2
SWH       12335.526 Hz
FIDRES    0.188225 Hz
AQ         2.6564426 sec
RG         64
DW         40.533 usec
DE         6.50 usec
TE         295.1 K
D1         1.00000000 sec
TDO        1
    
```

```

===== CHANNEL f1 =====
NUC1      1H
P1        13.00 usec
PL1       -2.00 dB
PL1W      18.91009140 W
SFO1      600.1337060 MHz
SI         32768
SF         600.1300221 MHz
WDW        EM
SSB        0
LB         0.30 Hz
GB         0
PC         1.00
    
```



```

NAME      KTM-207 GPC-3
EXPNO     10
PROCNO    1
Date_     20190122
Time      8.26
INSTRUM   spect
PROBHD    5 mm PABBO BB-
PULPROG   zg30
TD         65536
SOLVENT   CDC13
NS         16
DS         2
SWH       12335.526 Hz
FIDRES    0.188225 Hz
AQ         2.6564426 sec
RG         161
DW         40.533 usec
DE         6.50 usec
TE         294.3 K
D1         1.00000000 sec
TDO        1
    
```

```

===== CHANNEL f1 =====
NUC1      1H
P1        13.00 usec
PL1       -2.00 dB
PL1W      18.91009140 W
SFO1      600.1337060 MHz
SI         32768
SF         600.1300191 MHz
WDW        EM
SSB        0
LB         0.30 Hz
GB         0
PC         1.00
    
```

## Chapter 2.

# Electrochemical Synthesis of the Protected Cyclic (1,3;1,6)- $\beta$ -Glucan Dodecasaccharide

### Introduction

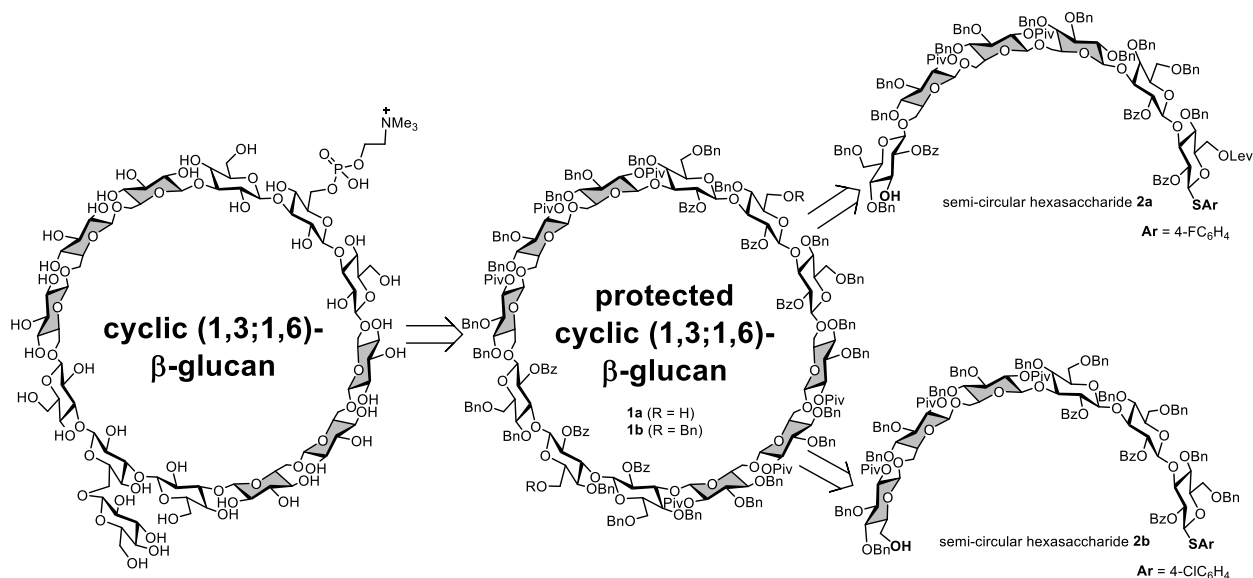
Electrochemical transformations of small molecules have been used as a powerful set of tools in organic synthesis for many decades.<sup>1</sup> Recent progress in this area has enabled the synthesis of complex molecules such as natural products,<sup>2</sup> peptides,<sup>3</sup> and oligosaccharides.<sup>4</sup> We have been interested in the automated synthesis of oligosaccharides using electrochemical methods and have developed a method named ‘automated electrochemical assembly’ (AEA), which is based on electrochemical generation of a glycosylation intermediate and its subsequent coupling with alcohols, including oligosaccharides.<sup>5</sup>

Cyclic oligosaccharides such as cyclodextrins (CDs), which contain 1,4- $\alpha$ -linked D-glucopyranose, have attracted the interest of researchers for more than a century because of their unique structures and properties.<sup>6</sup> To our knowledge,  $\delta$ -CD (nonasaccharide) is the largest CD that has been chemically synthesized to date.<sup>7</sup> With regard to cyclic oligosaccharides containing other glycosidic linkages and monosaccharides, cyclic oligo-1,6- $\beta$ -D-glucosamines up to the heptasaccharide were synthesized by the Nifantiev group<sup>8</sup> and our group.<sup>9</sup> More recently, our group reported the synthesis of cyclic oligo-1,4- $\alpha$ -N-acetylglucosamine ‘cyclokaodorin’ through an electrochemical polyglycosylation-isomerization-cyclization process.<sup>10</sup> In this case, however, only hexasaccharide and heptasaccharide were obtained. Therefore, the chemical synthesis of large cyclic oligosaccharides remains challenging.

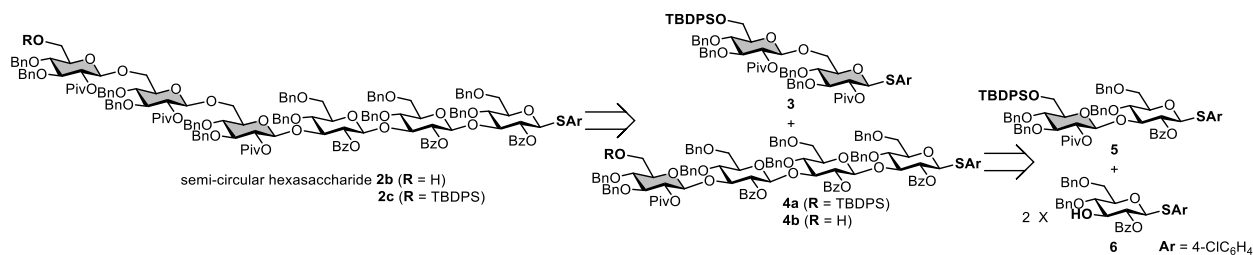
We then focused on a natural oligosaccharide isolated from *Bradyrhizobium japonicum* MTCC120.<sup>11</sup> The oligosaccharide has a cyclic dodecasaccharide structure that consists of two types of glucose trisaccharides with  $\beta$ -(1,3)- and  $\beta$ -(1,6)-glycosidic linkages. Here, we report the electrochemical synthesis of the protected cyclic (1,3;1,6)- $\beta$ -glucan dodecasaccharide as a potential precursor of the natural cyclic dodecasaccharide.

### Results and Discussion

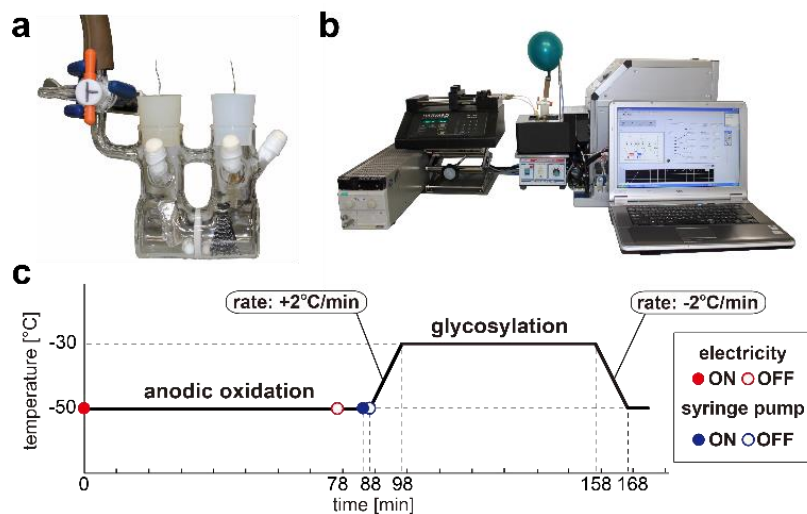
Protected cyclic dodecasaccharide **1** has a symmetric structure that consists of  $\beta$ -(1,3)- and  $\beta$ -(1,6)-glycosidic linkages (Figure 2-1). Thus, we envisioned that an ideal approach to synthesize protected cyclic dodecasaccharide **1** would be through dimerization of the semi-circular hexasaccharide building block **2** followed by cyclization in the same pot. These semi-circular hexasaccharides **2a** and **2b** were considered suitable building blocks because they both have a protecting-group-free hydroxy group (**2a**: 3-OH, **2b**: 6-OH) and thioaryl (SAr, **2a**: Ar = 4-FC<sub>6</sub>H<sub>4</sub>, **2b**: Ar = 4-ClC<sub>6</sub>H<sub>4</sub>) leaving group at the anomeric position (C-1). To examine the hypothesis, we synthesized the semi-circular hexasaccharide building block **2a**, bearing two  $\beta$ -(1,3)-glycosidic linkages and three  $\beta$ -(1,6)-glycosidic linkages.<sup>12</sup> Although **2a** was prepared under the electrochemical conditions, its total yield was very low. Moreover, **2a** had a protecting-group-free 3-OH which must be less reactive than the 6-OH group. Therefore, we designed semi-circular hexasaccharide **2b** as a building block equipped with a protecting-group-free 6-OH. Semi-circular hexasaccharide **2b** could be disconnected to disaccharide building block **3** and tetrasaccharide building block **4b** with  $\beta$ -(1,6)-glycosidic and  $\beta$ -(1,3)-glycosidic linkages, respectively (Figure 2-2). Tetrasaccharide **4a**, as the precursor of **4b** derived from disaccharide building block **5**, with a  $\beta$ -(1,3)-glycosidic linkage, and two equivalents of monosaccharide building block **6**, equipped with the protecting-group-free 3-OH.



**Figure 2-1.** Semi-circular hexasaccharide building blocks for the protected cyclic (1,3; 1,6)- $\beta$ -glucan.



**Figure 2-2.** Retrosynthesis of semi-circular hexasaccharide **2b** and its building blocks **3-6**.

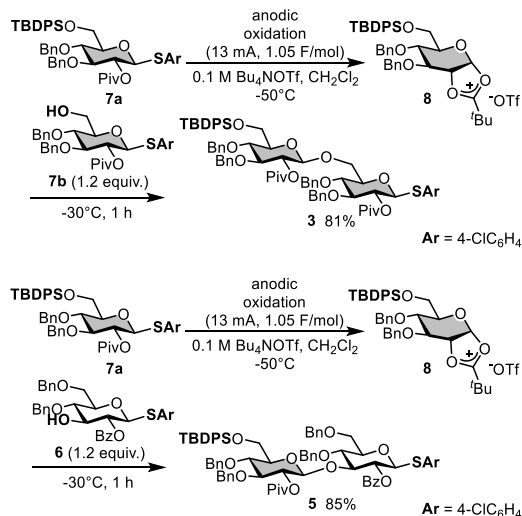


**Figure 2-3.** Devices for automated electrochemical assembly. a) Divided electrolysis cell equipped with platinum plate cathode and carbon fiber anode. b) The 1st generation automated electrochemical synthesizer. c) Schedule of synthesizer for a single cycle.

Automated electrochemical assembly was performed using a middle-size divided electrolysis cell (15 mL for anode and cathode) equipped with carbon fiber anode and platinum plate cathode under argon

atmosphere (Figure 2-3a). The electrolysis was placed in the cooling bath of the 1st generation automated electrochemical synthesizer (Figure 2-3b). The synthesizer was composed of chiller with a cooling bath, stable direct current (DC) power supply, syringe pump, magnetic stirrer, and personal computer (PC), and these devices were controlled by LabVIEW installed in the notebook PC. The schedule of a single cycle was shown in Figure 2-3c. The DC power supply applied a constant current (13 mA, 1.05 F/mol) during the anodic oxidation and the electrolysis time (4677 sec = ca 78 min) depended on both reaction scale (0.60 mmol) and current value (13 mA). The chiller kept two temperatures  $-50\text{ }^{\circ}\text{C}$  and  $-30\text{ }^{\circ}\text{C}$  during anodic oxidation and glycosylation, respectively. In some cases, we switched off the chiller before quenching of the reaction and raised the temperature up to  $0\text{ }^{\circ}\text{C}$  to complete the glycosylation. Two gastight syringes were filled with solution of a building block and solvent for anodic chamber and cathodic chamber, respectively. They were set to the syringe pump and solutions were added at rate 1.0 mL/min after electrolysis.

Monosaccharide building blocks **6**, **7a** and **7b** were prepared from D-glucose pentaacetate according to the reported procedures (Figure 2-4). Disaccharide building block **3**, with a  $\beta$ -(1,6)-glycosidic linkage, was prepared using AEA between **7a** and **7b** in the presence of tetrabutylammonium triflate ( $\text{Bu}_4\text{NOTf}$ ) as an electrolyte. Glycosyl dioxalenium ion intermediate **8** was generated by anodic oxidation of **7a** under constant current conditions at  $-50\text{ }^{\circ}\text{C}$ . Subsequent coupling of **8** and building block **7b** (1.2 equiv.) afforded disaccharide **3** in 81% yield. This is a standard AEA protocol, and the details of reaction conditions and structures of possible intermediates have been omitted from the following figures (see the ESI for details of reaction conditions). Although disaccharide building block **5**, with a  $\beta$ -(1,3)-glycosidic linkage, was also prepared using AEA of **7a** and **6** (1.2 equiv.) in 85% yield, the one-pot synthesis of tetrasaccharide **4a** from monosaccharide building block **7a** using AEA with three cycles was sluggish.



**Figure 2-4.** Synthesis of disaccharide building blocks **3** and **5**.

The electrolyte for AEA was optimized using the electrochemical formation of  $\beta$ -(1,3)-glycosidic linkages using monosaccharides **9** and **6** (1.2 equiv.) as model building blocks (Table 2-1). Whereas the use of tetraethylammonium triflate ( $\text{Et}_4\text{NOTf}$ ) afforded disaccharide **10** in moderate yield (entry 1),  $\text{Bu}_4\text{NOTf}$ , which has been used as a standard electrolyte for AEA, gave the product **10** in good yield (entry 2). We also examined the use of ionic liquids (entries 3–6). The initial voltage of anodic oxidation was significantly influenced by the electrolyte; however, there was no clear relationship between the initial voltage and the

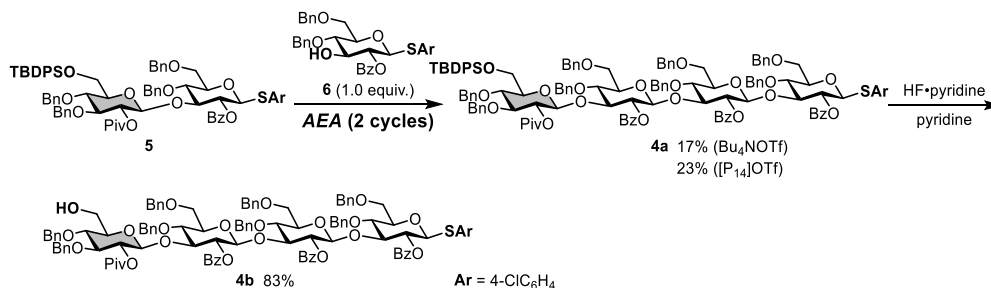
product yield. Amongst these ionic liquids, 1-butyl-1-methylpyrrolidinium triflate ([P<sub>14</sub>]OTf) afforded the desired disaccharide **10** in the highest yield (entry 6). Therefore, we used ionic liquid [P<sub>14</sub>]OTf as an electrolyte in the following glycosylation reactions. It has not been clarified why [P<sub>14</sub>]OTf gave the best yield; however, oxidation potential (E<sub>ox</sub>) of monosaccharide building block **9** measured with [P<sub>14</sub>]OTf (E<sub>ox</sub> = 1.67 V vs. SCE) was slightly lower than that measured with Bu<sub>4</sub>NOTf (E<sub>ox</sub> = 1.70 V vs. SCE). We assume that electrolytes may influence the structure of electrical double layer and the process of single electron transfer.

**Table 2-1.** Optimization of electrolyte of AEA

entry	electrolyte	initial	yield <sup>b</sup>
1	Et <sub>4</sub> NOTf	26 V	61%
2 <sup>a</sup>	Bu <sub>4</sub> NOTf	14 V	79%
3 <sup>a</sup>	[Bmim]OTf	59 V	70%
4	[P <sub>1MOM</sub> ]OTf	93 V	81%
5	[P <sub>1MEM</sub> ]OTf	26 V	89%
6	[P <sub>14</sub> ]OTf	51 V	97%

<sup>a</sup>Inter-electrode voltage. <sup>b</sup>Determined by NMR.

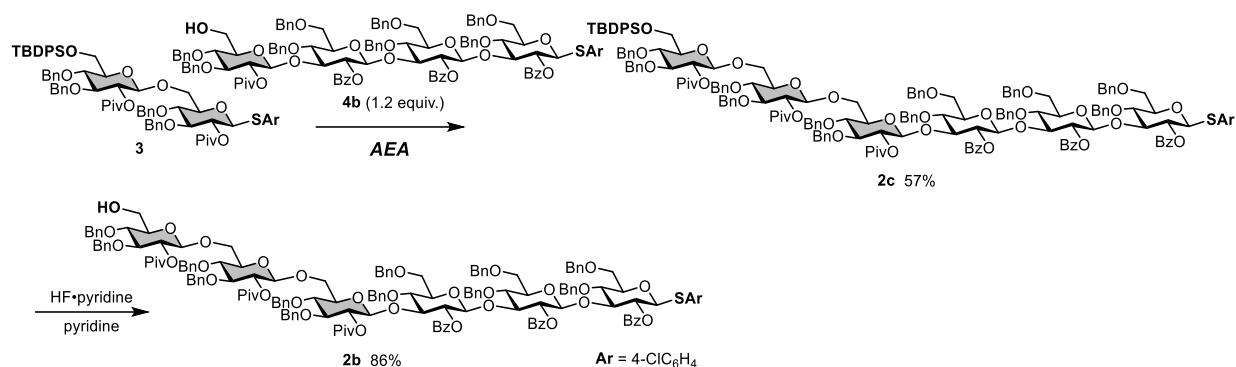
The synthesis of tetrasaccharide **4a**, with three β-(1,3)-glycosidic linkages, from disaccharide **5** was carried out using AEA with two consecutive glycosylation cycles with monosaccharide building block **6** (1.0 equiv). The process was still challenging; however, performing the reaction sequence in the presence of [P<sub>14</sub>]OTf gave a slightly better yield than with Bu<sub>4</sub>NOTf (Figure 2-5). Deprotection of the tert-butyl-diphenylsilyl (TBDPS) group of **4a** was achieved successfully in the presence of hydrogen fluoride pyridine complex (HF•pyridine) to obtain tetrasaccharide building block **4b**, equipped with a protecting-group-free 6-OH, in 83% yield.



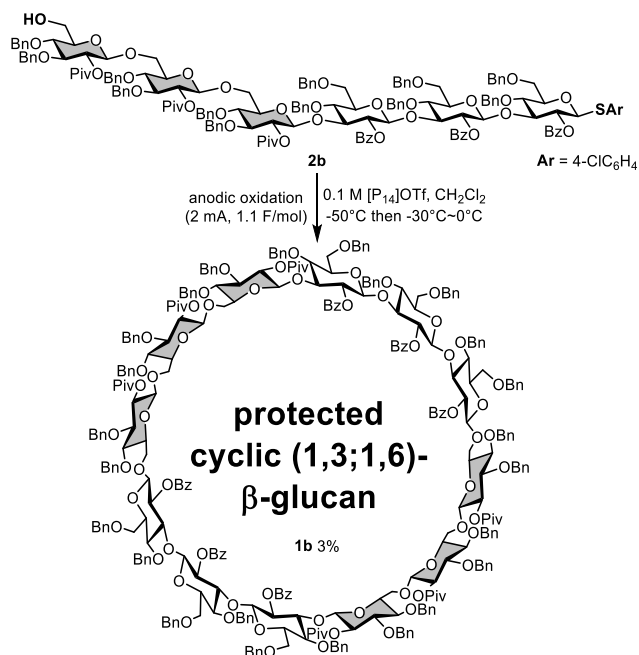
**Figure 2-5.** Synthesis of tetrasaccharide building block **4b**.

Semi-circular hexasaccharide building block **2b** was prepared using AEA and subsequent TBDPS deprotection (Figure 2-6). Disaccharide **3** and tetrasaccharide **4b** (1.2 equiv.) were assembled to prepare TBDPS-protected semi-circular hexasaccharide **2c** in the presence of [P<sub>14</sub>]OTf as an electrolyte.

Deprotection of the TBDPS group at 6-OH was carried out under the standard reaction conditions with HF•pyridine, and the desired semi-circular hexasaccharide **2b** was obtained in 86% yield. Thus-obtained **2b**, equipped with a protecting-group-free 6-OH, was used as a building block in the one-pot dimerization–cyclization process to synthesize protected cyclic dodecasaccharide **1b** (Scheme 2-1). Although the yield of **1b** was very low (3%), protected cyclic (1,3;1,6)- $\beta$ -glucan dodecasaccharide was obtained, together with by-products such as cyclic hexasaccharide and larger cyclic oligosaccharides, which were detected by MALDI-TOF-MS (see the experimental section for details).



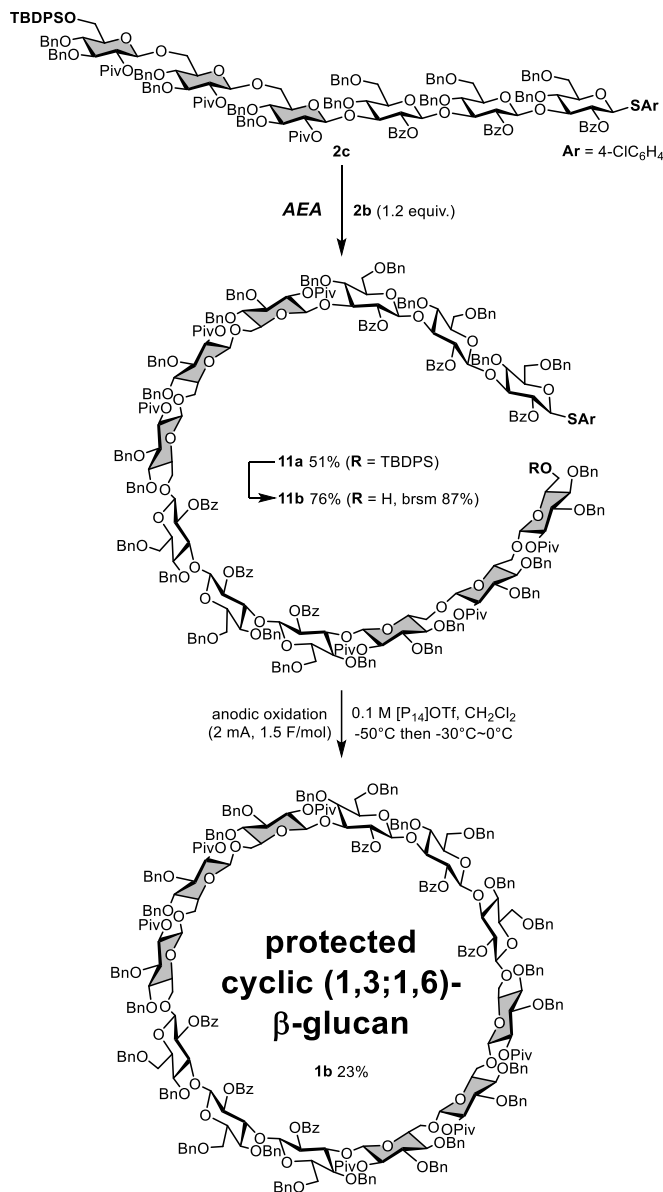
**Figure 2-6.** Synthesis of semi-circular hexasaccharide **2b**.



**Scheme 2-1.** One-pot dimerization-cyclization process for the preparation of protected cyclic dodecasaccharide **1b**.

The results of the one-pot dimerization–cyclization process encouraged us to synthesize protected cyclic dodecasaccharide **1b** using AEA (Scheme 2-2). Two semi-circular hexasaccharide building blocks, **2b** and **2c** (1.2 equiv.), were assembled using AEA to prepare linear dodecasaccharide **11a** in 51% yield. The major by-product of the reaction was hydroxy sugar of **2c**, which was detected by MALDI-TOF-MS. The TBDPS group on the 6-OH of **11a** was then deprotected to obtain **11b** as a precursor of protected cyclic dodecasaccharide **1b**. Finally, the intramolecular electrochemical glycosylation of **11b** was performed at a

low concentration (5 mM) to synthesize **1b** in 23% yield. The three-step yield of **1b** was ca. 10%, which was three times higher than that of the one-pot process shown in Scheme 2-1.



**Scheme 2-2.** Synthesis of protected cyclic dodecasaccharide **1b**.

## Conclusion

We have synthesized the protected precursor of cyclic (1,3;1,6)-β-glucan dodecasaccharide, which is the core structure of the natural oligosaccharide isolated from *Bradyrhizobium japonicum* MTCC120. We designed a semi-circular hexasaccharide and its reactivity was confirmed by the electrochemical one-pot dimerization–cyclization process. Finally, the linear precursor of cyclic dodecasaccharide was prepared using AEA of linear hexasaccharides and subsequent electrochemical intramolecular glycosylation afforded the protected cyclic dodecasaccharide in a higher yield. Further optimization of the electrochemical intramolecular glycosylation and global deprotection to obtain cyclic dodecasaccharide are in progress in our laboratory.



## Experimental

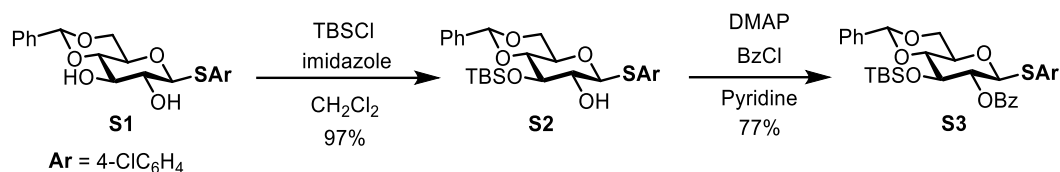
### 1. General

$^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were recorded on Bruker AVANCE II 600 ( $^1\text{H}$  600 MHz,  $^{13}\text{C}$  150 MHz). ESI-MS and MALDI-TOF-MS were recorded on Thermo Scientific Exactive spectrometer and Bruker ultrafleXtreme, respectively. Measurements of oxidation potentials of monosaccharides (conc. 4.0 mM) were carried out in 0.1 M of electrolyte in  $\text{CH}_2\text{Cl}_2$  using a glassy carbon disk working electrode, a platinum wire counter electrode, and a saturated calomel electrode (SCE) as a reference electrode with sweep rate of 10 mV/s at 2000 rpm. Preparative recycling gel permeation chromatography (PR-GPC) was performed on Japan Analytical Industry LC-5060. Kanto silica gel 60 N (spherical, neutral, 63-210  $\mu\text{m}$ ) was used for silica gel column chromatography. The automated synthesizer is consisting of the commercially available instruments such as the chiller with a cooling bath (UCR-150, Techno Sigma), the power supply for constant current electrolysis (PMC 350-0.2 A, KIKUSUI), the syringe pump (PHD 2000 infusion, Harvard apparatus), and the system controller (LabVIEW, National Instruments). Merck TLC (silica gel 60 F254) was used for TLC analysis. Starting material **S1** was prepared by the conventional method and characterized according to the reported method.<sup>13</sup> Unless otherwise noted, all materials were obtained from commercial suppliers and used without further purification.

### 2. Preparation of building blocks

#### 2-1. Preparation of 4-Chlorophenyl 2-*O*-benzoyl-4,6-di-*O*-benzyl-1-thio- $\beta$ -D-glucopyranoside (**6**)

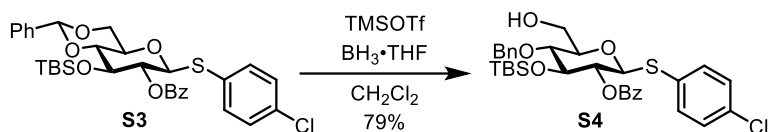
##### 2-1-1. 4-Chlorophenyl 2-*O*-benzoyl-4,6-*O*-benzylidene-3-*O*-*tert*-butyldimethylsilyl-1-thio- $\beta$ -D-glucopyranoside (**S3**)



To the solution of **S1** (25.05 mmol, 9.89 g) in  $\text{CH}_2\text{Cl}_2$  (48 mL), *tert*-butyldimethylsilyl chloride (30.1 mmol, 4.53 g) and imidazole (35.1 mmol, 2.39 g) were sequentially added at 0 °C and the reaction mixture was stirred at room temperature overnight. After the completion of the reaction determined by TLC (Hexane/EtOAc 1:1), the reaction mixture was quenched with MeOH. The mixture was washed with sat. aqueous  $\text{NaHCO}_3$  for three times and  $\text{H}_2\text{O}$  and extracted with EtOAc. The organic layer was dried over  $\text{Na}_2\text{SO}_4$ , and the solvent was removed under reduced pressure. The crude product was purified with silica gel chromatography to obtain **S2** in 97% yield (24.3 mmol, 12.4 g). To the solution of **S2** (8.02 mmol, 3.89 g) and DMAP (0.802 mmol, 99.2 mg) in pyridine (64 mL), benzoyl chloride (16.04 mmol, 1.85 mL) was added, and the reaction mixture was stirred at 55 °C overnight. The reaction was quenched by 1 N aqueous solution of hydrochloric acid and washed with deionized water three times. The organic layer was dried over  $\text{Na}_2\text{SO}_4$  and concentrated under the reduced pressure. Thus-obtained crude product was purified by silica gel chromatography (eluent: Hexane/EtOAc 7:3) to afford **S3** (6.15 mmol, 3.62 g) in 77% yield. TLC (Hexane/EtOAc 5:1)  $R_f = 0.70$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  8.06 (dd,  $J = 8.4, 1.2$  Hz, 2 H), 7.58 (td,  $J = 7.2, 1.2$  Hz, 1 H), 7.46 (td,  $J = 7.8, 1.8$  Hz, 4 H), 7.39–7.34 (m, 5 H), 7.25 (td,  $J = 8.4, 2.4$  Hz, 2 H), 5.53 (s, 1 H), 5.21 (*pseudo-t*,  $J = 9.6$  Hz, 1 H), 4.80 (d,  $J = 10.2$  Hz, 1 H), 4.39 (dd,  $J = 10.8, 4.8$  Hz, 1 H), 4.03 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.79 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.61–3.55 (m, 2 H), 0.67 (s, 9 H), -0.07 (s, 3

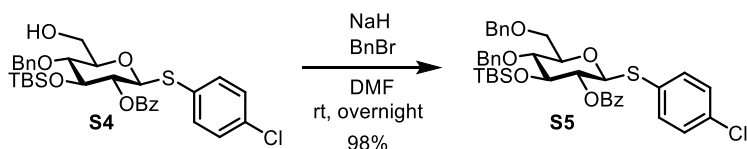
H), -0.15 (s, 3 H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  165.1, 136.9, 134.5, 133.2, 130.5, 129.9, 129.8, 129.1, 129.0, 128.4, 128.1, 126.3, 101.9, 86.7, 81.1, 74.3, 73.4, 70.8, 68.5, 25.5, 17.9, -4.2, -4.9; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{32}\text{H}_{37}\text{ClIKO}_6\text{SSi}$ ;  $[\text{M}+\text{K}]^+$  651.1400; found 651.1402.

2-1-2. 4-Chlorophenyl 2-*O*-benzoyl-4-*O*-benzyl-3-*O*-*tert*-butyldimethylsilyl-1-thio- $\beta$ -D-glucopyranoside (**S4**)



To the mixture of **S3** (2.60 mmol, 1.60 g) and MS4A (855 mg) in  $\text{CH}_2\text{Cl}_2$  (13 mL),  $\text{BH}_3\text{-THF}$  (1 M) (13 mmol, 13 mL) was added, and the reaction mixture was stirred at 0 °C for 10 min. Then TMSOTf (0.39 mmol, 0.07 mL) was added, and the reaction mixture was stirred at room temperature overnight. After the completion of the reaction determined by TLC (Hexane/EtOAc 5:1), the reaction mixture was diluted with  $\text{CH}_2\text{Cl}_2$  and quenched with sat. aqueous  $\text{NaHCO}_3$ . The mixture was washed with  $\text{H}_2\text{O}$  for three times and dried over  $\text{Na}_2\text{SO}_4$ . After filtration and removal of solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain **S4** in 79% yield (2.06 mmol, 1.27 g). TLC (Hexane/EtOAc 5:1)  $R_f$  = 0.34;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  8.06–8.03 (m, 2 H), 7.59 (*pseudo-t*,  $J$  = 6.0 Hz, 1 H), 7.47 (*pseudo-t*,  $J$  = 6.0 Hz, 2 H), 7.36–7.29 (m, 7 H), 7.24–7.23 (m, 2 H), 5.16 (*pseudo-t*,  $J$  = 9.6 Hz, 1 H), 4.85 (d,  $J$  = 11.4 Hz, 1 H), 4.74 (d,  $J$  = 10.2 Hz, 1 H), 4.63 (d,  $J$  = 12.0 Hz, 1 H), 3.95 (*pseudo-t*,  $J$  = 9.0 Hz, 1 H), 3.86 (ddd,  $J$  = 12.0, 6.0, 2.4 Hz, 1 H), 3.69–3.64 (m, 1 H), 3.55 (*pseudo-t*,  $J$  = 9.0 Hz, 1 H), 3.47 (ddd,  $J$  = 9.6, 4.8, 2.4 Hz, 1 H), 1.80 (dd,  $J$  = 7.8, 6.0 Hz, 1 H), 0.77 (s, 9 H), 0.004 (s, 3 H), -0.17 (s, 3 H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  165.4, 137.8, 134.3, 133.9, 133.3, 133.1, 130.04, 129.93, 129.1, 128.5, 127.8, 127.6, 86.3, 79.7, 78.1, 76.5, 75.1, 73.1, 62.0, 25.6, 17.8, -4.0, -4.3; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{32}\text{H}_{39}\text{ClIKO}_6\text{SSi}$ ;  $[\text{M}+\text{K}]^+$ , 653.1557; found 653.1556.

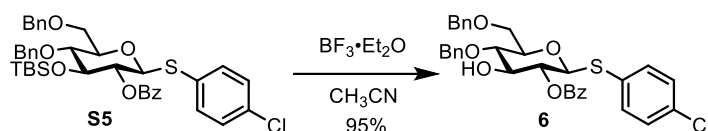
2-1-3. 4-Chlorophenyl 2-*O*-benzoyl-4,6-di-*O*-benzyl-3-*O*-*tert*-butyldimethylsilyl-1-thio- $\beta$ -D-glucopyranoside (**S5**)



To the mixture of **S4** (2.06 mmol, 1.27 g) and DMF (10 mL), benzyl bromide (7.4 mmol, 0.18 g) was added at 0 °C. NaH 60% in mineral oil (7.42 mmol, 298 mg) was dissolved in DMF (10 mL) and added to the reaction mixture in five portions (2.0 mL). After the completion of the reaction confirmed by TLC (Hexane/EtOAc 5:1), the reaction mixture was quenched with MeOH. The mixture was washed with  $\text{H}_2\text{O}$  for three times and dried over  $\text{Na}_2\text{SO}_4$ . After filtration and removal of solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain **S5** in 98% yield (2.01 mmol, 1.42 g). TLC (Hexane/EtOAc 5:1)  $R_f$  = 0.67;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  8.04 (d,  $J$  = 7.2 Hz, 2 H), 7.58 (*pseudo-t*,  $J$  = 7.2 Hz, 1 H), 7.46 (*pseudo-t*,  $J$  = 7.8 Hz, 2 H), 7.40 (d,  $J$  = 8.4 Hz, 2 H), 7.41–7.34 (m, 8 H), 7.33–7.27 (m, 2 H), 7.14 (td,  $J$  = 9.0, 2.4 Hz, 2 H), 5.16 (*pseudo-t*,  $J$  = 9.0 Hz, 1 H), 4.81 (d,  $J$  = 11.4 Hz, 1 H), 4.69 (d,  $J$  = 10.2 Hz, 1 H), 4.59 (d,  $J$  = 11.4 Hz, 1 H), 4.55 (d,  $J$  = 12.0 Hz, 1 H), 4.52 (d,  $J$  = 12.0 Hz, 1 H),

3.95–3.91 (m, 1 H), 3.75 (d,  $J = 10.2$  Hz, 1 H), 3.68 (dd,  $J = 10.8, 3.6$  Hz, 1 H), 3.58 (d,  $J = 6.0$  Hz, 2 H), 0.76 (s, 9 H), -0.02 (s, 3 H), -0.18 (s, 3 H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  165.3, 138.1, 138.0, 134.0, 133.2, 131.3, 130.1, 129.9, 128.9, 128.8, 128.41, 128.35, 127.7, 127.6, 127.5, 86.0, 79.4, 78.6, 76.8, 75.0, 73.4, 73.0, 69.0, 25.6, 17.7, -4.0, -4.3; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{39}\text{H}_{45}\text{ClNaO}_6\text{SSi}$ ;  $[\text{M}+\text{Na}]^+$ , 727.2287; found 727.2271.

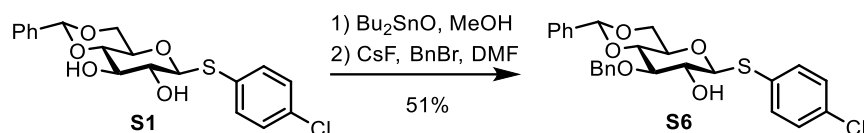
#### 2-1-4. 4-Chlorophenyl 2-*O*-benzoyl-4,6-di-*O*-benzyl-1-thio- $\beta$ -D-glucopyranoside (**6**)



To the solution of **S5** (3.88 mmol, 2.74 g) in  $\text{CH}_3\text{CN}$  (50 mL),  $\text{BF}_3\cdot\text{Et}_2\text{O}$  (5.82 mmol, 0.736 mL) was added, and the reaction mixture was stirred at  $0^\circ\text{C}$  for 30 min. After the completion of the reaction determined by TLC (hexane/EtOAc 9:1), the reaction mixture was quenched with 1 N HCl. The mixture was washed with  $\text{H}_2\text{O}$  for three times and dried over  $\text{Na}_2\text{SO}_4$ . After filtration and removal of solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain **6** in 96% yield (3.71 mmol, 2.19 g). TLC (Hexane/EtOAc 9:1)  $R_f$  0.086;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  8.06 (d,  $J = 7.2$  Hz, 2 H), 7.56 (*pseudo-t*,  $J = 7.8$  Hz, 1 H), 7.45–7.23 (m, 14 H), 7.13 (d,  $J = 8.4$  Hz, 2 H), 5.01 (*pseudo-t*,  $J = 7.8$  Hz, 1 H), 4.78 (d,  $J = 11.4$  Hz, 1 H), 4.72 (d,  $J = 9.6$  Hz, 1 H), 4.62 (d,  $J = 11.4$  Hz, 1 H), 4.59 (d,  $J = 11.4$  Hz, 1 H), 4.44 (d,  $J = 12.0$  Hz, 1 H), 3.91–3.88 (m, 1 H), 3.80 (d,  $J = 10.8$  Hz, 1 H), 3.72 (dd,  $J = 10.8, 4.2$  Hz, 1 H), 3.59–3.56 (m, 2 H), 2.73 (s, 1 H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  166.2, 138.14, 138.05, 134.4, 133.6, 130.7, 130.1, 129.5, 129.1, 128.62, 128.56, 128.51, 128.12, 128.06, 127.82, 127.79, 85.3, 79.2, 77.9, 77.2, 75.0, 73.5, 73.2, 69.0; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{33}\text{H}_{31}\text{ClKO}_6\text{S}$ ;  $[\text{M}+\text{K}]^+$ , 629.1161; found 629.1168.

#### 2-2. Preparation of 4-Chlorophenyl 3,4-di-*O*-benzyl-6-*O*-*tert*-butyldiphenylsilyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**7a**)

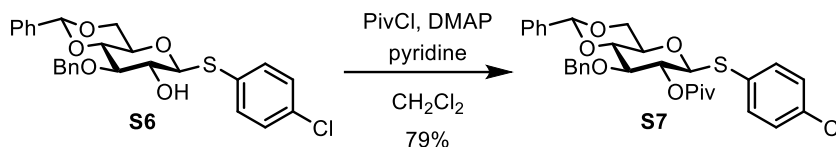
##### 2-2-1. 4-Chlorophenyl 3-*O*-benzyl-4,6-*O*-benzylidene-1-thio- $\beta$ -D-glucopyranoside (**S6**)



To the solution of **S1** (8.17 mmol, 3.23 g) in MeOH (33 mL), and dibutyltin oxide (10.2 mmol, 2.54 g) was added at room temperature and the reaction mixture was stirred at  $80^\circ\text{C}$  for 6 h. After removal of solvent under reduced pressure, DMF (63 mL), CsF (10.22 mmol, 1.55 g) and BnBr (10.2 mmol, 1.22 mL) were added, and the reaction mixture was stirred at room temperature for 16 h. After the completion of the reaction determined by TLC (Hexane/EtOAc 1:1). The reaction mixture was washed with  $\text{H}_2\text{O}$  for three times and dried over  $\text{Na}_2\text{SO}_4$ . After filtration and removal of solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain **S6** in 51% yield (4.15 mmol, 2.01 mg). TLC (Hexane/EtOAc 1:1)  $R_f = 0.89$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.48–7.46 (m, 4 H), 7.39–7.28 (m, 10 H), 5.56 (s, 1 H), 4.95 (d,  $J = 12.0$  Hz, 1 H), 4.77 (d,  $J = 12.0$  Hz, 1 H), 4.59 (d,  $J = 9.6$  Hz, 1 H), 4.36 (dd,  $J = 10.2, 4.8$  Hz, 1 H), 3.78 (*pseudo-t*,  $J = 10.8$  Hz, 1 H), 3.68 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.64 (*pseudo-t*,  $J =$

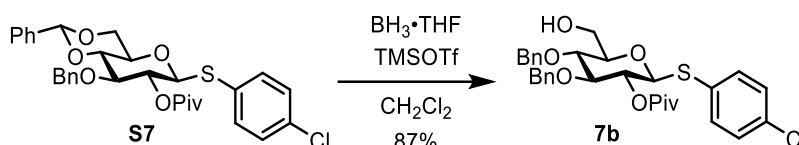
9.0 Hz, 1 H), 3.51 (dd,  $J = 9.6, 4.8$  Hz, 1 H), 3.47 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 2.56 (s, 1 H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  138.1, 137.2, 134.8, 134.7, 129.2, 129.1, 128.5, 128.3, 128.1, 128.0, 126.0, 101.3, 88.1, 81.6, 81.1, 74.8, 72.1, 70.8, 68.6, 29.7.

#### 2-2-2. 4-Chlorophenyl 3-*O*-benzyl-4,6-*O*-benzylidene-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**S7**)



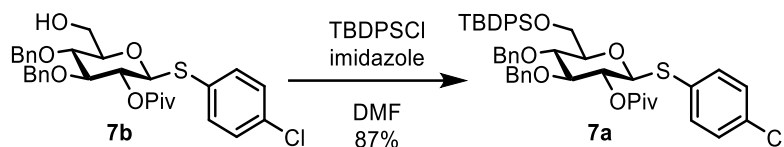
To the solution of **S6** (2.13 mmol, 1.00 g) in  $\text{CH}_2\text{Cl}_2$  (6.81 mL), DMAP (3.20 mmol, 391 mg), pyridine (10.7 mmol, 0.860 mL) and pivaloyl chloride (3.20 mmol, 0.395 mL) were added and the reaction mixture was stirred at 50 °C overnight. After the completion of the reaction determined by TLC (Hexane/EtOAc 4:1), the reaction mixture was quenched with 1 N HCl. The mixture was washed with  $\text{H}_2\text{O}$  for three times and dried over  $\text{Na}_2\text{SO}_4$ . After filtration and removal of solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain **S7** in 79% yield (1.68 mmol, 928 mg). TLC (Hexane/EtOAc 4:1)  $R_f = 0.68$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.47–7.46 (m, 2 H), 7.41 (dt,  $J = 9.0, 2.4$  Hz, 2 H), 7.39–7.36 (m, 3 H), 7.29–7.24 (m, 7 H), 5.56 (s, 1 H), 5.08–5.00 (m, 1 H), 4.86 (dd,  $J = 11.4, 3.0$  Hz, 1 H), 4.68 (d,  $J = 11.4$  Hz, 1 H), 4.64 (d,  $J = 11.4$  Hz, 1 H), 4.38 (dd,  $J = 10.8, 4.8$  Hz, 1 H), 3.82–3.77 (m, 2 H), 3.73 (*pseudo-t*,  $J = 9.6$  Hz, 1 H), 3.52 (td,  $J = 9.6, 5.4$  Hz, 1 H), 1.23 (s, 9 H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  176.5, 138.0, 137.1, 134.7, 134.5, 130.4, 129.13, 129.07, 128.3, 127.6, 127.5, 126.0, 101.3, 86.7, 81.2, 80.4, 74.6, 70.9, 70.6, 68.5, 38.8, 27.2; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{31}\text{H}_{33}\text{ClKO}_6\text{S}$  [ $\text{M}+\text{K}$ ] $^+$  607.1318; found 607.1328.

#### 2-2-3. 4-Chlorophenyl 3,4-di-*O*-benzyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**7b**)



To the solution of **S7** in  $\text{CH}_2\text{Cl}_2$  (8.12 mL),  $\text{BH}_3\text{-THF}$  (8.16 mmol, 9.06 mL) was added at 0 °C and the reaction mixture was stirred at 0 °C for 10 min. Then trimethylsilyl triflate (0.245 mmol, 0.0451 mL) was added at 0 °C and the reaction mixture was stirred at room temperature for 2.5 h. After the completion of the reaction determined by TLC (eluent: Hexane/EtOAc 4:1), the reaction mixture was quenched with sat. aqueous  $\text{NaHCO}_3$ . The mixture was washed with  $\text{H}_2\text{O}$  for three times and dried over  $\text{Na}_2\text{SO}_4$ . After filtration and removal of solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain **7b** in 87% yield (1.43 mmol, 815 mg). TLC (Hexane/EtOAc 4:1)  $R_f = 0.36$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.39 (dt,  $J = 8.4, 1.8$  Hz, 2 H), 7.33–7.24 (m, 12 H), 5.04 (*pseudo-t*,  $J = 9.6$  Hz, 1 H), 4.79 (d,  $J = 11.4$  Hz, 1 H), 4.78 (d,  $J = 10.8$  Hz, 1 H), 4.70 (d,  $J = 11.4$  Hz, 1 H), 4.62 (dd,  $J = 13.2, 10.8$  Hz, 2 H), 3.88 (ddd,  $J = 12.0, 6.0, 2.4$  Hz, 1 H), 3.74 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.73–3.68 (m, 1 H), 3.63 (*pseudo-t*,  $J = 9.6$  Hz, 1 H), 3.44 (ddd,  $J = 9.6, 4.8, 3.0$  Hz, 1 H), 2.47 (s, 1 H), 1.23 (s, 9 H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  176.8, 138.11, 138.07, 138.05, 137.94, 137.89, 134.3, 133.7, 131.5, 129.3, 128.6, 128.5, 128.1, 128.0, 127.8, 127.4, 86.4, 84.5, 79.90, 79.85, 75.3, 75.2, 71.6, 61.9, 38.9, 27.3; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{31}\text{H}_{35}\text{ClKO}_6\text{S}$  [ $\text{M}+\text{K}$ ] $^+$  609.1474; found 609.1480.

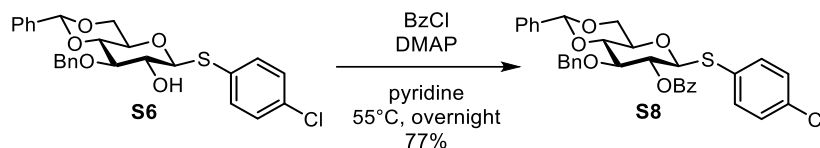
2-2-4. 4-Chlorophenyl 3,4-di-*O*-benzyl-6-*O*-*tert*-butyldiphenylsilyl-2-*O*-pivaloyl-1-thio-β-D-glucopyranoside (**7a**)



To the solution of **7b** (2.71 mmol, 1.55 g) in DMF (8.41 mL), imidazole (5.43 mmol, 370 mg) and *tert*-butylchlorodiphenylsilane (4.07 mmol, 1.05 mL) were added at 0 °C and the reaction mixture was stirred at room temperature overnight. After the completion of the reaction determined by TLC (Hexane/EtOAc 4:1), the reaction mixture was quenched with sat. aqueous NaHCO<sub>3</sub>. The mixture was washed with H<sub>2</sub>O for three times and dried over Na<sub>2</sub>SO<sub>4</sub>. After filtration and removal of solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain **7a** in 67% yield (1.83 mmol, 1.48 g). TLC (Hexane/EtOAc 4:1) R<sub>f</sub> = 0.67; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz) δ 7.75 (dd, *J* = 7.8, 1.2 Hz, 2 H), 7.69 (dd, *J* = 7.8, 1.2 Hz, 2 H), 7.45 (d, *J* = 9.0 Hz, 2 H), 7.42 (td, *J* = 7.2, 1.2 Hz, 2 H), 7.36–7.24 (m, 12 H), 7.15 (dt, *J* = 9.0, 2.4 Hz, 2 H), 7.09 (dd, *J* = 7.8, 3.6 Hz, 2 H), 5.09 (*pseudo-t*, *J* = 9.6 Hz, 1 H), 4.79 (d, *J* = 10.8 Hz, 1 H), 4.78 (d, *J* = 10.8 Hz, 1 H), 4.69 (d, *J* = 10.8 Hz, 1 H), 4.61 (dd, *J* = 7.8 Hz, 2 H), 3.98 (dd, *J* = 11.4, 1.2 Hz, 1 H), 3.92 (dd, *J* = 11.4, 4.2 Hz, 1 H), 3.80 (*pseudo-t*, *J* = 9.6 Hz, 1 H), 3.73 (*pseudo-t*, *J* = 9.0 Hz, 1 H), 3.44 (ddd, *J* = 9.6, 3.6, 1.2 Hz, 1 H), 1.24 (s, 9 H), 1.07 (s, 9 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz) δ 176.7, 138.1, 138.0, 136.0, 135.7, 134.1, 133.8, 133.4, 133.0, 131.8, 129.9, 129.1, 128.5, 128.0, 127.9, 127.8, 127.5, 86.4, 84.9, 80.4, 75.5, 75.2, 71.7, 62.8, 38.9, 27.3, 27.0, 19.4; HRMS (ESI) *m/z* calculated for C<sub>47</sub>H<sub>53</sub>ClKO<sub>6</sub>SSi [M+K]<sup>+</sup> 847.2652; found 847.2664.

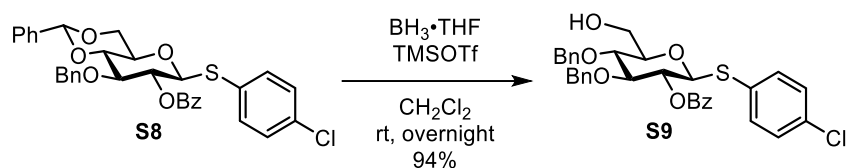
2-3. Preparation of 4-Chlorophenyl 2-*O*-benzoyl-3,4,6-*O*-tribenzyl-1-thio-β-D-glucopyranoside (**9**)

2-3-1. 4-Chlorophenyl 2-*O*-benzoyl-3-*O*-benzyl-4,6-*O*-benzylidene-1-thio-β-D-glucopyranoside (**S8**)



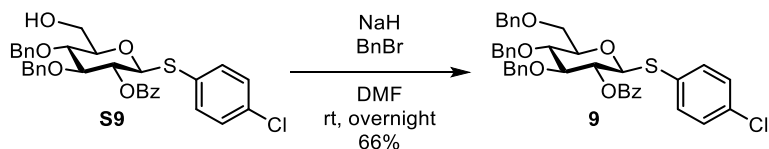
To the mixture of **S6** (8.02 mmol, 3.89 g) and DMAP (0.802 mmol, 97.98 mg) in pyridine (64 mL), benzoyl chloride (16.0 mmol, 1.86 mL) was added, and the reaction mixture was stirred at 55 °C overnight. After the completion of the reaction determined by TLC (Hexane/EtOAc 7:3), the reaction mixture was quenched with 1 N HCl. The mixture was diluted with CH<sub>2</sub>Cl<sub>2</sub>, washed with H<sub>2</sub>O for three times and dried over Na<sub>2</sub>SO<sub>4</sub>. After filtration and removal of solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain **S8** in 77% yield (6.15 mmol, 3.62 g). TLC (Hexane/EtOAc 4:1) R<sub>f</sub> = 0.67; <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.02–8.00 (m, 2 H), 7.62 (*pseudo-t*, *J* = 7.2 Hz, 1 H), 7.51–7.46 (m, 4 H), 7.42–7.38 (m, 5 H), 7.25 (s, 2 H), 7.14–7.04 (m, 5 H), 5.61 (s, 1 H), 5.24 (dd, *J* = 10.2, 9.0 Hz, 1 H), 4.80 (d, *J* = 12.0 Hz, 1 H), 4.78 (d, *J* = 9.6 Hz, 1 H), 4.66 (d, *J* = 12.0 Hz, 1 H), 4.42 (dd, *J* = 10.2, 5.4 Hz, 1 H), 3.88 (*pseudo-t*, *J* = 9.6 Hz, 1 H), 3.83 (*pseudo-t*, *J* = 10.2 Hz, 1 H), 3.79 (*pseudo-t*, *J* = 9.6 Hz, 1 H), 3.56 (td, *J* = 9.6, 4.8 Hz, 1 H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 164.9, 137.5, 137.0, 134.7, 133.3, 130.0, 129.0, 128.4, 128.2, 128.1, 128.0, 127.5, 125.9, 101.2, 86.4, 81.3, 79.1, 74.2, 71.8, 70.5, 68.4; HRMS (ESI) *m/z* calculated for C<sub>33</sub>H<sub>29</sub>ClKO<sub>6</sub>S [M+K]<sup>+</sup> 627.1005; found 627.1010.

### 2-3-2. 4-Chlorophenyl 2-*O*-benzoyl-3,4-di-*O*-benzyl-1-thio-β-D-glucopyranoside (**S8**)



To the mixture of **S8** (4.70 mmol, 2.77 g) and MS4A (1.5 g) in CH<sub>2</sub>Cl<sub>2</sub> (24 mL), BH<sub>3</sub>-THF (1.0 M, 24 mmol, 24 mL) and TMSOTf (0.703 mmol, 127 μL) were added at 0 °C and the reaction mixture was stirred at room temperature overnight. After the completion of the reaction determined by TLC (Hexane/EtOAc 7:3), the reaction mixture was diluted with CH<sub>2</sub>Cl<sub>2</sub> and quenched with sat. aqueous NaHCO<sub>3</sub>. The mixture was washed with H<sub>2</sub>O for three times and dried over Na<sub>2</sub>SO<sub>4</sub>. After filtration and removal of solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain **S9** in 81% yield (3.81 mmol, 2.25 g). TLC (Hexane/EtOAc 5:1) R<sub>f</sub> = 0.20; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz) δ 8.04–8.01 (m, 2 H), 7.62–7.59 (m, 1 H), 7.46 (*pseudo-t*, *J* = 7.8 Hz, 2 H), 7.38–7.29 (m, 8 H), 7.24 (s, 1 H), 7.14–7.09 (m, 5 H), 5.22 (*pseudo-t*, *J* = 9.0 Hz, 1 H), 4.84 (d, *J* = 10.8 Hz, 1 H), 4.76 (d, *J* = 9.6 Hz, 1 H), 4.74 (d, *J* = 11.4 Hz, 1 H), 4.66 (d, *J* = 10.8 Hz, 1 H), 4.65 (d, *J* = 10.8 Hz, 1 H), 4.62 (d, *J* = 10.8 Hz, 1 H), 3.91 (ddd, *J* = 12.0, 6.0, 3.0 Hz, 1 H), 3.86 (*pseudo-t*, *J* = 9.0 Hz, 1 H), 3.75–3.71 (m, 1 H), 3.68 (*pseudo-t*, *J* = 9.0 Hz, 1 H), 3.44 (ddd, *J* = 9.6, 4.8, 3.0 Hz, 1 H), 1.82 (dd, *J* = 7.8, 6.0 Hz, 1 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz) δ 165.20, 137.83, 137.72, 137.57, 134.48, 134.14, 133.98, 133.40, 133.25, 130.74, 129.89, 129.69, 129.24, 129.15, 128.59, 128.54, 128.33, 128.21, 128.14, 128.10, 128.07, 127.78, 86.53, 85.95, 83.97, 79.77, 79.47, 77.42, 75.86, 75.61, 75.40, 75.22, 72.43, 62.00; HRMS (ESI) *m/z* calculated for C<sub>33</sub>H<sub>31</sub>ClK<sub>6</sub>O<sub>6</sub>S [M+K]<sup>+</sup> 629.1161; found 629.1163.

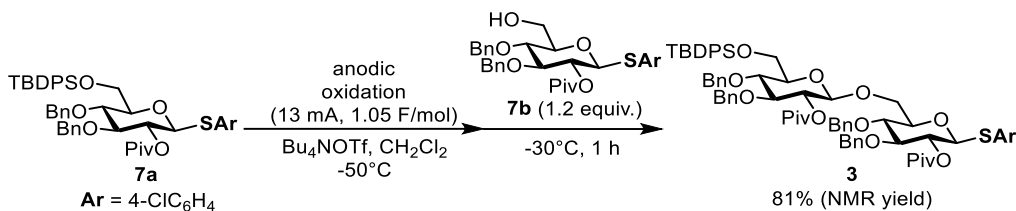
### 2-3-3. 4-Chlorophenyl 2-*O*-benzoyl-3,4,6-tri-*O*-benzyl-1-thio-β-D-glucopyranoside (**9**)



To the mixture of **S9** (3.81 mmol, 2.25 g) and DMF (30 mL), BnBr (13.7 mmol, 1.63 mL) was added at 0 °C. NaH 60% in mineral oil (13.7 mmol, 548 mg) was dissolved in DMF (10 mL) and added to the reaction mixture in five portions (2.0 mL×5). After the completion of the reaction determined by TLC (Hexane/EtOAc 4:1), the reaction mixture was quenched with MeOH, diluted with EtOAc. The mixture was washed with H<sub>2</sub>O for three times and dried over Na<sub>2</sub>SO<sub>4</sub>. After filtration and removal of solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain **9** in 60% yield (2.30 mmol, 1.57 g). TLC (Hexane/EtOAc 5:1) R<sub>f</sub> = 0.25; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz) δ 8.03 (d, *J* = 7.8 Hz, 2 H), 7.58 (*pseudo-t*, *J* = 7.8 Hz, 1 H), 7.45 (*pseudo-t*, *J* = 7.8 Hz, 2 H), 7.42 (d, *J* = 8.4 Hz, 2 H), 7.38–7.28 (m, 8 H), 7.21 (d, *J* = 7.2 Hz, 2 H), 7.15–7.07 (m, 7 H), 5.23 (*pseudo-t*, *J* = 9.6 Hz, 1 H), 4.80 (d, *J* = 10.8 Hz, 1 H), 4.72 (d, *J* = 10.2 Hz, 2 H), 4.63 (d, *J* = 10.8 Hz, 1 H), 4.60–4.53 (m, 3 H), 3.84 (d, *J* = 9.0 Hz, 1 H), 3.83–3.79 (m, 1 H), 3.75–3.69 (m, 2 H), 3.62–3.58 (m, 1 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz) δ 165.18, 138.15, 137.89, 137.64, 134.33, 134.26, 133.34, 130.85, 129.90, 129.81, 128.97, 128.62, 128.52, 128.48, 128.33, 128.07, 127.97, 127.77, 127.72, 85.67, 84.25, 79.45, 77.75, 75.43, 75.19, 73.52, 72.36, 68.95; HRMS (ESI) *m/z* calculated for C<sub>40</sub>H<sub>37</sub>ClK<sub>6</sub>O<sub>6</sub>S [M+K]<sup>+</sup> 719.1631; found 719.1630.

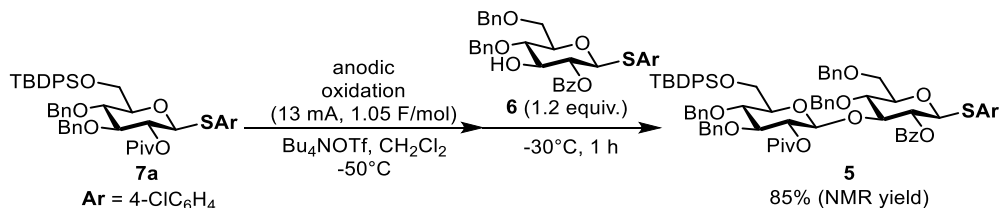
### 3. Synthesis of disaccharide building blocks

#### 3-1. Preparation of 4-Chlorophenyl 3,4-di-*O*-benzyl-6-*O*-*tert*-butyldiphenylsilyl-2-*O*-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-*O*-benzyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**3**)



The automated synthesis of **3** was carried out in an H-type divided cell (4G glass filter) equipped with a carbon felt anode (Nippon Carbon JF-20-P7) and a platinum plate cathode (10 mm×20 mm). In the anodic chamber were placed terminal building block **7a** (0.601 mmol, 486 mg), Bu<sub>4</sub>NOTf (1.50 mmol, 588 mg) and CH<sub>2</sub>Cl<sub>2</sub> (15 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.65 mmol, 57 μL), Bu<sub>4</sub>NOTf (1.50 mmol, 588 mg) and CH<sub>2</sub>Cl<sub>2</sub> (15 mL). The constant current electrolysis (13.0 mA) was carried out at -50 °C with magnetic stirring until 1.05 F/mol of electricity was consumed. After the electrolysis, building block **7b** (0.72 mmol, 418 mg) dissolved in CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL) was subsequently added by the syringe pump under an argon atmosphere at -30 °C, and kept for 60 min. After the cycle, Et<sub>3</sub>N (0.50 mL) was added, and the mixture was filtered through a short column (4×3 cm) of silica gel to remove Bu<sub>4</sub>NOTf. After removal of the solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain **3** (241 mg). NMR yield was determined using tetrachloroethane as internal standard (0.487 mmol, 81% yield). TLC (Hexane/EtOAc 4:1) R<sub>f</sub> 0.66; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz) δ 7.74–7.70 (*m*, 2 H), 7.67 (dd, *J* = 7.8, 1.2 Hz, 2 H), 7.42 (d, *J* = 8.4 Hz, 2 H), 7.38 (d, *J* = 7.2 Hz, 1 H), 7.34–7.22 (*m*, 21 H), 7.21 (d, *J* = 6.6 Hz, 2 H), 7.18 (ddd, *J* = 5.4, 2.4, 1.2 Hz, 2 H), 7.15 (dd, *J* = 6.0, 2.4 Hz, 2 H), 5.09 (dd, *J* = 9.6, 8.4 Hz, 1 H), 4.99 (*pseudo-t*, *J* = 10.2 Hz, 1 H), 4.84–4.66 (*m*, 7 H), 4.54 (*pseudo-t*, *J* = 9.6 Hz, 2 H), 4.48 (d, *J* = 7.8 Hz, 1 H), 4.01 (d, *J* = 9.6 Hz, 1 H), 3.86 (*pseudo-t*, *J* = 9.0 Hz, 1 H), 3.71 (*pseudo-t*, *J* = 9.0 Hz, 1 H), 3.67–3.61 (*m*, 3 H), 3.36–3.32 (*m*, 2 H), 1.22 (s, 9 H), 1.15 (s, 9 H), 1.04 (s, 9 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz) δ 176.8, 176.7, 138.23, 138.16, 137.95, 137.69, 135.9, 135.6, 134.3, 134.1, 133.6, 133.1, 131.4, 129.8, 129.3, 128.5, 128.1, 128.0, 127.84, 127.77, 127.73, 127.6, 127.4, 101.1, 86.3, 84.7, 83.4, 79.8, 78.0, 77.7, 76.2, 75.34, 75.27, 75.1, 75.0, 73.2, 71.5, 67.8, 62.7, 38.9, 38.8, 27.3, 19.4; HRMS (ESI) *m/z* calculated for C<sub>72</sub>H<sub>83</sub>ClKO<sub>12</sub>SSi [M+K]<sup>+</sup> 1273.4695; found 1273.4636.

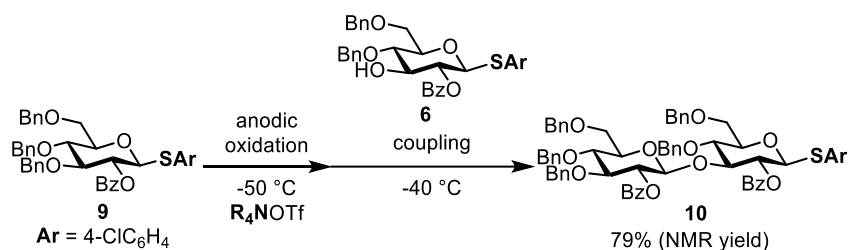
#### 3-2. Preparation of 4-Chlorophenyl 3,4-di-*O*-benzyl-6-*O*-*tert*-butyldiphenylsilyl-2-*O*-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-*O*-benzoyl-4,6-di-*O*-benzyl-1-thio- $\beta$ -D-glucopyranoside (**5**)



The automated synthesis of **5** was carried out in an H-type divided cell (4G glass filter) equipped with a carbon felt anode (Nippon Carbon JF-20-P7) and a platinum plate cathode (10 mm×20 mm). In the anodic chamber were placed terminal building block **7a** (0.603 mmol, 489 mg), Bu<sub>4</sub>NOTf (1.50 mmol, 595 mg) and CH<sub>2</sub>Cl<sub>2</sub> (15 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.67 mmol, 59

$\mu\text{L}$ ),  $\text{Bu}_4\text{NOTf}$  (1.50 mmol, 595 mg) and  $\text{CH}_2\text{Cl}_2$  (15 mL). The constant current electrolysis (13.0 mA) was carried out at  $-50\text{ }^\circ\text{C}$  with magnetic stirring until 1.05 F/mol of electricity was consumed. After the electrolysis, building block **6** (0.741 mmol, 438 mg) dissolved in  $\text{CH}_2\text{Cl}_2$  (2.0 mL) was subsequently added by the syringe pump under an argon atmosphere at  $-30\text{ }^\circ\text{C}$ , and kept for 60 min. After the cycle,  $\text{Et}_3\text{N}$  (0.50 mL) was added, and the mixture was filtered through a short column ( $4\times 3\text{ cm}$ ) of silica gel to remove  $\text{Bu}_4\text{NOTf}$ . After removal of the solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain **5** (276 mg). NMR yield was determined using tetrachloroethane as internal standard (0.513 mmol, 85% yield). TLC (Hexane/EtOAc 4:1)  $R_f$  0.63;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  8.02 (dd,  $J = 8.4, 1.2\text{ Hz}$ , 2 H), 7.74 (dd,  $J = 5.4, 2.4\text{ Hz}$ , 2 H), 7.68 (dd,  $J = 7.8, 1.2\text{ Hz}$ , 2 H), 7.60 (*pseudo-t*,  $J = 7.2\text{ Hz}$ , 1 H), 7.45–7.20 (m, 28 H), 7.10 (dt,  $J = 9.0, 2.4\text{ Hz}$ , 2 H), 7.05 (dd,  $J = 7.2, 1.2\text{ Hz}$ , 2 H), 5.21 (*pseudo-t*,  $J = 9.0\text{ Hz}$ , 1 H), 5.05 (dd,  $J = 9.6, 7.8\text{ Hz}$ , 1 H), 4.96 (d,  $J = 12.0\text{ Hz}$ , 1 H), 4.72 (d,  $J = 10.8\text{ Hz}$ , 1 H), 4.68–4.56 (m, 6 H), 4.52 (d,  $J = 10.8\text{ Hz}$ , 1 H), 4.50 (d,  $J = 10.8\text{ Hz}$ , 1 H), 4.48 (d,  $J = 12.0\text{ Hz}$ , 1 H), 3.92 (dd,  $J = 10.8, 1.2\text{ Hz}$ , 1 H), 3.77 (dt,  $J = 11.4, 1.8\text{ Hz}$ , 2 H), 3.69 (*pseudo-t*,  $J = 9.0\text{ Hz}$ , 1 H), 3.60 (ddd,  $J = 9.6, 6.6, 1.8\text{ Hz}$ , 1 H), 3.52–3.48 (m, 2 H), 3.44 (*pseudo-t*,  $J = 9.0\text{ Hz}$ , 1 H), 3.19 (dd,  $J = 9.6, 3.6\text{ Hz}$ , 1 H), 1.21 (s, 9 H), 1.02 (s, 9 H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  177.3, 164.9, 138.3, 138.1, 137.8, 136.0, 135.7, 134.1, 133.8, 133.2, 133.1, 131.8, 129.8, 129.6, 129.0, 128.9, 128.8, 128.6, 128.5, 128.3, 128.2, 127.99, 127.95, 127.8, 127.63, 127.56, 98.8, 86.3, 83.2, 79.4, 78.2, 77.8, 76.7, 75.31, 75.25, 74.8, 74.0, 73.5, 69.6, 62.6, 39.0, 27.3, 26.9, 19.3; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{74}\text{H}_{79}\text{ClKO}_{12}\text{SSi}$   $[\text{M}+\text{K}]^+$  1293.4382; found 1293.4398.

#### 4. Optimization of electrolyte

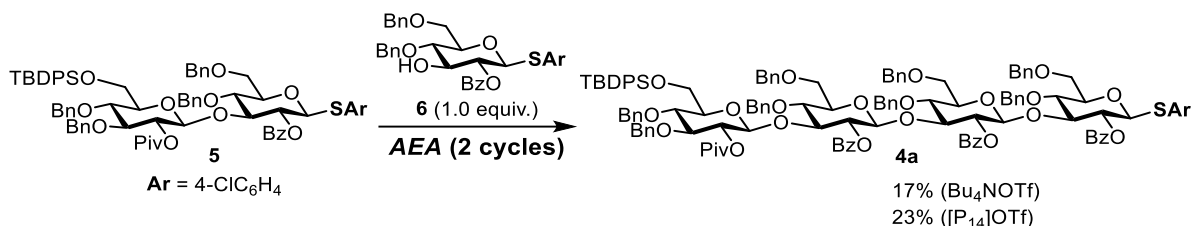


The automated synthesis of disaccharide **10** was carried out in an H-type divided cell (4G glass filter) equipped with a carbon felt anode (Nippon Carbon JF-20-P7) and a platinum plate cathode (10 mm $\times$ 20 mm). In the anodic chamber were placed terminal building block **9** (0.101 mmol, 68.9 mg),  $\text{Bu}_4\text{NOTf}$  (0.5 mmol, 196 mg) and  $\text{CH}_2\text{Cl}_2$  (5.0 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.09 mmol, 8.0  $\mu\text{L}$ ),  $\text{Bu}_4\text{NOTf}$  (0.5 mmol, 196 mg) and  $\text{CH}_2\text{Cl}_2$  (5.0 mL). The constant current electrolysis (3.0 mA) was carried out at  $-50\text{ }^\circ\text{C}$  with magnetic stirring until 1.2 F/mol of electricity was consumed. After the electrolysis, building block **6** (0.120 mmol, 71.2 mg) dissolved in  $\text{CH}_2\text{Cl}_2$  (0.60 mL) was subsequently added by the syringe pump under an argon atmosphere at  $-40\text{ }^\circ\text{C}$ , and kept for 60 min. After the cycle,  $\text{Et}_3\text{N}$  (0.20 mL) was added, and the mixture was filtered through a short column ( $4\times 3\text{ cm}$ ) of silica gel to remove  $\text{Bu}_4\text{NOTf}$ . After removal of the solvent under reduced pressure, NMR yield was determined using tetrachloroethane as internal standard (0.079 mmol, 79% yield). **4-Chlorophenyl 2-O-benzoyl-3,4,6-tri-O-benzyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-O-benzoyl-4,6-di-O-benzyl-1-thio- $\beta$ -D-glucopyranoside (10)**. TLC (Hexane/EtOAc 7:3)  $R_f = 0.53$ ;  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  7.99 (dd,  $J = 7.2, 1.2\text{ Hz}$ , 2 H), 7.80 (d,  $J = 7.2\text{ Hz}$ , 2 H), 7.66 (*pseudo-t*,  $J = 7.2\text{ Hz}$ , 1 H), 7.59 (*pseudo-t*,  $J = 7.2\text{ Hz}$ , 1 H), 7.52 (*pseudo-t*,  $J = 7.8\text{ Hz}$ , 2 H), 7.43–7.41 (m, 2 H), 7.35–7.33 (m, 2 H), 7.30–7.27 (m, 13 H), 7.22–7.21 (m, 4 H), 7.14–7.08 (m, 8 H), 7.00 (d,  $J = 6.6\text{ Hz}$ , 2 H), 5.24 (dd,  $J = 9.6, 7.8\text{ Hz}$ , 1 H), 5.16 (*pseudo-t*,  $J = 9.6\text{ Hz}$ , 1 H), 5.06 (d,  $J = 11.4\text{ Hz}$ , 1 H), 4.80 (d,  $J = 7.8\text{ Hz}$ , 1 H), 4.73 (d,  $J = 10.8\text{ Hz}$ , 1 H), 4.65 (d,  $J = 5.4\text{ Hz}$ , 1 H), 4.55 (d,  $J =$

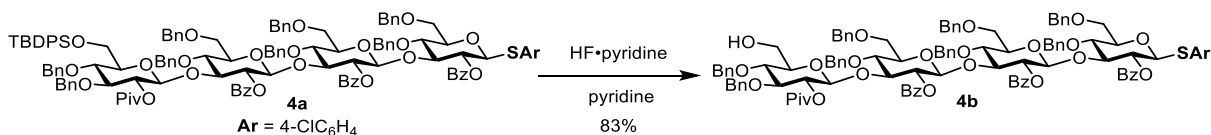


12.0 Hz, 2 H), 4.53–4.48 (m, 4 H), 4.44 (*pseudo-t*,  $J = 11.4$  Hz, 2 H), 4.28 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.82 (dd,  $J = 11.4, 1.8$  Hz, 1 H), 3.75 (dd,  $J = 10.8, 1.8$  Hz, 1 H), 3.62–3.57 (m, 5 H), 3.52 (ddd,  $J = 10.2, 6.0, 3.6$  Hz, 1 H), 3.44 (ddd,  $J = 10.8, 5.4, 1.8$  Hz, 1 H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  165.45, 164.60, 138.45, 138.42, 138.20, 137.71, 137.58, 133.73, 133.61, 133.18, 133.07, 132.16, 129.96, 129.90, 129.77, 129.56, 129.36, 128.93, 128.71, 128.60, 128.53, 128.45, 128.42, 128.41, 128.40, 128.27, 128.24, 128.16, 128.01, 128.00, 127.91, 127.84, 127.76, 127.69, 127.59, 127.54, 127.48, 127.31, 100.27, 86.23, 83.00, 80.41, 79.22, 78.14, 75.81, 75.69, 75.25, 75.15, 75.04, 74.00, 73.57, 73.50, 73.44, 73.01, 69.32, 69.12, 29.76; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{67}\text{H}_{63}\text{Cl}_{10}\text{O}_{12}\text{S}$   $[\text{M}+\text{K}]^+$  1165.3360; found 1165.3311.

## 5. Synthesis of tetrasaccharide building block



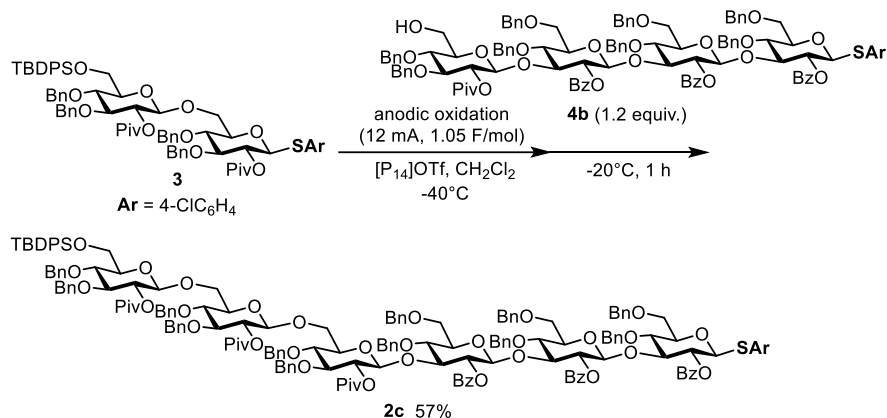
The automated synthesis of tetrasaccharide **4a** was carried out in an H-type divided cell equipped with a carbon felt anode and a platinum plate cathode (10 mm×10 mm). In the anodic chamber were placed disaccharide building block **5** (0.10 mmol, 126 mg),  $[\text{P}_{14}]\text{OTf}$  (0.50 mmol, 0.12 mL) and  $\text{CH}_2\text{Cl}_2$  (5 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.1 mmol, 9  $\mu\text{L}$ ),  $[\text{P}_{14}]\text{OTf}$  (0.50 mmol, 0.12 mL) and  $\text{CH}_2\text{Cl}_2$  (5 mL). The constant current electrolysis (3.0 mA) was carried out at  $-50$  °C with stirring until 1.2 F/mol of electricity was consumed. After the electrolysis, building block **6** (0.10 mmol, 59 mg) dissolved in  $\text{CH}_2\text{Cl}_2$  (0.6 mL) was subsequently added by the syringe pump under an argon atmosphere at  $-50$  °C and then  $-30$  °C kept for 60 min. This process was repeated two cycles. After the second cycle,  $\text{Et}_3\text{N}$  (0.2 mL) was added, and the reaction mixture was filtered through a short column (4×3 cm) of silica gel to remove electrolyte. Removal of the solvent under reduced pressure and the crude product was purified with silica gel chromatography (eluent: Hexane/EtOAc 4:1) and preparative recycling GPC (eluent:  $\text{CHCl}_3$ ). Target tetrasaccharide **4a** was obtained in 23% isolated yield (0.023 mmol, 49 mg). Thus-obtained **4a** was used as a starting material for the next step without detailed structural characterization.



Tetrasaccharide **4a** (0.46 mmol, 0.98 g) was dissolved in pyridine (3.5 mL) and the solution was cooled to  $0$  °C. 70%  $\text{HF}\cdot\text{pyridine}$  (0.35 mL) was added to the solution and the reaction mixture was stirred at  $0$  °C to room temperature for overnight. Conversion of **4a** was confirmed by TLC (Hexane/EtOAc 3:1) and aqueous sodium bicarbonate solution was added to quench the reaction. The aqueous solution was extracted with chloroform and the combined organic layer was washed with aqueous sodium bicarbonate solution and 1 N aqueous hydrochloric acid. The reaction mixture was dried over  $\text{Na}_2\text{SO}_4$  and concentrated under reduced pressure to obtain crude product (1.45 g). Thus-obtained crude product was purified by silica gel chromatography (eluent: Hexane/EtOAc 4:1) and tetrasaccharide **4b** (0.38 mmol, 723 mg) in 83% yield. **4-Chlorophenyl 3,4-di-O-benzyl-2-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-O-benzoyl-4,6-di-O-**

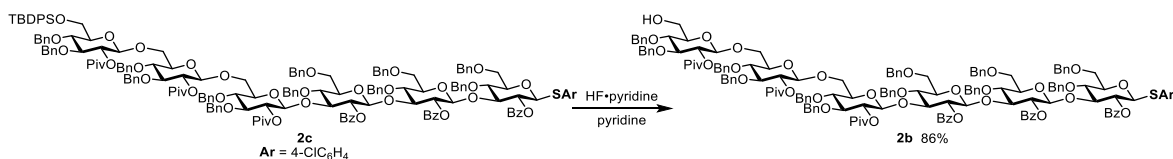
**benzyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-O-benzoyl-4,6-di-O-benzyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-O-benzoyl-4,6-di-O-benzyl-1-thio- $\beta$ -D-glucopyranoside (4b).** TLC (Hexane/EtOAc 3:1)  $R_f$  = 0.19;  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  7.87 (dd,  $J$  = 8.4, 1.8 Hz, 2 H), 7.73 (dd,  $J$  = 8.4, 1.2 Hz, 2 H), 7.67–7.62 (m, 3 H), 7.61–7.57 (m, 1 H), 7.55–7.51 (m, 1 H), 7.46–7.43 (m, 2 H), 7.38–7.35 (m, 4 H), 7.34–7.18 (m, 40 H), 7.16–7.14 (m, 2 H), 7.07–7.04 (m, 2 H), 5.05 (dd,  $J$  = 9.0, 7.8 Hz, 1 H), 4.98–4.90 (m, 4 H), 4.89 (dd,  $J$  = 9.6, 7.8 Hz, 1 H), 4.81 (d,  $J$  = 10.8 Hz, 1 H), 4.71 (d,  $J$  = 7.8 Hz, 1 H), 4.67 (d,  $J$  = 10.8 Hz, 1 H), 4.59 (d,  $J$  = 10.8 Hz, 1 H), 4.57 (d,  $J$  = 7.8 Hz, 1 H), 4.55 (d,  $J$  = 9.6 Hz, 1 H), 4.51 (d,  $J$  = 10.2 Hz, 1 H), 4.47–4.38 (m, 10 H), 4.33 (d,  $J$  = 12.0 Hz, 1 H), 4.11 (*pseudo-t*,  $J$  = 9.0 Hz, 1 H), 3.90 (*pseudo-t*,  $J$  = 7.8 Hz, 1 H), 3.88 (*pseudo-t*,  $J$  = 9.0 Hz, 1 H), 3.73–3.62 (m, 4 H), 3.55–3.36 (m, 10 H), 3.30 (ddd,  $J$  = 9.6, 4.8, 1.8 Hz, 1 H), 3.27 (*pseudo-t*,  $J$  = 9.0 Hz, 1 H), 3.00 (ddd,  $J$  = 9.6, 4.8, 1.8 Hz, 1 H), 1.03 (s, 9 H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  177.1, 164.5, 164.4, 164.3, 138.49, 138.46, 138.31, 138.14, 137.8, 137.6, 133.6, 133.4, 133.2, 133.1, 132.1, 129.88, 123.83, 129.79, 129.76, 129.53, 129.37, 129.35, 128.79, 128.46, 128.38, 128.31, 128.25, 128.13, 128.05, 127.81, 127.65, 127.62, 127.45, 127.16, 100.3, 100.2, 99.5, 86.1, 82.7, 80.7, 79.6, 79.1, 78.1, 77.6, 76.1, 76.0, 75.8, 75.4, 75.3, 75.2, 75.0, 74.9, 74.8, 74.6, 74.2, 74.0, 73.39, 73.38, 73.31, 73.0, 72.9, 69.6, 69.2, 68.9, 61.4, 38.7, 27.0; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{112}\text{H}_{113}\text{ClKO}_{24}\text{S}$   $[\text{M}+\text{K}]^+$  1947.6663; found 1947.6721.

## 6. Synthesis of semi-circular hexasaccharide



The automated synthesis of semi-circular hexasaccharide **2c** was carried out in an H-type divided cell equipped with a carbon felt anode and a platinum plate cathode (20 mm $\times$ 20 mm). In the anodic chamber were placed disaccharide building block **3** (0.75 mmol, 930 mg),  $[\text{P}_{14}]\text{OTf}$  (1.6 mmol, 0.63 g) and  $\text{CH}_2\text{Cl}_2$  (15 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.79 mmol, 70  $\mu\text{L}$ ),  $[\text{P}_{14}]\text{OTf}$  (0.50 mmol, 0.12 mL) and  $\text{CH}_2\text{Cl}_2$  (15 mL). The constant current electrolysis (12 mA) was carried out at -40  $^\circ\text{C}$  with stirring until 1.05 F/mol of electricity was consumed. After the electrolysis, tetrasaccharide building block **4b** (0.90 mmol, 1.71 g) dissolved in  $\text{CH}_2\text{Cl}_2$  (3.5 mL) was subsequently added by the syringe pump under an argon atmosphere at -40  $^\circ\text{C}$  and then -20  $^\circ\text{C}$  kept for 60 min. Then  $\text{Et}_3\text{N}$  (0.75 mL) was added, and solvent was removed under reduced pressure. The crude product was purified with silica gel chromatography (eluent: Hexane/EtOAc 5:1). Target semi-circular hexasaccharide **2c** was obtained in 57% isolated yield (0.423 mmol, 1.27 g). **4-Chlorophenyl 3,4-di-O-benzyl-6-O-tert-butylidiphenylsilyl-2-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-O-benzyl-2-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-O-benzyl-2-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-O-benzoyl-4,6-di-O-benzyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-O-benzoyl-4,6-di-O-benzyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-O-benzoyl-4,6-di-O-benzyl-1-thio- $\beta$ -D-glucopyranoside (2c);** (Hexane/EtOAc 3:1)  $R_f$  = 0.50;  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )

$\delta$  7.82 (d,  $J = 7.2$  Hz, 2 H), 7.78 (d,  $J = 7.2$  Hz, 2 H), 7.70–7.69 (m, 2 H), 7.63 (d,  $J = 7.2$  Hz, 2 H), 7.55–7.48 (m, 4 H), 7.37–7.08 (m, 75 H), 7.06 (d,  $J = 8.4$  Hz, 2 H), 5.09–5.04 (m, 3 H), 4.94–4.85 (m, 5 H), 4.83 (d,  $J = 10.8$  Hz, 1 H), 4.76–4.66 (m, 8 H), 4.64–4.51 (m, 10 H), 4.49–4.44 (m, 4 H), 4.41 (d,  $J = 12.6$  Hz, 1 H), 4.38 (d,  $J = 11.4$  Hz, 1 H), 4.34–4.29 (m, 3 H), 4.26 (d,  $J = 12.0$  Hz, 1 H), 4.20 (d,  $J = 12.4$  Hz, 1 H), 4.10 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 4.07 (*pseudo-t*,  $J = 8.4$  Hz, 1 H), 3.99 (d,  $J = 10.8$  Hz, 1 H), 3.95–3.90 (m, 3 H), 3.85–3.79 (m, 2 H), 3.75 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.71 (dd,  $J = 12.0, 5.4$  Hz, 1 H), 3.67–3.64 (m, 4 H), 3.59 (d,  $J = 10.8$  Hz, 1 H), 3.35–3.26 (m, 13 H), 3.21 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.13 (ddd,  $J = 9.6, 3.6, 1.8$  Hz, 1 H), 1.14 (s, 9 H), 1.13 (s, 9 H), 1.06 (s, 9 H), 1.00 (s, 9 H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  177.1, 176.64, 176.59, 164.51, 164.47, 164.45, 138.5, 138.4, 138.3, 138.2, 138.1, 138.0, 137.98, 137.91, 137.66, 135.8, 135.7, 135.5, 135.4, 133.54, 133.45, 133.26, 133.0, 132.9, 132.1, 129.77, 129.72, 129.64, 129.60, 129.45, 129.24, 129.16, 128.75, 128.42, 128.35, 128.27, 128.20, 128.13, 128.01, 127.91, 127.84, 127.62, 127.57, 127.51, 127.46, 127.41, 127.35, 127.30, 127.26, 126.9, 100.9, 100.3, 99.80, 99.77, 99.5, 86.0, 83.13, 83.11, 82.5, 80.3, 79.04, 78.98, 78.6, 78.2, 77.8, 77.4, 76.2, 76.1, 76.0, 75.8, 75.7, 75.14, 75.05, 74.89, 74.85, 74.79, 74.76, 74.73, 74.65, 74.55, 74.52, 74.46, 74.35, 74.03, 73.8, 73.3, 73.2, 73.1, 72.7, 72.6, 69.8, 69.2, 69.1, 67.5, 66.4, 62.5, 38.7, 38.6, 27.23, 27.18, 27.0, 26.7, 19.2; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{178}\text{H}_{191}\text{ClKO}_{36}\text{SSi}$   $[\text{M}+\text{K}]^+$  3038.1925; found 3038.2100.

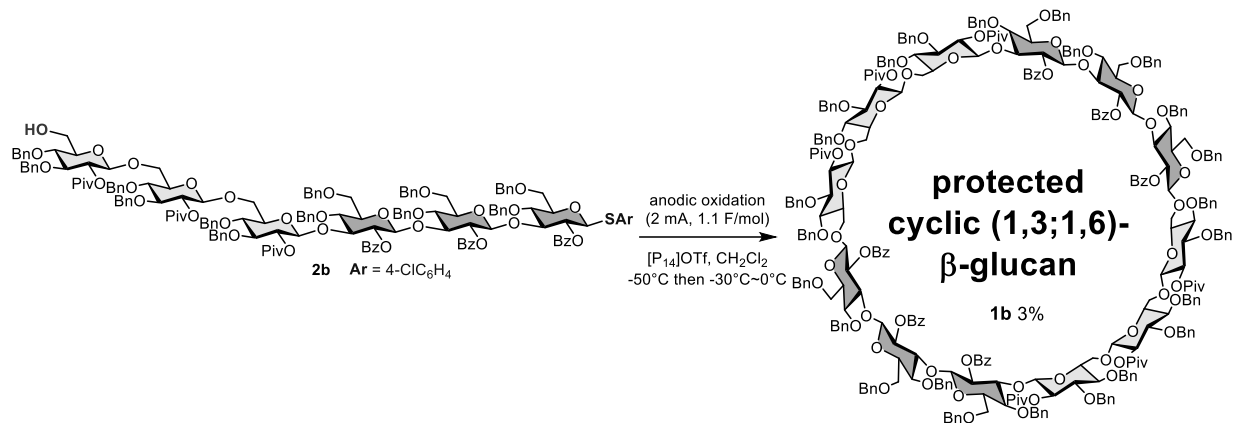


Semi-circular hexasaccharide **2c** (0.26 mmol, 790 mg) was dissolved in pyridine (2.0 mL) and the solution was cooled to 0 °C. 70% HF·pyridine (0.35 mL) was added to the solution and the reaction mixture was stirred at 0 °C to room temperature for 4 h. Conversion of **2c** was confirmed by TLC (Hexane/EtOAc 7:3) and aqueous sodium bicarbonate solution was added to quench the reaction. The aqueous solution was extracted with chloroform and the combined organic layer was washed with aqueous sodium bicarbonate solution and 1 N aqueous hydrochloric acid. The reaction mixture was dried over  $\text{Na}_2\text{SO}_4$  and concentrated under reduced pressure to obtain crude product. Thus-obtained crude product was purified by silica gel chromatography (eluent: Hexane/EtOAc 3:1) and semi-circular hexasaccharide **2b** (0.227 mmol, 628 mg) in 86% yield. **4-Chlorophenyl 3,4-di-O-benzyl-2-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-O-benzyl-2-O-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-O-benzoyl-4,6-di-O-benzyl- $\beta$ -glucopyranosyl-(1 $\rightarrow$ 3)-2-O-benzoyl-4,6-di-O-benzyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-O-benzoyl-4,6-di-O-benzyl-1-thio- $\beta$ -D-glucopyranoside (2b)**; TLC (Hexane/EtOAc 7:3)  $R_f = 0.50$ ;  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  7.83 (dd,  $J = 7.2, 4.2$  Hz, 4 H), 7.55 (*pseudo-t*,  $J = 7.2$  Hz, 1 H), 7.49 (*pseudo-t*,  $J = 7.2$  Hz, 1 H), 7.45 (d,  $J = 7.8$  Hz, 2 H), 7.38–7.14 (m, 63 H), 7.12–7.07 (m, 8 H), 5.12 (*pseudo-t*,  $J = 8.4$  Hz, 1 H), 5.01–4.89 (m, 3 H), 4.88–4.70 (m, 7 H), 4.69–4.58 (m, 6 H), 4.58–4.44 (m, 11 H), 4.42 (d,  $J = 12.0$  Hz, 1 H), 4.37 (d,  $J = 12.0$  Hz, 1 H), 4.32 (d,  $J = 12.0$  Hz, 1 H), 4.29–4.27 (m, 2 H), 4.24 (d,  $J = 12.0$  Hz, 1 H), 4.21 (d,  $J = 12.0$  Hz, 1 H), 4.12 (*pseudo-t*,  $J = 7.8$  Hz, 1 H), 4.09 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 3.93 (*pseudo-t*,  $J = 7.2$  Hz, 1 H), 3.88 (d,  $J = 10.2$  Hz, 1 H), 3.74–3.42 (m, 20 H), 3.37–3.31 (m, 5 H), 3.26 (ddd,  $J = 10.2, 4.8, 2.4$  Hz, 1 H), 3.11 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 2.15 (*pseudo-t*, 1 H), 1.18 (s, 9 H), 1.11 (s, 9 H), 1.09 (m, 9 H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  177.2, 176.6, 176.5, 164.53, 164.49, 138.52, 138.47, 138.26, 138.23, 138.20, 138.15, 138.09, 138.07, 138.02, 137.9, 137.8, 133.7, 133.6, 133.56, 133.32, 133.25, 133.0, 132.9, 132.1, 129.8, 129.7, 129.6, 129.5, 129.2, 129.1, 128.77, 128.65, 128.49, 128.36, 128.29, 128.24, 128.21, 128.19, 128.13, 128.08, 128.01, 127.95, 127.92, 127.78, 127.74, 127.64, 127.57, 127.47, 127.41, 127.38, 127.33, 127.28, 127.25, 127.22, 127.14, 127.04,

126.99, 101.4, 100.6, 99.71, 99.68, 99.63, 86.0, 83.0, 82.8, 82.4, 80.3, 79.1, 78.8, 78.7, 77.8, 77.7, 77.4, 76.2, 76.0, 75.7, 75.5, 75.1, 74.90, 74.78, 74.76, 74.62, 74.58, 74.55, 74.49, 74.37, 74.29, 73.8, 73.7, 73.3, 73.2, 73.1, 72.84, 72.79, 72.57, 70.0, 69.2, 68.5, 67.3, 61.8, 38.73, 38.69, 38.66, 27.25, 27.08, 27.00. 26.9; HRMS (ESI)  $m/z$  calculated for  $C_{162}H_{173}ClKO_{36}S$   $[M+K]^+$  2800.0753; found 2800.0688.

## 7. Synthesis of protected cyclic dodecasaccharide

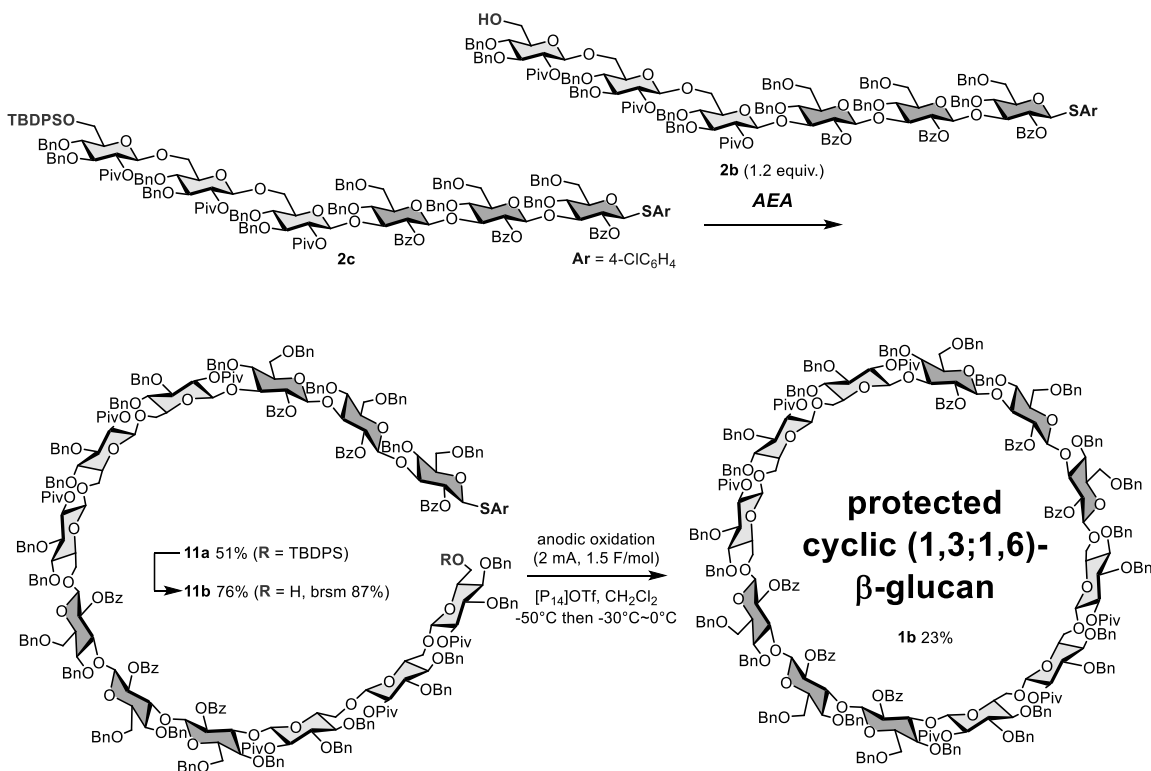
### 7-1. One-pot dimerization-cyclization process



The dimerization and cyclization of linear dodecasaccharide **2b** was carried out in an H-type divided cell equipped with a carbon felt anode and a platinum plate cathode (10 mm $\times$ 10 mm). In the anodic chamber were placed protected linear dodecasaccharide **2b** (0.135 mmol, 374 mg),  $[P_{14}]OTf$  (0.63 mmol, 0.15 mL) and  $CH_2Cl_2$  (5.0 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.15 mmol, 13  $\mu$ L),  $[P_{14}]OTf$  (0.50 mmol, 0.12 mL) and  $CH_2Cl_2$  (4.2 mL). The constant current electrolysis (2.0 mA) was carried out at -50  $^{\circ}C$  with stirring until 1.1 F/mol of electricity was consumed and then -30  $^{\circ}C$  kept for 60 min. After elevation of the reaction temperature to 0  $^{\circ}C$ ,  $Et_3N$  (0.2 mL) was added to both chambers, and the reaction mixture was dissolved in  $CHCl_3$  and washed with water to remove electrolyte  $[P_{14}]OTf$ . Thus-obtained organic layer was dried over  $Na_2SO_4$  and concentrated under reduced pressure to obtain the crude product (479 mg). Silica gel chromatography (eluent: Hexane/ $EtOAc$  4:1) and preparative recycling GPC (eluent:  $CHCl_3$ ) afforded target protected cyclic dodecasaccharide **1b** in 3% yield (2.3  $\mu$ mol, 12 mg). **Cyclobis-(1 $\rightarrow$ 6)-(3,4-di-*O*-benzyl-2-*O*-pivaloyl- $\beta$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)-(3,4-di-*O*-benzyl-2-*O*-pivaloyl- $\beta$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)-(3,4-di-*O*-benzyl-2-*O*-pivaloyl- $\beta$ -D-glucopyranosyl)-(1 $\rightarrow$ 3)-(2-*O*-benzoyl-4,6-di-*O*-benzyl- $\beta$ -D-glucopyranosyl)-(1 $\rightarrow$ 3)-(2-*O*-benzoyl-4,6-di-*O*-benzyl- $\beta$ -D-glucopyranosyl) (**1b**); TLC (Hexane/ $EtOAc$  3:1)  $R_f$  = 0.30;  $^1H$  NMR (600 MHz,  $CDCl_3$ )  $\delta$  7.79 (*pseudo-t*,  $J$  = 6.6 Hz, 4 H), 7.67–7.62 (m, 4 H), 7.59 (d,  $J$  = 7.2 Hz, 4 H), 7.46–7.43 (m, 6 H), 7.36–7.14 (m, 126 H), 7.05 (d,  $J$  = 6.0 Hz, 2 H), 5.09–5.00 (m, 12 H), 4.96–4.88 (m, 6 H), 4.74 (d,  $J$  = 7.8 Hz, 2 H), 4.70 (dd,  $J$  = 10.8, 3.0 Hz, 2 H), 4.66–4.61 (m, 6 H), 4.56–4.42 (m, 26 H), 4.38–4.25 (m, 12 H), 4.15 (d,  $J$  = 7.8 Hz, 2 H), 4.09 (*pseudo-t*,  $J$  = 9.0 Hz, 2 H), 4.02 (*pseudo-t*,  $J$  = 9.0 Hz, 2 H), 4.00–3.96 (m, 4 H), 3.79–3.74 (m, 4 H), 3.67–3.27 (m, 48 H), 3.21–3.15 (m, 4 H), 1.11 (s, 36 H), 1.08 (s, 18 H);  $^{13}C$  NMR (150 MHz,  $CDCl_3$ )  $\delta$  177.0, 176.5, 176.4, 164.7, 164.4, 163.8, 138.7, 138.6, 138.43, 138.38, 138.2, 138.1, 138.04, 137.95, 137.91, 133.30, 133.24, 133.20, 129.82, 129.75, 129.5, 129.4, 129.3, 128.8, 128.61, 128.56, 128.43, 128.35, 128.25, 128.20, 128.15, 128.10, 128.05, 128.01, 127.96, 127.7, 127.6, 127.45, 127.39, 127.35, 127.31, 127.26, 127.21, 127.13, 127.0, 126.9, 126.8, 100.7, 100.5, 100.43, 100.35, 100.14, 99.3, 83.0, 82.9, 82.8, 82.7, 79.75, 79.66, 78.3,**

78.2, 78.1, 77.6, 77.5, 76.3, 76.1, 75.6, 75.3, 75.2, 75.1, 74.85, 74.80, 74.72, 74.59, 74.54, 74.48, 74.33, 74.21, 73.84, 73.76, 73.3, 73.2, 73.1, 72.9, 72.2, 69.7, 69.15, 69.07, 67.7, 66.9, 38.65, 38.63, 38.59, 27.3, 27.1, 26.9; MS (MALDI)  $m/z$  calculated for  $C_{312}H_{336}KO_{72}$   $[M+K]^+$  5273.22; found 5273.04.

## 7-2. Stepwise process via AEA



The automated synthesis of linear dodecasaccharide **11a** was carried out in an H-type divided cell equipped with a carbon felt anode and a platinum plate cathode (20 mm $\times$ 20 mm). In the anodic chamber were placed hexasaccharide building block **2c** (0.135 mmol, 405 mg),  $[P_{14}]OTf$  (0.76 mmol, 0.175 mL) and  $CH_2Cl_2$  (3.9 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.20 mmol, 18  $\mu$ L),  $[P_{14}]OTf$  (0.50 mmol, 0.12 mL) and  $CH_2Cl_2$  (4.9 mL). The constant current electrolysis (2.0 mA) was carried out at  $-50^\circ C$  with stirring until 1.1 F/mol of electricity was consumed. After the electrolysis, hexasaccharide building block **2b** (0.162 mmol, 450 mg) dissolved in  $CH_2Cl_2$  (0.9 mL) was subsequently added by the syringe pump under an argon atmosphere at  $-50^\circ C$  and then  $-30^\circ C$  kept for 60 min. After elevation of the reaction temperature to  $0^\circ C$ ,  $Et_3N$  (0.4 mL) was added, and the reaction mixture was filtered through a short column (4 $\times$ 3 cm) of silica gel to remove electrolyte  $Bu_4NOTf$ . Removal of the solvent under reduced pressure and the crude product was purified with silica gel chromatography (eluent: Hexane/ $EtOAc$  3:1) and preparative recycling GPC (eluent:  $CHCl_3$ ). Target linear dodecasaccharide **11a** was obtained in 51% isolated yield (0.069 mmol, 389 mg). Thus-obtained **11a** was used as a starting material for the next step without detailed structural characterization.

Linear dodecasaccharide **11a** (0.069 mmol, 389 mg) was dissolved in pyridine (0.53 mL) and the solution was cooled to  $0^\circ C$ . 70%  $HF\cdot$ pyridine (0.10 mL) was added to the solution and the reaction mixture was stirred at  $0^\circ C$  to room temperature for 2 h. Conversion of **11a** was confirmed by TLC (Hexane/ $EtOAc$  7:3) and aqueous sodium bicarbonate solution was added to quench the reaction. The aqueous solution was

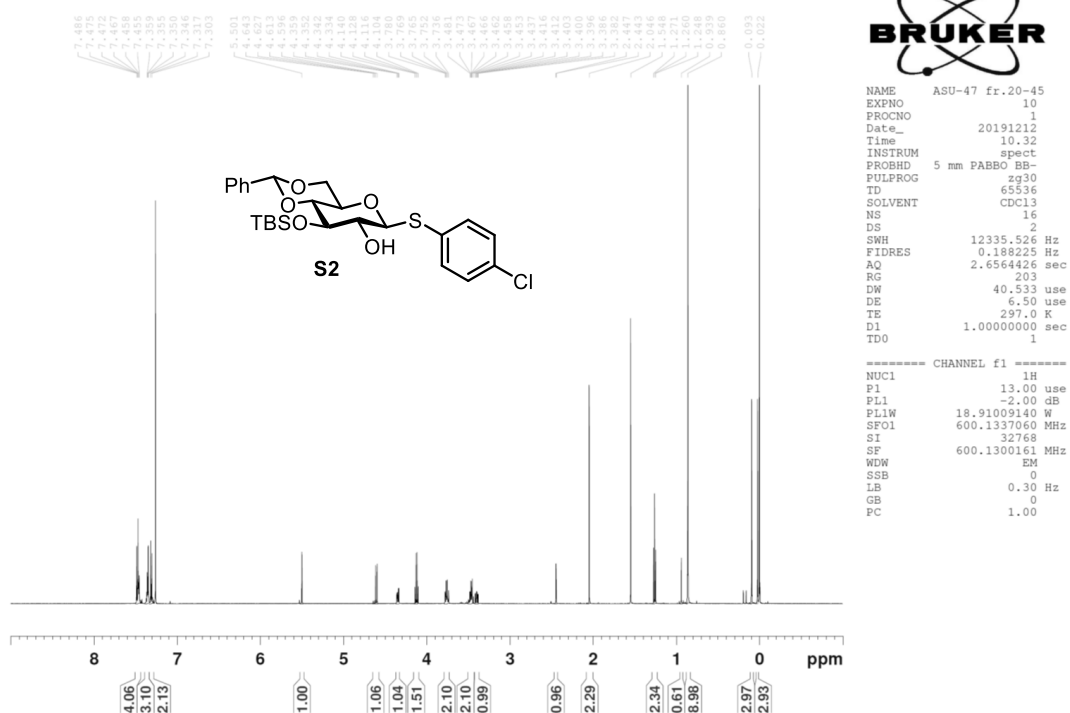


## References

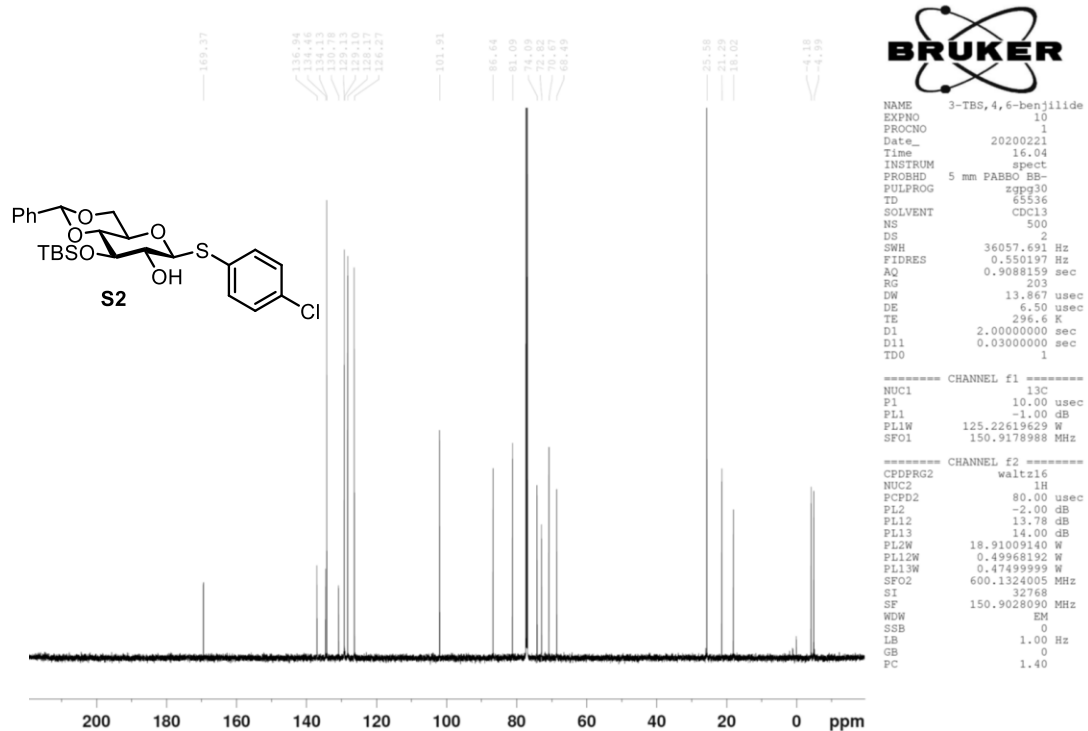
- [1] (a) J. Yoshida, K. Kataoka, R. Horcajada and A. Nagaki, *Chem. Rev.*, **2008**, *108*, 2265; (b) J. Yoshida, Y. Ashikari, K. Matsumoto and T. Nokami, *J. Synth. Org. Chem. Jpn.*, **2013**, *71*, 1136; (c) J. Yoshida, A. Shimizu, Y. Ashikari, T. Morofuji, R. Hayashi, T. Nokami and A. Nagaki, *Bull. Chem. Soc. Jpn.*, **2015**, *88*, 763; (d) M. Yan, Y. Kawamata and P. S. Baran, *Chem. Rev.*, **2017**, *117*, 13230; (e) J. Yoshida, A. Shimizu and R. Hayashi, *Chem. Rev.*, **2018**, *118*, 4702; (f) K. D. Moeller, *Chem. Rev.*, **2018**, *118*, 4817; (g) C. Zhu, N. W. J. Ang, T. H. Meyer, Y. Qiu and L. Ackermann, *ACS Cent. Sci.*, **2021**, *7*, 415.
- [2] (a) P. Hu, B. K. Peters, C. A. Malapit, J. C. Vantourout, P. Wang, J. Li, L. Mele, P.-G. Echeverria, S. D. Minter and P. S. Baran, *J. Am. Chem. Soc.*, **2020**, *142*, 20979; (b) Y. Gao, D. E. Hill, W. Hao, B. J. McNicholas, J. C. Vantourout, R. G. Hadt, S. E. Reisman, D. G. Blackmond and P. S. Baran, *J. Am. Chem. Soc.*, **2021**, *143*, 9478; (d) K. Hayashi, J. Griffin, K. C. Harper, Y. Kawamata and P. S. Baran, *J. Am. Chem. Soc.*, **2022**, *144*, 5762; (e) S. J. Harwood, M. D. Palkowitz, C. N. Gannett, P. Perez, Z. Yao, L. Sun, H. D. Abruña, S. L. Anderson and P. S. Baran, *Science*, **2022**, *375*, 745.
- [3] S. Nagahara, Y. Okada, Y. Kitano and K. Chiba, *Chem. Sci.*, **2021**, *12*, 12911.
- [4] (a) S. Manmode, K. Matsumoto, T. Itoh and T. Nokami, *Asian J. Org. Chem.*, **2018**, *7*, 1719; (b) A. Shibuya and T. Nokami, *Chem. Rec.*, **2021**, *21*, 2389; (c) K. Yano, N. Sasaki, T. Itoh and T. Nokami, *J. Synth. Org. Chem. Jpn.*, **2021**, *79*, 839.
- [5] (a) T. Nokami, R. Hayashi, Y. Saigusa, A. Shimizu, C.-Y. Liu, K.-K. Mong and J. Yoshida, *Org. Lett.*, **2013**, *15*, 4520; (b) T. Nokami, Y. Isoda, N. Sasaki, A. Takaiso, S. Hayase, T. Itoh, R. Hayashi, A. Shimizu and J. Yoshida, *Org. Lett.*, **2015**, *17*, 1525.
- [6] (a) M. Davis and M. Brewster, *Nat. Rev. Drug Discov.*, **2004**, *3*, 1023; (b) G. Crini, *Chem. Rev.*, **2014**, *114*, 10940.
- [7] M. Wakao, K. Fukase and S. Kusumoto, *J. Org. Chem.*, **2002**, *67*, 8182.
- [8] (a) M. L. Gening, D. V. Titov, A. A. Grachev, A. G. Gerbst, O. N. Yudina, A. S. Shashkov, A. O. Chizhov, Y. E. Tsvetkov and N. E. Nifantiev, *Eur. J. Org. Chem.*, **2010**, 2465; (b) D. V. Titov, M. L. Gening, A. G. Gerbst, A. O. Chizhov, Y. E. Tsvetkov and N. E. Nifantiev, *Carbohydr. Res.*, **2013**, *381*, 161; (c) M. L. Gening, Y. E. Tsvetkov, D. V. Titov, A. G. Gerbst, O. N. Yudina, A. A. Grachev, A. S. Shashkov, S. Vidal, A. Imberty, T. Saha, D. Kand, P. Talukdar, G. B. Pier and N. E. Nifantiev, *Pure Appl. Chem.*, **2013**, *85*, 1879.
- [9] S. Manmode, S. Tanabe, T. Yamamoto, N. Sasaki, T. Nokami and T. Itoh, *ChemistryOpen*, **2019**, *8*, 869.
- [10] H. Endo, M. Ochi, M. A. Rahman, T. Hamada, T. Kawano and T. Nokami, *Chem. Commun.*, **2022**, *58*, 7948.
- [11] (a) M. W. Breedveld and K. J. Miller, *Microbiol. Rev.*, **1994**, *58*, 145; (b) A. V. Nair, S. N. Gummadi and M. Doble, *Biotechnol. Lett.*, **2016**, *38*, 1519; (c) E. Cho, D. Jeong, Y. Choi and S. Jung, *J. Incl. Phenom. Macrocycl. Chem.*, **2016**, *85*, 175.
- [12] (a) S. Manmode, M. Kato, T. Ichiyanagi, T. Nokami and T. Itoh, *Asian J. Org. Chem.*, **2018**, *7*, 1802; (b) A. Shibuya, M. Kato, A. Saito, S. Manmode, N. Nishikori, T. Itoh, A. Nagaki and T. Nokami, *Eur. J. Org. Chem.*, **2022**, *19*, e202200135.
- [13] N. Basu, S. K. Maity, S. Roy, S. Singha and R. Ghosh, *Carbohydr. Res.*, **2011**, *346*, 534.

# <sup>1</sup>H and <sup>13</sup>C NMR spectra of synthetic intermediates and monosaccharide building blocks

## <sup>1</sup>H NMR

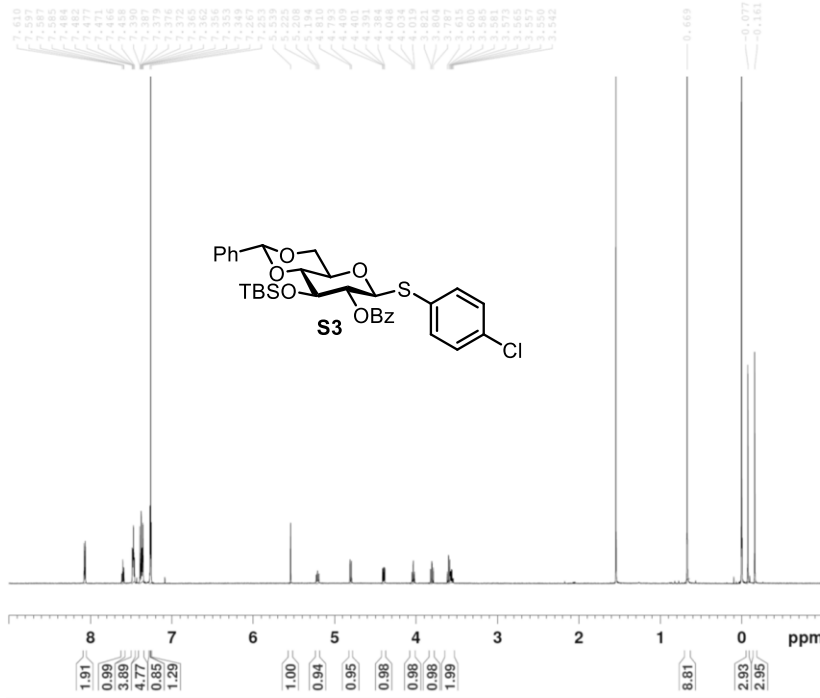


## <sup>13</sup>C NMR





<sup>1</sup>H NMR

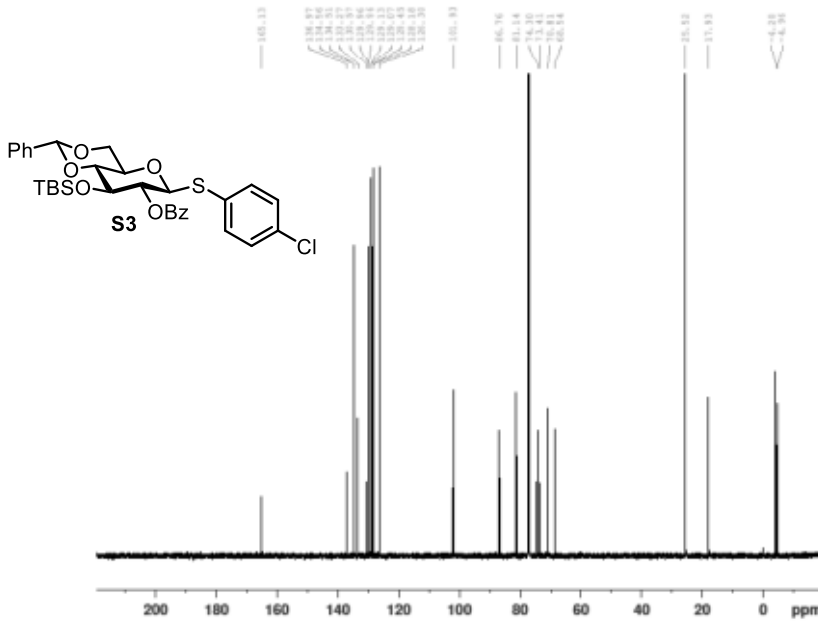


NAME	ASU-48-3
EXPNO	10
PROCNO	1
Date_	20200123
Time	17.31
INSTRUM	spect
PROBHD	5 mm PABBO BB-
PULPROG	zg30
TD	65536
SOLVENT	CDCl3
NS	16
DS	2
SWH	12335.526 Hz
FIDRES	0.188225 Hz
AQ	2.6564426 sec
RG	203
DW	40.533 use
DE	6.50 use
TE	296.2 K
D1	1.0000000 sec
TDO	1

==== CHANNEL f1 =====

NUC1	1H
P1	13.00 use
PL1	-2.00 dB
PL1W	18.91009140 W
SFO1	600.1337060 MHz
SI	32768
SF	600.1300159 MHz
WDW	EM
SSB	0
LB	0.30 Hz
GB	0
PC	1.00

<sup>13</sup>C NMR



NAME	2-Bz,3-TBS
EXPNO	10
PROCNO	1
Date_	20200221
Time	16.52
INSTRUM	spect
PROBHD	5 mm PABBO BB-
PULPROG	zgpg30
TD	65536
SOLVENT	cdcl3
NS	500
DS	2
SWH	36057.691 Kz
FIDRES	0.1550197 Hz
AQ	0.9088158 sec
RG	203
SM	13.867 use
SE	6.50 use
TE	296.6 K
D1	2.0000000 sec
D11	0.0300000 sec
TDO	1

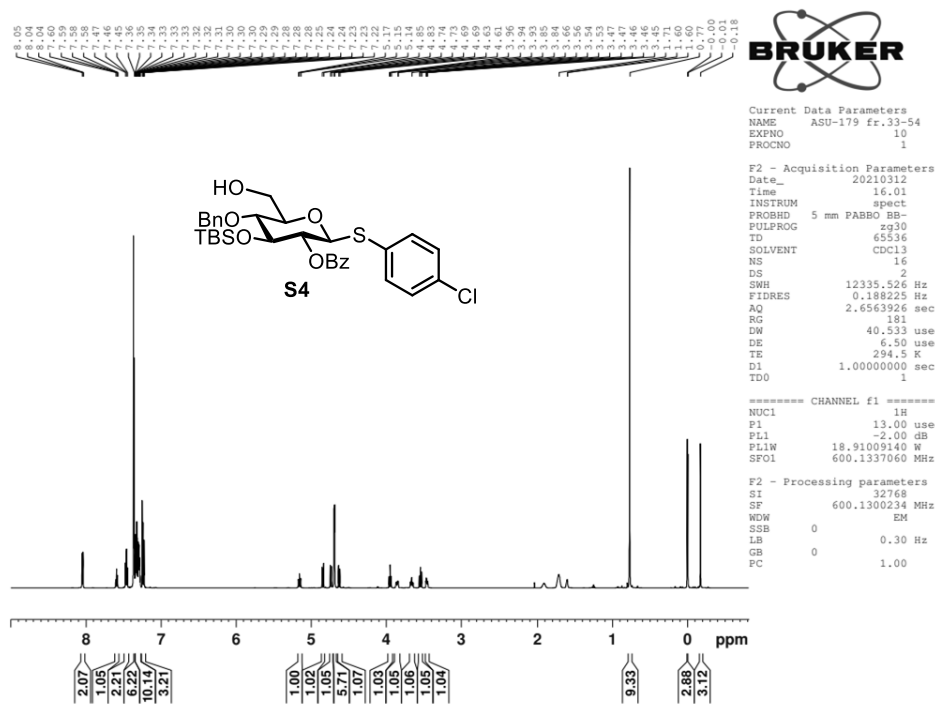
==== CHANNEL F1 =====

NUC1	13C
P1	10.00 use
PL1	-1.00 dB
PL1W	125.22419629 W
SFO1	150.9178988 MHz

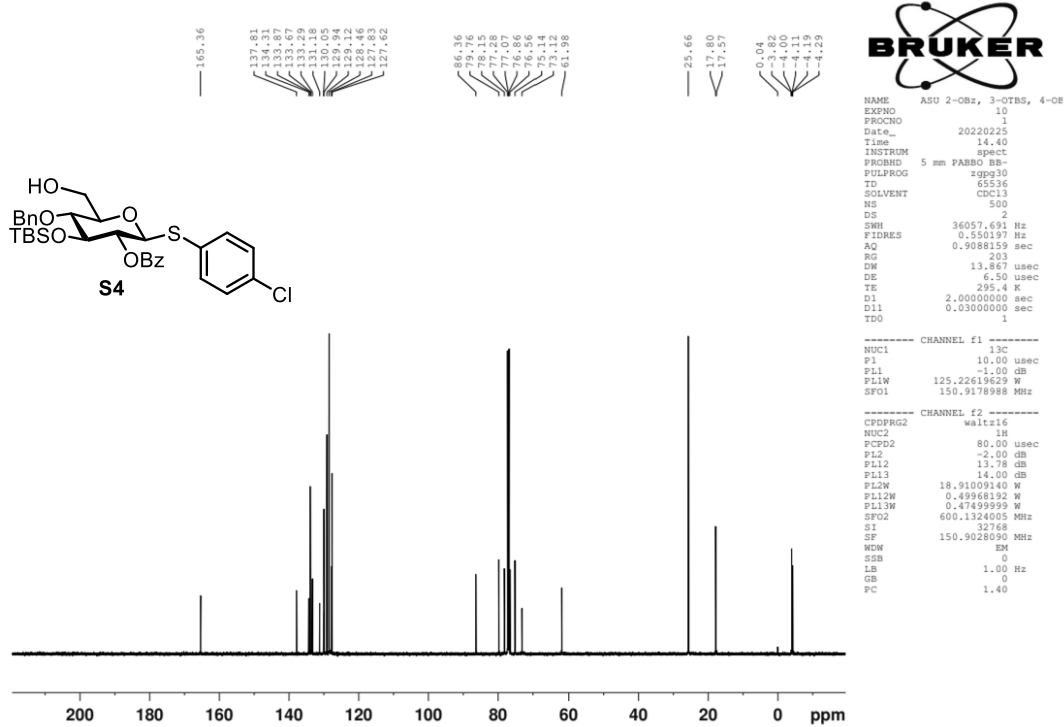
==== CHANNEL F2 =====

CPDPRG2	waltz16
NUC2	1H
PCP2	80.00 use
PC2	-2.00 dB
PL12	13.78 dB
PL13	14.00 dB
PL2W	18.91009140 W
PL3W	0.49468192 W
PL3W	0.47499489 W
SFO2	600.1324000 MHz
SI	32768
SF	150.9028990 MHz
WDW	EM
SSB	0
LB	1.00 Kz
GB	0
PC	1.40

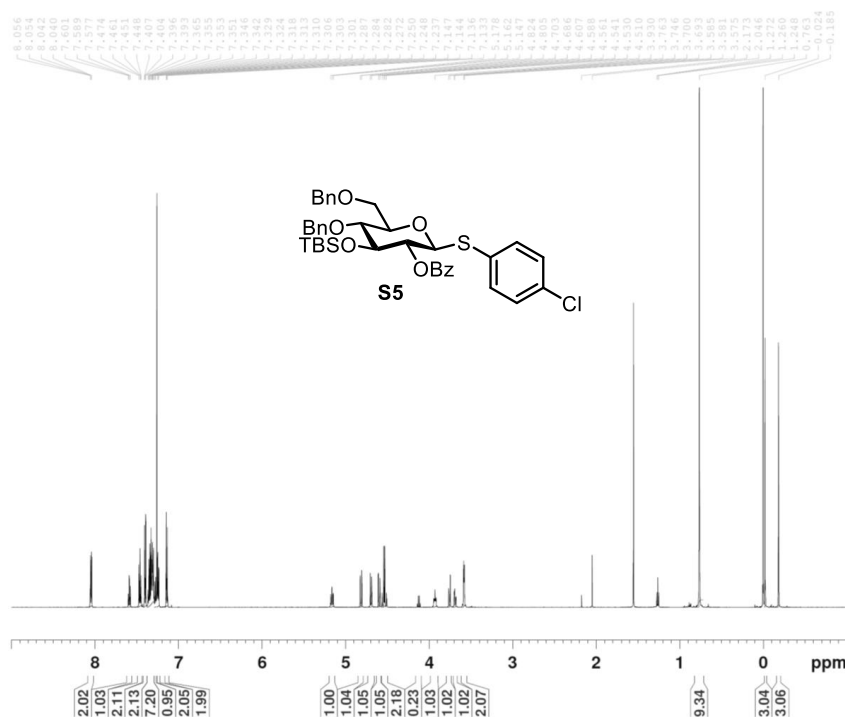
<sup>1</sup>H NMR



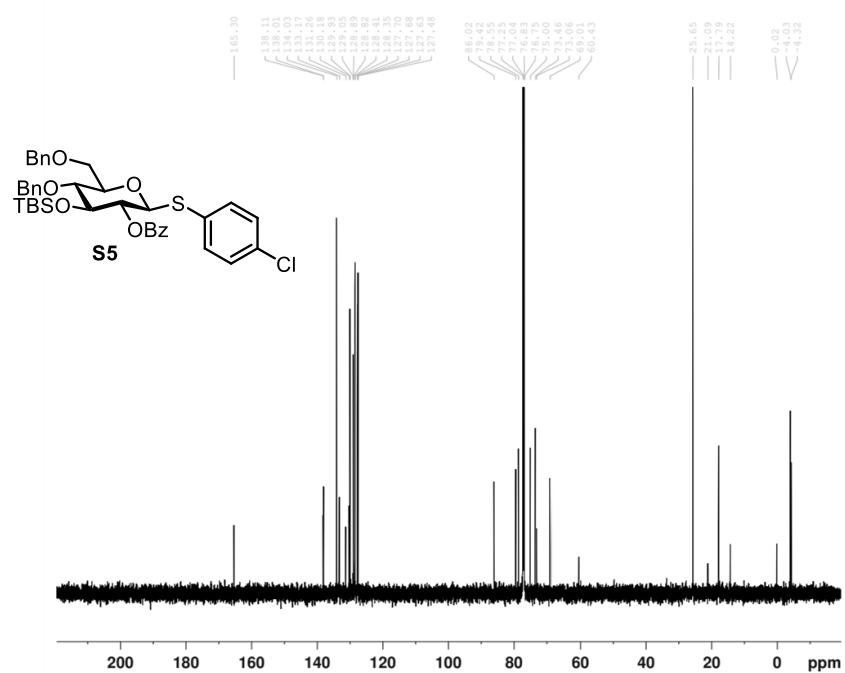
<sup>13</sup>C NMR



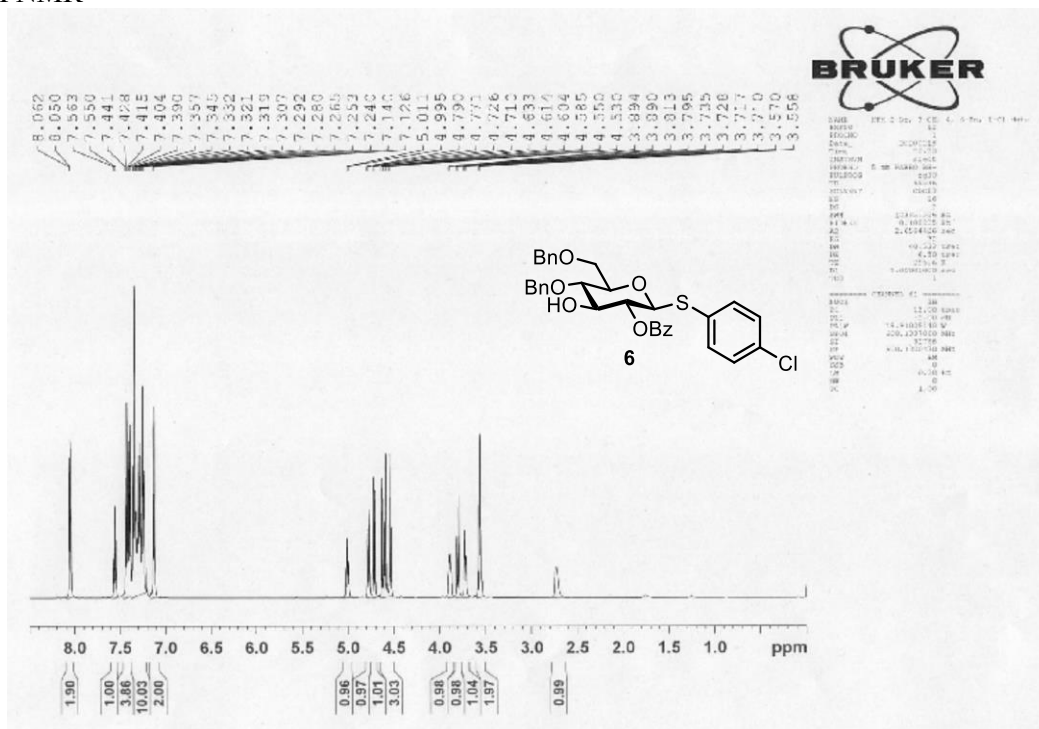
# <sup>1</sup>H NMR



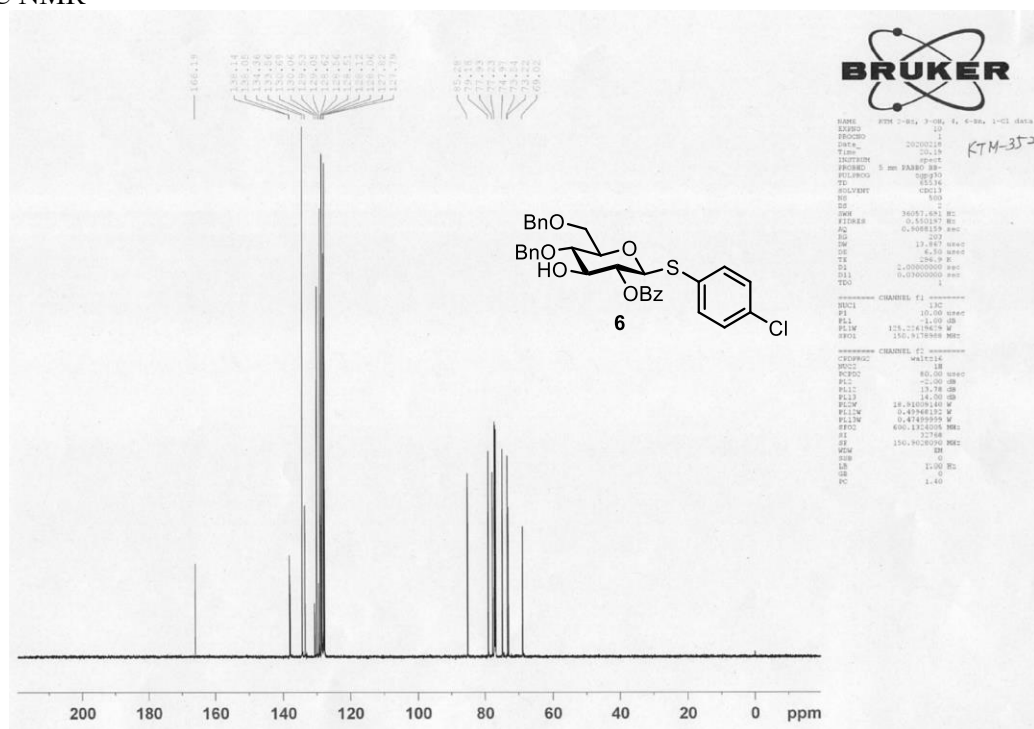
# <sup>13</sup>C NMR



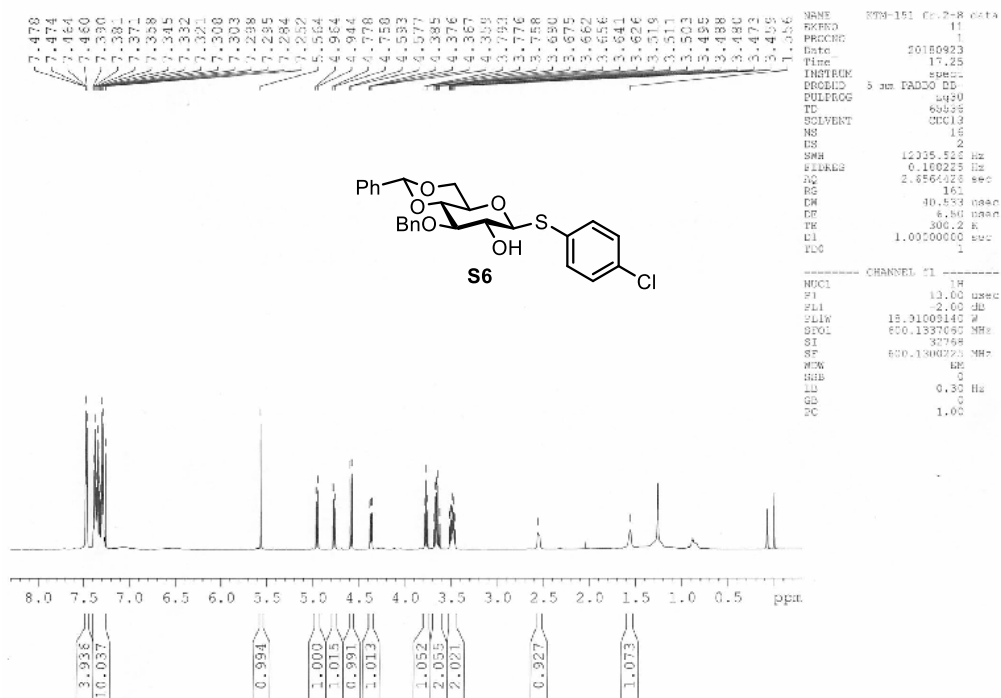
<sup>1</sup>H NMR



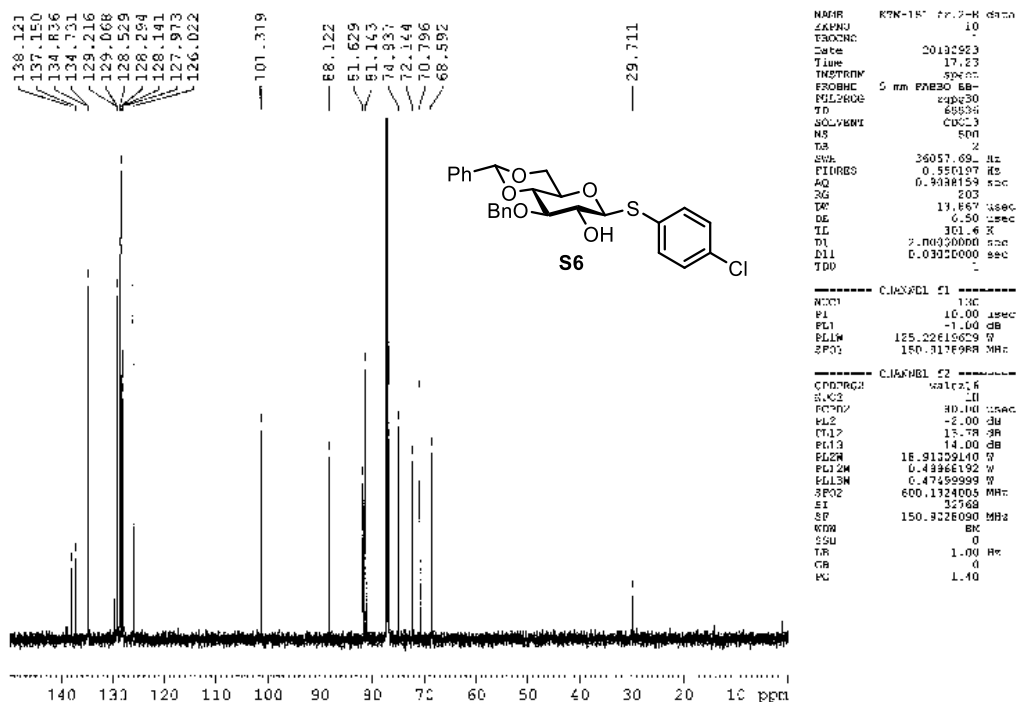
<sup>13</sup>C NMR



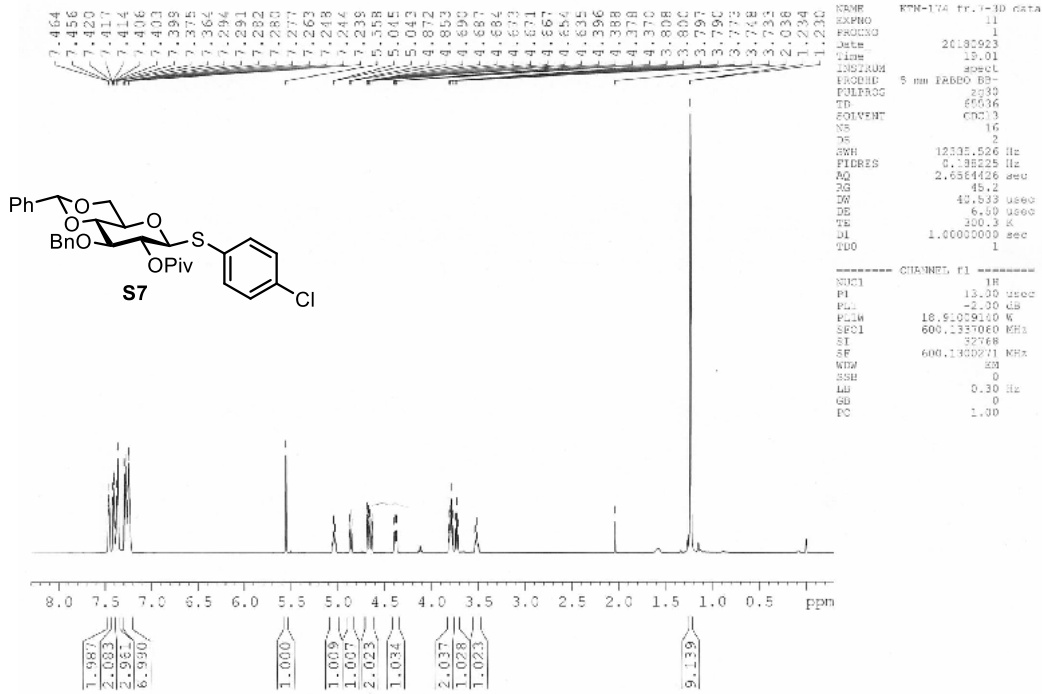
<sup>1</sup>H NMR



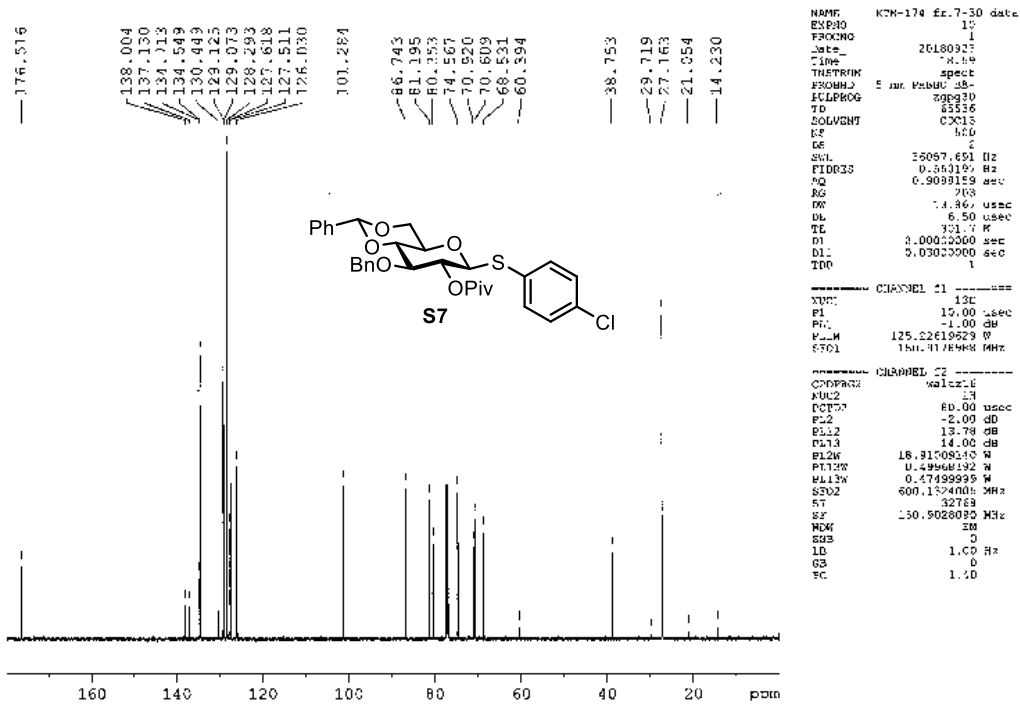
<sup>13</sup>C NMR



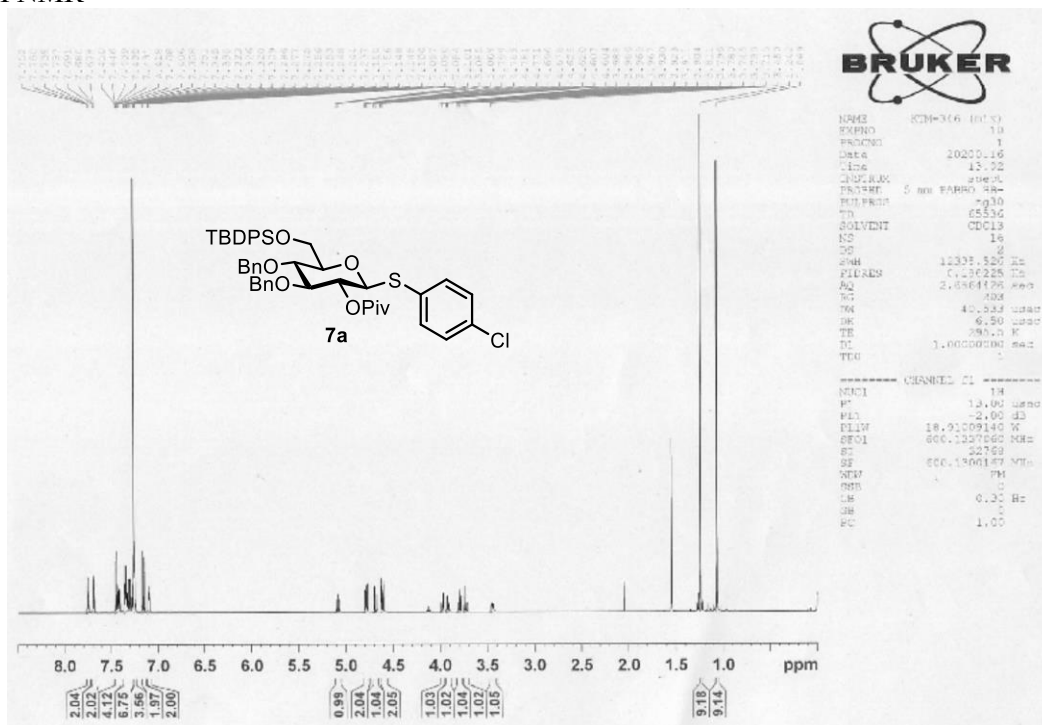
<sup>1</sup>H NMR



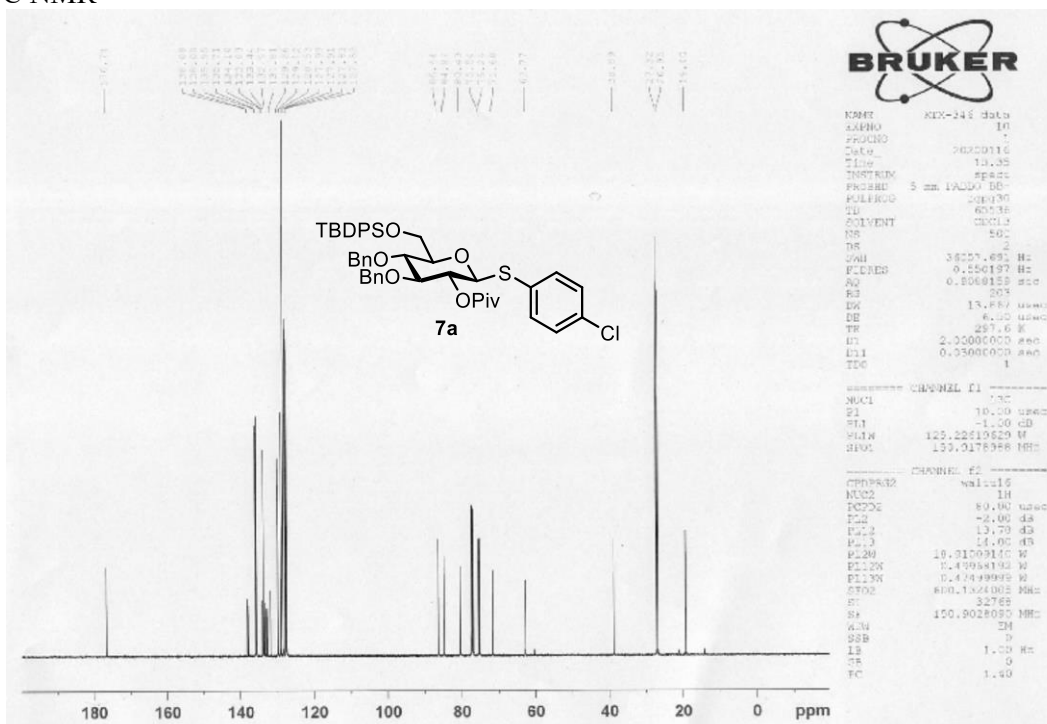
<sup>13</sup>C NMR



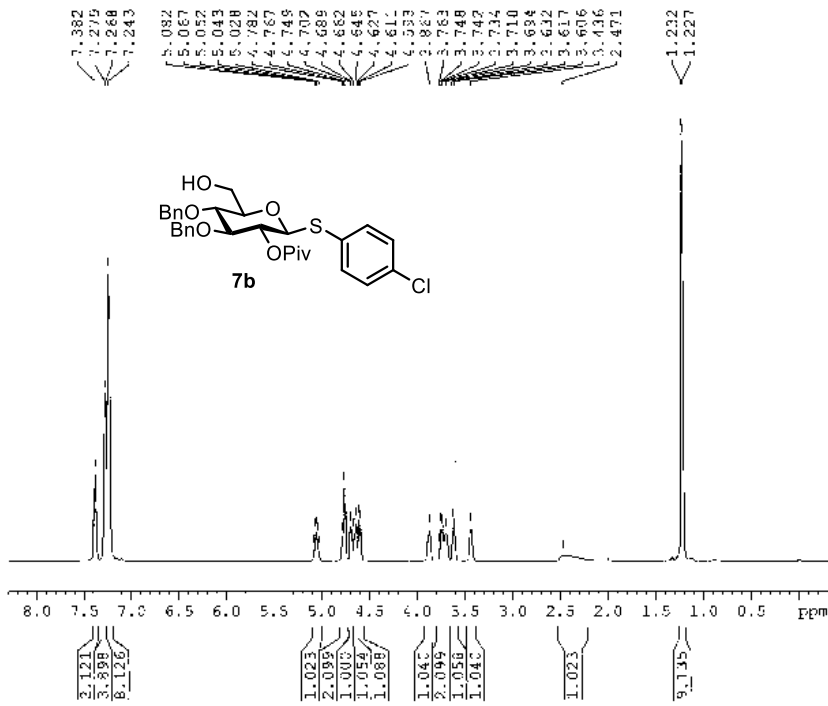
<sup>1</sup>H NMR



<sup>13</sup>C NMR



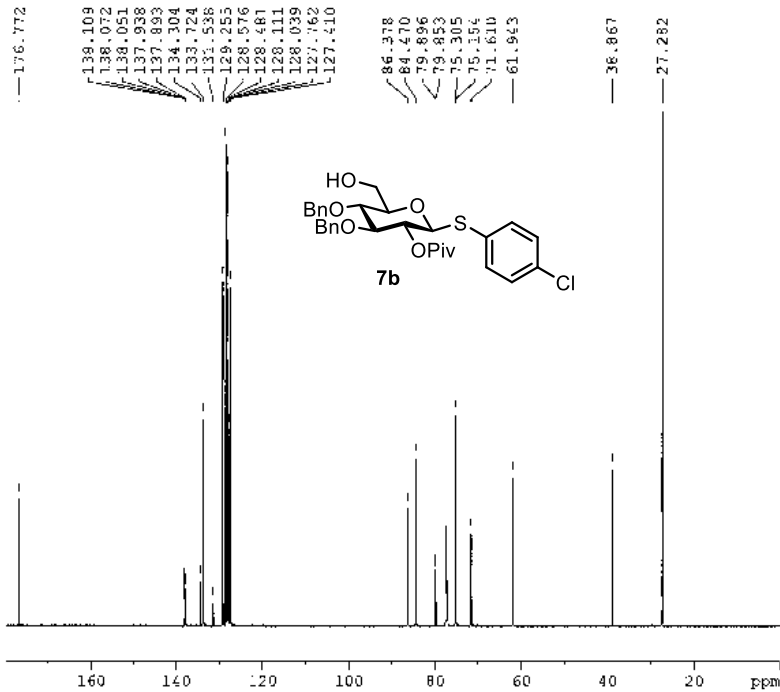
<sup>1</sup>H NMR



```

NAME      R7M-176 fr.-16-25 data
EXPNO    11
PROCNO    1
Date_     20180923
Time      14.59
INSTRUM   spect
PROBHD    5 mm F4000 DD-
PULPROG   zg30
TD         65536
SOLVENT   CDCl3
NS         18
DS         2
SWH        12315.525 Hz
FIDRES     0.180225 Hz
AQ         2.5584425 sec
RG         15
DA         10.533 usec
DE         6.50 usec
TE         299.4 K
D1         1.0020000 sec
TD0        1
----- CHANNEL f1 -----
NUC1       1H
P1         14.00 usec
PL1        -2.00 dB
PL12       19.91203140 W
SFO1       400.1337550 MHz
SC         32768
SF         600.1300800 MHz
WDW        EM
SSB        0
LB         0.30 Hz
GB         0
PC         1.00
  
```

<sup>13</sup>C NMR

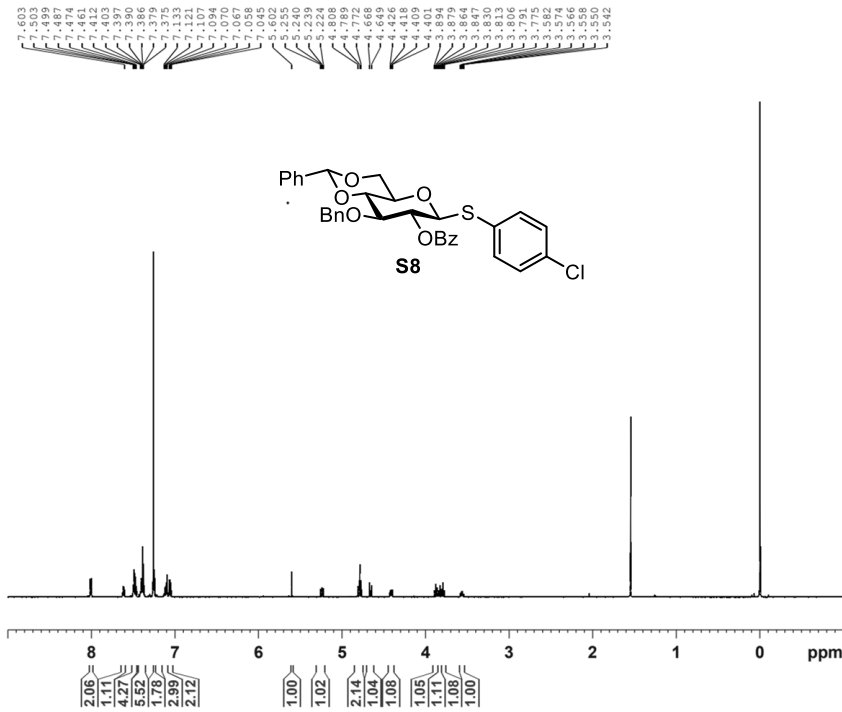


```

NAME      R7M-176 fr.-16-25 data
EXPNO    11
PROCNO    1
Date_     20180923
Time      14.59
INSTRUM   spect
PROBHD    5 mm F4000 DD-
PULPROG   zgpg30
TD         65536
SOLVENT   CDCl3
NS         18
DS         2
SWH        16057.621 Hz
FIDRES     0.560397 Hz
AQ         0.9068239 sec
RG         703
DA         13.464 usec
DE         6.50 usec
TE         302.3 K
D1         2.0000000 sec
D11        0.03500000 sec
TD0        1
----- CHANNEL f1 -----
NUC1       13C
P1         10.00 usec
PL1        -1.00 dB
PL12       129.22610429 W
SFO1       100.6189844 MHz
----- CHANNEL f2 -----
CPDPRG2   waltz16
NUC2       1H
PCPD2     80.00 usec
RG2        2.00 dB
PL2        13.74 dB
PL23       14.00 dB
PL24       18.81309840 W
PL25       0.48958232 W
PL26       0.47999995 W
SFO2       600.1324005 MHz
SF         150.9528000 MHz
WDW        EM
SSB        0
LB         1.00 Hz
GB         0
PC         1.00
  
```



# <sup>1</sup>H NMR



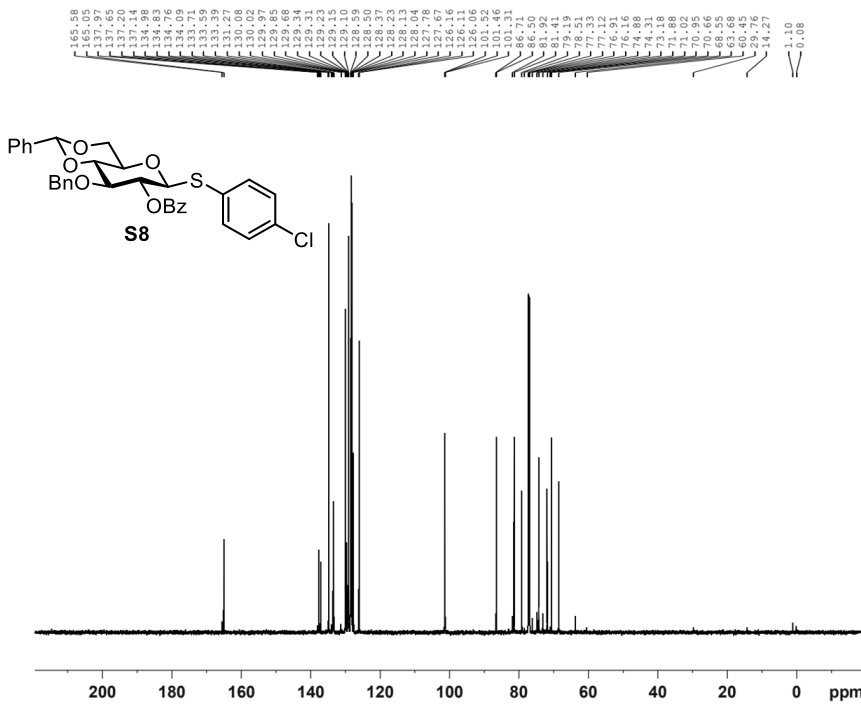
Current Data Parameters  
NAME YIS-data S8 0216  
EXPNO 10  
PROCNO 1

F2 - Acquisition Parameters  
Date\_ 20230216  
Time 9.19  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zg30  
TD 65536  
SOLVENT CDCl3  
NS 16  
DS 2  
SWH 12335.526 Hz  
FIDRES 0.188225 Hz  
AQ 2.6563926 sec  
RG 203  
DW 40.533 usec  
DE 6.50 usec  
TE 293.8 K  
D1 1.00000000 sec  
TDO 1

===== CHANNEL f1 =====  
SF01 600.1337060 MHz  
NUC1 1H  
P1 14.00 usec  
PLW1 13.50000000 W

F2 - Processing parameters  
SI 32768  
SF 600.1300145 MHz  
WDW EM  
SSB 0  
LB 0.30 Hz  
GB 0  
FC 1.00

# <sup>13</sup>C NMR



Current Data Parameters  
NAME YIS-data S8 0216  
EXPNO 20  
PROCNO 1

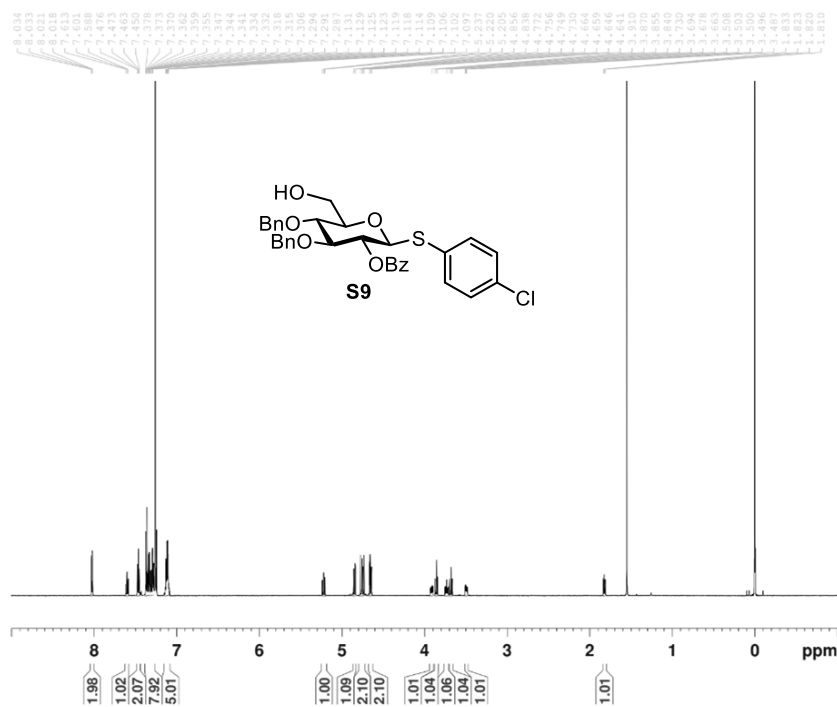
F2 - Acquisition Parameters  
Date\_ 20230216  
Time 10.08  
INSTRUM spect  
PROBHD 5 mm PABBO BB-  
PULPROG zgpg30  
TD 65536  
SOLVENT CDCl3  
NS 800  
DS 2  
SWH 36057.691 Hz  
FIDRES 0.550197 Hz  
AQ 0.9087659 sec  
RG 203  
DW 13.867 usec  
DE 6.50 usec  
TE 294.5 K  
D1 2.00000000 sec  
D11 0.03000000 sec  
TDO 1

===== CHANNEL f1 =====  
SF01 150.9178988 MHz  
NUC1 13C  
P1 12.00 usec  
PLW1 79.40000153 W

===== CHANNEL f2 =====  
SF02 600.1324005 MHz  
NUC2 1H  
CPDPRG2 waltz16  
PCPD2 70.00 usec  
PLW2 13.50000000 W  
PLW12 0.54000002 W  
PLW13 0.26460001 W

F2 - Processing parameters  
SI 32768  
SF 150.9028090 MHz  
WDW EM  
SSB 0  
LB 1.00 Hz  
GB 0  
FC 1.40

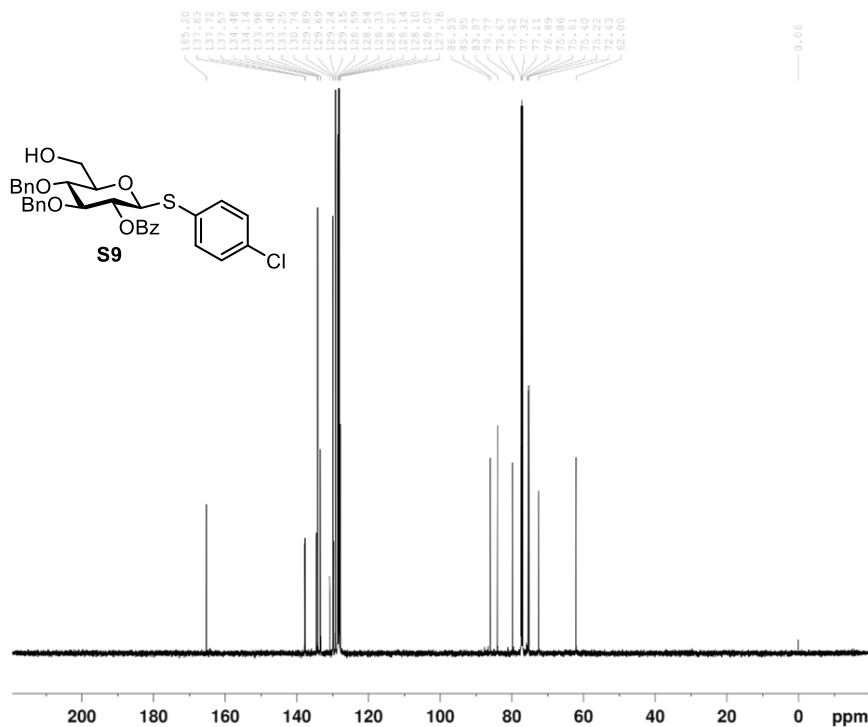
<sup>1</sup>H NMR



```

NAME VIS-model donor substrate prot
EXPNO 1
PROCNO 1
Date_ 20210222
Time 11:14
INSTRUM spect
PROBHD 5 mm PABBO HD-
PULPROG zg30
TD 65536
SOLVENT CDCl3
NS 14
DS 2
SWH 12335.500 Hz
FIDRES 0.188225 Hz
AQ 2.6564436 sec
RG 203
OR 40.533 usec
DE 6.50 usec
TE 295.4 K
D1 1.00000000 sec
TD0 1
----- CHANNEL f1 -----
NUC1 1H
P1 13.00 usec
PL1 -2.00 dB
PL1W 18.91009140 W
SFO1 600.137060 MHz
SI 32768
SF 600.1300141 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00
    
```

<sup>13</sup>C NMR



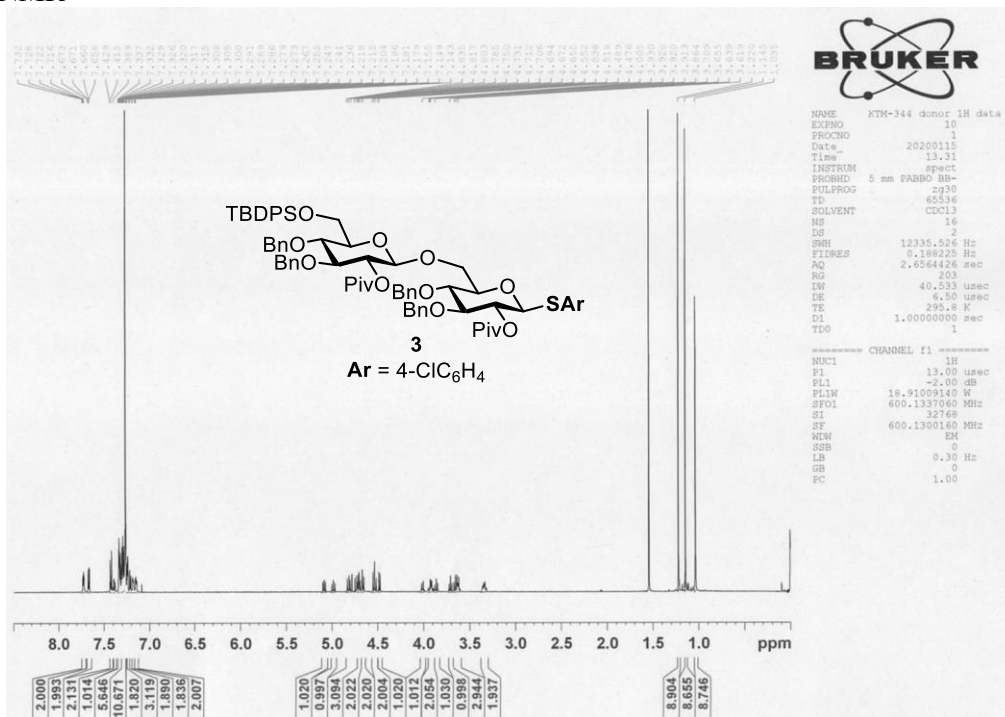
```

NAME VIS-model donor substrate carb
EXPNO 1
PROCNO 1
Date_ 20210222
Time 12:11
INSTRUM spect
PROBHD 5 mm PABBO HD-
PULPROG zgpg30
TD 65536
SOLVENT CDCl3
RG 300
DS 2
SW 36057.601 Hz
FIDRES 0.550197 Hz
AQ 0.9088109 sec
RG 203
OR 13.867 usec
DE 6.50 usec
TE 296.5 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1
----- CHANNEL f1 -----
NUC1 13C
P1 80.00 usec
PL1 -1.00 dB
PL1W 125.26209629 W
SFO1 100.917888 MHz
----- CHANNEL f2 -----
CPDPRG2 waltz16
NUC2 1H
PCPD2 80.00 usec
PL2 -2.00 dB
PL2W 13.78 dB
PL13 141.00 dB
PL2W 18.91009140 W
P11W 18.99981392 W
P11W 0.47498999 W
SFO2 400.1324005 MHz
L2 22768
SF 150.9028090 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40
    
```

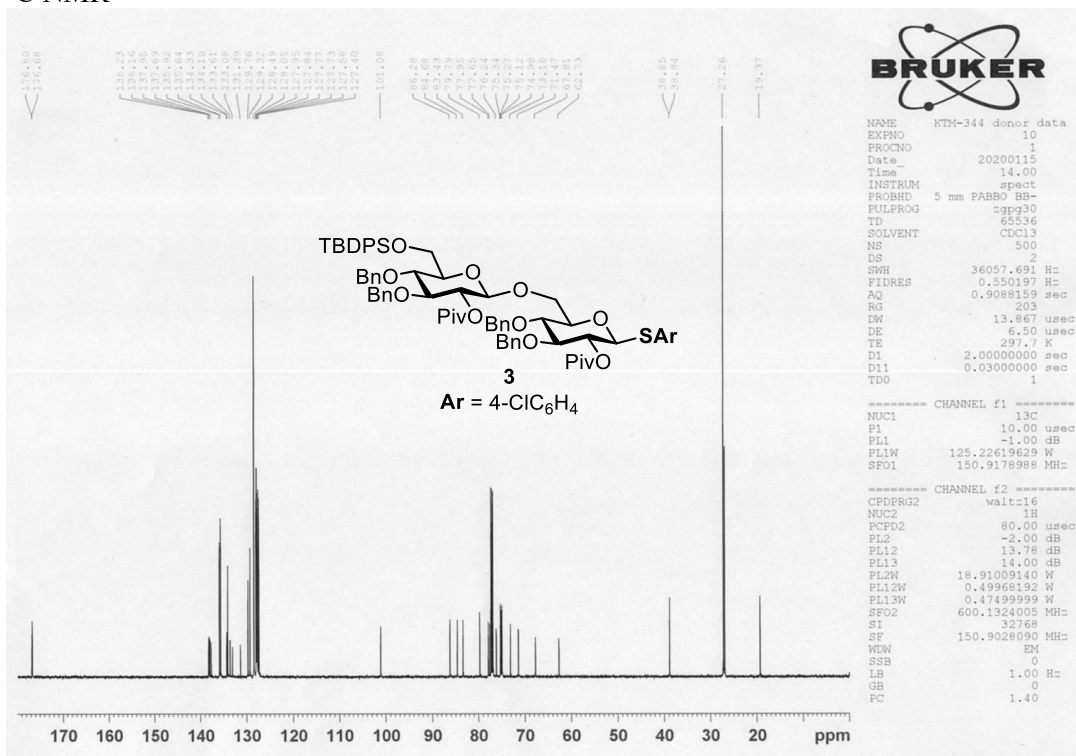


<sup>1</sup>H, <sup>13</sup>C NMR, H-H COSY and HMQC spectra of disaccharides.

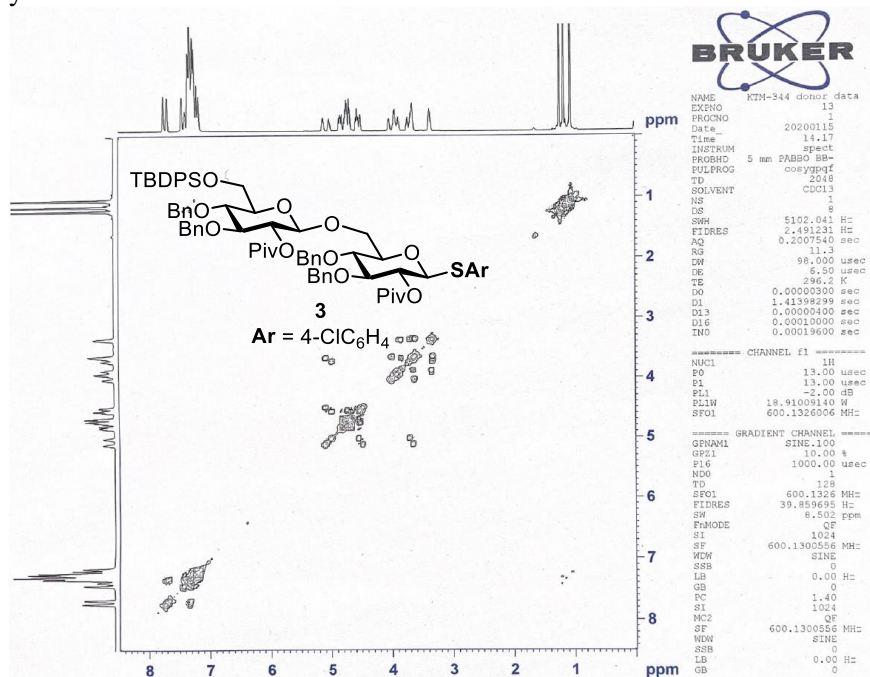
<sup>1</sup>H NMR



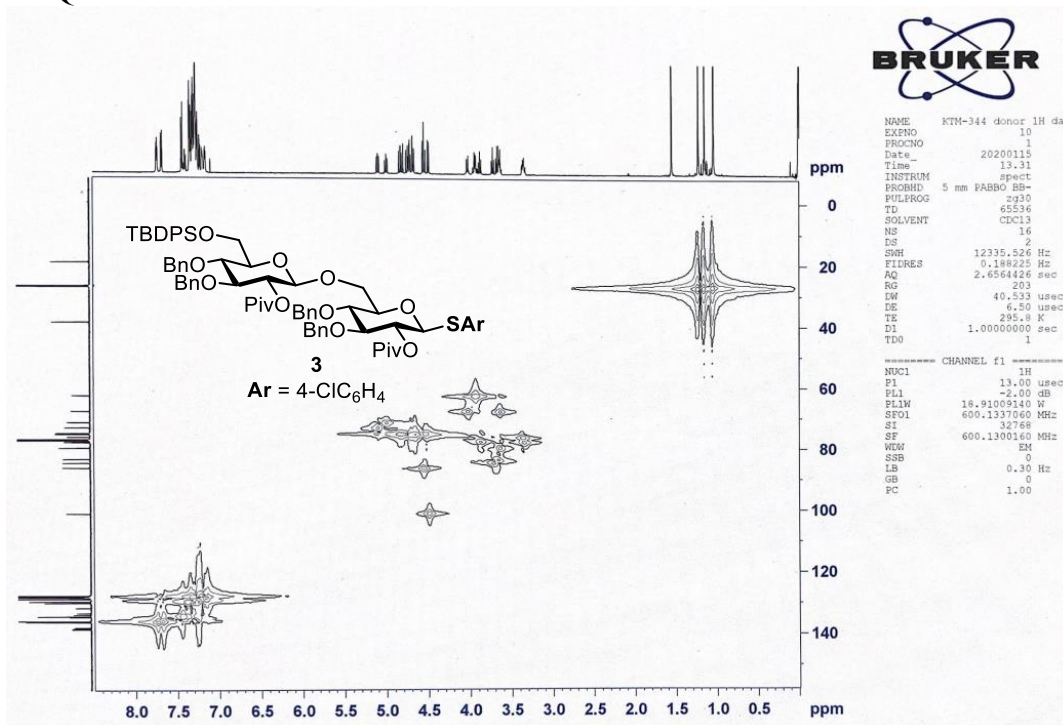
<sup>13</sup>C NMR



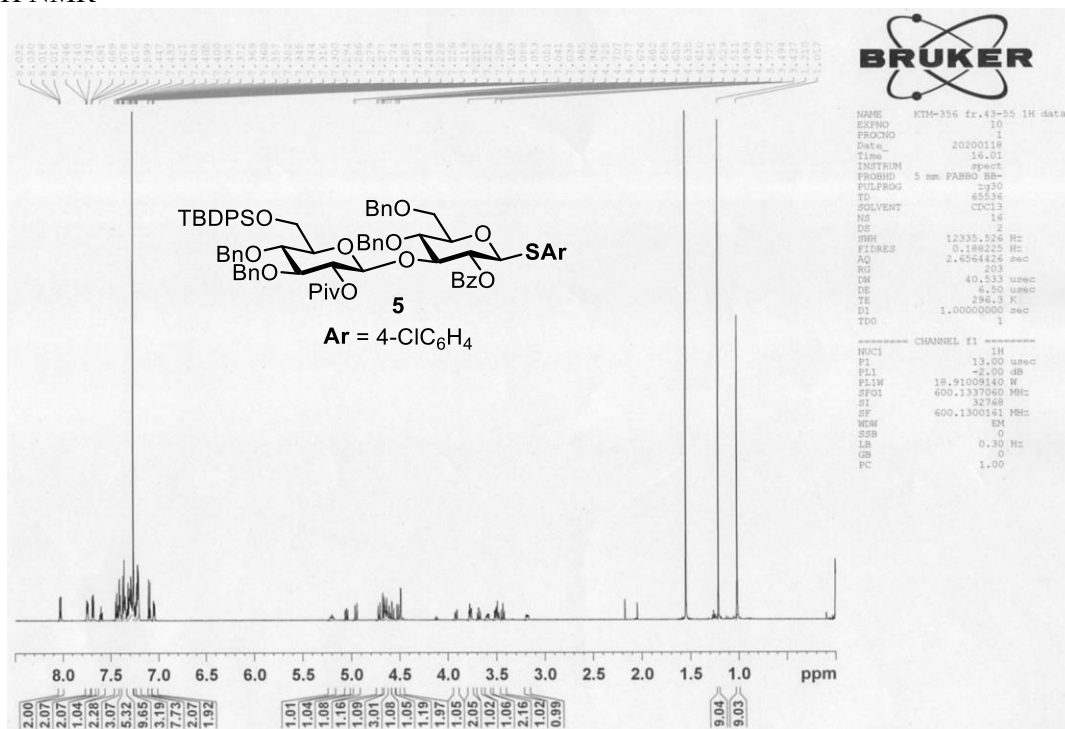
H-H cosy



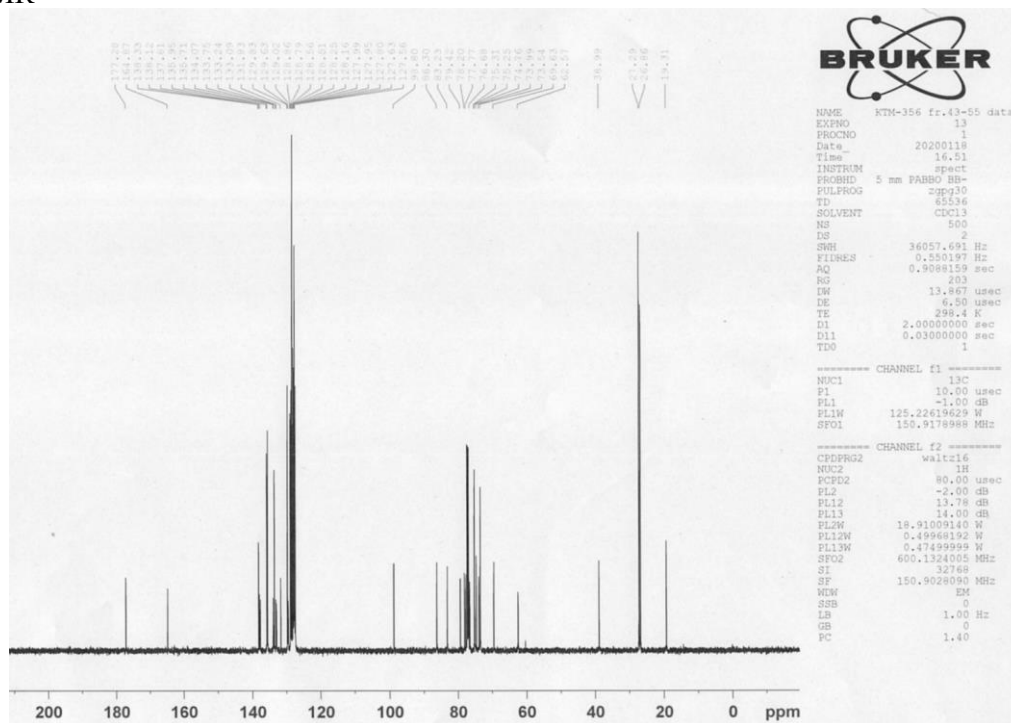
HMQC



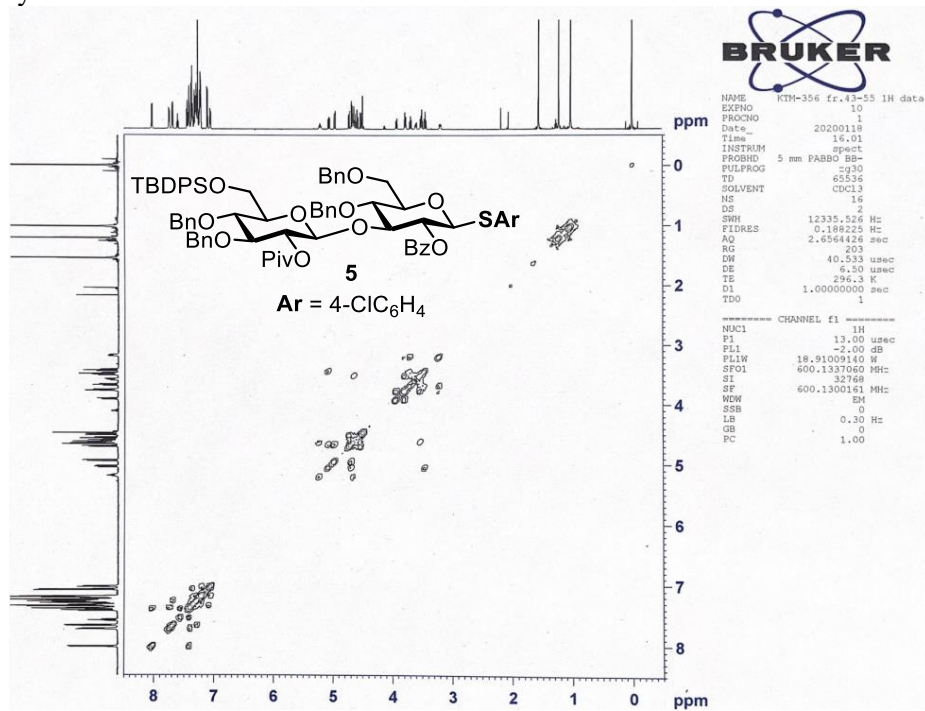
<sup>1</sup>H NMR



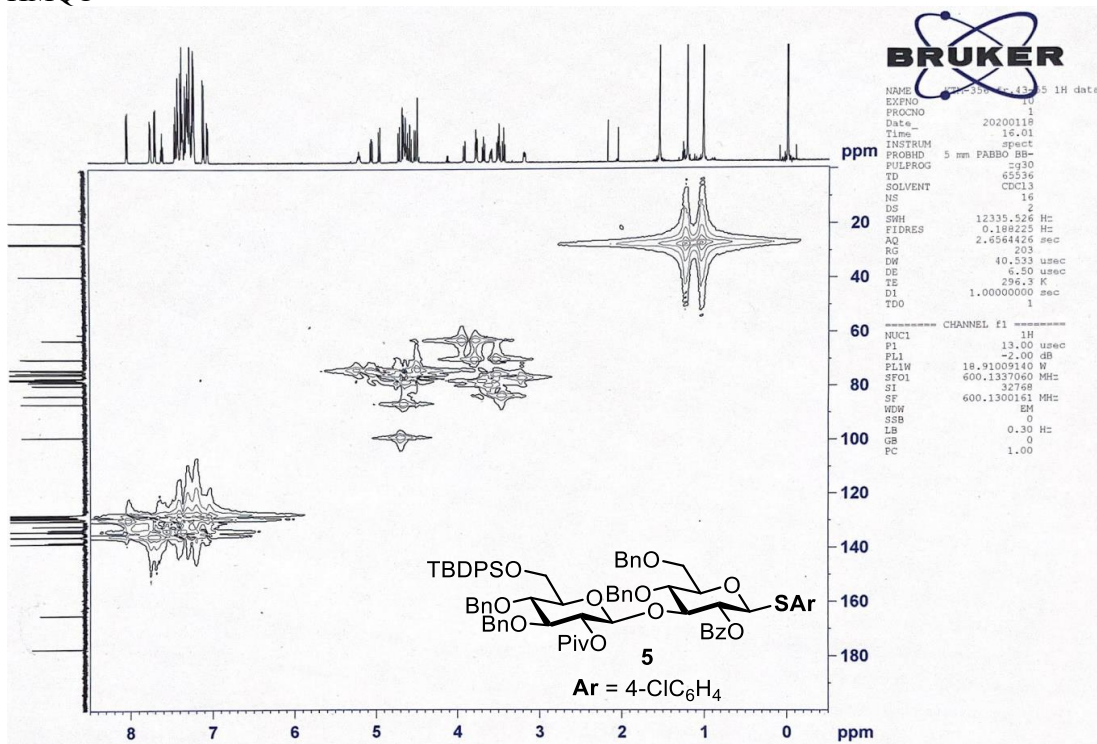
<sup>13</sup>C NMR



H-H cosy



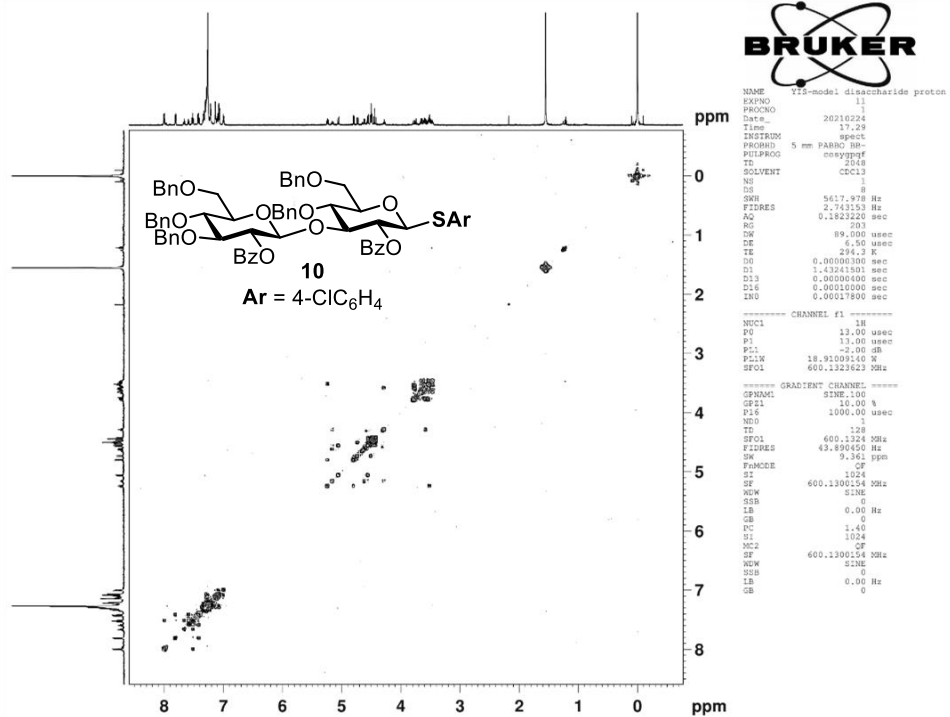
HMQC



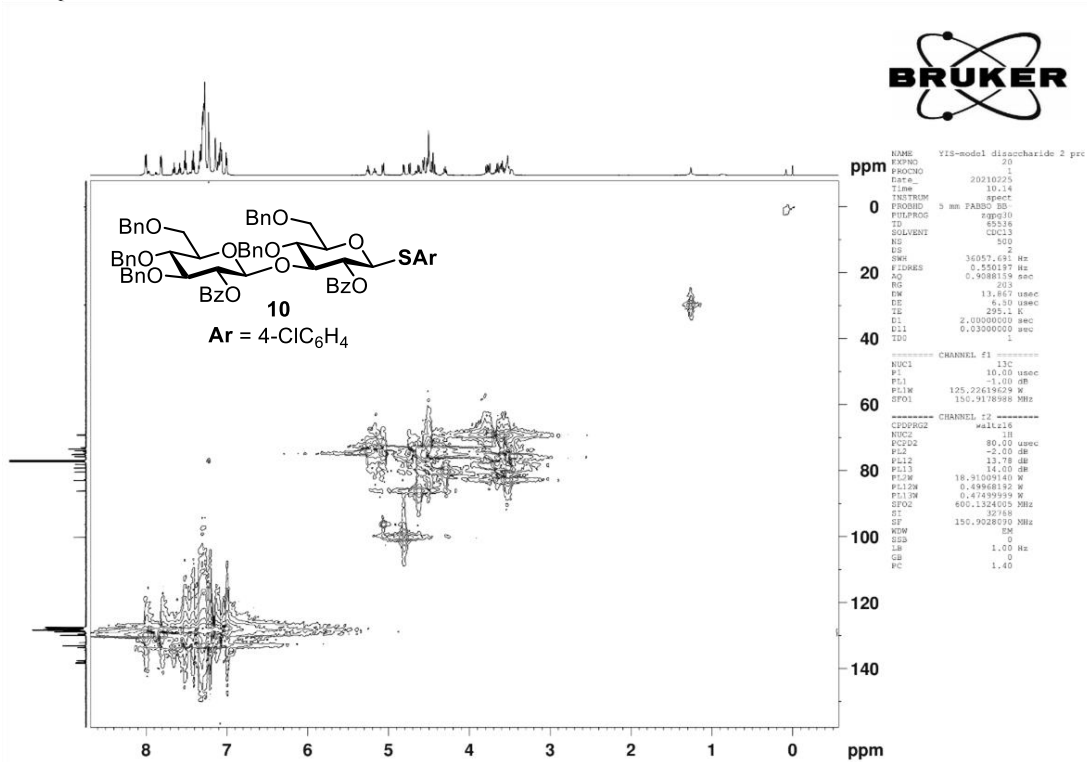




H-H cosy

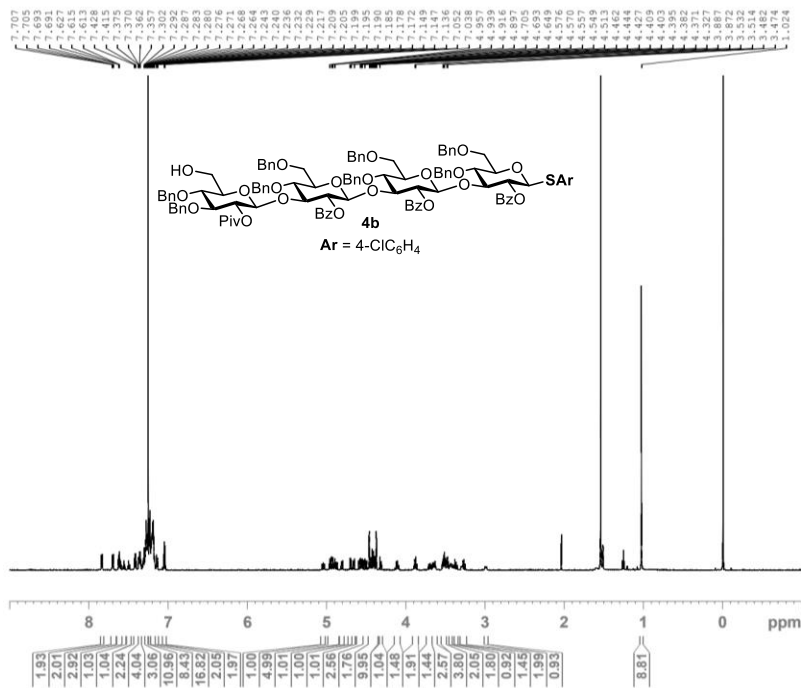


HMQC



# <sup>1</sup>H, <sup>13</sup>C NMR, H-H COSY and HMQC spectra of tetrasaccharide building block

## <sup>1</sup>H NMR



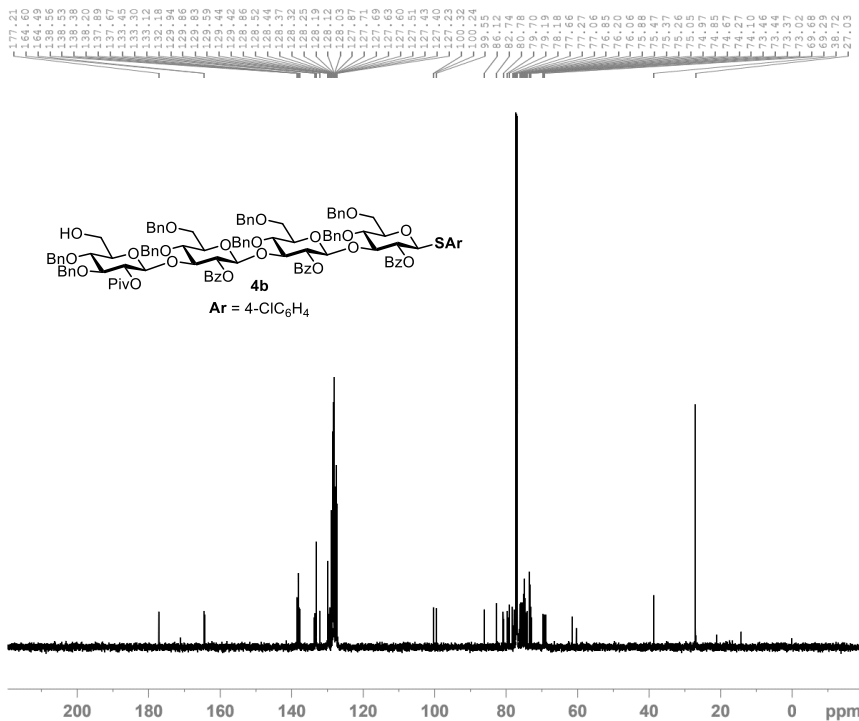
```
Current Data Parameters
NAME YIS-219 COSY
EXPNO 10
PROCNO 1

F2 - Acquisition Parameters
Date_ 20220819
Time_ 14.43
INSTRUM spect
PROBHD 5 mm PABBO BB-
PULPROG zg30
TD 65536
SOLVENT CDCl3
NS 16
DS 2
SWH 12335.526 Hz
FIDRES 0.188225 Hz
AQ 2.6563926 sec
RG 203
DW 40.533 usec
DE 6.50 usec
TE 302.4 K
D1 1.00000000 sec
TDO 1

===== CHANNEL f1 =====
SFO1 600.1337060 MHz
NUC1 1H
P1 14.00 usec
PLW1 13.50000000 W

F2 - Processing parameters
SI 32768
SF 600.1300184 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00
```

## <sup>13</sup>C NMR



```
Current Data Parameters
NAME YIS-219 carbon
EXPNO 10
PROCNO 1

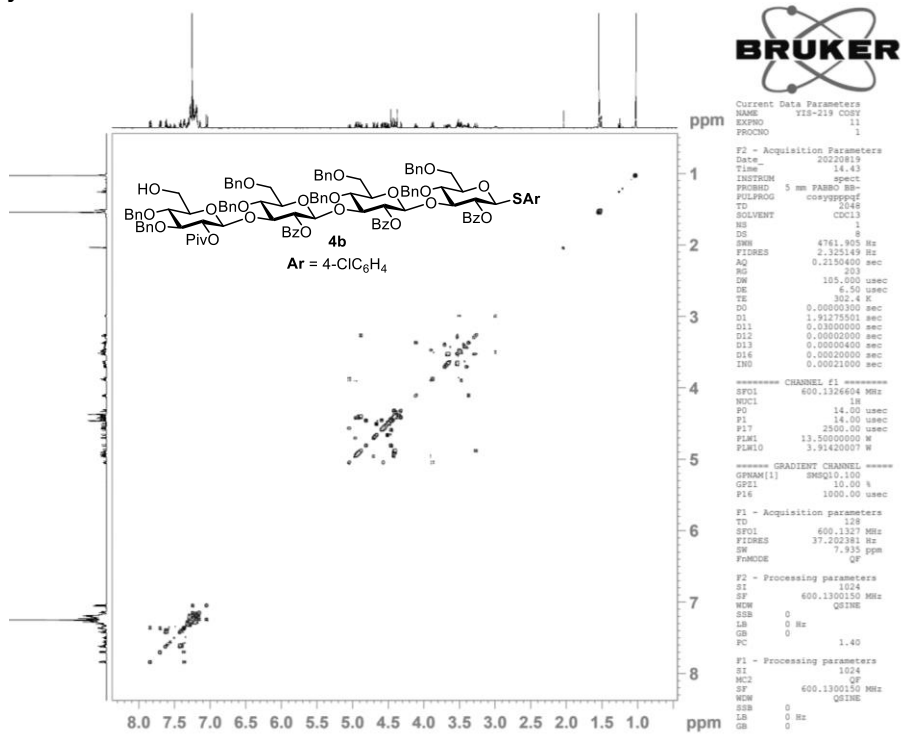
F2 - Acquisition Parameters
Date_ 20220819
Time_ 15.17
INSTRUM spect
PROBHD 5 mm PABBO BB-
PULPROG zgpg30
TD 65536
SOLVENT CDCl3
NS 500
DS 2
SWH 36057.691 Hz
FIDRES 0.550197 Hz
AQ 0.9087659 sec
RG 203
DW 13.867 usec
DE 6.50 usec
TE 303.3 K
D1 2.00000000 sec
D11 0.03000000 sec
TDO 1

===== CHANNEL f1 =====
SFO1 150.9178988 MHz
NUC1 13C
P1 12.00 usec
PLW1 79.40000153 W

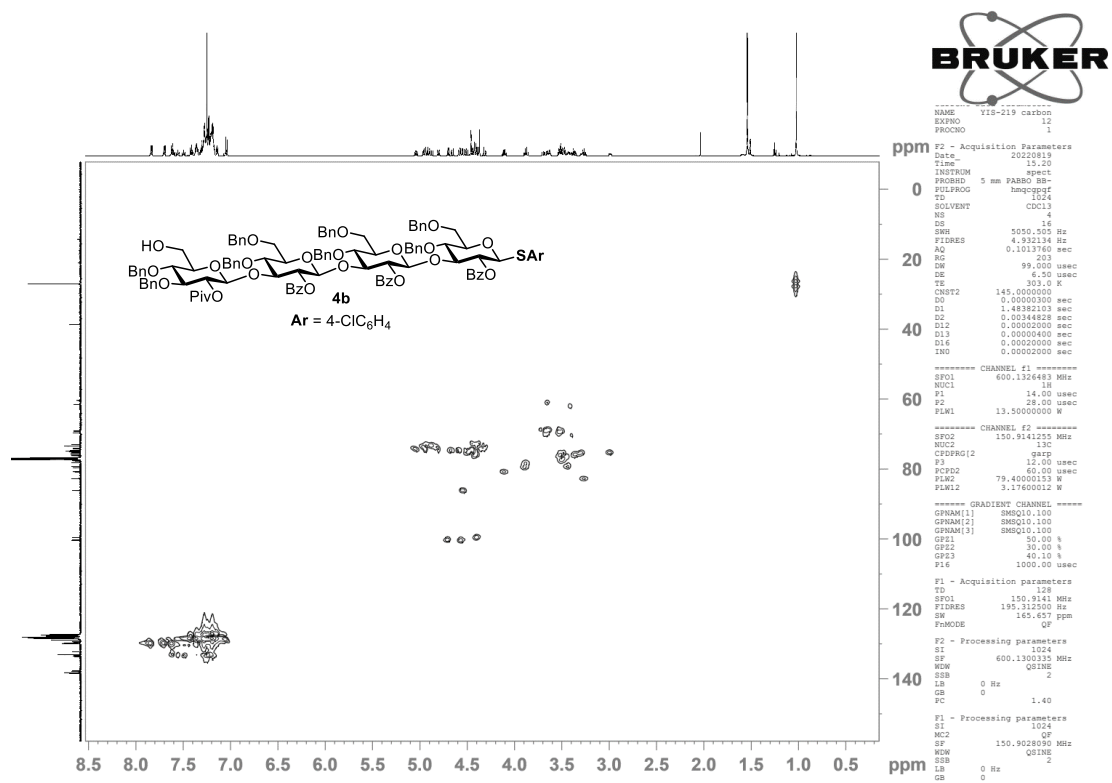
===== CHANNEL f2 =====
SFO2 600.1324005 MHz
NUC2 1H
CPDPRG2 waltz16
PCPD2 70.00 usec
PLW2 13.50000000 W
PLW12 0.54000002 W
PLW13 0.26460001 W

F2 - Processing parameters
SI 32768
SF 150.9028090 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40
```

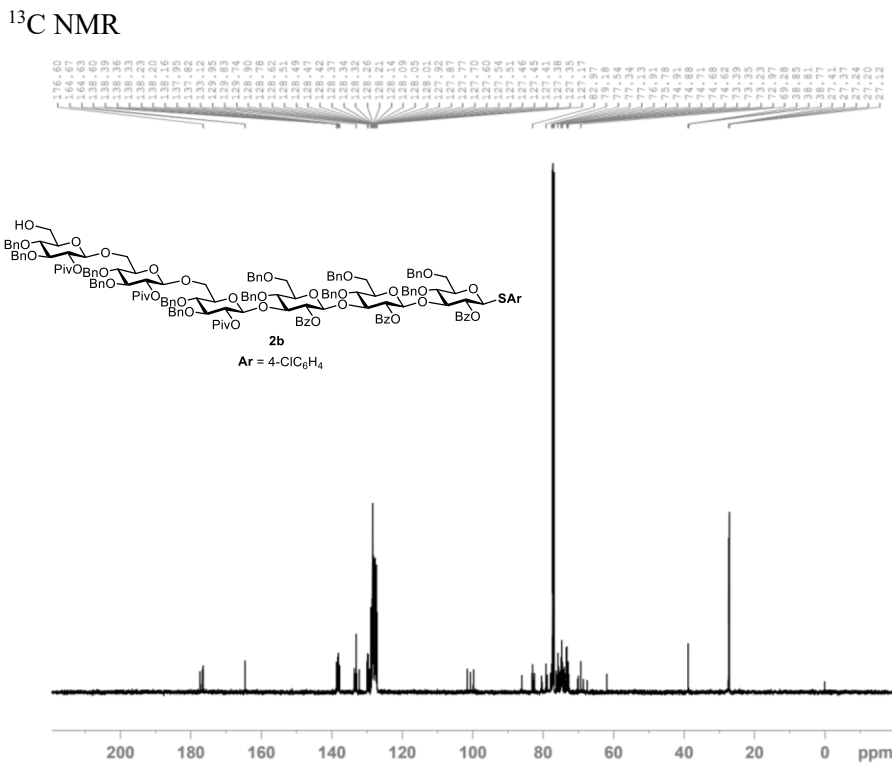
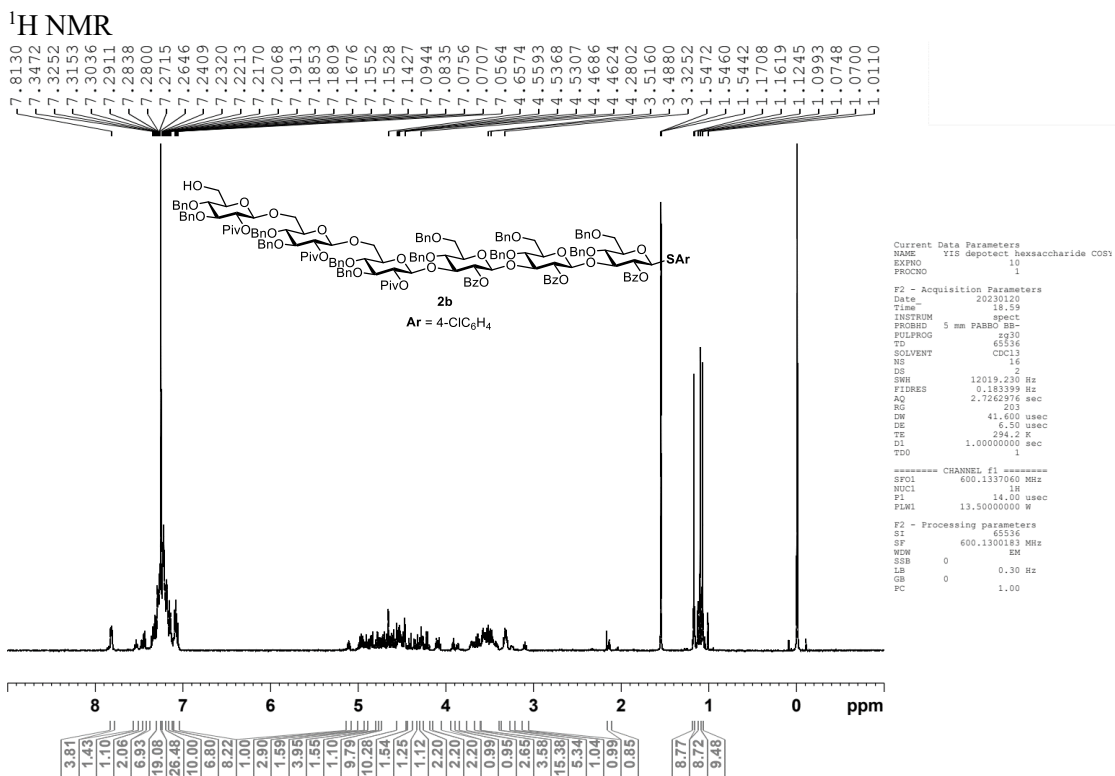
H-H cosy



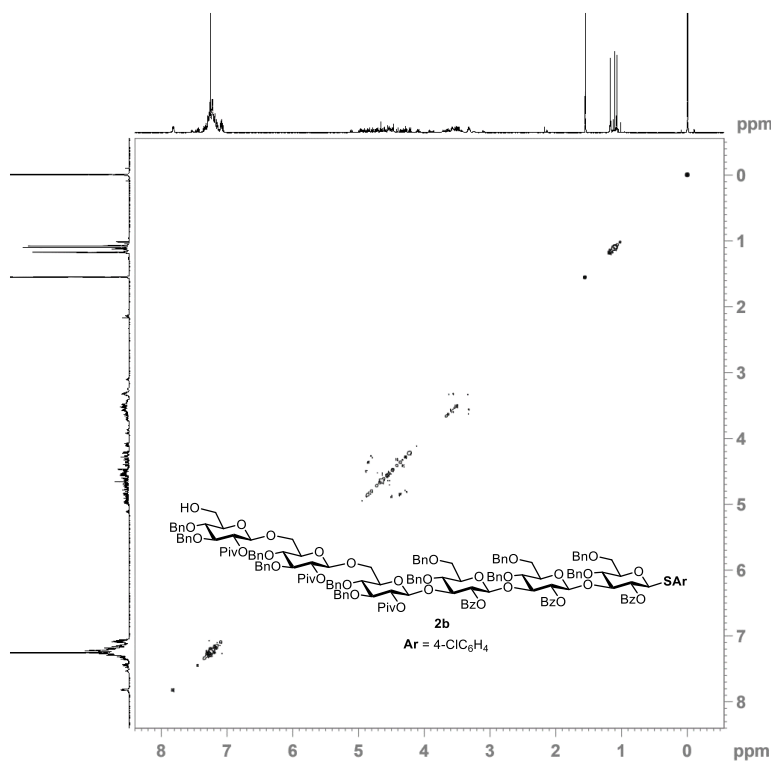
HMQC



# <sup>1</sup>H, <sup>13</sup>C NMR, H-H COSY and HMQC spectra of semi-circular hexasaccharide



H-H cosy



```

Current Data Parameters
NAME      V18 depotect hexsacaride COSY
EXPNO    11
PROCNO   1

F2 - Acquisition Parameters
Date_    20230120
Time     19.00
INSTRUM  spect
PROBHD   5 mm PABBO BB-
PULPROG  cosypppqcf
TD        2048
SOLVENT  CDCl3
NS        8
DS        1
SWH       5376.344 Hz
FIDRES   2.425168 Hz
AQ        0.1904640 sec
RG        181
DE        6.50 usec
TE        294.2 K
DO        0.0000300 sec
D1        1.3713196 sec
D11       0.0300000 sec
D12       0.0002000 sec
D13       0.0000400 sec
D16       0.0002000 sec
INO       0.00018600 sec

===== CHANNEL f1 =====
SF01     600.1323662 MHz
NUC1     1H
P1        14.00 usec
P11       14.00 usec
P17       2500.00 usec
PLM1     13.5000000 W
PLM10    3.9342000 W

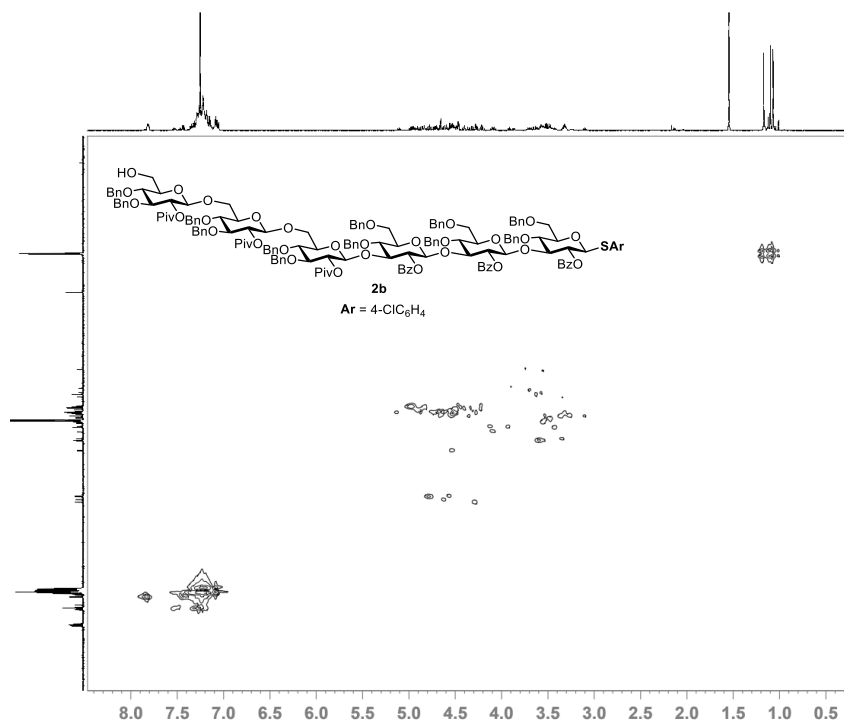
===== GRADIENT CHANNEL =====
GPMAM[1] SMSQ10.100
GFS1     10.00 %
P16      1000.00 usec

F1 - Acquisition parameters
TD        128
SF01     600.1324 MHz
FIDRES   42.002689 Hz
AQ        8.959 ppm
F2MODE    QF

F2 - Processing parameters
SI        1024
SF        600.1300131 MHz
WDW       QSIKINE
SSB       0
LB        0 Hz
GB        0
PC        1.40

F1 - Processing parameters
SI        1024
MC2       QF
SF        600.1300131 MHz
WDW       QSIKINE
SSB       0
LB        0 Hz
GB        0
    
```

HMQC



```

Current Data Parameters
NAME      V18 depotect hexsacaride HMQC
EXPNO    11
PROCNO   1

F2 - Acquisition Parameters
Date_    20230120
Time     19.46
INSTRUM  spect
PROBHD   5 mm PABBO BB-
PULPROG  hmqcspcf
TD        1024
SOLVENT  CDCl3
NS        8
DS        1
SWH       5000.000 Hz
FIDRES   4.882813 Hz
AQ        0.1024000 sec
RG        203
DE        100.000 usec
TE        294.2 K
CMT2     145.0000000 sec
DO        0.0000300 sec
D1        1.48779703 sec
D2        0.00348281 sec
D12       0.0002000 sec
D13       0.0000400 sec
D16       0.0002000 sec
INO       0.00002000 sec

===== CHANNEL f1 =====
SF01     600.1327125 MHz
NUC1     1H
P1        14.00 usec
P2        28.00 usec
PLM1     13.5000000 W

===== CHANNEL f2 =====
SF02     150.914125 MHz
NUC2     13C
CFPRG[2] hmqc
P3        12.00 usec
P3PC2    60.00 usec
PLM2     79.40000153 W
PLM12    3.27600012 W

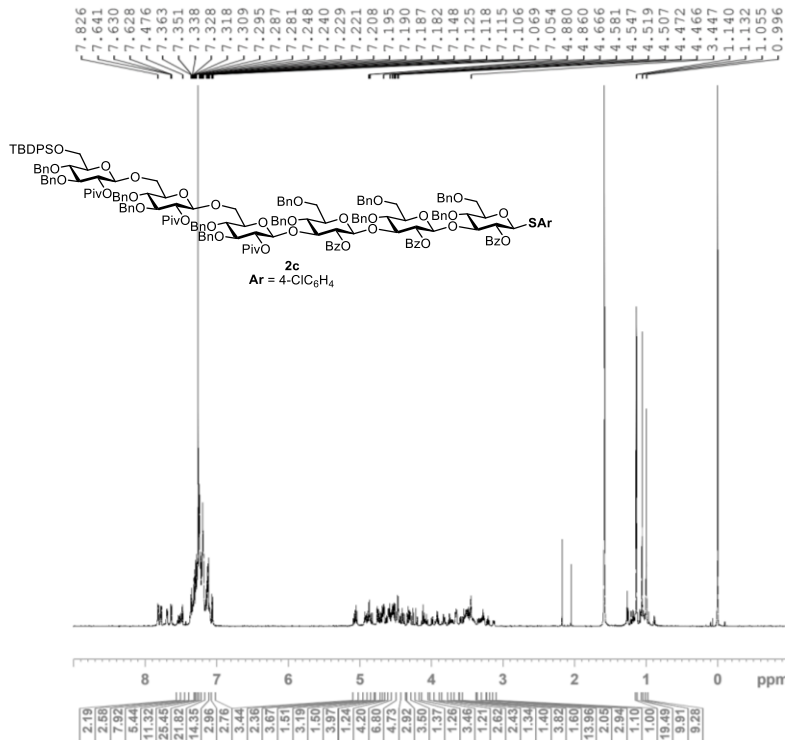
===== GRADIENT CHANNEL =====
GPMAM[1] SMSQ10.100
GPMAM[2] SMSQ10.100
GPMAM[3] SMSQ10.100
GFS1     50.00 %
GFS2     30.00 %
GFS3     40.10 %
P16      1000.00 usec

F1 - Acquisition parameters
TD        128
SF01     150.9141 MHz
FIDRES   195.312500 Hz
AQ        145.455 ppm
F2MODE    QF

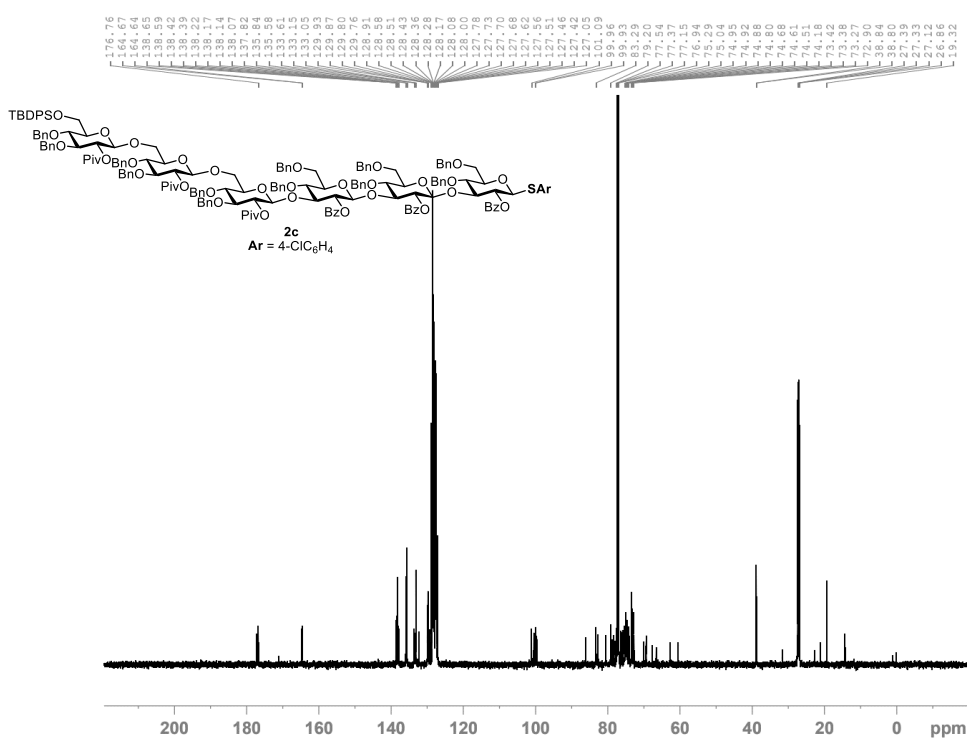
F2 - Processing parameters
SI        1024
SF        600.1300410 MHz
WDW       QSIKINE
SSB       2
LB        0 Hz
GB        0
PC        1.40

F1 - Processing parameters
SI        1024
MC2       QF
SF        150.9028090 MHz
WDW       QSIKINE
SSB       2
LB        0 Hz
GB        0
    
```

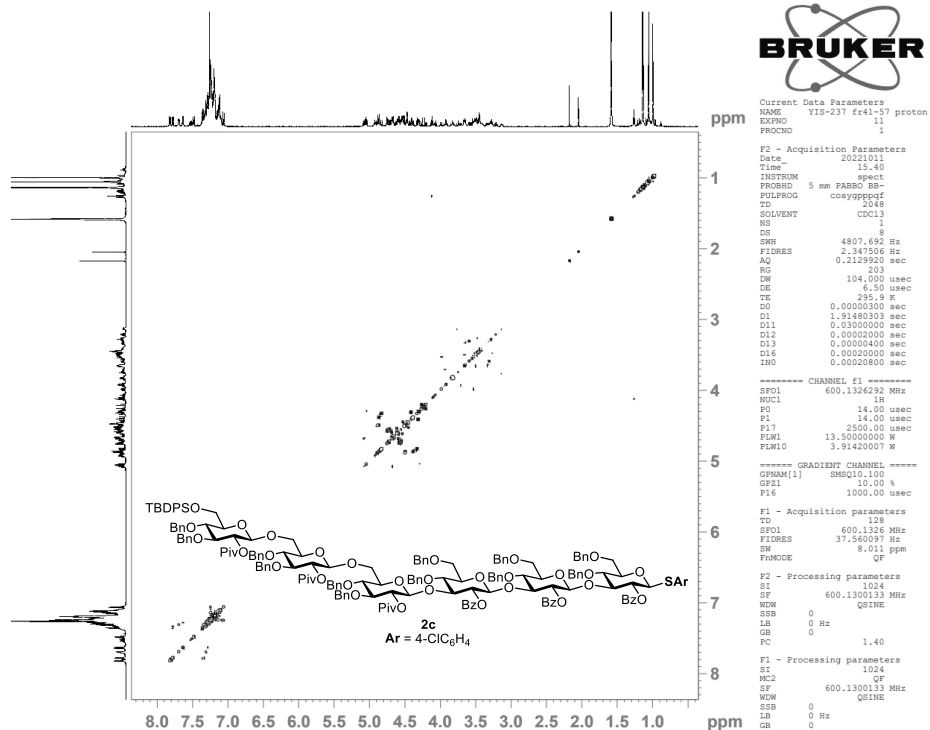
<sup>1</sup>H NMR



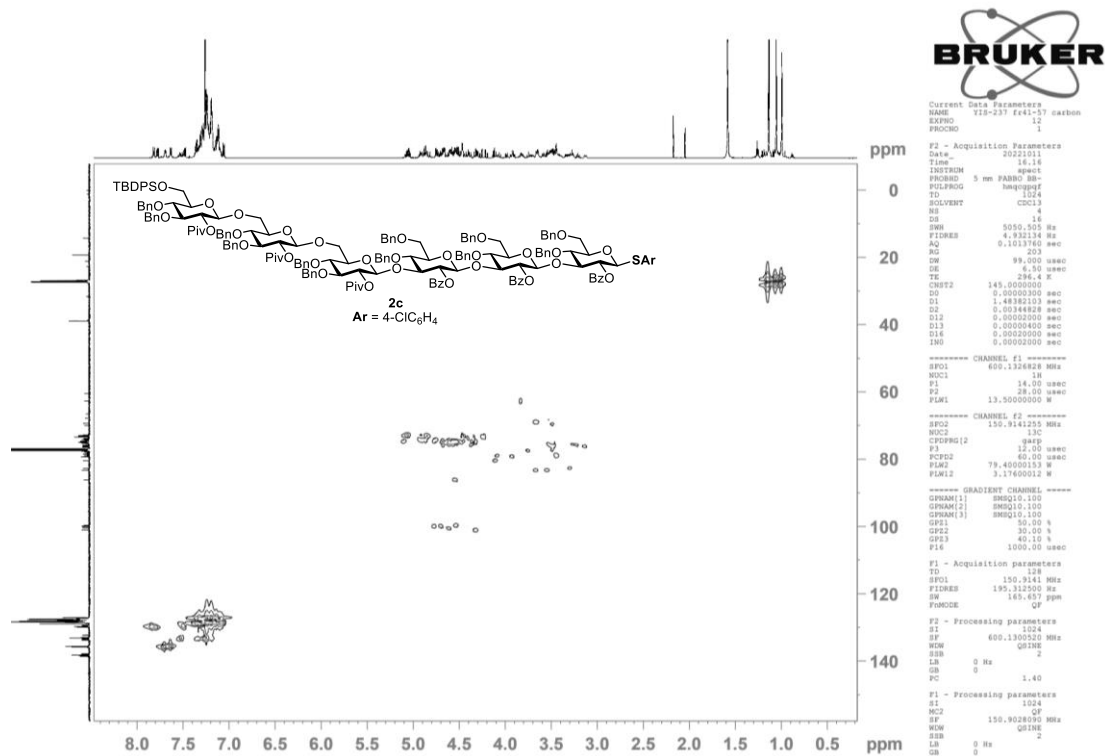
<sup>13</sup>C NMR



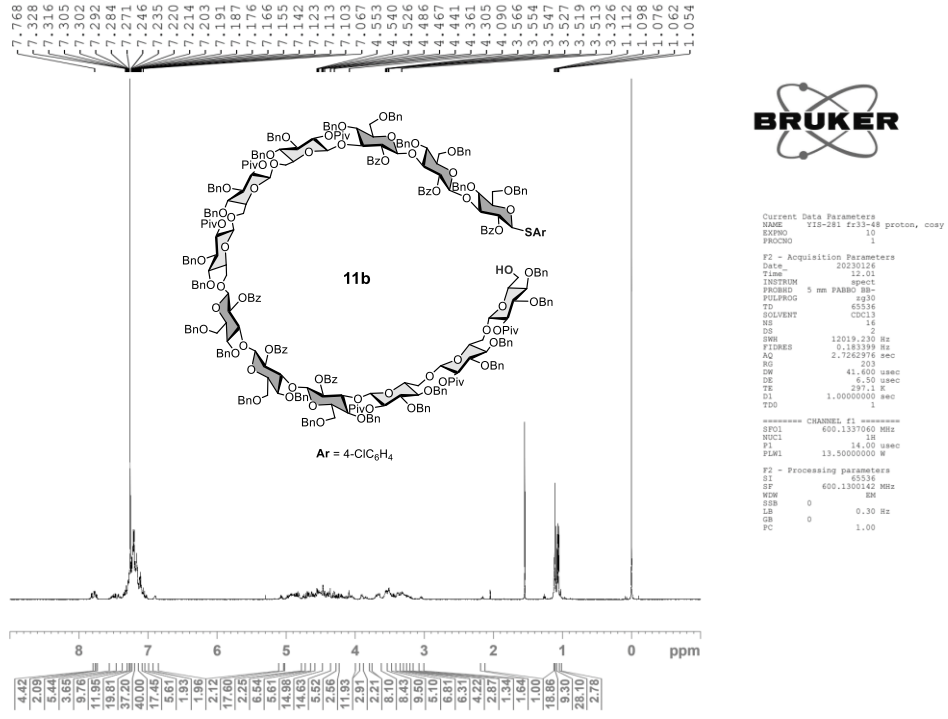
# H-H cosy



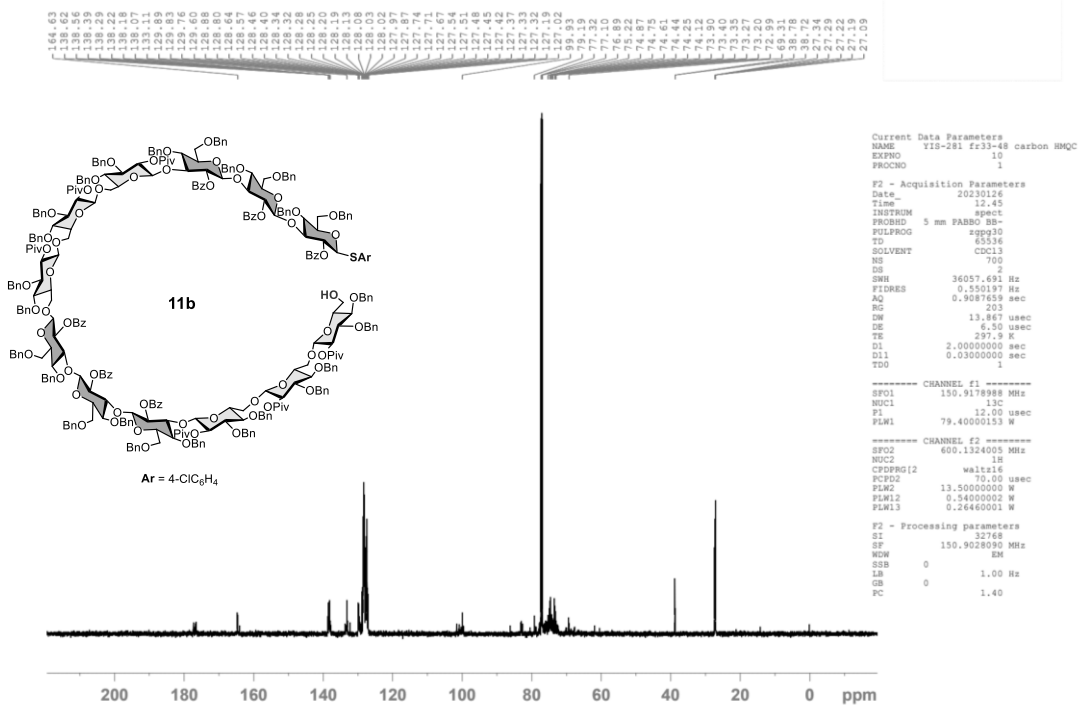
# HMQC



<sup>1</sup>H, <sup>13</sup>C NMR, H-H COSY and HMQC spectra of linear and cyclic dodecasaccharides  
<sup>1</sup>H NMR

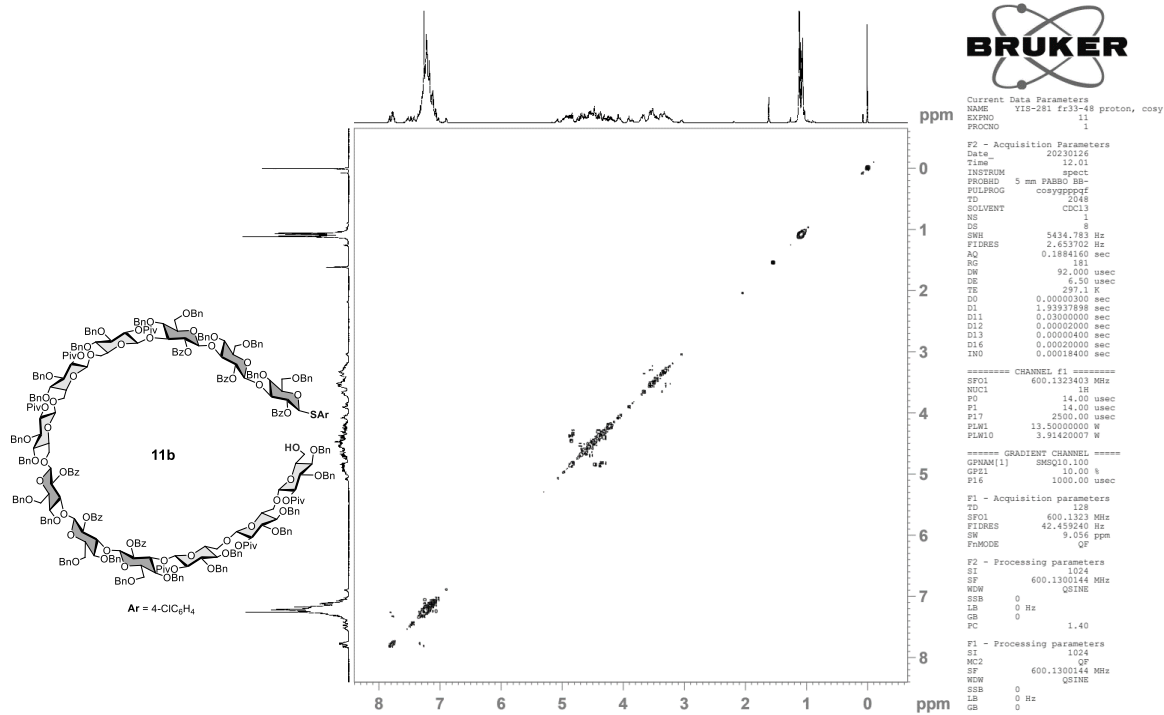


<sup>13</sup>C NMR





# H-H cosy



Current Data Parameters  
 NAME Y15-281 fr33-48 proton\_cosy  
 EXPNO 11  
 PROCNO 1

F2 - Acquisition Parameters  
 Date\_ 20230226  
 Time 12.01  
 INSTRUM spect  
 PROBHD 5 mm PABBO BB-  
 PULPROG cosypppgp2  
 TD 2048  
 SOLVENT CDCl3  
 NS 1  
 DS 8  
 SWH 5434.783 Hz  
 FIDRES 2.653702 Hz  
 AQ 0.1864160 sec  
 RG 181  
 DE 6.50 usec  
 TE 297.1 K  
 DO 0.0000300 sec  
 D1 1.93937898 sec  
 D11 0.0300000 sec  
 D12 0.00002000 sec  
 D13 0.0000400 sec  
 D16 0.0000000 sec  
 INO 0.00018400 sec

===== CHANNEL f1 =====  
 SF01 600.1323403 MHz  
 NUC1 1H  
 P1 14.00 usec  
 P17 2500.00 usec  
 SFO1 13.5000000 M  
 FLM10 3.91420007 M

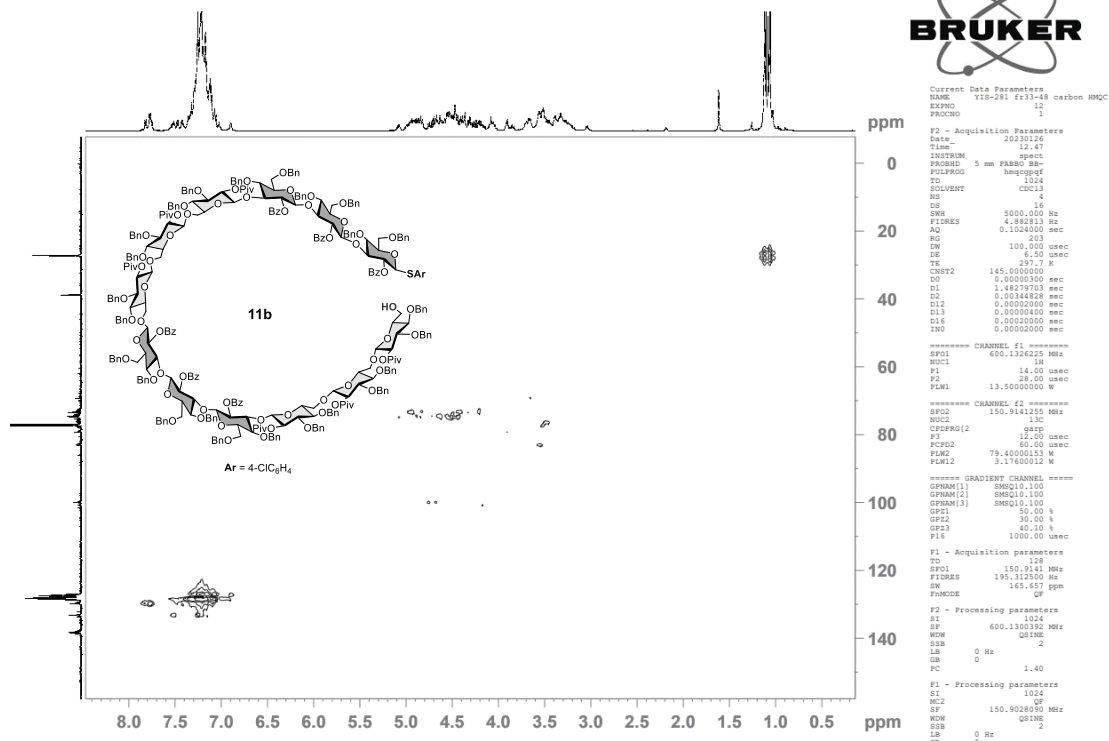
===== GRADIENT CHANNEL =====  
 GPRM[1] SMSQ10.100  
 GP1 10.00 %  
 P16 1000.00 usec

F1 - Acquisition parameters  
 TD 128  
 SF01 600.1323 MHz  
 FIDRES 42.459240 Hz  
 SW 9.056 ppm  
 FMODE QF

F2 - Processing parameters  
 SI 1024  
 SF 600.1300144 MHz  
 WDW QSI  
 SSB 0  
 LB 0 Hz  
 GB 0  
 FC 1.40

F1 - Processing parameters  
 SI 1024  
 MC2 QF  
 SF 600.1300144 MHz  
 WDW QSI  
 SSB 0  
 LB 0 Hz  
 GB 0

# HMQC



Current Data Parameters  
 NAME Y15-281 fr33-48 carbon\_hmhc  
 EXPNO 12  
 PROCNO 1

F2 - Acquisition Parameters  
 Date\_ 20230226  
 Time 12.47  
 INSTRUM spect  
 PROBHD 5 mm PABBO BB-  
 PULPROG hmqcpgp2  
 TD 1024  
 SOLVENT CDCl3  
 NS 4  
 DS 14  
 SWH 5000.000 Hz  
 FIDRES 4.86813 Hz  
 AQ 0.1024000 sec  
 RG 203  
 DE 6.50 usec  
 TE 297.1 K  
 CNST2 145.0000000 sec  
 D1 1.48279703 sec  
 D2 0.00344828 sec  
 D12 0.00002000 sec  
 D13 0.00000000 sec  
 D16 0.00002000 sec  
 INO 0.00000000 sec

===== CHANNEL f1 =====  
 SF01 600.1326225 MHz  
 NUC1 1H  
 P1 14.00 usec  
 P2 28.00 usec  
 FLM1 13.5000000 M

===== CHANNEL f2 =====  
 SF02 150.8141225 MHz  
 NUC2 13C  
 P3 12.00 usec  
 P3R02 60.00 usec  
 FLM2 79.40000153 M  
 FLM12 3.17460012 M

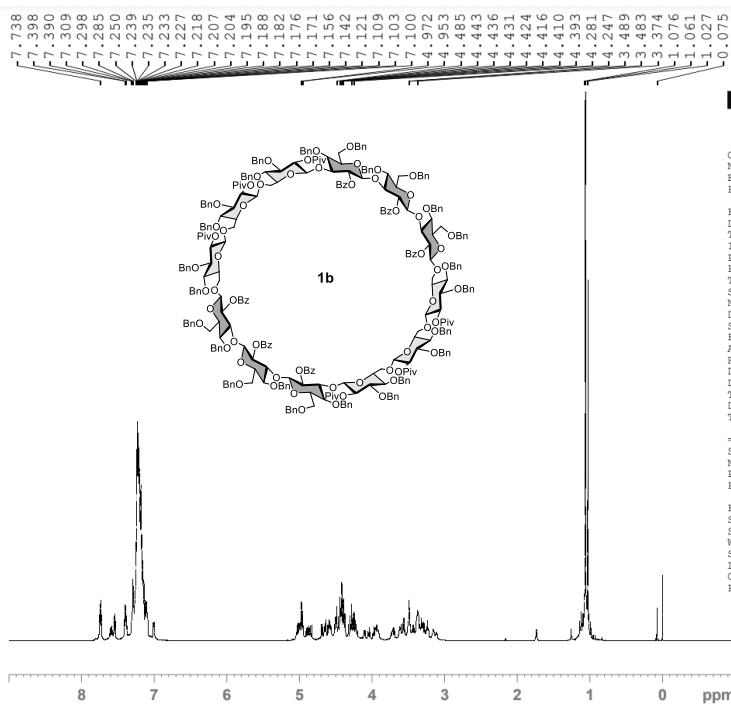
===== GRADIENT CHANNEL =====  
 GPRM[1] SMSQ10.100  
 GPRM[2] SMSQ10.100  
 GPRM[3] SMSQ10.100  
 GP1 50.00 %  
 GP2 30.00 %  
 GP3 40.10 %  
 P16 1000.00 usec

F1 - Acquisition parameters  
 TD 128  
 SF01 150.8141 MHz  
 FIDRES 195.312500 Hz  
 SW 145.557 ppm  
 FMODE QF

F2 - Processing parameters  
 SI 1024  
 SF 600.1300192 MHz  
 WDW QSI  
 SSB 2  
 LB 0 Hz  
 GB 0  
 FC 1.40

F1 - Processing parameters  
 SI 1024  
 MC2 QF  
 SF 150.8028090 MHz  
 WDW QSI  
 SSB 2  
 LB 0 Hz  
 GB 0

# <sup>1</sup>H NMR



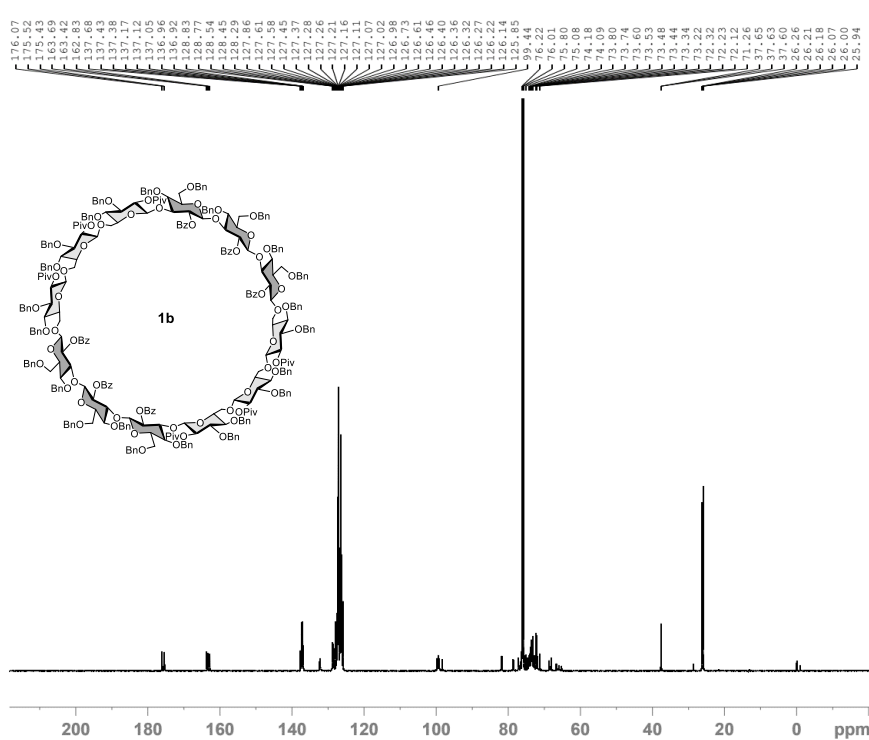
Current Data Parameters  
 NAME YIS-cyclic12\_0214  
 EXPNO 10  
 PROCNO 1

F2 - Acquisition Parameters  
 Date 20230214  
 Time 15.22  
 INSTRUM spect  
 PROBHD 5 mm PABBO BB-  
 PULPROG zg30  
 TD 65536  
 SOLVENT CDCl3  
 NS 16  
 DS 2  
 SWH 12019.230 Hz  
 FIDRES 0.183399 Hz  
 AQ 2.7262976 sec  
 RG 71.8  
 DW 41.600 usec  
 DE 6.50 usec  
 TE 293.8 K  
 D1 1.00000000 sec  
 TDO 1

==== CHANNEL f1 =====  
 SF01 600.1337060 MHz  
 NUC1 1H  
 P1 14.00 usec  
 PLW1 13.50000000 W

F2 - Processing parameters  
 SI 65536  
 SF 600.1300297 MHz  
 WDW EM  
 SSB 0  
 LB 0 0.30 Hz  
 GB 0  
 FC 1.00

# <sup>13</sup>C NMR



Current Data Parameters  
 NAME YIS-cyclic12\_0219  
 EXPNO 20  
 PROCNO 1

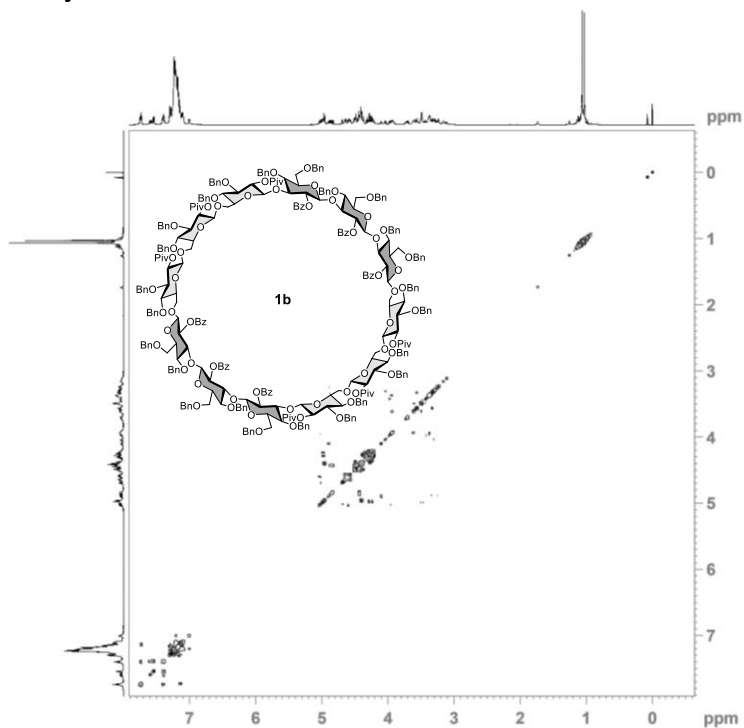
F2 - Acquisition Parameters  
 Date 20230220  
 Time 8.52  
 INSTRUM spect  
 PROBHD 5 mm PABBO BB-  
 PULPROG zgpg30  
 TD 65536  
 SOLVENT CDCl3  
 NS 17000  
 DS 2  
 SWH 36057.691 Hz  
 FIDRES 0.550197 Hz  
 AQ 0.9087659 sec  
 RG 203  
 DW 13.867 usec  
 DE 6.50 usec  
 TE 295.0 K  
 D1 2.00000000 sec  
 D11 0.03000000 sec  
 TDO 1

==== CHANNEL f1 =====  
 SF01 150.9178988 MHz  
 NUC1 13C  
 P1 12.00 usec  
 PLW1 79.40000153 W

==== CHANNEL f2 =====  
 SF02 600.1324005 MHz  
 NUC2 1H  
 CPDPRG2 waltz16  
 PCPD2 70.00 usec  
 PLW2 13.50000000 W  
 PLW12 0.54000002 W  
 PLW13 0.26460001 W

F2 - Processing parameters  
 SI 32768  
 SF 150.9029732 MHz  
 WDW EM  
 SSB 0  
 LB 0 1.00 Hz  
 GB 0  
 FC 1.40

H-H cosy



Current Data Parameters  
 NAME Vis-cyclic12\_0214  
 EXPRNO 1  
 PROCNO 1

F2 - Acquisition Parameters  
 Date\_ 20230214  
 Time\_ 15.23  
 INSTRUM spect  
 PROBHD 5 mm PABBO BB-  
 PULPROG cosypppqr  
 TD 2048  
 SOLVENT CDCl3  
 NS 1  
 DS 8  
 SWH 5555.556 Hz  
 FIDRES 2.712474 Hz  
 AQ 0.1843200 sec  
 RG 36  
 SW 90.000 usec  
 DE 6.50 usec  
 TE 293.2 K  
 DO 0.00000000 sec  
 D1 1.94247501 sec  
 D11 0.03000000 sec  
 D12 0.00002000 sec  
 D13 0.00000000 sec  
 D16 0.00020000 sec  
 INO 0.00018000 sec

----- CHANNEL f1 -----  
 SF01 600.1324254 MHz  
 NUC1 1H  
 P0 14.00 usec  
 P1 14.00 usec  
 P17 2500.00 usec  
 PLW1 13.50000000 W  
 PLW2 3.91420007 W

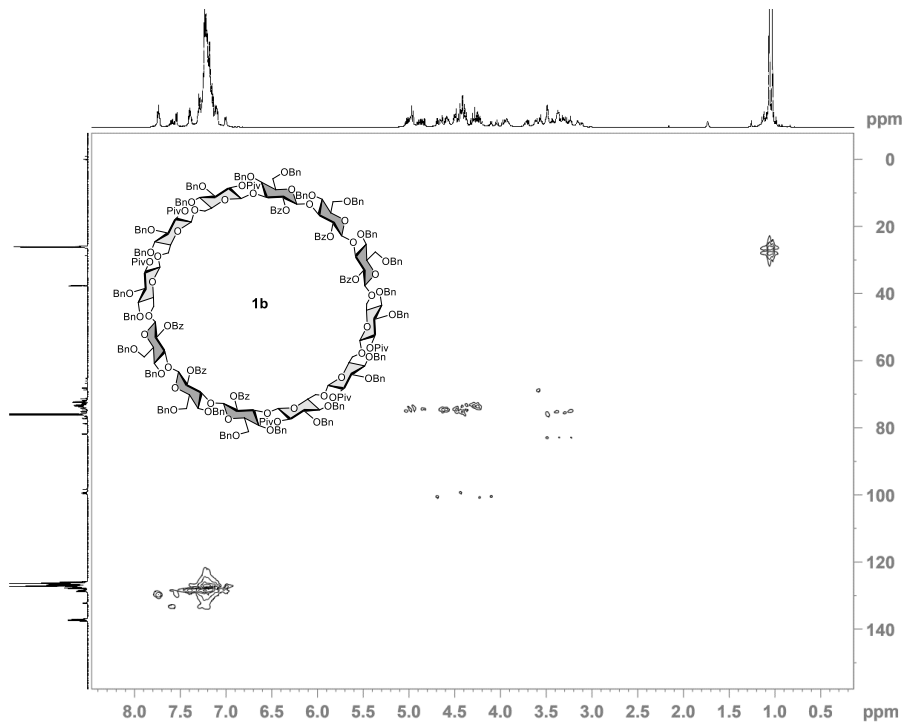
----- GRADIENT CHANNEL -----  
 GPMAX[1] SMQ10.100  
 GPE1 10.00 %  
 P16 1000.00 usec

F1 - Acquisition parameters  
 TD 128  
 SF01 600.1324 MHz  
 FIDRES 43.402779 Hz  
 SW 9.257 ppm  
 FwMODE QF

F2 - Processing parameters  
 SI 1024  
 SF 600.1300297 MHz  
 WDR 0  
 SSB 0  
 LB 0 Hz  
 GB 0  
 PC 1.40

F1 - Processing parameters  
 SI 1024  
 WCF QF  
 SF 600.1300297 MHz  
 WDR 0  
 SSB 0  
 LB 0 Hz  
 GB 0

HMQC



Current Data Parameters  
 NAME Vis-cyclic12\_0214  
 EXPRNO 22  
 PROCNO 1

F2 - Acquisition Parameters  
 Date\_ 20230214  
 Time\_ 17.15  
 INSTRUM spect  
 PROBHD 5 mm PABBO BB-  
 PULPROG hmqcqpqr  
 TD 1024  
 SOLVENT CDCl3  
 NS 4  
 DS 15  
 SWH 5494.505 Hz  
 FIDRES 5.365728 Hz  
 AQ 0.0931840 sec  
 RG 201  
 SW 91.000 usec  
 DE 6.50 usec  
 TE 294.2 K  
 CNTR2 145.000000  
 DC 0.00000000 sec  
 D1 1.49201298 sec  
 D2 0.00348218 sec  
 D12 0.00002000 sec  
 D13 0.00000000 sec  
 D16 0.00020000 sec  
 INO 0.00002000 sec

----- CHANNEL f1 -----  
 SF01 600.1324043 MHz  
 NUC1 1H  
 P1 14.00 usec  
 P2 25.00 usec  
 PLW1 13.50000000 W

----- CHANNEL f2 -----  
 SF02 150.9142125 MHz  
 NUC2 13C  
 CPDPRG2 gptp  
 P3 12.00 usec  
 P4 60.00 usec  
 PLW2 79.40000133 W  
 PLW3 3.17800012 W

----- GRADIENT CHANNEL -----  
 GPMAX[1] SMQ10.100  
 GPMAX[2] SMQ10.100  
 GPMAX[3] SMQ10.100  
 GPE1 50.00 %  
 GPE2 20.00 %  
 GPE3 80.10 %  
 P16 1000.00 usec

F1 - Acquisition parameters  
 TD 128  
 SF01 150.9141 MHz  
 FIDRES 195.312500 Hz  
 SW 185.652 ppm  
 FwMODE QF

F2 - Processing parameters  
 SI 1024  
 SF 600.1300298 MHz  
 WDR 0  
 SSB 2  
 LB 0 Hz  
 GB 0  
 PC 1.40

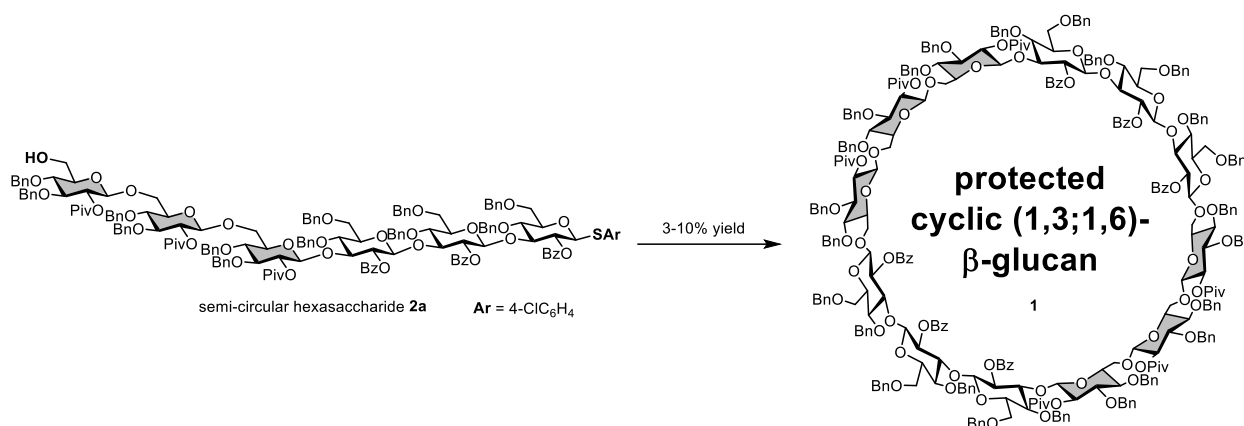
F1 - Processing parameters  
 SI 1024  
 WCF QF  
 SF 150.9028090 MHz  
 WDR 0  
 SSB 2  
 LB 0 Hz  
 GB 0

## Chapter 3.

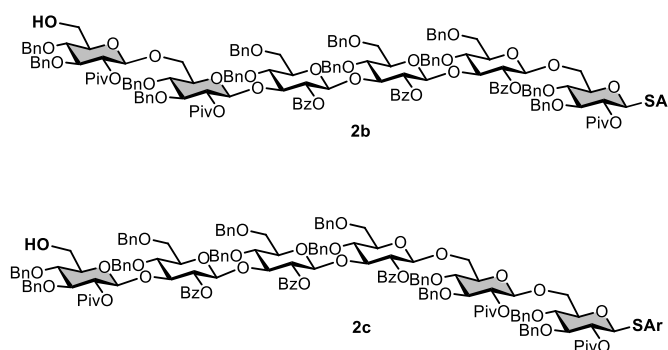
### Towards Rational Design of Oligosaccharide Building Blocks of Cyclic $\beta$ -Glucans

#### Introduction

As described in chapter 2, protected cyclic dodecasaccharide **1** was successfully synthesized from semi-circular hexasaccharide **2a** (Figure 3-1). To explore the possibility of further improvement of the yield, we planned to synthesize the cyclic dodecasaccharide **1** from other hexasaccharide precursors **2b** and **2c** (Figure 3-2). Among potential precursors, semi-circular hexasaccharide **2c** was synthesized because it also has protecting-group free 6-OH as a reaction site and can be synthesized by automated electrochemical assembly of building blocks in hand.



**Figure 3-1.** Dimerization-cyclization of semi-circular hexasaccharide precursor **2a**.

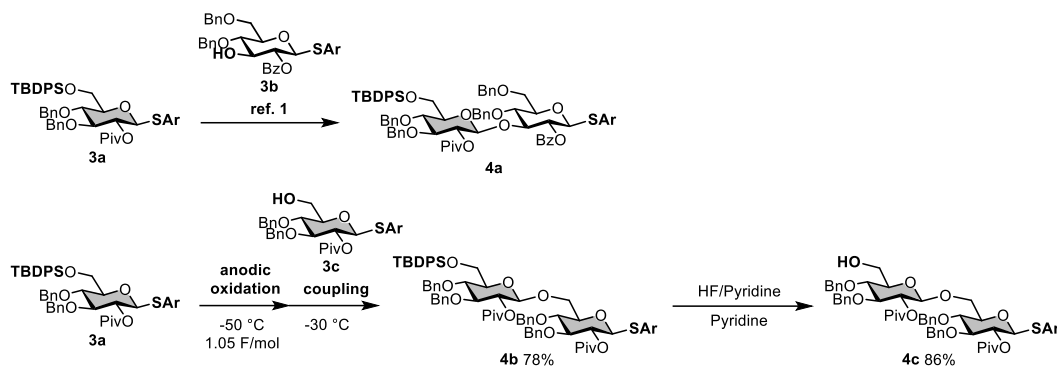


**Figure 3-2.** Potential hexasaccharide precursors of cyclic dodecasaccharide **1**.

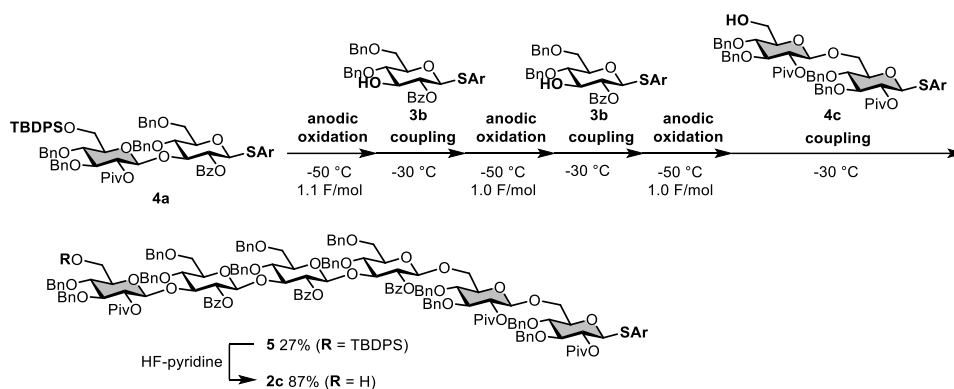
#### Results and Discussion

First, disaccharide building blocks **4a** and **4c** were synthesized (Scheme 3-1). Disaccharide **4a** and monosaccharides **3a-c** were prepared according to the reported procedures.<sup>1</sup> Glycosylation reaction of **3a** and **3c** and following deprotection of TBDPS group afforded disaccharide building block **4c**. Then, hexasaccharide **5** was synthesized by three steps of AEA using disaccharides **4a** and **4c** and two equivalents

of monosaccharide **3b**. Hexasaccharide **5** was converted to hexasaccharide precursor **2c** by deprotection of TBDPS group (Scheme 3-2).

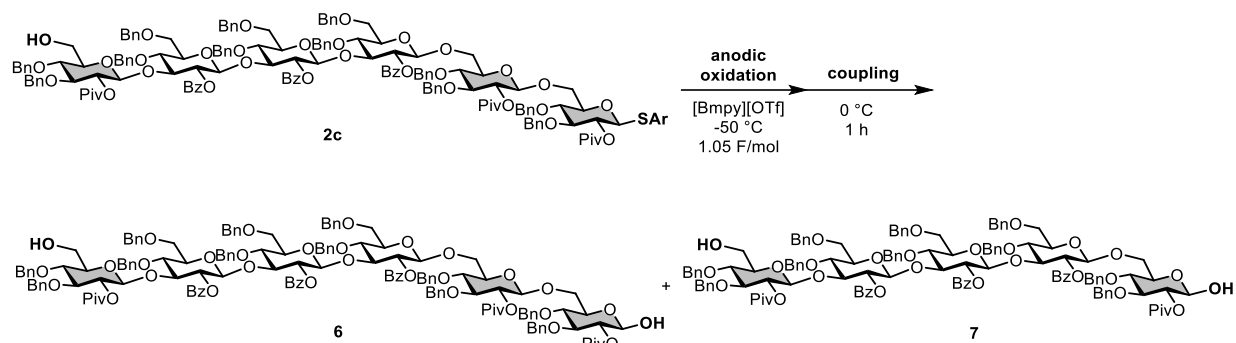


**Scheme 3-1.** Synthesis of disaccharide building blocks **4a** and **4c**.

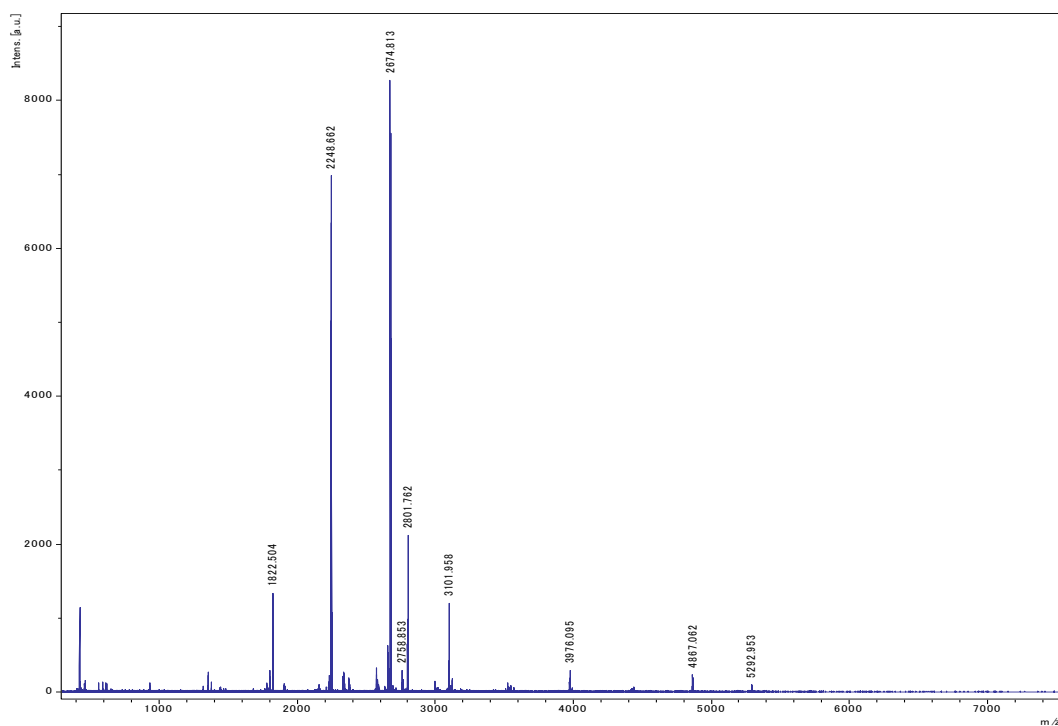


**Scheme 3-2.** Synthesis of semi-circular hexasaccharide precursor **2c**.

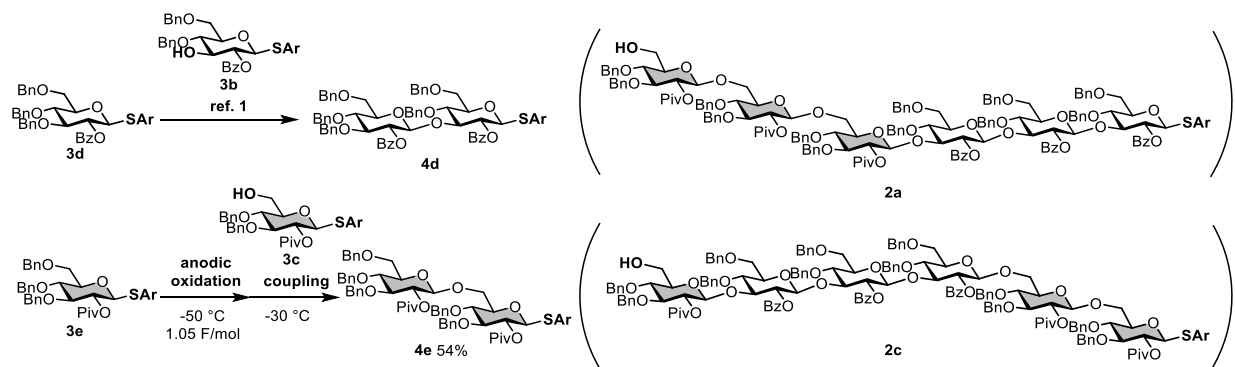
Contrary to our expectation, electrochemical oxidation of precursor **2c** afforded neither cyclic dodecasaccharide nor linear dodecasaccharide. Then we carefully checked by-products of the reaction and found formation of hydroxysugars of hexasaccharide **6** and pentasaccharide **7** (Figure 3-2). MALDI-TOF-MS showed two major molecular ion peaks  $[M+K]^+$  which are derived from hydroxysugars **6** and **7** (Figure 3-3). Although hydroxysugars are typical by-products of glycosylation, the formation of shorter oligosaccharide **7** is rare. By comparison of the structures of precursors **2a** and **2c**, glycosidic linkage of the disaccharide unit of the reducing end was supposed to be a key factor for the side reactions. For better understanding of the relationship between the disaccharide structure and its reactivity, we planned to investigate using model disaccharides **4d-e** (Scheme 3-3).  $\beta$ -1,3-Disaccharide **4d** is a model compound of hexasaccharide **2a** and  $\beta$ -1,6-disaccharide **4e** is a model compound of hexasaccharide **2c**, respectively. Disaccharide **4d** was synthesized according to the reported procedure.<sup>1</sup> Disaccharide **4e** was synthesized by glycosylation reaction of monosaccharides **3c** and **3e**.



**Figure 3-2.** By-products of electrochemical oxidation of hexasaccharide **2c**.



**Figure 3-3.** MALDI-TOF MS of crude products of electrochemical oxidation of **2c**.



**Scheme 3-3.** Synthesis of model disaccharides of the hexasaccharide precursors.

To explore information which can reveal the relationship, we focused on RRV (Relative Reactivity Value). RRV, developed by C.-H. Wong and co-workers, is an index of reactivity of glycosyl donors.<sup>2</sup> RRV is calculated based on ratio of reaction rate between a target thioglycoside and a standard thioglycoside (Figure 3-4). Competitive reaction of them with methanol gives relative reactivity of the target thioglycoside. Reactivities of multiple thioglycosides can be compared by comparing RRVs of them. We examined RRVs of model disaccharides **4d** and **4e**. RRV values of **4d** and **4e** are 108.9 and 57.7, respectively (Figure 3-5). Although RRV of **4d** was larger than that of **4e** as we expected, the difference was not enough to explain the relationship between the structure and the reactivity.

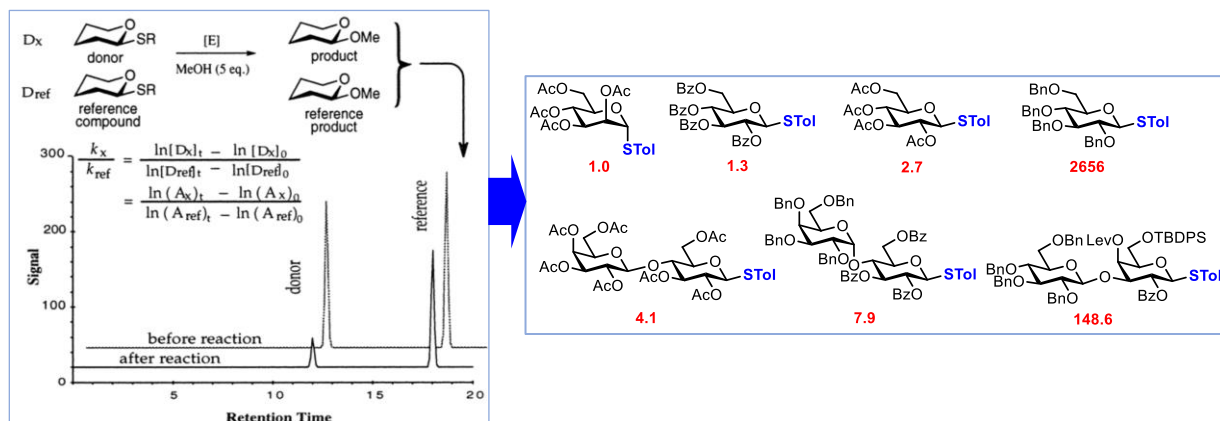


Figure 3-4. Principle of RRV.

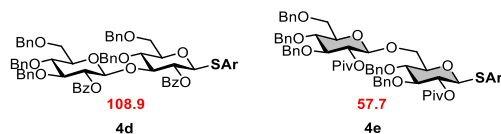
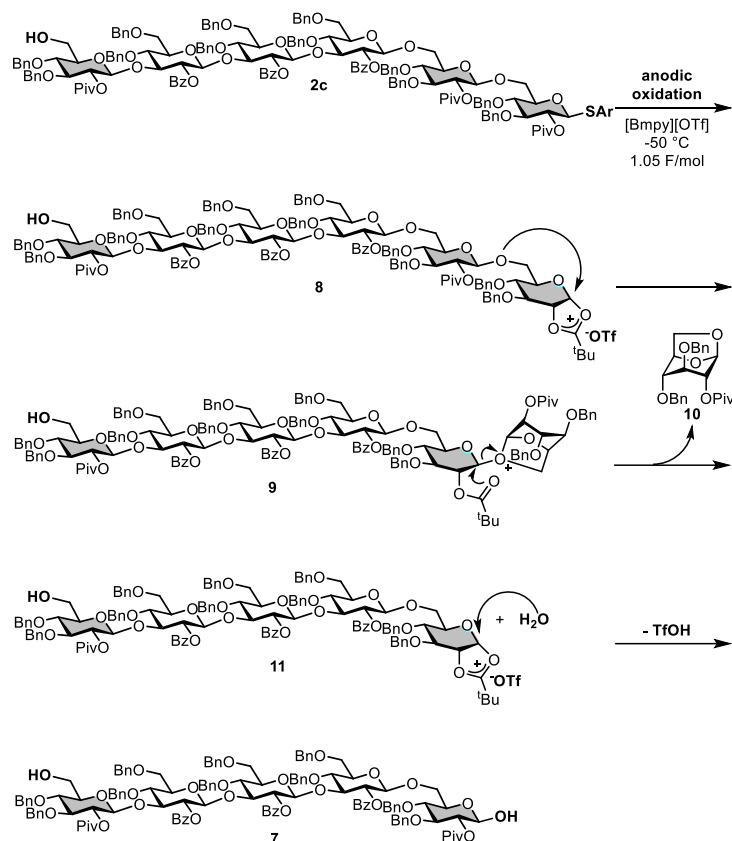


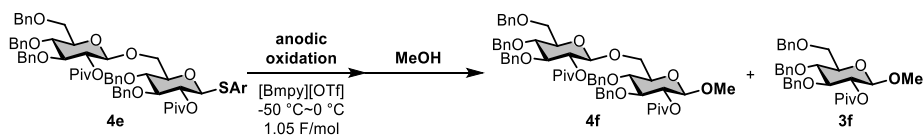
Figure 3-5. RRV values of model disaccharides **4d** and **4e**.

We turned our attention to mechanism of formation of pentasaccharide hydroxysugar **7**. Because 1,6-anhydrosugar **10** was often observed as a by-product in glycosylation reaction, we presumed that leaving of 1,6-anhydrosugar **10** caused the fragmentation. Our plausible mechanism is depicted in Figure 3-6. Electrochemical oxidation of **2c** afford the corresponding dioxalenium ion **8**. The 6-position oxygen in the reducing end attacks the anomer carbon of **8**. The following attack of 2-pivaloyl oxygen on the anomer carbon leads to leaving of 1,6-anhydrosugar **10**. Thus-obtained pentasaccharide dioxalenium ion **11** is finally hydrolysed to form pentasaccharide hydroxysugar **7**. Although 1,6-anhydrosugar **10** was not isolated from the reaction mixture, we regarded it as a potential by-product.



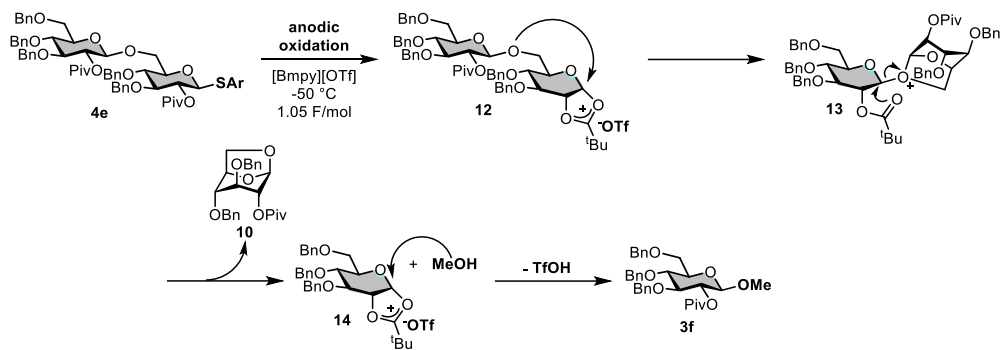
**Figure 3-6.** Plausible mechanism of the shortening reaction.

Next,  $\beta$ -1,6-disaccharide **4e** was electrochemically oxidized, followed by addition of methanol. Not only 1-methylglycoside disaccharide **4f** but also 1-methylglycoside monosaccharide **3f** was observed (Figure 3-7). In other words, the fragmentation also occurred in the reaction using disaccharide **4e**. We presumed that monosaccharide **3f** was produced by the mechanism shown in Figure 3-8 via leaving of 1,6-anhydrosugar **10**. Although 1,6-anhydrosugar **10** was not isolated, it was detected by mass analysis of the reaction mixture.



**Figure 3-7.** Model reaction using  $\beta$ -1,6-disaccharide **4e** as a glycosyl donor.





**Figure 3-8.** Plausible mechanism of formation of 1-methylglycoside monosaccharide **3f**.

We have focused on the reducing end of the hexasaccharide precursors. However, there is still room for investigation in other moieties. For example, glycosidic linkage of the disaccharide unit of the non-reducing end could affect the dimerization-cyclization and the side reactions. Deeper understanding of glycosylation of semi-circular hexasaccharides is expected to lead to rational design of oligosaccharide building blocks for the cyclic dodecasaccharide.

### Conclusion

In the course of synthetic study of protected cyclic dodecasaccharide **1**, two hexasaccharide precursors **2a** and **2c** exhibited different reactivity. Dimerization-cyclization of hexasaccharide **2c** was failed. In this reaction, hydroxysugars of hexasaccharide **6** and pentasaccharide **7** were obtained. By comparison of the structures of the precursors, glycosidic linkage of the disaccharide unit of the reducing end was supposed to be a key factor for the side reactions. In the reaction using disaccharide **4e**, which is a model compound of hexasaccharide **2c**, fragmentation was also observed. Monosaccharide **3f** and 1,6-anhydrosugar **10** were detected by mass analysis. The result was consistent with our plausible mechanism. More detailed knowledge regarding structure-reactivity relationship is expected to lead to rational design of oligosaccharide building blocks in future.

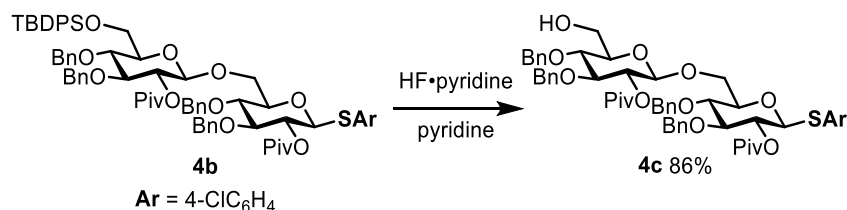
## Experimental

### 1. General

$^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were recorded on Bruker AVANCE II 600 ( $^1\text{H}$  600 MHz,  $^{13}\text{C}$  150 MHz). ESI-MS and MALDI-TOF-MS were recorded on Thermo Scientific Exactive spectrometer and Bruker ultrafleXtreme, respectively. Preparative recycling gel permeation chromatography (PR-GPC) was performed on Japan Analytical Industry LC-5060. Kanto silica gel 60 N (spherical, neutral, 63-210  $\mu\text{m}$ ) was used for silica gel column chromatography. The automated synthesizer is consisting of the commercially available instruments such as the chiller with a cooling bath (UCR-150, Techno Sigma), the power supply for constant current electrolysis (PMC 350-0.2 A, KIKUSUI), the syringe pump (PHD 2000 infusion, Harvard apparatus), and the system controller (LabVIEW, National Instruments). Merck TLC (silica gel 60 F254) was used for TLC analysis. Monosaccharide **S1**, monosaccharides **3a-d** and disaccharides **4a**, **4b**, and **4d** were prepared according to the reported procedures.<sup>1</sup> Unless otherwise noted, all materials were obtained from commercial suppliers and used without further purification.

### 2. Preparation of building blocks

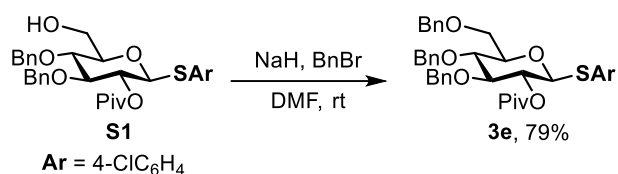
2-1. 4-Chlorophenyl 3,4-di-*O*-benzyl-2-*O*-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-*O*-benzyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**4c**)



Disaccharide **4b** (0.880 mmol, 1.09 g) was dissolved in pyridine (8.8 mL) and the solution was cooled to 0 °C. 70% HF-pyridine (0.70 mL) was added to the solution and the reaction mixture was stirred at 0 °C to room temperature for 5 h. Conversion of **4b** was confirmed by TLC (Hexane/EtOAc 7:3) and aqueous sodium bicarbonate solution was added to quench the reaction. The aqueous solution was extracted with chloroform and the combined organic layer was washed with aqueous sodium bicarbonate solution and 1 N aqueous hydrochloric acid. The reaction mixture was dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure to obtain crude product. Thus-obtained crude product was purified by silica gel chromatography to obtain **4c** (0.761 mmol, 759 mg) in 86% yield. TLC (Hexane/EtOAc 4:1) R<sub>f</sub> 0.21; [ $\alpha$ ]<sub>D</sub><sup>22</sup> = 28.181 (*c* = 1.1, CHCl<sub>3</sub>);  $^1\text{H}$  NMR (CDCl<sub>3</sub>, 600 MHz)  $\delta$  7.41 (d, *J* = 8.4 Hz, 2 H), 7.34-7.27 (m, 16 H), 7.24 (d, *J* = 7.2 Hz, 4 H), 5.06 (*pseudo-t*, *J* = 7.8 Hz, 1 H), 5.02 (*pseudo-t*, *J* = 9.6 Hz, 1 H), 4.81 (d, *J* = 9.6 Hz, 1 H), 4.77 (d, *J* = 10.4 Hz, 2 H), 4.74 (d, *J* = 9.6 Hz, 1 H), 4.70 (d, *J* = 10.4 Hz, 1 H), 4.68 (d, *J* = 10.4 Hz, 1 H), 3.91 (dd, *J* = 10.2, 1.6 Hz, 1 H), 3.82 (ddd, *J* = 16.0, 8.8, 4.0 Hz, 1 H), 3.72-3.64 (m, 5 H), 3.56-3.53 (m, 1 H), 3.47 (*Pseudo-t*, *J* = 9.6 Hz, 1 H), 3.40-3.37 (m, 1 H), 1.22 (s, 9 H), 1.12 (s, 9 H);  $^{13}\text{C}$  NMR (CDCl<sub>3</sub>, 150 MHz)  $\delta$  176.71, 138.06, 137.89, 137.82, 137.74, 134.19, 133.58, 131.83, 129.31, 128.54, 128.43, 128.13, 128.06, 128.01, 127.74, 127.69, 127.45, 127.36, 101.20, 86.73, 84.58, 83.11, 79.37, 77.58, 77.51, 75.56, 75.30, 75.05, 75.01, 74.99, 72.84, 71.36, 68.30, 62.02.

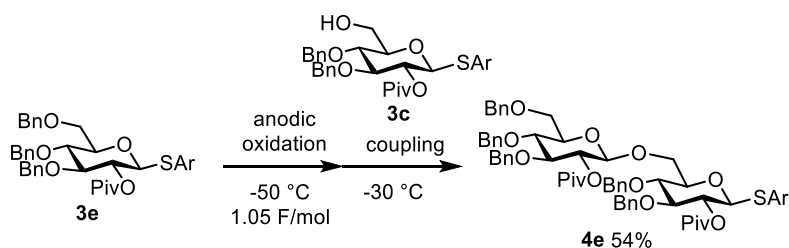
2-2. 4-Chlorophenyl 3,4,6-tri-*O*-benzyl-2-*O*-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-*O*-benzyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**4e**)

2-2-1. 4-Chlorophenyl 3,4,6-tri-*O*-benzyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**3e**)



4-Chlorophenyl 3,4-di-*O*-benzyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside **S1** (4.90 mmol, 2.80 g) was dissolved in DMF (35 mL) and the solution was cooled to 0 °C. BnBr (17.6 mmol, 2.10 g) was added to the solution, followed by addition of a solution of NaH (17.9 mmol, 0.430 g) in DMF (15 mL). The reaction mixture was stirred at 0 °C to room temperature for 5 h. Conversion of **S1** was confirmed by TLC (Hexane/EtOAc 4:1) and methanol was added to quench the reaction. The solution was diluted with hexane/EtOAc (4:1), washed with water, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure to obtain crude product. Thus-obtained crude product was purified by silica gel chromatography to obtain **3e** (3.90 mmol, 2.58 g) in 79% yield. TLC (Hexane/EtOAc 4:1) R<sub>f</sub> 0.56; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz)  $\delta$  7.44 (d, *J* = 8.4 Hz, 2 H), 7.37-7.28 (m, 11 H), 7.23 (d, *J* = 7.2 Hz, 2 H), 7.17 (d, *J* = 8.4 Hz, 4 H), 5.05 (*pseudo-t*, *J* = 9.0 Hz, 1 H), 4.76 (d, *J* = 4.8 Hz, 1 H), 4.75 (d, *J* = 4.8 Hz, 1 H), 4.68 (d, *J* = 10.8 Hz, 1 H), 4.59 (*pseudo-t*, *J* = 6.6 Hz, 1 H), 4.56 (d, *J* = 7.8 Hz, 1 H), 4.53 (d, *J* = 12.0 Hz, 2 H), 3.78 (dd, *J* = 10.8, 1.8 Hz, 1 H), 3.73-3.70 (m, 2 H), 3.66 (*pseudo-t*, *J* = 9.0 Hz, 1 H), 1.23 (s, 9 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz)  $\delta$  176.74, 138.21, 138.07, 137.98, 134.23, 134.13, 133.94, 133.77, 133.52, 132.15, 131.49, 131.33, 129.26, 129.23, 128.55, 128.53, 128.51, 128.07, 127.81, 127.79, 127.77, 127.20, 86.05, 84.74, 80.75, 77.07, 75.59, 75.16, 74.89, 73.52, 71.55, 69.01, 38.87, 38.82, 27.29, 27.17; HRMS (ESI) *m/z* calculated for C<sub>87</sub>H<sub>83</sub>ClKO<sub>18</sub>S; [M+K]<sup>+</sup> 699.1944; found 699.1951.

2-2-2. 4-Chlorophenyl 3,4,6-tri-*O*-benzyl-2-*O*-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-*O*-benzyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**4e**)



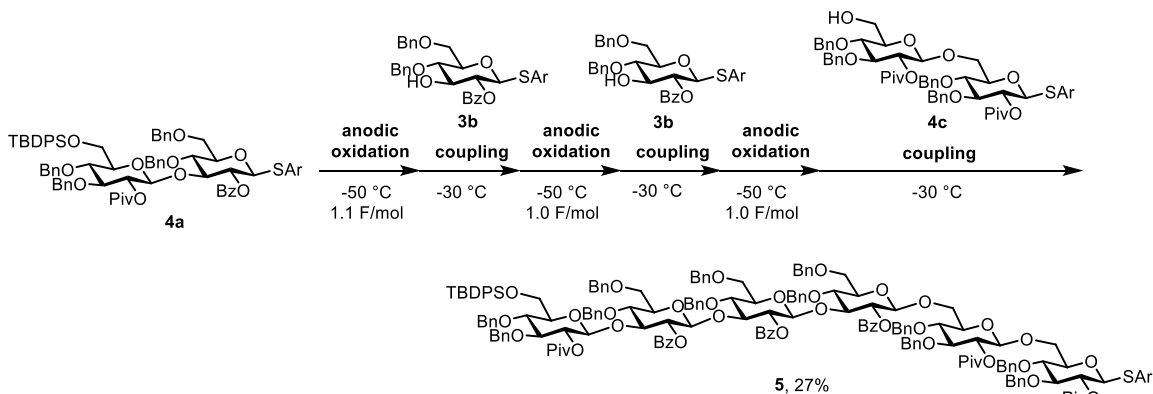
The automated synthesis of **4e** was carried out in an H-type divided cell (4G glass filter) equipped with a carbon felt anode (Nippon Carbon JF-20-P7) and a platinum plate cathode (10 mm $\times$ 20 mm). In the anodic chamber were placed terminal building block **3e** (0.60 mmol, 497 mg), Bu<sub>4</sub>NOTf (1.50 mmol, 587 mg) and CH<sub>2</sub>Cl<sub>2</sub> (15 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.63 mmol, 56  $\mu$ L), Bu<sub>4</sub>NOTf (1.50 mmol, 587 mL) and CH<sub>2</sub>Cl<sub>2</sub> (15 mL). The constant current electrolysis (13.0 mA) was carried out at -50 °C with magnetic stirring until 1.05 F/mol of electricity was consumed. After the electrolysis, building block **3c** (0.72 mmol, 411 mg) dissolved in CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL) was subsequently added by the syringe pump under an argon atmosphere at -30 °C, and kept for 60 min. After the cycle, Et<sub>3</sub>N (0.50 mL) was added to quench the mixture. After removal of the solvent under reduced pressure, the crude

product was purified with silica gel chromatography to obtain **4e** (0.52 mmol, 570 mg) in 54% yield. TLC (Hexane/EtOAc 4:1)  $R_f$  0.56;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 600 MHz)  $\delta$  7.41 (d,  $J = 8.4$  Hz, 2 H), 7.33-7.27 (m, 20 H), 7.22 (d,  $J = 7.2$  Hz, 5 H), 7.15-7.13 (m, 2 H), 5.10 (*pseudo-t*,  $J = 8.4$  Hz, 1 H), 5.02 (*pseudo-t*,  $J = 9.0$  Hz, 1 H), 4.76 (t,  $J = 14.4$  Hz, 3 H), 4.70 (*pseudo-t*,  $J = 10.8$  Hz, 1 H), 4.66 (d,  $J = 10.8$  Hz, 1 H), 4.49-4.57 (m, 5 H), 4.46 (*pseudo-t*,  $J = 7.8$  Hz, 1 H), 4.00 (dd,  $J = 10.4, 1.8$  Hz, 1 H), 3.75-3.66 (m, 6 H), 3.60 (td,  $J = 7.2, 1.8$  Hz, 1 H), 3.50-3.43 (m, 2 H), 1.21 (s, 9 H), 1.12 (s, 9 H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 150 MHz)  $\delta$  176.69, 176.65, 138.15, 138.11, 137.98, 137.98, 137.78, 134.09, 133.64, 131.86, 129.28, 128.48, 128.45, 128.42, 128.09, 128.02, 127.85, 127.83, 127.70, 127.65, 127.46, 127.32, 101.22, 86.55, 84.57, 83.36, 79.28, 77.77, 77.10, 76.89, 75.27, 75.21, 75.02, 74.99, 73.52, 72.86, 71.39, 68.76, 68.08, 38.81, 38.77, 27.20, 27.16; HRMS (ESI)  $m/z$  calculated for  $\text{C}_{87}\text{H}_{83}\text{ClKO}_{18}\text{S}$ ;  $[\text{M}+\text{K}]^+$  1125.3986; found 1125.3959.

### 3. Synthesis of hexasaccharide precursor **2c** by AEA

3-1. 4-Chlorophenyl 3,4-di-*O*-benzyl-2-*O*-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-*O*-benzoyl-4,6-di-*O*-benzyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-*O*-benzoyl-4,6-di-*O*-benzyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-*O*-benzoyl-4,6-di-*O*-benzyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-*O*-benzyl-2-*O*-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-*O*-benzyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**2c**)

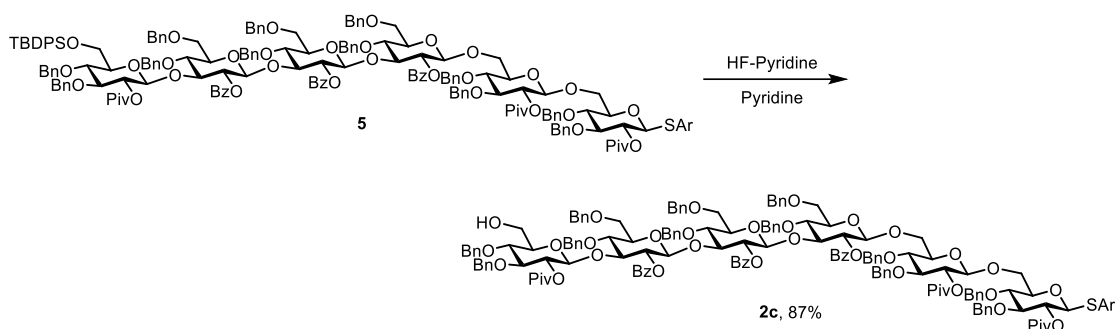
3-1-1. 4-Chlorophenyl 3,4-di-*O*-benzyl-6-*O*-*tert*-butyldiphenylsilyl-2-*O*-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-*O*-benzoyl-4,6-di-*O*-benzyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-2-*O*-benzoyl-4,6-di-*O*-benzyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-*O*-benzyl-2-*O*-pivaloyl- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-3,4-di-*O*-benzyl-2-*O*-pivaloyl-1-thio- $\beta$ -D-glucopyranoside (**5**)



The automated synthesis of **5** was carried out in an H-type divided cell (4G glass filter) equipped with a carbon felt anode (Nippon Carbon JF-20-P7) and a platinum plate cathode (10 mm $\times$ 20 mm). In the anodic chamber were placed terminal building block **4a** (0.30 mmol, 377 mg), [Bmpy][OTf] (1.50 mmol, 0.35 mL) and  $\text{CH}_2\text{Cl}_2$  (15 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.32 mmol, 28  $\mu\text{L}$ ), [Bmpy][OTf] (1.50 mmol, 0.35 mL) and  $\text{CH}_2\text{Cl}_2$  (15 mL). The constant current electrolysis (12.0 mA) was carried out at  $-50$   $^\circ\text{C}$  with magnetic stirring until 1.1 F/mol of electricity was consumed. After the electrolysis, building block **3b** (0.30 mmol, 178 mg) dissolved in  $\text{CH}_2\text{Cl}_2$  (2.0 mL) was subsequently added by the syringe pump under an argon atmosphere at  $-30$   $^\circ\text{C}$ , and kept for 60 min. After cooling to  $-50$   $^\circ\text{C}$ , The constant current electrolysis (12.0 mA) was carried out at  $-50$   $^\circ\text{C}$  with magnetic stirring until 1.1 F/mol of electricity was consumed. After the electrolysis, building block **3b** (0.30 mmol, 178 mg) dissolved in  $\text{CH}_2\text{Cl}_2$  (2.0 mL) was subsequently added by the syringe pump under an argon atmosphere at  $-30$   $^\circ\text{C}$ , and kept for 60 min. After cooling to  $-50$   $^\circ\text{C}$ , The constant current electrolysis (12.0 mA) was carried out at  $-50$   $^\circ\text{C}$  with magnetic stirring until 1.1 F/mol of electricity was consumed. After the electrolysis, building

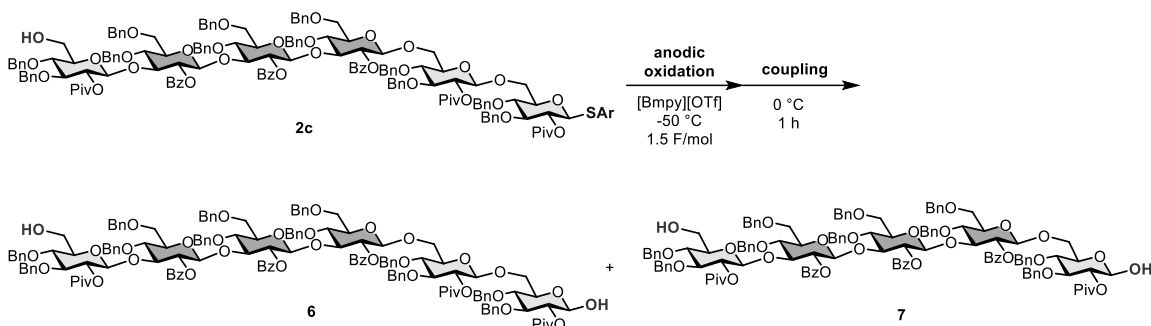
block **4c** (0.30 mmol, 300 mg) dissolved in CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL) was subsequently added by the syringe pump under an argon atmosphere at -30 °C, and kept for 60 min. After the cycle, Et<sub>3</sub>N (0.50 mL) was added, and the mixture was filtered through a short column (4×3 cm) of silica gel to remove [Bmpy][OTf]. After removal of the solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain 486 mg of a mixture containing **5**. The mixture was purified by PR-GPC to obtain **5** (0.0822 mmol, 247 mg) in 27% yield.

3-1-2. 4-Chlorophenyl 3,4-di-*O*-benzyl-2-*O*-pivaloyl-β-D-glucopyranosyl-(1→3)-2-*O*-benzoyl-4,6-di-*O*-benzyl-β-D-glucopyranosyl-(1→3)-2-*O*-benzoyl-4,6-di-*O*-benzyl-β-D-glucopyranosyl-(1→6)-3,4-di-*O*-benzyl-2-*O*-pivaloyl-β-D-glucopyranosyl-(1→6)-3,4-di-*O*-benzyl-2-*O*-pivaloyl-1-thio-β-D-glucopyranoside (**2c**)



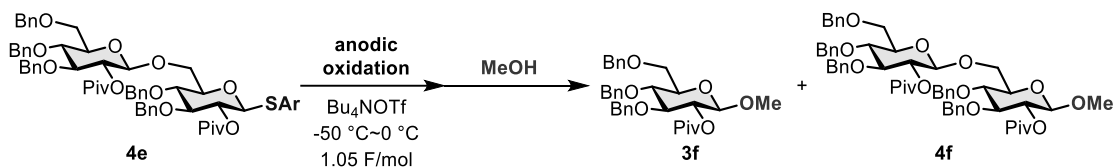
Hexasaccharide **5** (0.121 mmol, 365 mg) was dissolved in pyridine (1.0 mL) and the solution was cooled to 0 °C. 70% HF-pyridine (0.10 mL) was added to the solution and the reaction mixture was stirred at 0 °C to room temperature for 5 h. Conversion of **5** was confirmed by TLC (Hexane/EtOAc 7:3) and aqueous sodium bicarbonate solution was added to quench the reaction. The aqueous solution was extracted with chloroform and the combined organic layer was washed with aqueous sodium bicarbonate solution and 1 N aqueous hydrochloric acid. The reaction mixture was dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure to obtain crude product. Thus-obtained crude product was purified by silica gel chromatography to obtain **2c** (0.105 mmol, 289 mg) in 87% yield. TLC (Hexane/EtOAc 7:3) R<sub>f</sub>: 0.65; [α]<sub>D</sub><sup>22</sup> = -16.363 (*c* = 1.1, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz) δ 7.78 (d, *J* = 7.2 Hz, 2 H), 7.66 (d, *J* = 7.2 Hz, 2 H), 7.58-7.55 (m, 4 H), 7.31-7.27 (m, 7 H), 7.25-7.13 (m, 46 H), 7.09-6.94 (m, 2 H), 5.00 (dd, *J* = 19.5, 19.5, 9.0 Hz, 4 H), 4.95-4.92 (m, 3 H), 4.87 (dd, *J* = 9.0, 7.8 Hz, 1 H), 4.80 (d, *J* = 10.8 Hz, 1H), 4.68 (ddd, *J* = 8.7, 8.7, 2.4 Hz, 3 H), 4.61 (dd, *J* = 10.8, 4.8 Hz, 3 H), 4.58 (d, *J* = 13.8 Hz, 2 H), 4.52 (dd, *J* = 10.2, 6.0 Hz, 3 H), 4.45-4.35 (m, 12 H), 4.31 (*pseudo-t*, *J* = 12.0 Hz, 2 H), 4.26 (d, *J* = 7.8 Hz, 1 H), 4.21 (dd, *J* = 23.4, 10.8 Hz, 2 H), 4.07 (*pseudo-t*, 18 Hz, 1 H), 4.00 (d, *J* = 9.6 Hz, 1 H), 3.88 (*pseudo-t*, *J* = 16.8 Hz, 1 H), 3.81 (dd, *J* = 18.0, 9.0 Hz, 2 H), 3.72 (d, *J* = 10.2 Hz, 1 H), 3.63-3.57 (m, 4 H), 3.54-3.47 (m, 7 H), 3.45-3.38 (m, 6 H), 3.36 (m, 1 H), 3.32 (*pseudo-t*, *J* = 18.0 Hz, 1 H), 3.28-3.23 (m, 4 H), 2.95 (m, 1 H), 1.21 (s, 9 H), 1.06 (s, 9 H), 1.01 (s, 9 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz) δ 177.18, 176.57, 176.48, 164.52, 138.61, 138.38, 138.35, 138.24, 138.15, 138.00, 137.89, 137.80, 137.72, 137.57, 133.79, 129.83, 129.75, 129.64, 129.18, 128.48, 128.41, 128.39, 128.31, 128.28, 128.26, 128.19, 128.17, 128.13, 128.12, 128.09, 128.03, 127.99, 127.86, 127.80, 127.73, 127.68, 127.63, 127.53, 127.50, 127.47, 127.38, 127.29, 127.19, 127.16, 100.37, 84.41, 82.64, 77.51, 77.49, 75.23, 75.06, 75.01, 74.96, 74.80, 74.78, 74.74, 73.37, 73.34, 73.33, 38.71, 38.64, 38.60, 27.13, 27.05, 26.94.

#### 4. Electrochemical oxidation of hexasaccharide precursor **2c**



The one-pot synthesis of **2c** was carried out in an H-type divided cell (4G glass filter) equipped with a carbon felt anode (Nippon Carbon JF-20-P7) and a platinum plate cathode (10 mm×10 mm). In the anodic chamber were placed hexasaccharide building block **2c** (0.10 mmol, 277 mg), [Bmpy][OTf] (0.50 mmol, 117  $\mu$ L) and  $\text{CH}_2\text{Cl}_2$  (5.0 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.10 mmol, 8  $\mu$ L), [Bmpy][OTf] (0.50 mmol, 117  $\mu$ L) and  $\text{CH}_2\text{Cl}_2$  (5.0 mL). The constant current electrolysis (3.0 mA) was carried out at  $-50\text{ }^\circ\text{C}$  with magnetic stirring until 1.5 F/mol of electricity was consumed. After the electrolysis, the reaction temperature was raised to  $0\text{ }^\circ\text{C}$ , and kept for 60 min. Then,  $\text{Et}_3\text{N}$  (0.20 mL) was added to quench the reaction. After removal of the solvent under reduced pressure, electrolyte was removed by silica gel short pad (eluent: hexane/EtOAc 1:1). Crude product (333 mg) was analyzed by MALDI-TOF-MS and formation of hydroxy sugars **6** and **7** was confirmed.

#### 5. Electrochemical glycosylation of model disaccharide **4e** with methanol



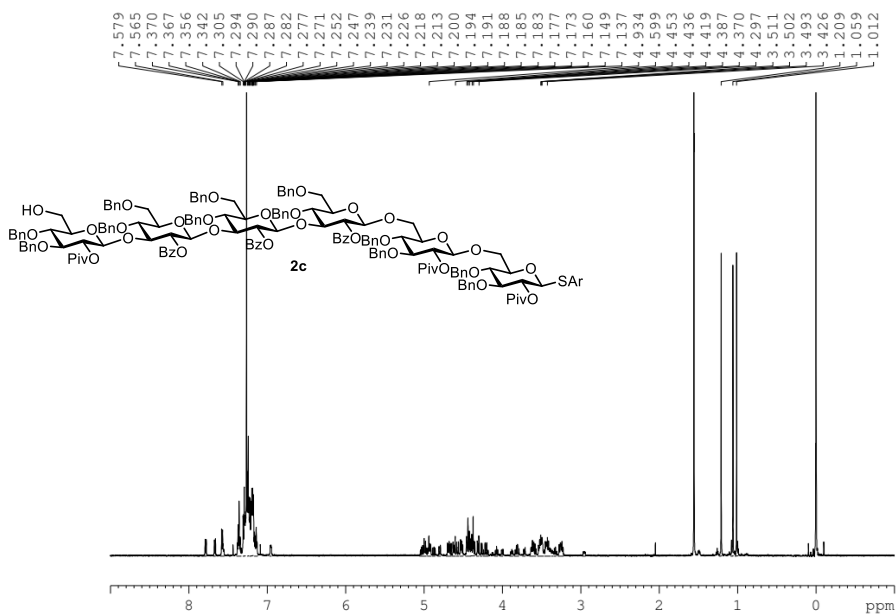
Electrochemical activation of **4e** was carried out in an H-type divided cell (4G glass filter) equipped with a carbon felt anode (Nippon Carbon JF-20-P7) and a platinum plate cathode (10 mm×10 mm). In the anodic chamber were placed disaccharide building block **4e** (0.10 mmol, 109 mg),  $\text{Bu}_4\text{NOTf}$  (0.50 mmol, 210 mg) and  $\text{CH}_2\text{Cl}_2$  (5.0 mL). In the cathodic chamber were placed trifluoromethanesulfonic acid (0.11 mmol, 10  $\mu$ L),  $\text{Bu}_4\text{NOTf}$  (0.50 mmol, 203 mL) and  $\text{CH}_2\text{Cl}_2$  (5.0 mL). The constant current electrolysis (4.0 mA) was carried out at  $-50\text{ }^\circ\text{C}$  with magnetic stirring until 1.05 F/mol of electricity was consumed. Then the reaction temperature was raised to  $0\text{ }^\circ\text{C}$  and kept for 1 h. Methanol (1.0 mmol, 40  $\mu$ L) was added at  $0\text{ }^\circ\text{C}$  and kept for another 1 h. After the cycle,  $\text{Et}_3\text{N}$  (0.10 mL) was added to quench the mixture. After removal of the solvent under reduced pressure, the crude product was purified with silica gel chromatography to obtain methyl glycoside **3f**<sup>3</sup> and disaccharide **4f** have been detected by MALDI-TOF-MS.

## References

- [1] A. Shibuya, Y. Ishisaka, A. Saito, M. Kato, S. Manmode, H. Komatsu, M. A. Rahman, N. Sasaki, T. Itoh, T. Nokami, *Faraday Discuss.*, **2023**, *247*, 59-69.
- [2] Z. Zhang, I. R. Ollmann, X.-S. Ye, R. Wischnat, T. Baasov, C.-H. Wong, *J. Am. Chem. Soc.*, **1999**, *121*, 734-753.
- [3] J. A. Turner, N. Rosano, D. J. Gorelik, M. S. Taylor, *ACS Catal.* **2021**, *11*, 11171-11179.

# $^1\text{H}$ , $^{13}\text{C}$ NMR, H-H cosy and HMQC spectra

## $^1\text{H}$ NMR

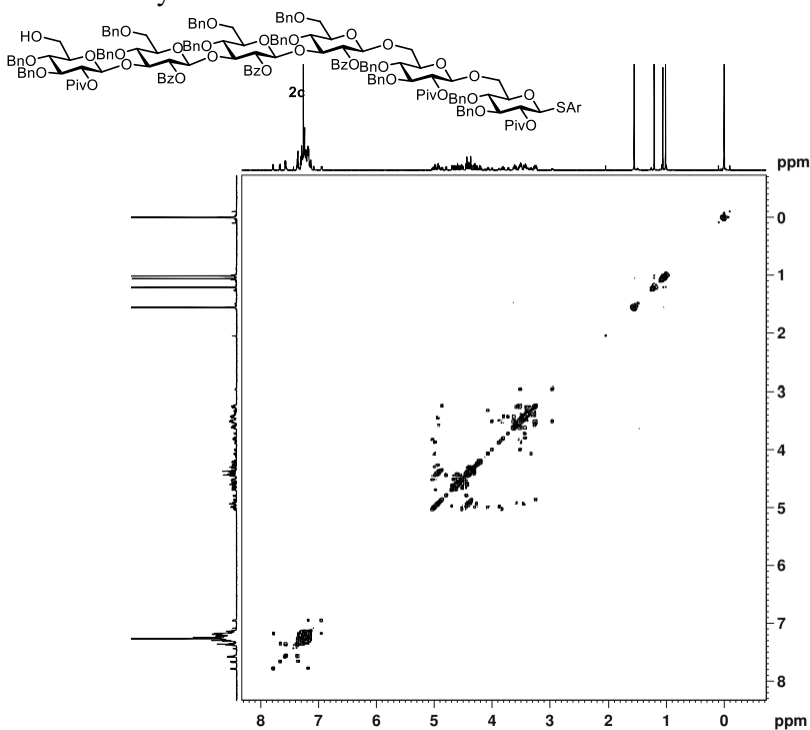


```

NAME ASU-256 fr.31-55
EXPNO 10
PROCNO 1
Date_ 20220112
Time 18.38
INSTRUM spect
PROBHD 5 mm PABBO BB-
PULPROG zg30
TD 65536
SOLVENT CDCl3
NS 16
DS 2
SWH 12335.526 Hz
FIDRES 0.188225 Hz
AQ 2.6564426 sec
RG 203
DW 40.533 usec
DE 6.50 usec
TE 294.0 K
D1 1.0000000 sec
TD0 1

===== CHANNEL f1 =====
NUC1 1H
P1 13.00 usec
PL1 -2.00 dB
PL1W 18.91009140 W
SFO1 600.137060 MHz
SI 32768
SF 600.1300155 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00
    
```

## H-H cosy



```

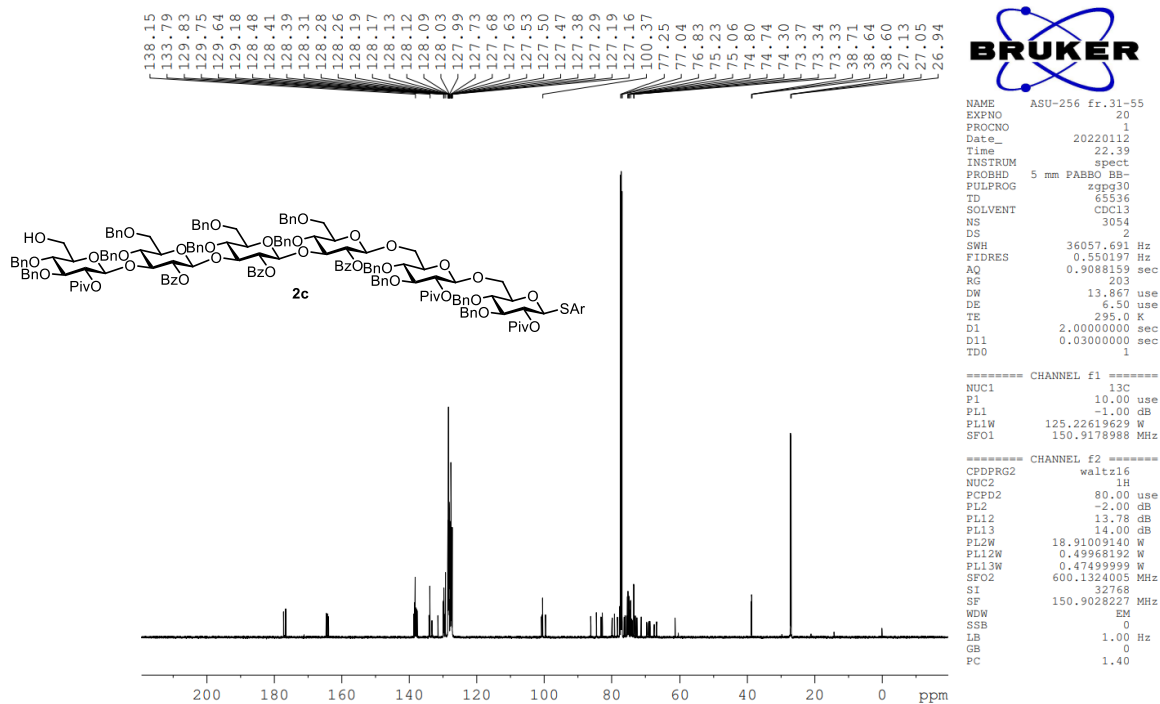
NAME ASU-256 fr.31-55
EXPNO 12
PROCNO 1
Date_ 20220112
Time 18.41
INSTRUM spect
PROBHD 5 mm PABBO BB-
PULPROG cosyppgf
TD 2048
SOLVENT CDCl3
NS 1
DS 8
SWH 5434.783 Hz
FIDRES 2.653702 Hz
AQ 0.188460 sec
RG 203
DW 92.000 usec
DE 6.50 usec
TE 294.0 K
DO 0.00000300 sec
D1 1.42627096 sec
D13 0.00000400 sec
D16 0.00018000 sec
INO 0.00018400 sec

===== CHANNEL f1 =====
NUC1 1H
P1 13.00 usec
PL1 13.00 usec
PL1 -2.00 dB
PL1W 18.91009140 W
SFO1 600.132296 MHz

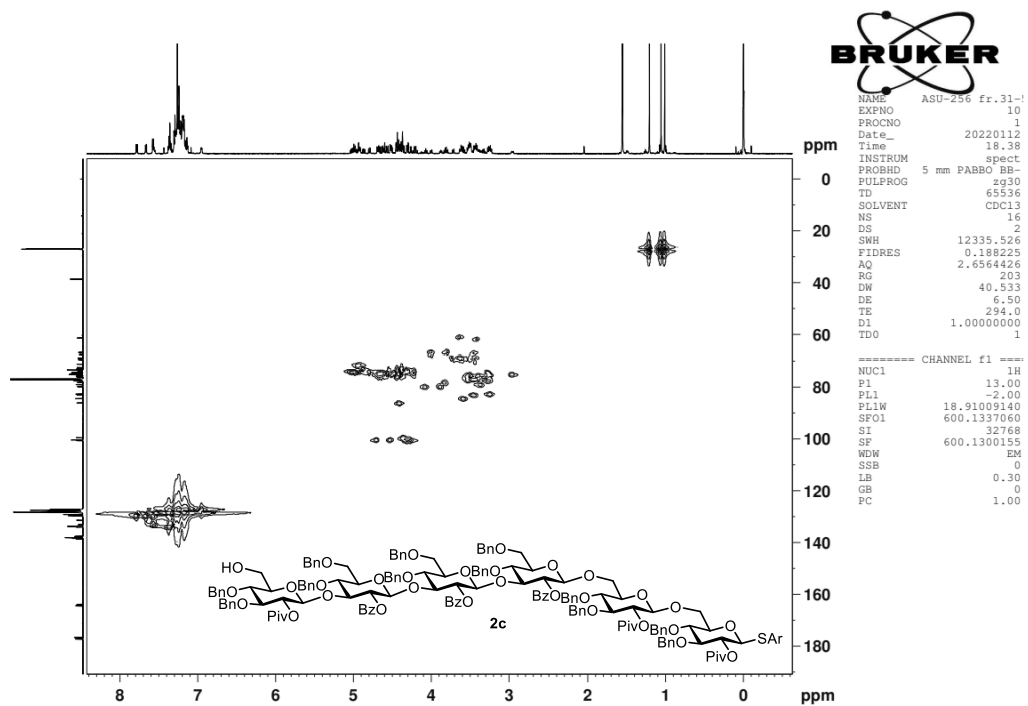
===== GRADIENT CHANNEL =====
GPNAM1 SINE.100
GPZ1 10.00 %
P16 1000.00 usec
ND0 1
TD 128
SFO1 600.1323 MHz
FIDRES 42.459240 Hz
SW 9.056 ppm
FWDWDE QF
SI 1024
SF 600.1300155 MHz
WDW SINE
SSB 0
LB 0.00 Hz
PC 1.40
SI 1024
MC2 QF
SF 600.1300155 MHz
WDW SINE
SSB 0
LB 0.00 Hz
GB 0
    
```

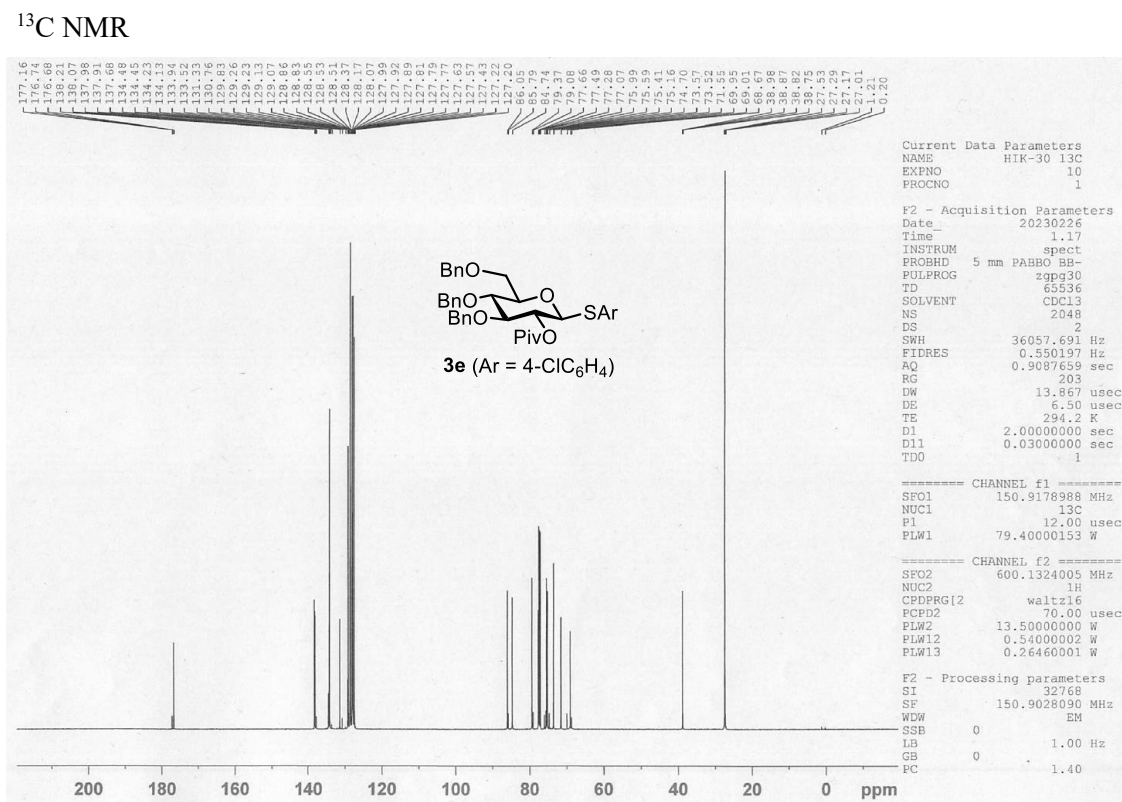
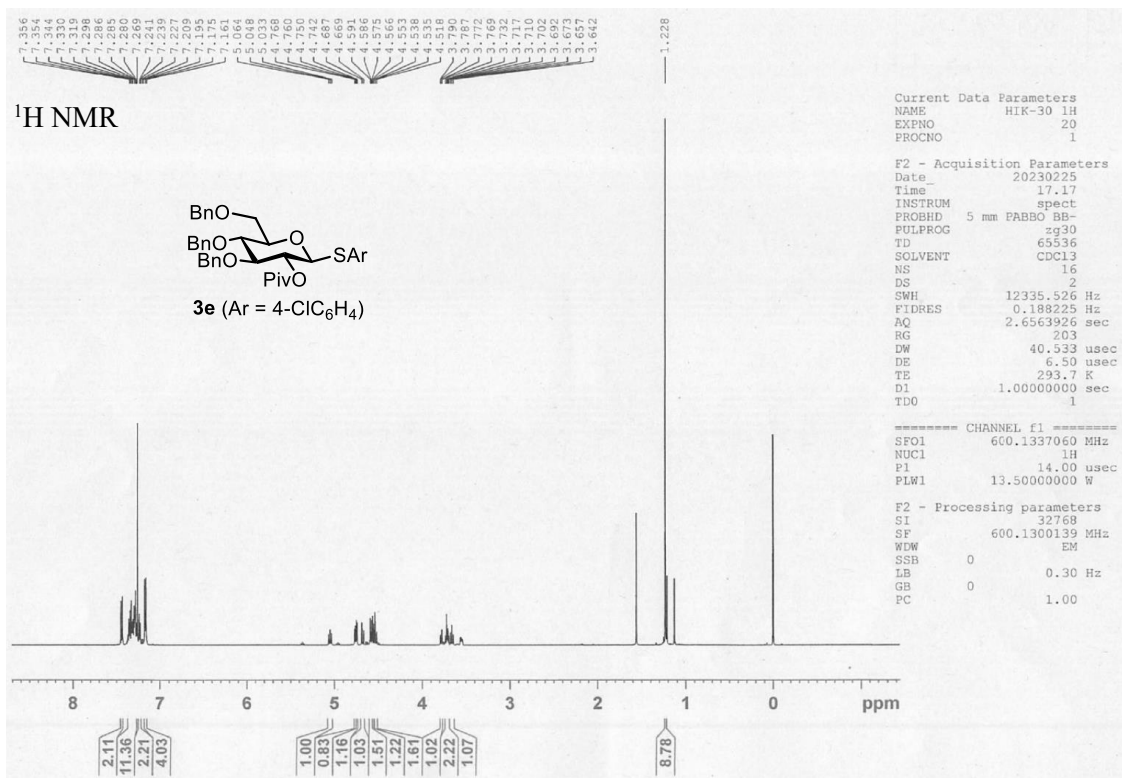


<sup>13</sup>C NMR

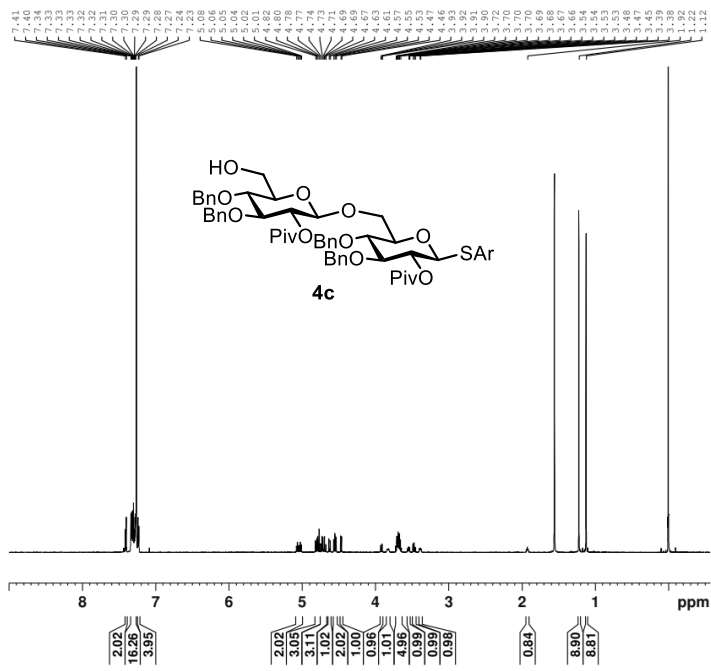


HMQC





<sup>1</sup>H NMR



```

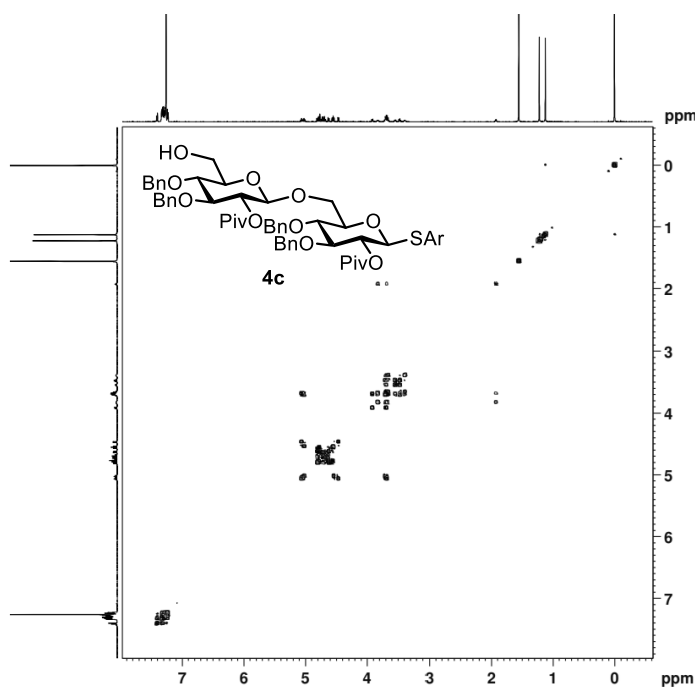
Current Data Parameters
NAME ASU b-1,6-disaccharide
EXPNO 10
PROCNO 1

F2 - Acquisition Parameters
Date_ 20220316
Time 6.40
INSTRUM spect
PROBHD 5 mm PABBO BB-
PULPROG zg30
TD 65536
SOLVENT CDCl3
NS 16
DS 2
SWH 12335.526 Hz
FIDRES 0.188225 Hz
AQ 2.6563926 sec
RG 203
DW 40.533 usec
DE 6.50 usec
TE 296.4 K
D1 1.00000000 sec
TD0 1

===== CHANNEL f1 =====
SFO1 600.1337060 MHz
NUC1 1H
P1 14.00 usec
PLM1 13.50000000 W

F2 - Processing parameters
SI 32768
SF 600.1301123 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00
    
```

H-H cosy



```

Current Data Parameters
NAME ASU b-1,6-disaccharide
EXPNO 10
PROCNO 1

F2 - Acquisition Parameters
Date_ 20220316
Time 7.58
INSTRUM spect
PROBHD 5 mm PABBO BB-
PULPROG cosypppqr
TD 2048
SOLVENT CDCl3
NS 1
DS 2
SWH 5154.639 Hz
FIDRES 2.516914 Hz
AQ 0.198566 sec
RG 203
DW 97.000 usec
DE 6.50 usec
TE 296.4 K
D0 0.00000000 sec
D1 1.99199000 sec
D11 0.04000000 sec
D12 0.00000000 sec
D13 0.00000400 sec
D16 0.00000000 sec
D18 0.00013400 sec

===== CHANNEL f1 =====
SFO1 600.1321198 MHz
NUC1 1H
P0 14.00 usec
P1 14.00 usec
P17 2500.00 usec
PLM1 13.50000000 W
PLM10 4.23360014 W

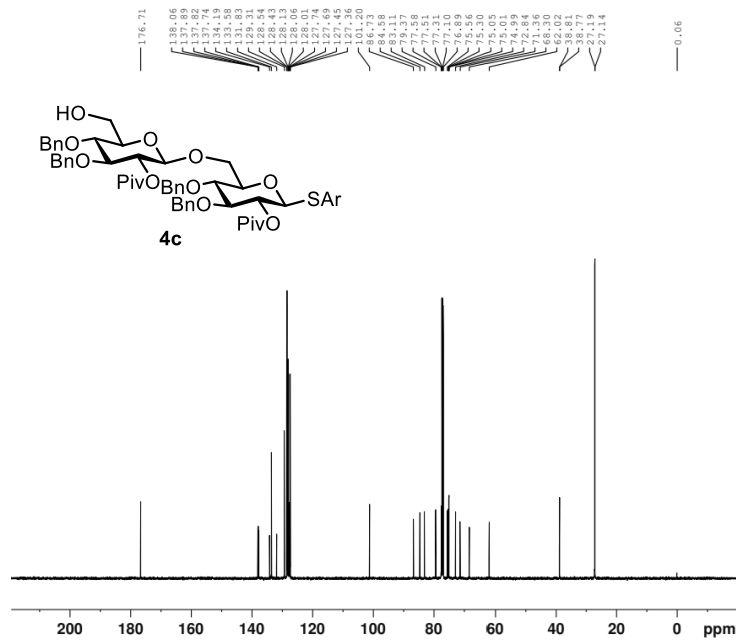
===== GRADIENT CHANNEL =====
GPRAM[1] SMSQ10.100
GR21 10.00 %
P16 1000.00 usec

F1 - Acquisition Parameters
TD 128
SFO1 600.1322 MHz
FIDRES 40.270618 Hz
SFO2 8.589 MHz
F1MODE QF

F2 - Processing parameters
SI 32768
SF 600.1301123 MHz
WDW Q9INE
SSB 0
LB 0 Hz
GB 0
PC 1.40

F1 - Processing parameters
SI 32768
SF 600.1301123 MHz
WDW Q9INE
SSB 0
LB 0 Hz
GB 0
PC 0
    
```

<sup>13</sup>C NMR



Current Data Parameters  
 NAME ASU b-1,6-diaccharide  
 EXPNO 20  
 PROCNO 1

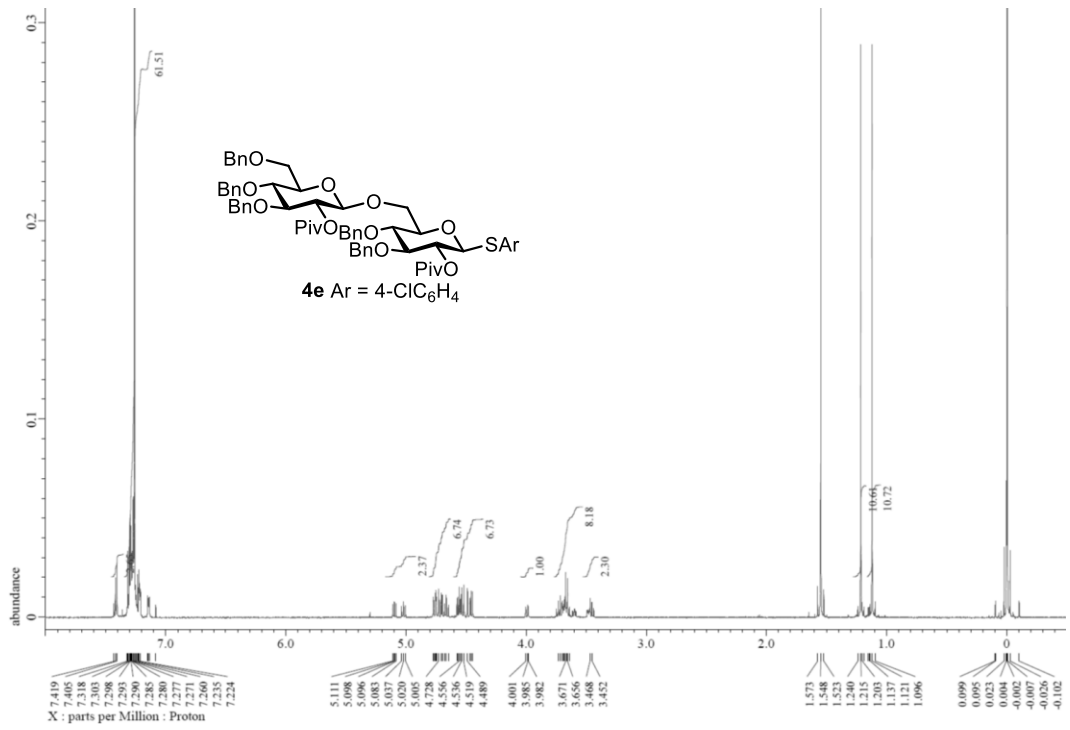
F2 - Acquisition Parameters  
 Date\_ 20220316  
 Time 7:33  
 INSTRUM spect  
 PROBHD 5 mm FAPBO BB-  
 PULPROG zgpg30  
 TD 65536  
 SOLVENT CDCl3  
 NS 883  
 DS 2  
 SWH 36057.691 Hz  
 FIDRES 0.550197 Hz  
 AQ 0.9087659 sec  
 RG 203  
 DW 13.867 usec  
 DE 6.50 usec  
 TE 297.4 K  
 D1 2.00000000 sec  
 D11 0.03000000 sec  
 TD0 1

==== CHANNEL f1 =====  
 SF01 150.9178988 MHz  
 NUC1 13C  
 P1 12.00 usec  
 PLM1 79.40000153 W

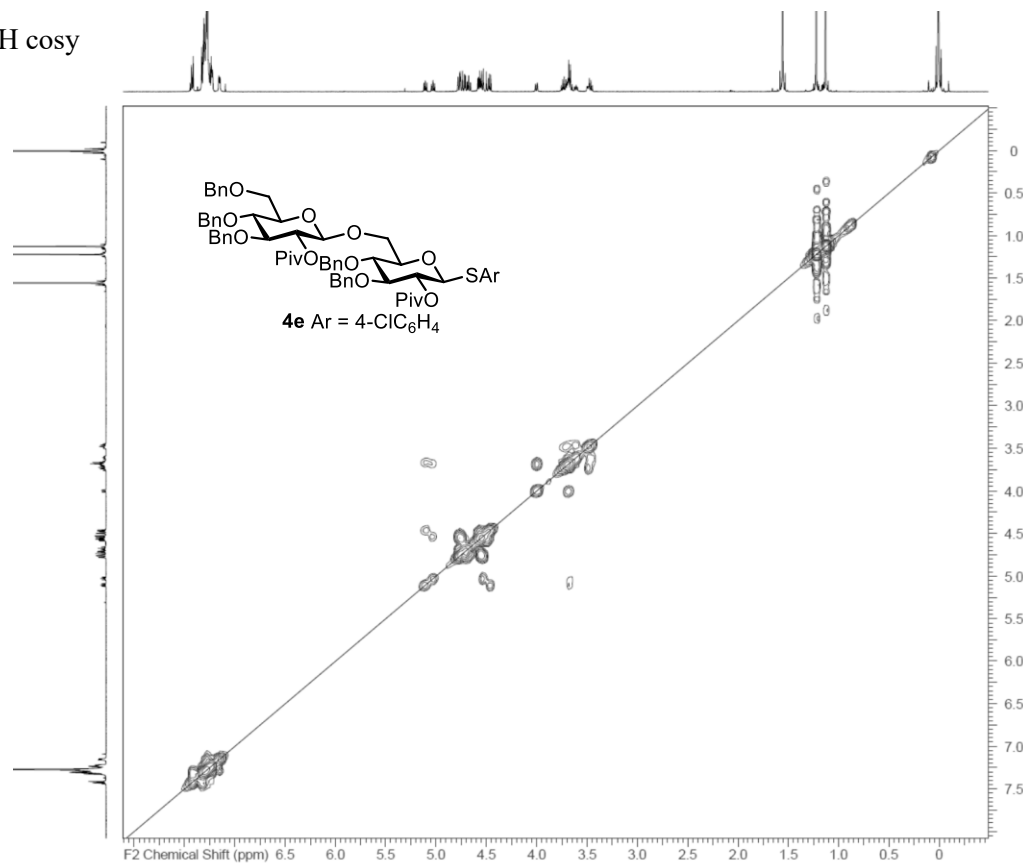
==== CHANNEL f2 =====  
 SF02 600.1324005 MHz  
 NUC2 1H  
 CPOPRG[2] waltz16  
 FCPD2 70.00 usec  
 PLM2 13.50000000 W  
 PLM12 0.54000002 W  
 PLM13 0.27162001 W

F2 - Processing parameters  
 SI 32768  
 SF 150.9028090 MHz  
 MW DM  
 SSB 0  
 LB 1.00 Hz  
 GB 0  
 PC 1.40

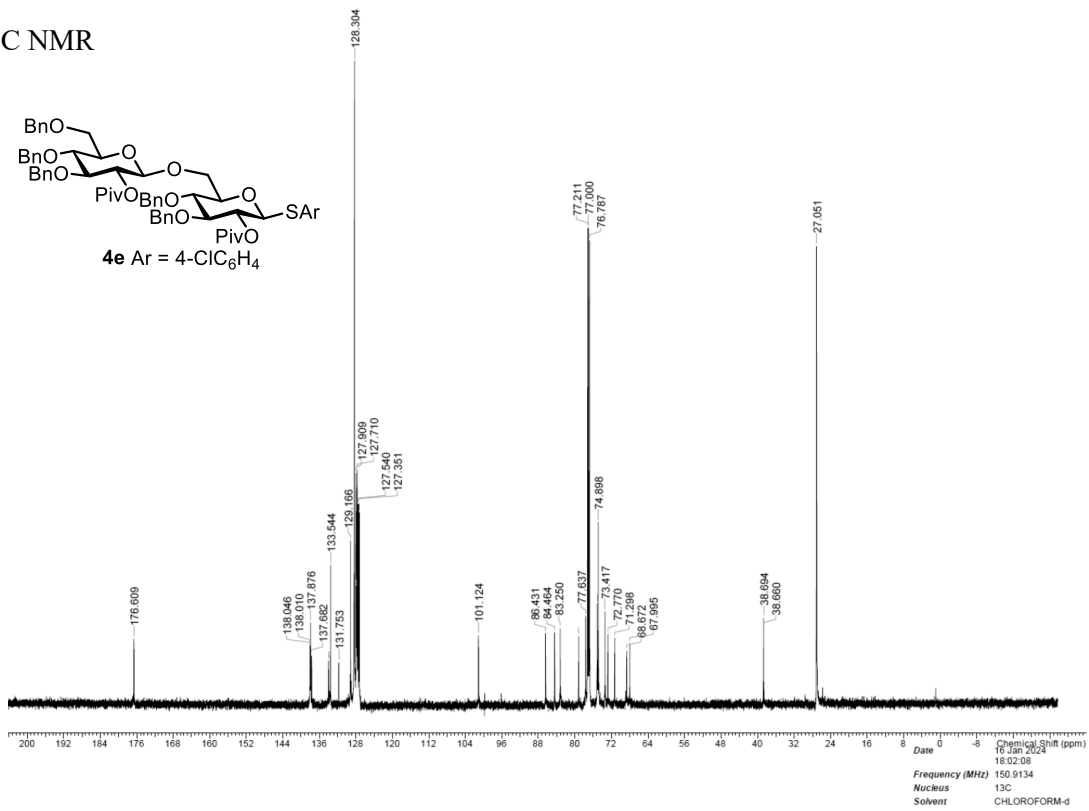
<sup>1</sup>H NMR



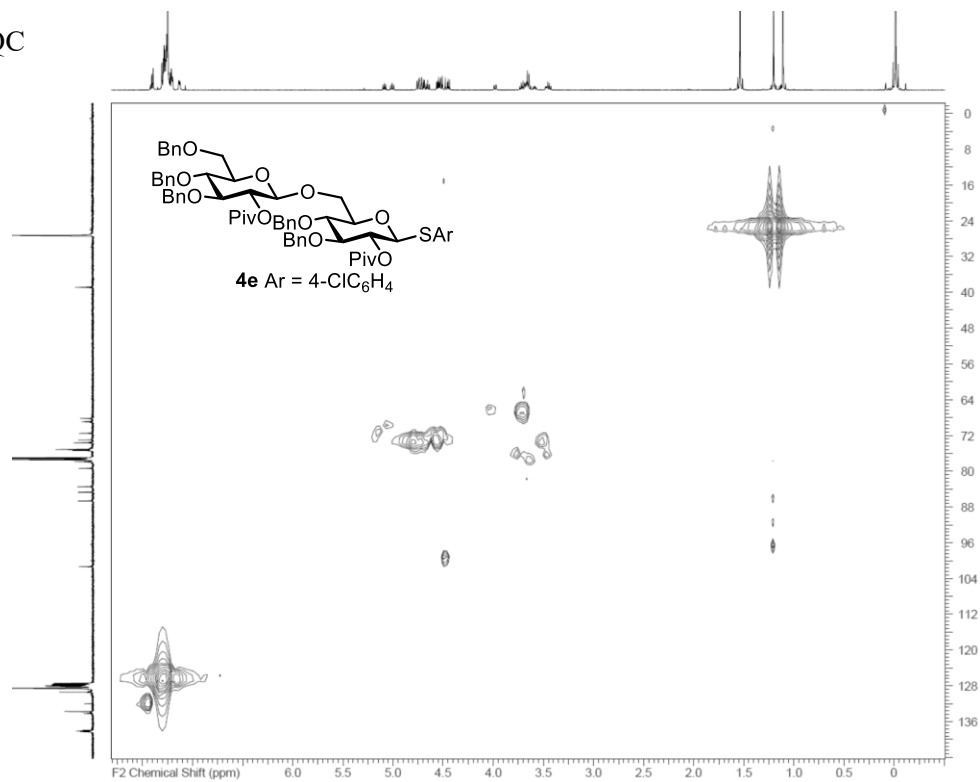
H-H cosy



<sup>13</sup>C NMR



HMQC



## Summary

We have been interested in electrochemical glycosylation, which can computationally control activation of thioglycosides, and developed ‘automated electrochemical assembly’ (AEA). A variety of oligosaccharides have been synthesized with AEA method. In this study, we then focused on a cyclic (1,3;1,6)- $\beta$ -glucan, which is produced by root-nodulating bacteria *Bradyrhizobium japonicum* MTCC 120.

In Chapter 1, for efficient glycosylation, we analyzed the reactive intermediates by VT-NMR. As a result, the glycosyl dioxalenium ion intermediates were identified and their thermal stability was evaluated. Based on the thus-obtained knowledge, reaction conditions were optimized and a  $\beta$ -(1,6)-glucan trisaccharide was synthesized in good yield.

In Chapter 2, our target compound had a symmetric dodecasaccharide structure that consists of  $\beta$ -(1,3)- and  $\beta$ -(1,6)-glycosidic linkages. As a result of our investigation, it was found that one of the semi-circular hexasaccharide precursors can be dimerized and cyclized in one-pot manner to form the cyclic dodecasaccharide. Moreover, the yield can be increased by separation of dimerization and cyclization.

In Chapter 3, in the course of synthetic study of the cyclic dodecasaccharide, it was found that side reactions vary depending on monosaccharide sequences of the hexasaccharide precursors. In particular, the glycosidic linkage of the disaccharide unit of the reducing end has great effect on reactivity. Therefore, we prepared model disaccharides of the hexasaccharides. One of the hexasaccharides and its model disaccharide caused similar fragmentation reaction. More detailed knowledge regarding structure-reactivity relationship is expected to lead to rational design of oligosaccharide building blocks in future.

## Acknowledgement

The studies presented in this thesis have been carried out under the direction of Professor Toshiki Nokami at Graduate School of Tottori University.

The author would like to express my sincerest gratitude to Professor Nokami for his kind guidance and fruitful discussions throughout this work. Professor Nokami not only directed the research in doctoral course at Tottori University but also led my training as a chemist in master's course at Kyoto University. Since the author ordinarily works at the company far from the university, the author has received a lot of special support from Professor Nokami.

The author wishes to thank Dr. Sasaki for his support. Dr. Sasaki checked his papers and led students who experimented and collected the data. The author also wishes to thank to Ms. Moeko Kato, Ms. Asuka Saito, Ms. Yui Ishisaka, Mr. Hiroto Komatsu and Ms. Yu-Cong Sun. The great data in this thesis was collected by their hard work. As the author ordinarily works as a corporate researcher, it is valuable to talk with younger students and enjoy the academic atmosphere. Communication with younger researchers motivates an older researcher.

The author must make special mention of Yoshida Laboratory at Kyoto University. The author learned the knowledge and the techniques of electrochemical glycosylation and low-temperature NMR analysis under the direction of the late Professor Jun-ichi Yoshida, Professor Seiji Suga (Okayama University), Professor Toshiki Nokami and Professor Aiichiro Nagaki (Hokkaido University). The VT-NMR experiment shown in this thesis was carried out at Kyoto University with cooperation with Professor Nagaki.

The author heartily thanks to his colleagues in Yoshida Laboratory at Kyoto University. The author learned a lot of things by spending time with Dr. Yutaka Tomida, Mr. Koji Ueoka, Mr. Ikuo Shimizu, Mr. Hirotsugu Usutani, Mr. Kazuya Soga, Mr. Hiroaki Tsuyama, Mr. Kosuke Ohata, Mr. Masafumi Inoue and all other members of Yoshida Laboratory.

The author is thankful to Cardurion Pharmaceuticals Inc. for financial support. All members of the company understood his situation and kindly allowed him to spend his time for the activities of the university.

Finally, the author would like to express deep appreciation to his wife Miho Shibuya and his parents Noriaki Shibuya and Hiromi Shibuya for their constant assistance and encouragement.



## List of publications

Title: Electrochemical synthesis of the protected cyclic (1,3;1,6)- $\beta$ -glucan dodecasaccharide

Author name: Akito Shibuya, Yui Ishisaka, Asuka Saito, Moeko Kato, Sujit Manmode, Hiroto Komatsu, Md Azadur Rahman, Norihiko Sasaki, Toshiyuki Itoh and Toshiki Nokami

Journal name: Faraday Discussions (volume 247, issue 01 November 2023, page 59-69)

Publication date: 2023/05/16

Title: Glycosyl Dioxalenium Ions as Reactive Intermediates of Automate Electrochemical Assembly

Author name: Akito Shibuya, Moeko Kato, Asuka Saito, Sujit Manmode, Naoto Nishikori, Toshiyuki Itoh, Aiichiro Nagaki and Toshiki Nokami

Journal name: European Journal of Organic Chemistry (volume 2022, issue 19, page e202200135)

Publication date: 2022/05/18

Title: Electrochemical Assembly for Synthesis of Middle-Sized Organic Molecules

Author name: Akito Shibuya and Toshiki Nokami

Journal name: The Chemical Record (volume 21, issue 9, page 2389-2396)

Publication date: 2021/06/08

## List of publications before 2019

Title: Electrochemical Generation of Glycosyl Triflate Pools

Author name: T. Nokami, A. Shibuya, H. Tsuyama, S. Suga, A. A. Bowers, D. Crich and J. Yoshida

Journal name: Journal of American Chemical Society (volume 129, issue 35, page 10922-10928)

Publication date: 2007/08/17

Title: Iterative Molecular Assembly Based on the Cation-Pool Method. Convergent Synthesis of Dendritic Molecules

Author name: T. Nokami, K. Ohata, M. Inoue, H. Tsuyama, A. Shibuya, K. Soga, M. Okajima, S. Suga and J. Yoshida

Journal name: Journal of American Chemical Society (volume 130, issue 33, page 10864–10865)

Publication date: 2008/07/29

Title: Oligosaccharide Synthesis Based on a One-pot Electrochemical Glycosylation–Fmoc Deprotection Sequence

Author name: T. Nokami, H. Tsuyama, A. Shibuya, T. Nakatsutsumi and J. Yoshida

Journal name: Chemistry Letters (volume 39, issue 9, 942-943)

Publication date: 2008/08/02

Title: Electrochemical Conversion of Thioglycosides to Glycosyl Triflates

Author name: T. Nokami, A. Shibuya, and J. Yoshida

Journal name: Trends in Glycoscience and Glycotechnology (volume 20, issue 114, 175-185)

Publication date: 2008/09/01

Title:  $\alpha$ - and  $\beta$ -Glycosyl Sulfonium Ions: Generation and Reactivity

Author name: T. Nokami, A. Shibuya, S. Manabe, Y. Ito and J. Yoshida

Journal name: Chemistry A European Journal (volume 15, issue 10, page 2252 – 2255)

Publication date: 2009/02/12

Title: Glycosyl Sulfonium Ions as Storable Intermediates for Glycosylations

Author name: T. Nokami, Y. Nozaki, Y. Saigusa, A. Shibuya, S. Manabe, Y. Ito and J. Yoshida

Journal name: Organic Letters (volume 13, issue 6, page 1544–1547)

Publication date: 2011/02/16

Title: Electrochemical generation of 2,3-oxazolidinone glycosyl triflates as an intermediate for stereoselective glycosylation

Author name: T. Nokami, A. Shibuya, Y. Saigusa, S. Manabe, Y. Ito and J. Yoshida

Journal name: Beilstein Journal of Organic Chemistry (volume 8, page 456–460)

Publication date: 2012/03/28